

Supporting Information for

Crystal Facet Engineering of TiO₂ Nanostructures for Enhancing Photoelectrochemical Water Splitting with BiVO₄ Nanodots

Mi Gyoung Lee^{1,†}, Jin Wook Yang^{1,†}, Hoonkee Park¹, Cheon Woo Moon², Dinsefa M. Andoshe¹, Jongseong Park¹, Chang-Ki Moon¹, Tae Hyung Lee¹, Kyoung Soon Choi³, Woo Seok Cheon¹, Jang-Joo Kim¹, and Ho Won Jang^{1,4,*}

¹Department of Materials Science and Engineering, Research Institute of Advanced Materials, Seoul National University, Seoul 08826, Republic of Korea

²Department of Chemistry and Nanoscience, Ewha Womans University, Seoul 03693, Republic of Korea

³National research Facilities & Equipment Center, Korea Basic Science Institute, Daejeon 34133, Republic of Korea

⁴Advanced Institute of Convergence Technology, Seoul National University, Suwon 16229, Republic of Korea

[†]M. G. L. and J. W. Y. contributed equally to this work

*Corresponding author. E-mail: hwjang@snu.ac.kr (Ho Won Jang)

Supplementary Tables and Figures

Table S1 Equivalent circuit analysis of EIS spectra for TiO₂ NFs, TiO₂ NRs, BiVO₄/TiO₂ NFs, and BiVO₄/TiO₂ NRs with different deposition cycles of BiVO₄

Photoanodes	R_s (Ω cm ²)	R_{ct1} (Ω cm ²)	R_{ct2} (Ω cm ²)
TiO ₂ NFs	4.99	3.16	4770.75
6 BiVO ₄ /TiO ₂ NFs	2.12	1.05	588.16
12 BiVO ₄ /TiO ₂ NFs	3.26	16.68	599.46
18 BiVO ₄ /TiO ₂ NFs	2.67	0.84	475.11
24 BiVO ₄ /TiO ₂ NFs	2.41	1.01	743.04
36 BiVO ₄ /TiO ₂ NFs	3.33	1.62	743.04
TiO ₂ NRs	2.11	2.06	259.89
BiVO ₄ /TiO ₂ NRs	3.34	5.18	1948.06

Table S2 Transient decay times of TiO₂ NFs, TiO₂ NRs, BiVO₄/TiO₂ NFs, and BiVO₄/TiO₂ NRs

Photoanodes	Transient decay time (s)
TiO ₂ NFs	6.36
TiO ₂ NRs	2.58
BiVO ₄ /TiO ₂ NFs	26.35
BiVO ₄ /TiO ₂ NRs	1.57

Table S3 TRPL lifetimes of TiO₂ NFs, TiO₂ NRs, BiVO₄/TiO₂ NFs, and BiVO₄/TiO₂ NRs

Photoanodes	τ_1 (ns)	τ_2 (ns)
TiO ₂ NFs	0.7397	6.437
TiO ₂ NRs	0.5978	4.314
BiVO ₄ /TiO ₂ NFs	0.8009	5.2046
BiVO ₄ /TiO ₂ NRs	0.7577	6.102

Table S4 a Energy level of valence band and work function of FTO, TiO₂ NFs, TiO₂ NRs, BiVO₄/TiO₂ NFs, and BiVO₄/TiO₂ NRs investigated by ultraviolet photoemission spectroscopy. **b** Optical band gap of TiO₂ NFs, TiO₂ NRs, and BiVO₄ measured by UV-visible spectroscopy

a

Materials	$E_F - E_{\text{VBM}}$ (eV)	E_F (eV)
FTO	N/A	4.4
TiO ₂ NFs	2.62	4.46
TiO ₂ NRs	2.67	3.58
BiVO ₄ /TiO ₂ NFs	2.39	4.18
BiVO ₄ /TiO ₂ NRs	2.39	4.18

b

Materials	Optical band gap (eV)
FTO	N/A
TiO ₂ NFs	2.84
TiO ₂ NRs	2.75
BiVO ₄	2.4

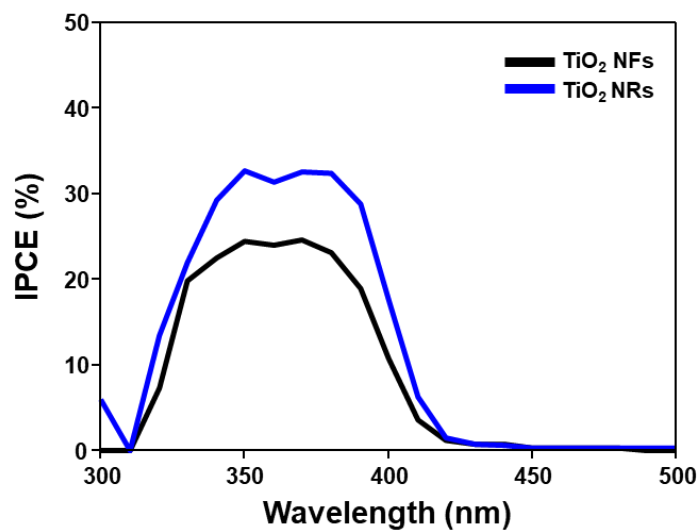


Fig. S1 IPCE at 1.23 V (vs. RHE) of TiO₂ NFs and TiO₂ NRs

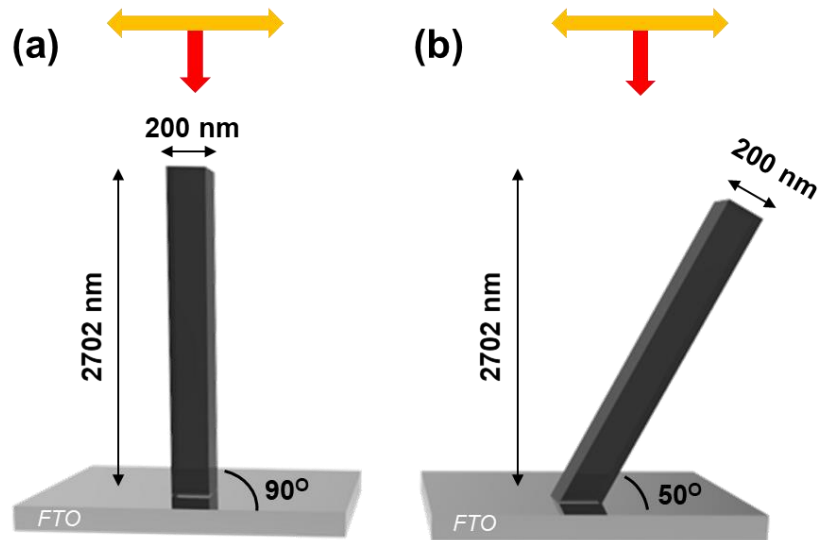


Fig. S2 Model used in finite-difference time-domain (FDTD) simulations of **a** TiO₂ NRs and **b** TiO₂ NFs

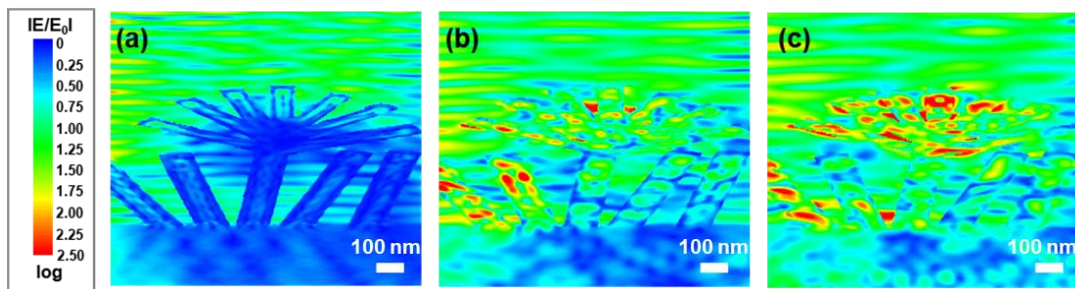


Fig. S3 FDTD simulations of near field enhancement of TiO₂ NFs on FTO under **a** 350 nm, **b** 450 nm, **c** 550 nm photons

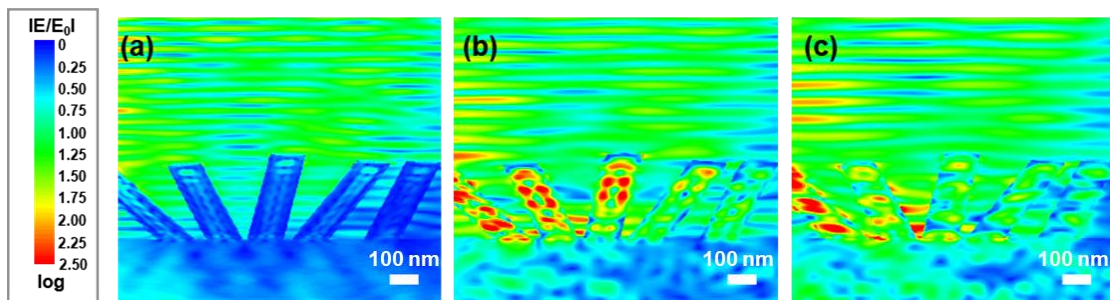


Fig. S4 FDTD simulations of near field enhancement of TiO₂ NRs on FTO under **a** 350 nm, **b** 450 nm, **c** 550 nm photons

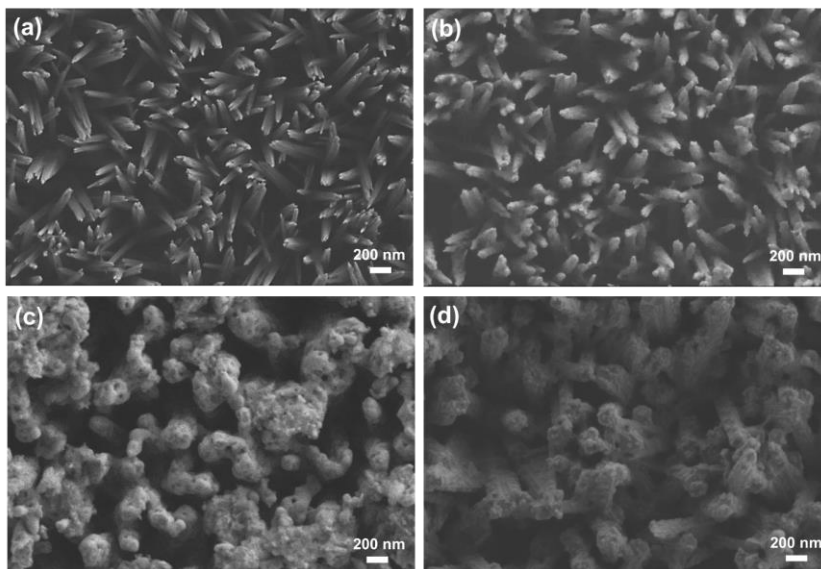


Fig. S5 SEM images of BiVO₄/TiO₂ NRs with different deposition cycles of BiVO₄: **a** 6 cycles, **b** 12 cycles, **c** 18 cycles, and **d** 36 cycles

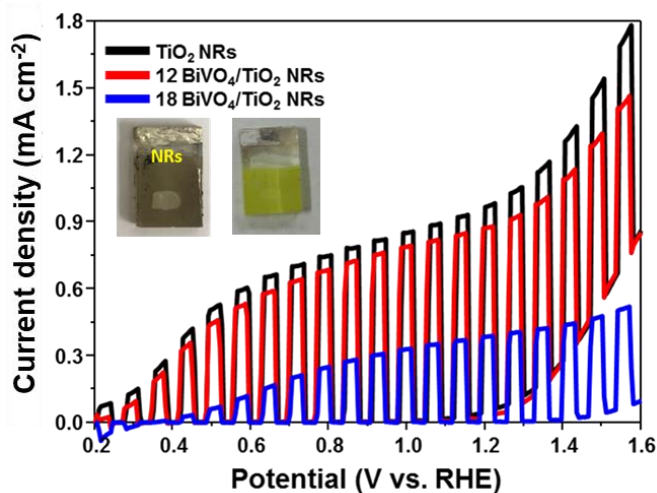


Fig. S6 LSVs of TiO₂ NRs and BiVO₄/TiO₂ NRs with different deposition cycles of BiVO₄ in 0.5 M K-P_i buffer with 1 M Na₂SO₃

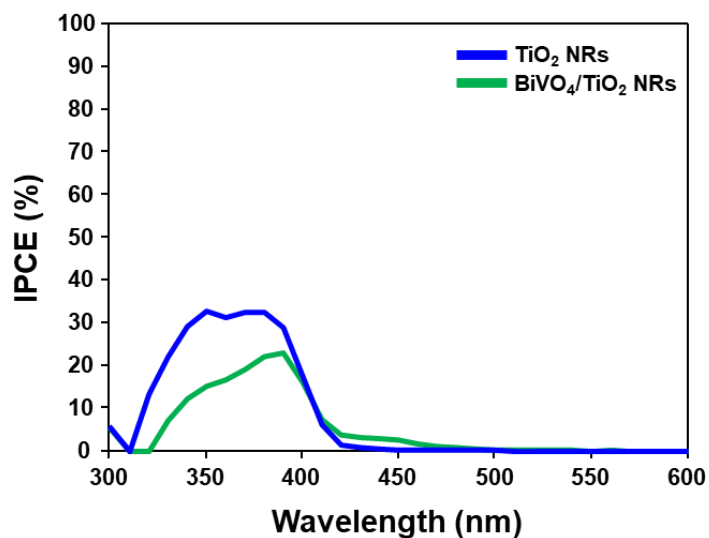


Fig. S7 IPCE at 1.23 V (vs. RHE) of TiO₂ NRs and BiVO₄/TiO₂ NRs

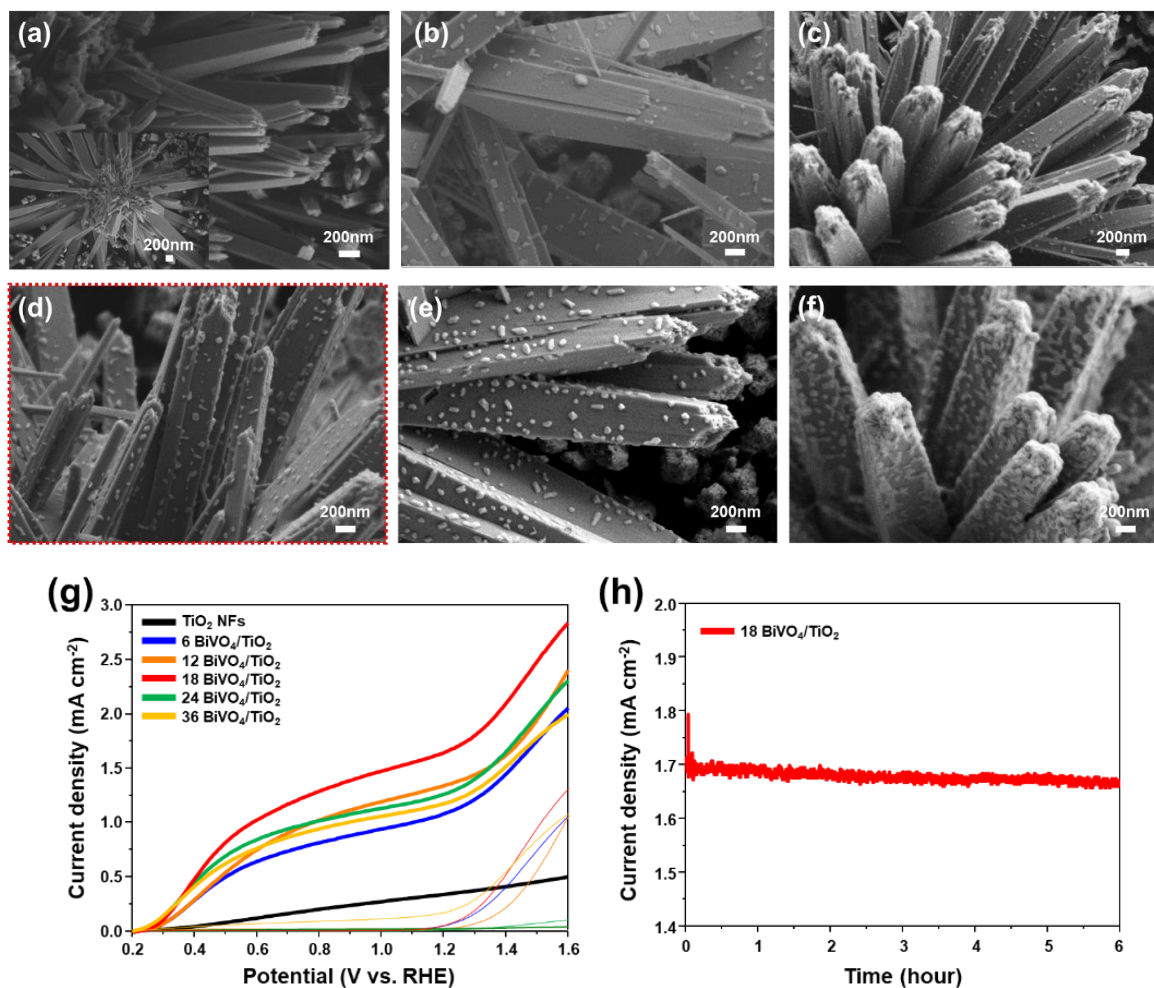


Fig. S8 SEM images of **a** TiO₂ NFs and BiVO₄/TiO₂ NFs with different deposition cycles of BiVO₄: **b** 6 cycles, **c** 12 cycles, **d** 18 cycles, **e** 24 cycles, and **f** 36 cycles. **g** LSV of TiO₂ NFs and BiVO₄/TiO₂ NFs with different deposition cycles of BiVO₄ in 0.5 M K-P_i buffer with 1 M Na₂SO₃. **h** Chronoamperometry of BiVO₄/TiO₂ NFs deposited by 18 cycles

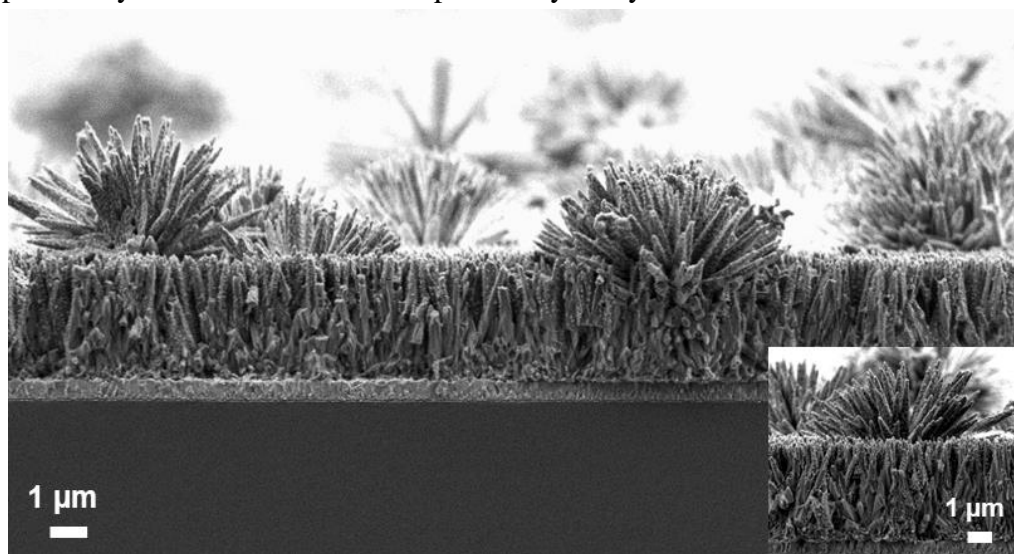


Fig. S9 Cross-sectional SEM images of BiVO₄/TiO₂ NFs deposited by 18 cycles. Inset is the enlarged image of BiVO₄/TiO₂ NFs

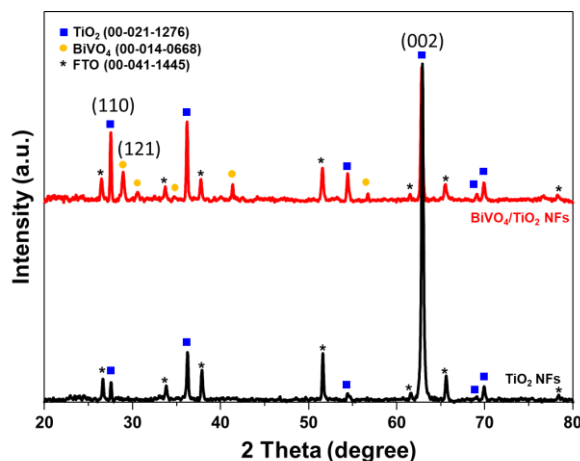


Fig. S10 XRD of TiO₂ NFs and BiVO₄/TiO₂ NFs

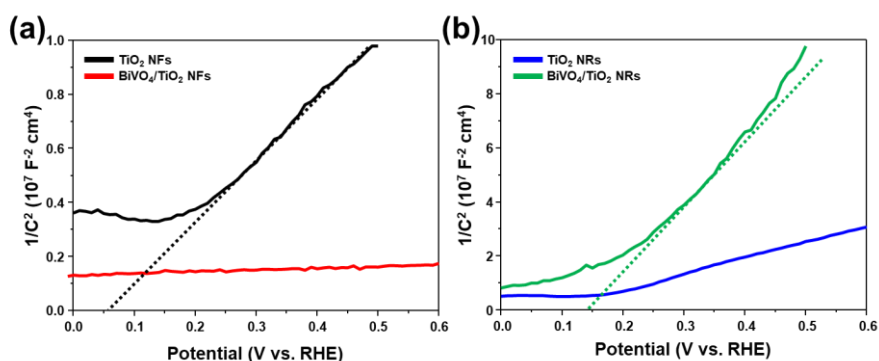


Fig. S11 Comparison of Mott-Schottky plot before and after deposition of BiVO₄ nanodots on the **a** TiO₂ NFs and **b** TiO₂ NRs. (Frequency: 1 kHz, Amplitude: 10 mV, under light off)

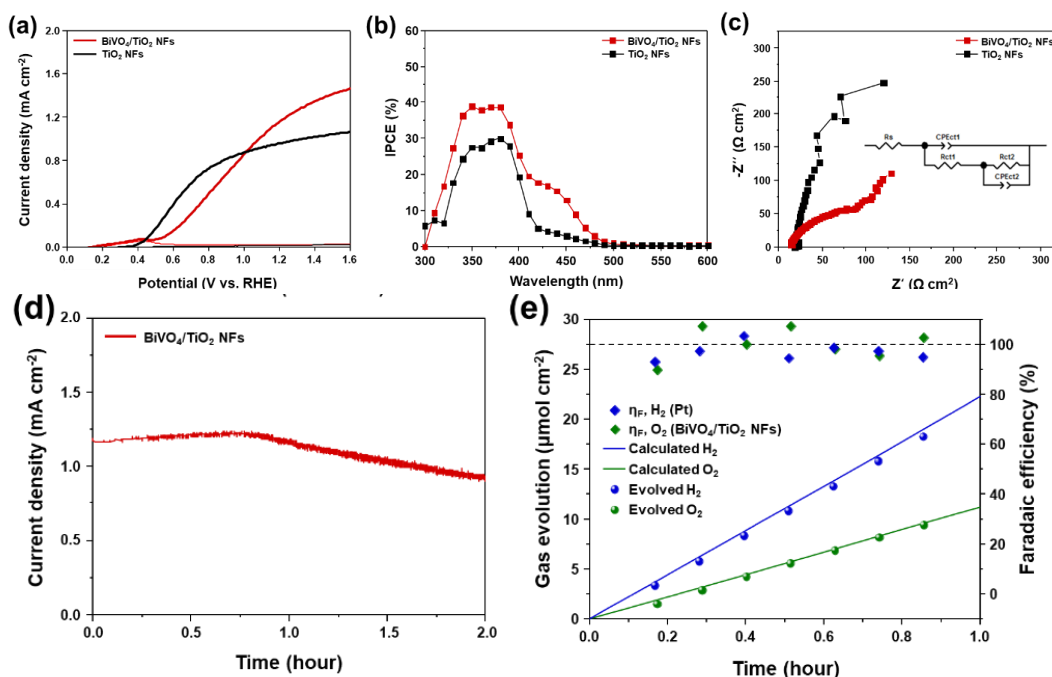


Fig. S12 **a** LSV (water oxidation) of TiO₂ NFs and 18 BiVO₄/TiO₂ NFs in 0.5 M K-P_i buffer without Na₂SO₃. **b** IPCE and **c** EIS of TiO₂ NFs and 18 BiVO₄/TiO₂ NFs at 1.23 V (vs. RHE) in 0.5 M K-P_i buffer without Na₂SO₃. **d** Chronoamperometry of 18 BiVO₄/TiO₂ NFs photoanode in 0.5 M K-P_i buffer without Na₂SO₃. **e** Gas evolution and Faradaic efficiency of 18 BiVO₄/TiO₂ NFs photoanode

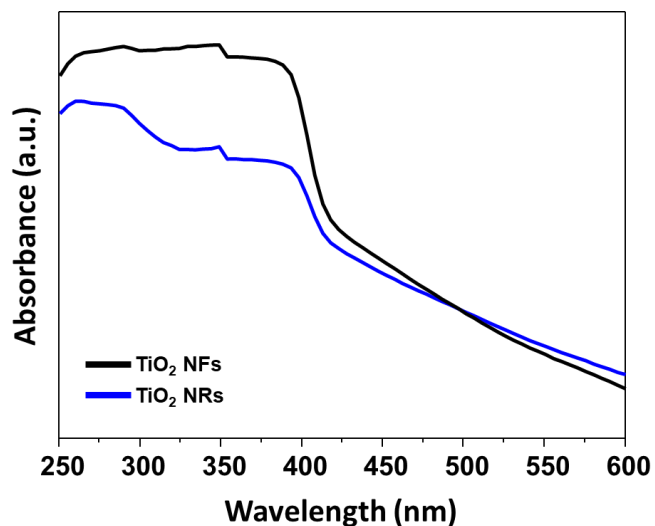


Fig. S13 UV-vis absorption spectra of TiO₂ NFs and TiO₂ NRs

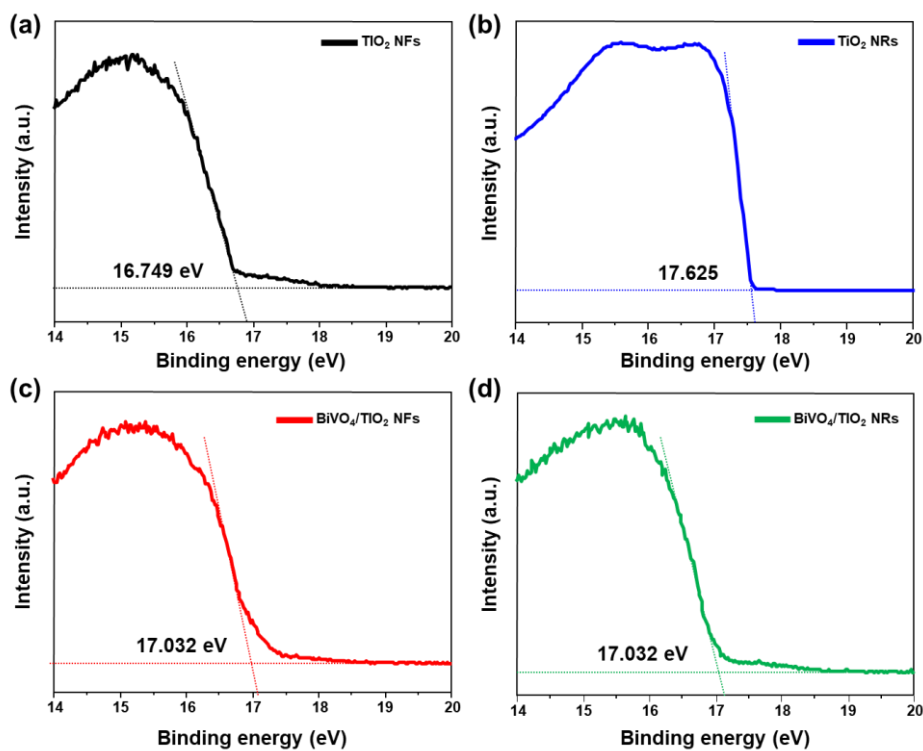


Fig. S14 Secondary electron emission spectra of **a** TiO₂ NFs, **b** TiO₂ NRs **c** BiVO₄/TiO₂ NFs, and **d** BiVO₄/TiO₂. $E_F = E_{cut\ off} - 21.21$ eV (He I source)

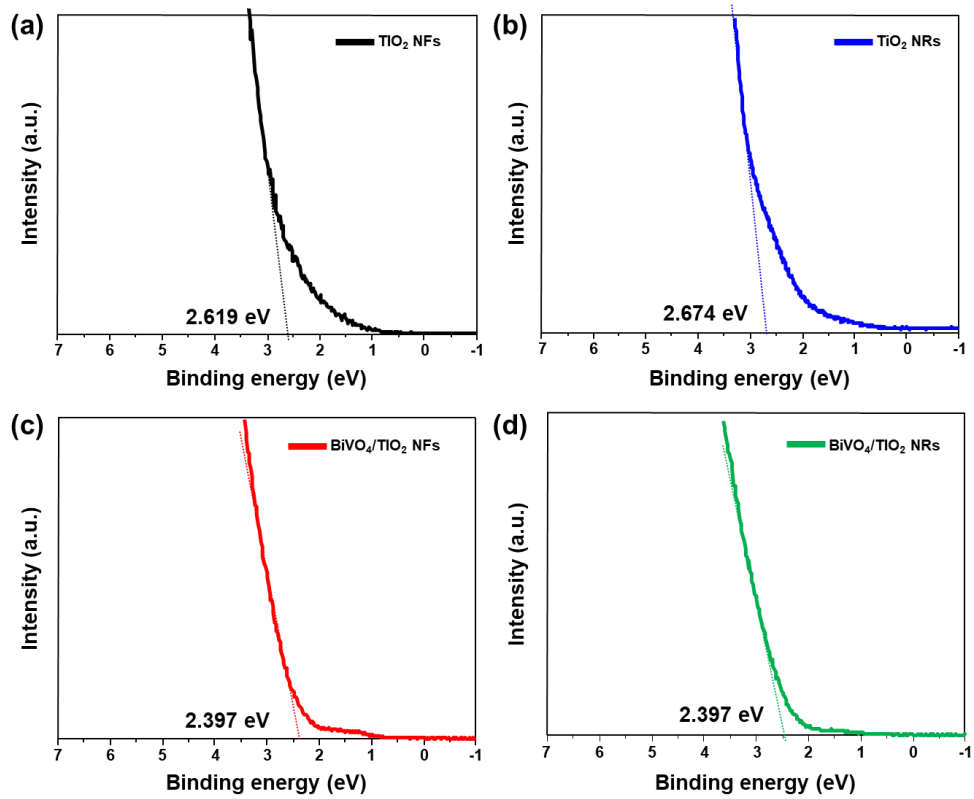


Fig. S15 Valence band spectra and the energy difference between Fermi level and valence band maximum ($E_F - E_{VBM}$) of **a** TiO_2 NFs, **b** TiO_2 NRs, **c** $\text{BiVO}_4/\text{TiO}_2$ NFs, and **d** $\text{BiVO}_4/\text{TiO}_2$ NRs. $E_{VBM} = E_F + E_{edge}$