Supporting Information for

# Hetero-Interfaces on Cu Electrode for Enhanced Electrochemical

# **Conversion of CO<sub>2</sub> to Multi-Carbon Products**

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## **Supplementary Figures and Tables**



Fig. S1 SEM images of UiO-66 with different sizes of (a) 100 nm, (b) 300 nm, and (c) 600 nm



Fig. S2 Photo of the H-cell setup used in this work



Fig. S3 Schematic of the synthetic process of UiO-66 modified Cu foil (X-UiO/Cu)



Fig. S4 SEM images of (a) UiO-66 nanoparticles and (b) mechanically polished Cu foil



Fig. S5 Cu LMM AES spectrum of 0.5-UiO/Cu-bare before CO<sub>2</sub>RR



**Fig. S6** (a) C 1s XPS spectrum, (b) O 1s XPS spectrum and (c) Zr 3d XPS spectrum of 0.5-UiO/Cu. (d) Zr 3d XPS spectrum of UiO-66 nanoparticles. All samples in Fig. S6 are characterized before CO<sub>2</sub>RR



**Fig. S7** (a) Linear fitting of capacitive currents and (b) the corresponding electrochemical double-layer capacitance ( $C_{dl}$ ) (the slopes of the fitting curves in (a)) of X-UiO/Cu. The ECSA-corrected current density for H<sub>2</sub>, C<sub>1</sub>, and C<sub>2+</sub> products as a function of potential on (c) Cu and (d) 0.5-UiO/Cu



Fig. S8 FEs of CO<sub>2</sub>RR and HER products on Cu foil as a function of potential



Fig. S9 Formation rates of C2+ products (r<sub>C2+</sub>) on Cu foil and 0.5-UiO/Cu as a function of potential



**Fig. S10** Top-view SEM images of 0.5-UiO/Cu electrode after CV: (**a**) with and (**b**) without the surface coating layer. (**c**) Top-view SEM image of 0.5-UiO/Cu electrode after CA without the surface coating layer



**Fig. S11** (a) C 1s spectrum and (b) O 1s spectrum of 0.5-UiO/Cu after CO<sub>2</sub>RR at -1.05 V vs. RHE for 1 h



**Fig. S12** Schematic of possible evolution process of UiO-66 coating on X-UiO/Cu electrode under CO<sub>2</sub>RR



**Fig. S13** The *in situ* surface-enhanced Raman spectra recorded between (**a**) 100-700 cm<sup>-1</sup> and (**b**) 1700-2400 cm<sup>-1</sup> on Cu foil at OPC, after CV, and at the selected potential range of -0.2 V to -1.2 V vs. RHE with a potential interval of 0.2 V for 10 min

Sample	C <sub>dl</sub> (μF cm <sup>-2</sup> )	$R_{ m f}$	ECSA	ECSA-corrected j <sub>C2+</sub>
Cu	35.37	1	1	1.99
0.1-UiO/Cu	187.39	5.3	5.3	1.76
0.25-UiO/Cu	198.41	5.61	5.61	3.35
0.5-UiO/Cu	245.9	6.95	6.95	3.82
1-UiO/Cu	325.32	9.2	9.2	2.90
2-UiO/Cu	408.26	11.54	11.54	2.45

**Table S1** The  $C_{dl}$ ,  $R_f$ , ECSA, and ECSA-corrected  $j_{C2+}$  of Cu and X-UiO/Cu

<b>Table 52</b> Companyon of CO <sub>2</sub> KK performance on 0.5-010/Cu with state-or-me-art Cu-based cataryst evaluated in ri-type c	Table S2 Comparison of CO <sub>2</sub> RR	performance on 0.5-UiO/Cu w	ith state-of-the-art Cu-based cat	alyst evaluated in H-type cell
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Refs.	catalyst	cell type	electrolyte	Ewe/(RHE)	FE of C <sub>2+</sub>	ECSA- collected jC <sub>2+</sub> (mA cm <sup>-2</sup> )	geometric area normalized jC <sub>2+</sub> (mA cm <sup>-2</sup> )	r <sub>C2+</sub> : Formation rate of C2+ (μmol s <sup>-1</sup> m <sup>-2</sup> )	stability (h)
This work	0.5-UiO/Cu	H-cell	<b>0.1 M KHCO</b> <sub>3</sub>	-1.05	<b>74.17%</b>	-3.822	-26.57	228.08	32
ACS Catal. 2021, 11, 2473-2482	Cu@N <sub>x</sub> C	home-made H-cell	0.1 M KHCO <sub>3</sub>	-1.1	76.8%	-5	-14.9	128.69	2.7
Joule. 2021 5, 429-440	Cu-DS	H-cell	0.1 M KHCO3	-1.08	78%	N/A	-23.4	186.56	30
Small. 2021, 2102293	Cu@Ag-2	Flow-cell	1 M KOH	-1.1	67.6%	N/A	-22.7	193.53	14
Nano Res. 2021									
doi.org/10.1007/s12274-	Cu-s	H-cell	0.1 M KHCO <sub>3</sub>	-1.1	55.8%	N/A	-26.69	230.52	8
021-3532-7									
Electrochim Acta. 2021,	n-Cu	H-cell	0.1 M KHCO <sub>2</sub>	-13	57.2%	N/A	-22.65	195 69	10
388, 138552	p Cu	II con	0.1 10 101003	1.5	37.270	1.1.1	22.05	195.09	10
Green Chem. 2020, 22,	CuO-	H-cell	0.1 M KHCO3	-1.1	50%	N/A	-3.77	32.56	9
6540-6546	CeO <sub>2</sub> /CB		•••••••		/ -				-
Angew. Chem. Int. Ed. 2021, 60, 7426-7435	5-Ag/Cu <sub>2</sub> O	H-cell	0.1 M KHCO <sub>3</sub>	-0.98	65%	-0.41	-6.09	51.25	12
Angew. Chem. Int. Ed.	Cu/CuSiOa	H_cell	0.1 M KHCO	_1 1	60 64%	N/A	_12.25	105 80	6
2021, 60, 15344-15347		II-cell	0.1 W KHCO <sub>3</sub>	-1.1	00.0470	$\mathbf{N}\mathbf{A}$	-12.25	105.00	0
ACS Appl. Nano Mater. 2020, 3, 257-263	Cu GNC-VL	H-cell	0.5 M KHCO <sub>3</sub>	-0.87	70.5%	N/A	-7.33	63.33	12
Chem. Mater. 2020, 32, 3304-3311	Cu <sub>3</sub> N	H-cell	0.1 M CsHCO <sub>3</sub>	-1	68%	-0.714	-12.58	102.15	3.33
ACS Energy Lett. 2021, 6, 437-444	CuBr-DDT	H-cell	0.1 M KCl	-1.25	72%	-8.75	-9.02	76.90	15
ACS Catal. 2020, 10, 4103-4111	Cu/PANI	H-cell	0.1 M KHCO <sub>3</sub>	-1.2	66%	-5.17	-14.9	127.91	20

Some data in **Table S2** is collected from figures in the corresponding literature, which may be less precise.

trail/at%	С	0	Cu	Zr
UiO/Cu NPs	67.89	27.47	0.83	3.80
UiO/Cu-CV NPs	49.71	41.86	1.04	7.39
UiO/Cu-CA NPs	33.71	49.26	2.72	14.32

Table S3 EDS results of 0.5-UiO/Cu NPs, 0.5-UiO/Cu-CV NPs, and 0.5-UiO/Cu-CA NPs

Table S4 FEs of H<sub>2</sub> and various CO<sub>2</sub>RR products as well as the geometric current density on Cu foil as a function of potential

E <sub>we</sub> (RHE)	j (mA cm <sup>-2</sup> )	$H_2$	СО	CH <sub>4</sub>	НСООН	C2H4	C <sub>2</sub> H <sub>5</sub> OH	СН3СООН	C <sub>3</sub> H <sub>7</sub> OH	СН <sub>3</sub> СНО
-0.85	1.92	60.01	7.12	0	27	0	0	0	0	0
-0.90	2.45	$60.5 \pm 2.91$	$8.69 \pm 1.03$	$0.18\pm0.18$	$27.84 \pm 3.66$	0	0	0	0	0
-0.95	3.29	$45.49 \pm 2.97$	$10.86 \pm 1.62$	$2.46\pm0.44$	$26.85 \pm 2.72$	4.3±1.13	0	0	0	0
-1	5.53	35.14±1.99	$10.72 \pm 0.89$	$11.4 \pm 2.85$	23.17±2.44	$11.88 \pm 1.53$	$1.06 \pm 1.83$	$0.14 \pm 0.25$	0	0
-1.05	9.63	23.3±2.52	$5.52 \pm 0.38$	$22.01{\pm}1.44$	15.73±2.22	$15.5 \pm 1.57$	4.66±1.72	$0.41 \pm 0.36$	0	0
-1.1	19.67	$20.72 \pm 4.04$	2.52±0.06	$32.59 \pm 0.83$	$10.79{\pm}1.48$	16.72±2.19	3.99±0.49	$1.09\pm0.40$	$1.56{\pm}1.08$	$1.57 \pm 2.22$

Table S5 FEs of H<sub>2</sub> and various CO<sub>2</sub>RR products as well as the geometric current density on 0.5-UiO/Cu electrode as a function of potential

Ewe (RHE)	j (mA cm <sup>-2</sup> )	$\mathbf{H}_2$	СО	CH <sub>4</sub>	нсоон	CH <sub>3</sub> OH	C <sub>2</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> OH	СН <sub>3</sub> СООН	C <sub>3</sub> H <sub>7</sub> OH	CH <sub>3</sub> CHO
-0.85	3.06	29.16	19.1	0	36.58	0	7.18	7	0	3.94	0
-0.9	5.81	$19.94 \pm 2.39$	$17.68 \pm 0.91$	0	$28.73 \pm 2.69$	0	$15.29 \pm 1.12$	7.07±3.12	0	$5.25 \pm 0.62$	0
-0.95	9.72	$17.76 \pm 3.47$	$10.5 \pm 1.2$	$1.33\pm0.31$	$18.01 \pm 1.91$	$0.84 \pm 0.78$	$28.44 \pm 2.47$	9.63±1.02	$0.54\pm0.19$	8.68±0.79	$0.56 \pm 0.47$
-1	23.95	$15.06 \pm 1.85$	$3.15 \pm 0.28$	$3.38 \pm 0.58$	8.73±0.99	$0.58 \pm 0.25$	$40.68 \pm 1.4$	16.8±1.33	$0.75\pm0.42$	8.21±0.31	$1.94\pm0.93$
-1.05	35.78	$11.98 \pm 2.81$	$1.41\pm0.28$	4.45±1	$5.7 \pm 2.07$	$0.53 \pm 0.58$	$42.2 \pm 1.92$	20.97±0.73	$0.86\pm0.21$	8.2±0.25	$1.95 \pm 0.63$
-1.1	56.71	$22.88{\pm}1.48$	0.83±0.19	6.35±0.9	3.06±0.55	$0.5 \pm 0.36$	33.02±1.77	22.41±2.44	0.5±0.19	3.77±0.74	$0.75 \pm 0.05$

Loading (mg cm <sup>-2</sup> )	j (mA cm <sup>-2</sup> )	$H_2$	CO	CH <sub>4</sub>	нсоон	CH <sub>3</sub> OH	C <sub>2</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> OH	СН <sub>3</sub> СООН	C <sub>3</sub> H <sub>7</sub> OH	CH <sub>3</sub> CHO
0.1	16.5	15.84	2.25	9.36	10.66	0	33.3	16.32	0.13	5.88	0.92
0.25	26.34	9.45±0.9	$1.66\pm0.14$	$5.56 \pm 0.62$	5.91±0.94	0.17±0.3	43.32±2.01	18.39±0.94	$0.72\pm0.18$	$6.58 \pm 0.28$	1.77±0.51
0.5	35.78	$11.98 \pm 2.81$	$1.41\pm0.28$	4.45±1	$5.7{\pm}2.07$	$0.53 \pm 0.58$	42.2±1.92	20.97±0.73	0.86±0.21	8.2±0.25	1.95±0.63
1	44.34	22.56±2.93	1.49±0.16	4.8±0.97	$5.84{\pm}1.23$	$0.7 \pm 0.82$	34.11±1.6	17.61±1.7	$0.75 \pm 0.24$	$6.27 \pm 2.94$	1.22±0.59
2	56.72	27.36	1.08	4.58	6.08	0.78	27.67	13.3	0.43	8.2	0.32

Table S6 FEs of H<sub>2</sub> and various CO<sub>2</sub>RR products as well as the geometric current density on various X-UiO/Cu electrodes at -1.05 V vs. RHE