

Supporting Information for

Highly Thermally Conductive Polymer/Graphene Composites with Rapid Room-Temperature Self-Healing Capacity

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S1 Structural Analysis and Characterization

The morphology and microstructure of PBA_x-PDMS and PBA_x-PDMS/FGf composites were characterized by field-emission scanning electron microscopy (FESEM; S4800-15kV, Hitachi, Japan). The chemical structures and components of the copolymer samples were examined by Fourier transformation infrared spectroscopy (FT-IR; Tensor27, Bruker, Germany) and X-ray photoelectron spectroscopy (XPS; 250Xi, Thermo Fischer, USA), respectively. All C1s line was corrected at 284.6 eV. The crystallinity of the polymer and the phase of the composite were characterized by X-ray diffraction (XRD, D8 Advance, Bruker, Germany) with Cu K α radiation ($\lambda = 1.54 \text{ \AA}$). The glass transition temperature (T_g) and of the samples was evaluated by differential scanning calorimetry (DSC; TA660, TA instrument, Japan) in an Ar atmosphere at heating and cooling rates of $10 \text{ }^\circ\text{C min}^{-1}$. ¹H-NMR spectra were recorded using a 400-MHz spectrometer (NMR; AVANCE III, Bruker, Swiss). Specifically, the content of the copolymerization unit is tested using the internal standard method, and the signal for quantitative integration and the number of protons generating the signal are selected to be found based on the molecular weight of the monomer. In order to ensure the reproducibility of the experiment, the molar ratio of the feed is used for its description. The CDCl₃ (¹H at 7.26 ppm) was used as the solvent, and the splitting patterns of NMR were as follows: s, singlet; t, triplet; q, quartet; m, multiplet.

S2 Supplementary Figures and Tables

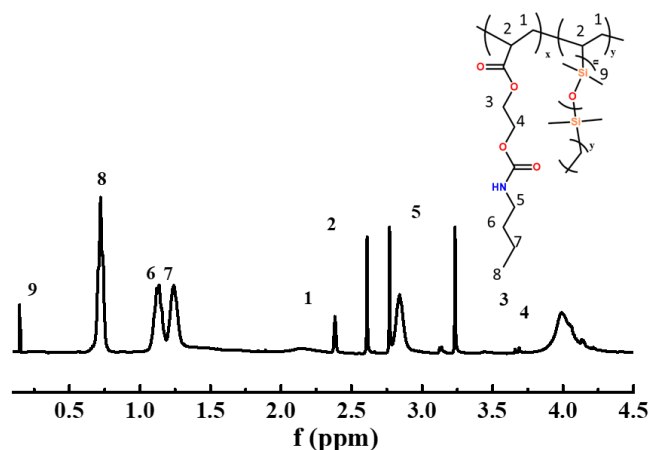


Fig. S1 ¹H NMR spectra of PBA-PDMS in CDCl₃

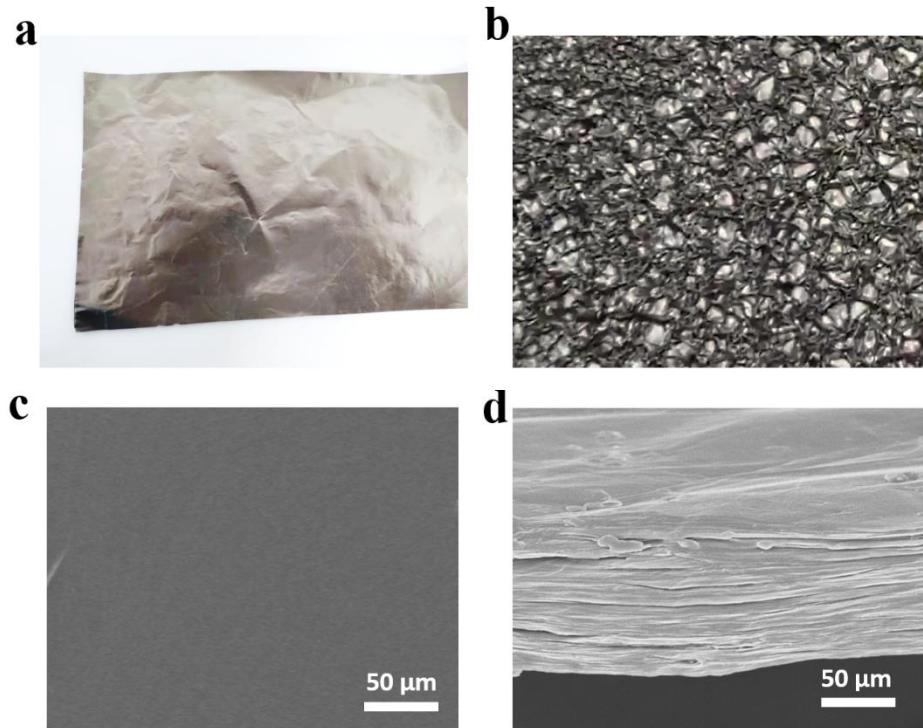


Fig. S2 (a) Digital electro-photograph of (a) graphene film and (b) FGf. The (c) surface and (d) cross sectional SEM image of FGf

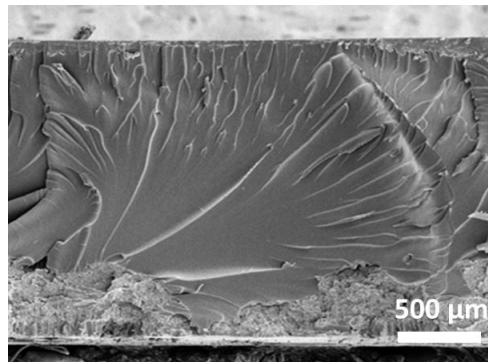


Fig. S3 Cross sectional SEM image of PBA-PDMS/FGf

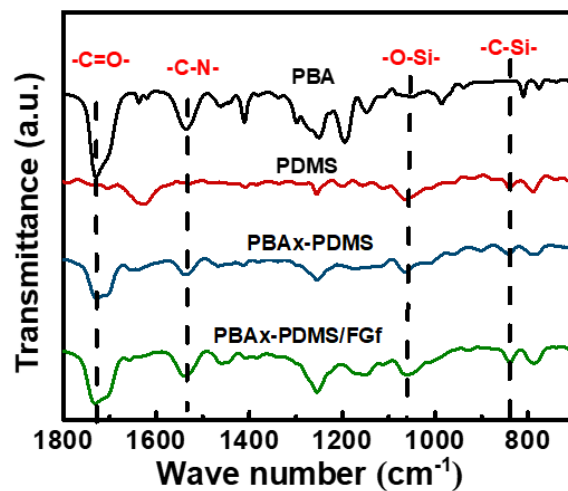


Fig. S4 FTIR spectra of PBA, PDMS, PBA_x-PDMS, and PBA_x-PDMS/FGf

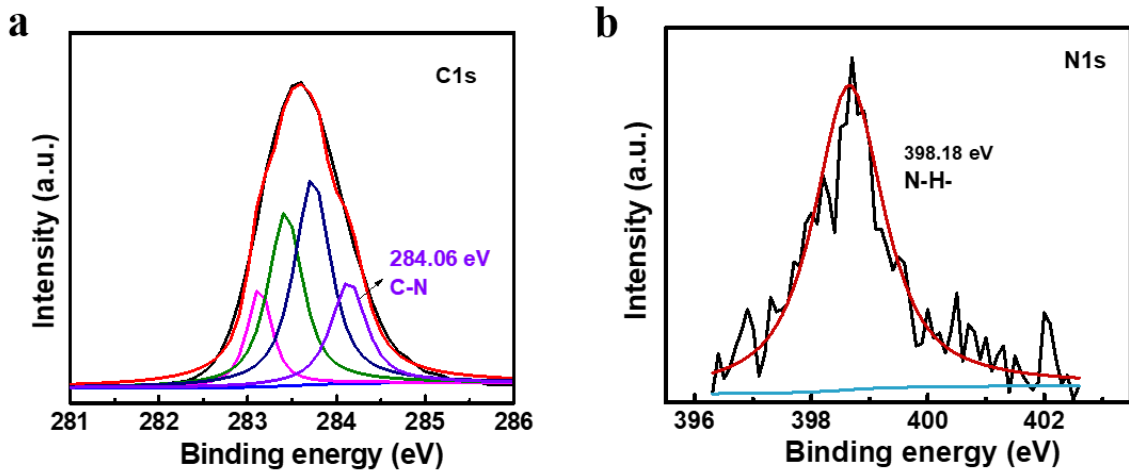


Fig. S5 Structural characterization of polymers. Magnification of (a) C1s, and (b) N1s of PBA-PDMS

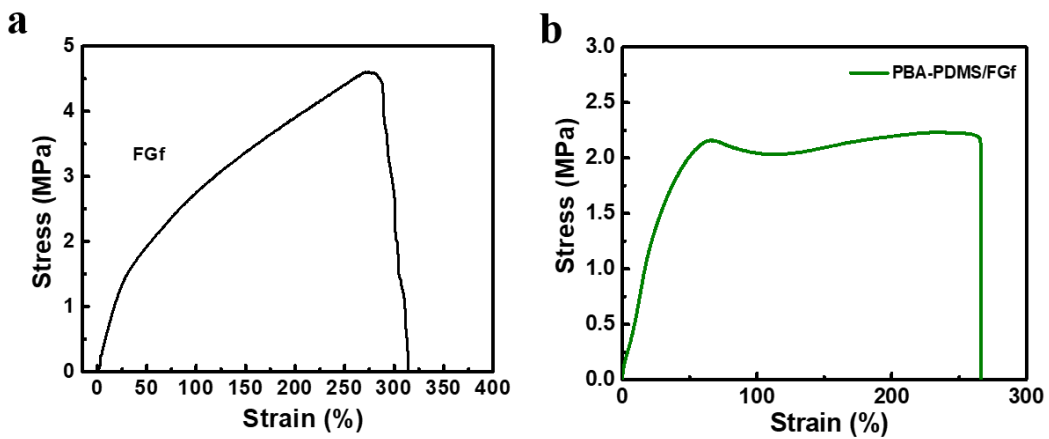


Fig. S6 (a) Stress-strain behavior of GFf and (b) PBA-PDMS/GFf at room temperature

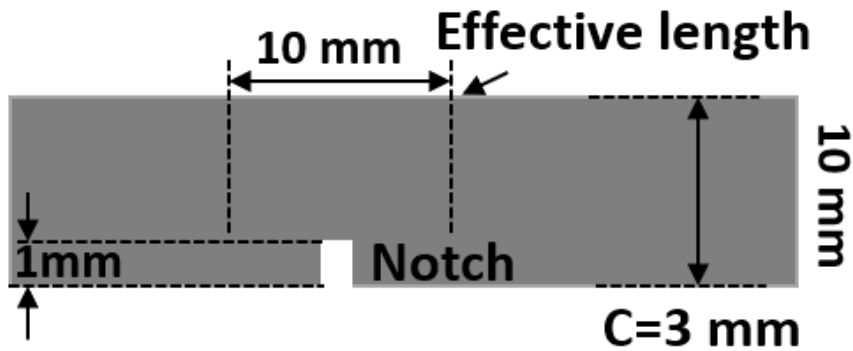


Fig. S7 Fracture energy test sample model

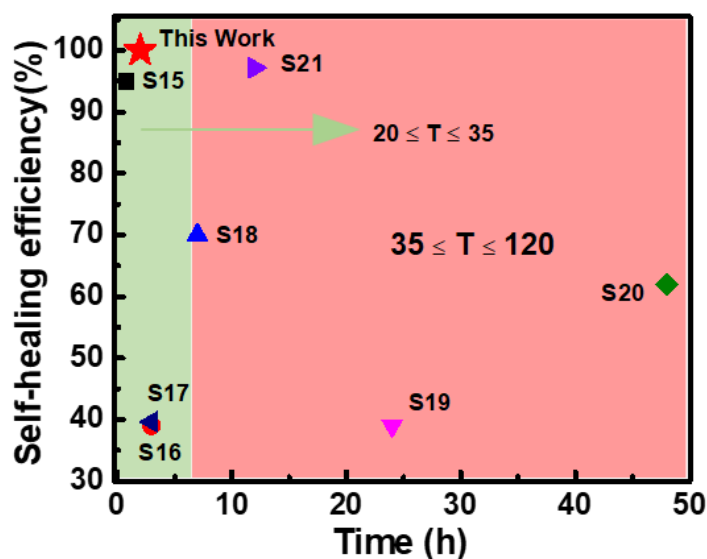


Fig. S8 Comparison of the self-healing efficiencies of the reported polymer compositions

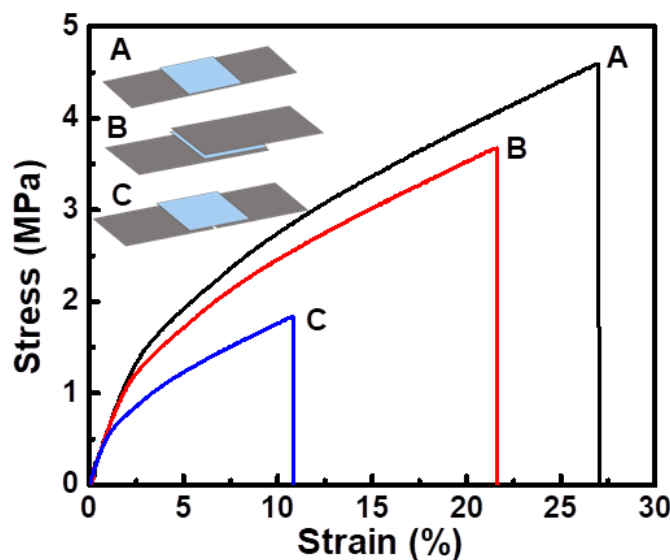


Fig. S9 Mechanical strength of composites with different composite methods

First, three graphene membranes with the same morphology, structure and length were selected in the initial state. In the initial state, the polymer composite is obtained by covering the graphene film with polymer, and its mechanical tensile curve is A. The same length of graphene is folded in half and filled with polymer in the middle, and its stretching curve is set to B. The graphene film breaks, and then the polymer is covered at the fracture, and its mechanical tensile curve is set as C. The results show that the highest mechanical strength of A in the intact state is 4.8 MPa. The mechanical strength of the composite in the B state is 3.65 MPa and the self-healing efficiency is 76%. The mechanical strength of the composite in the C state is 1.7 MPa and the self-healing efficiency is 35%. This indicates that the self-healing of the composite is mainly caused by the strong adhesion of the polymer to the graphene. And the data provide direct evidence of the primary mode of self-healing.

Table S1 Summary of the self-healing polymer materials reported in literatures

Sample	E (MPa)	T (°C)	Time (h)	η_{σ} (%)	Refs.
Py-PDMS-Co-1	-	25	24	71	[S1]
Boroxine-PDMS	182 ± 15.8	70	12	95	[S2]
PDMS-1	2.34 ± 0.22	25	4	95	[S3]
U-PDMS0.9K-Es	0.23 ± 0.07	40	2	100	[S4]
PDMS-MPU _{0.4} -IU _{0.6}	0.62 ± 0.06	25	48	72	[S5]
PDMS (A ₄)	-	100	2	85	[S6]
(UP) ₃ T	47.39 ± 1.03	40	12	77.5	[S7]
Zn (Hbimcp) ₂ -PDMS	43.68 ± 3.27	25	24	98.9 ± 1.9	[S8]
HPUrea	~ 0.12	25	0.5	97.2 ± 2.2	[S9]
BE-PDMS _{1:3} -UPy	130.46 ± 10	25	6	97.69 ± 0.33	[S10]
D-PDMS	0.51	25	24	100	[S11]
PDMS-2S	0.32	25	12	95	[S12]
GC-PDMS-10k	0.17	80	24	75	[S13]
PDMSPU	0.0747	60	10	76	[S14]
PBA-PDMS	0.23 ± 0.1	25	10 (min)	100	This work

The default room temperature is 25 °C.

Table S2 Summary of the graphene/polymers composites materials reported in literatures

Sample	σ (MPa)	T (°C)	Time (h)	η_{σ} (%)	Refs.
graphene/PDMS-urea	17.5	50	0.8	95	[S15]
SHPU/grapheme	4	30	3	39	[S15]
NR/Graphene-2	4.59 ± 0.08	70	7	70	[S16]
PANDA/Gr-0.6	22.3 ± 1.9	75	24	39 ± 4	[S17]
PU-EDM/rmGO-1	31.16 ± 3.30	65	48	62	[S18]
PU/MG050	7.10	30	3	39.63	[S19]
AgNW/EVA	4	50	12	97.17	[S20]
This work	2.23 ± 0.15	25	2	100	

Supplementary References

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