

Supporting Information for

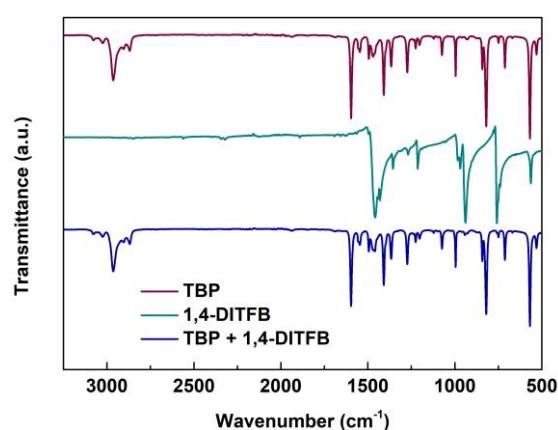
## Overcoming Perovskite Corrosion and De-Doping through Chemical Binding of Halogen Bonds towards Efficient and Stable Perovskite Solar Cells

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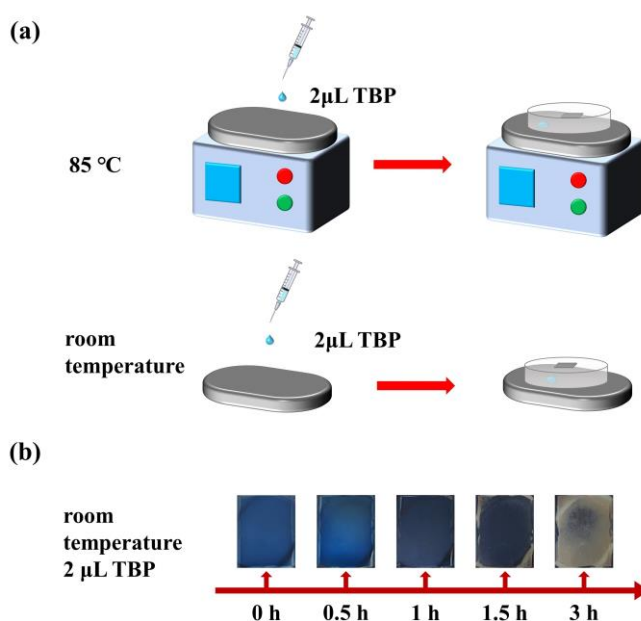
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### Supplementary Figures and Tables



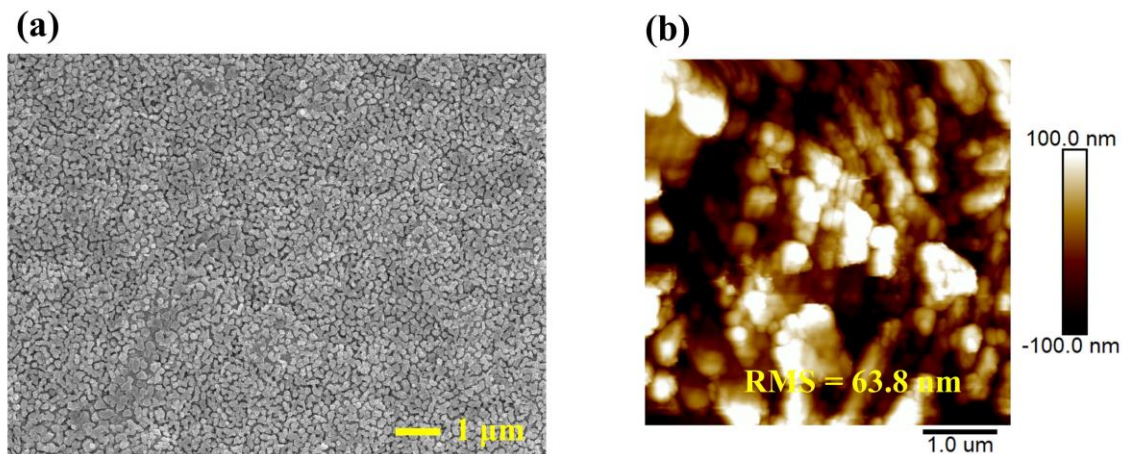
**Fig. S1** Full range FTIR of TBP, 1,4-DITFB and 1,4-DITFB-TBP solution



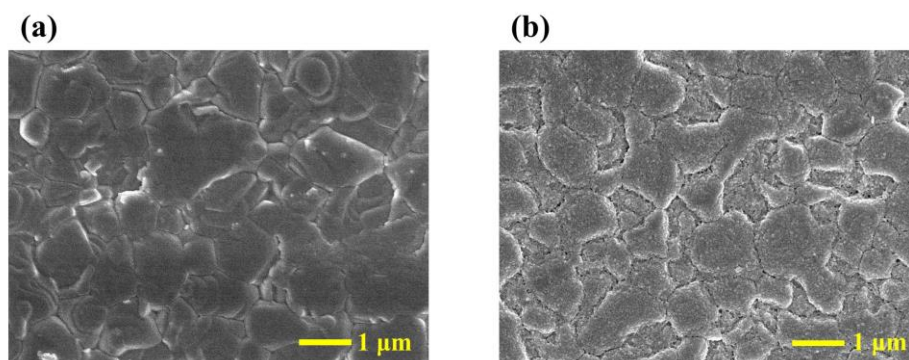
**Fig. S2 a** Schematic diagram depicting the evolution of FA<sub>1-x</sub>MA<sub>x</sub>PbI<sub>3</sub> perovskite films under different treatment conditions and times. **b** Photographs of the evolution of FA<sub>1-x</sub>MA<sub>x</sub>PbI<sub>3</sub> perovskite films at room temperature with 2 μL TBP

**At high temperature:** 2  $\mu\text{L}$  TBP was dropped on a hot stage at 85  $^{\circ}\text{C}$ , then a glass dish with the perovskite film on the top was immediately buckled on the TBP. Here 85  $^{\circ}\text{C}$  was chosen because the local temperature can reach this temperature during device operation [S1]. The control sample was treated in the similar way, but without TBP.

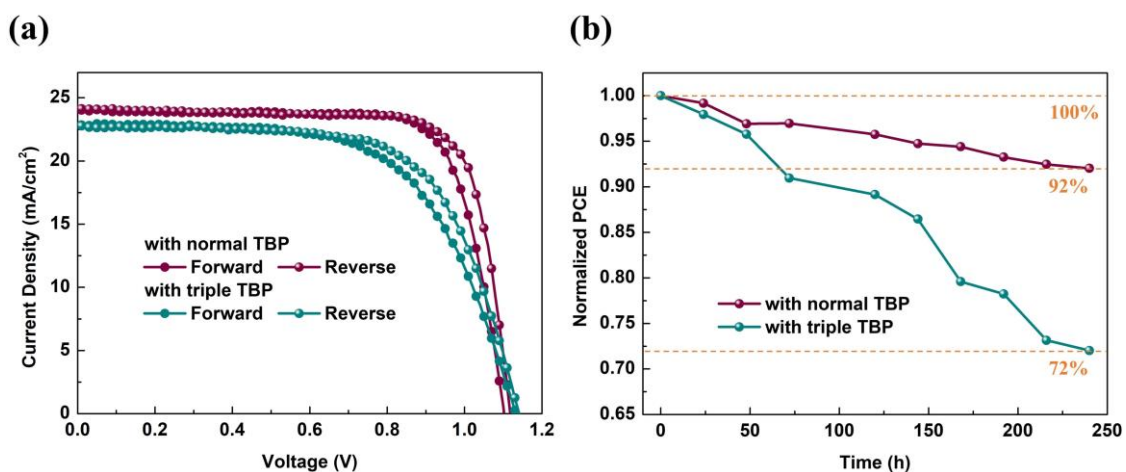
**At room temperature:** 2  $\mu\text{L}$  TBP was dropped on a metal substrate at room temperature, and the glass dish with the perovskite film on the top was immediately buckled on the TBP.



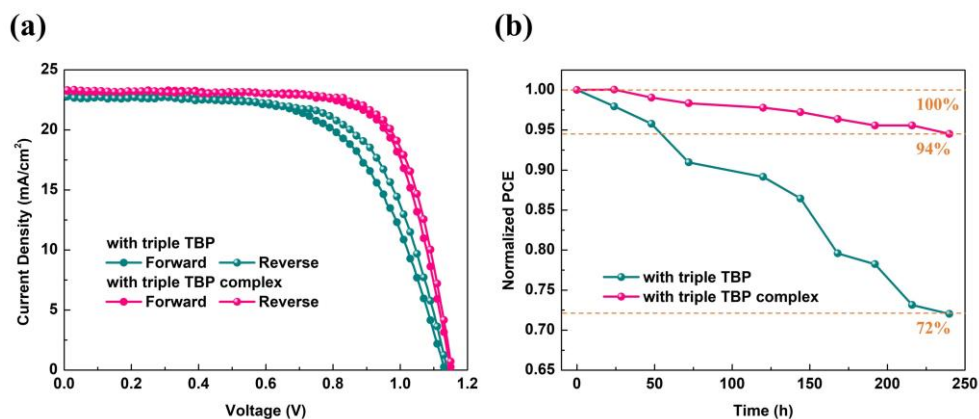
**Fig. S3** **a** SEM and **b** AFM images of the perovskite film corroded by TBP



**Fig. S4** SEM images of perovskite layers once **a** without and **b** with spiro spin-coating

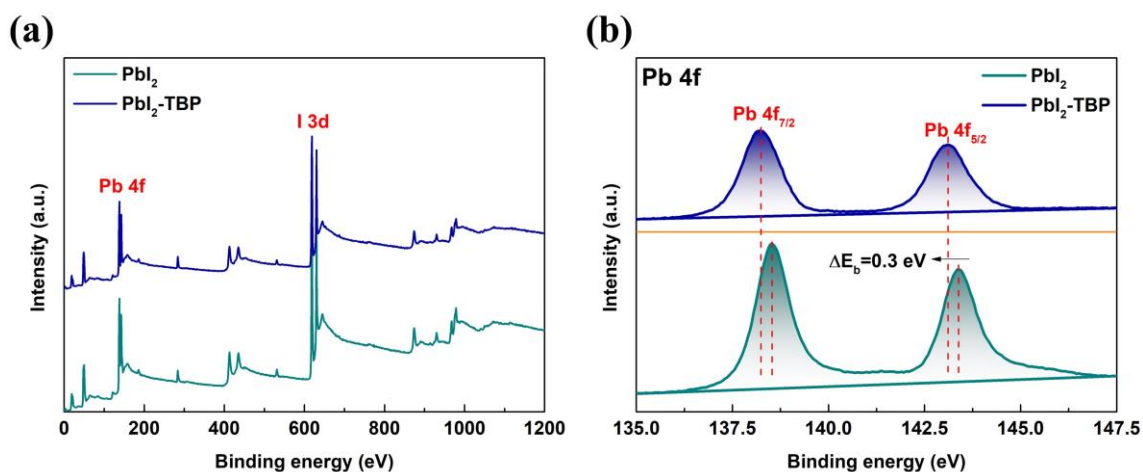


**Fig. S5** **a** J-V curves of the devices with normal and triple TBP measured by forward and reverse scans. **b** PCE evolution of the unencapsulated devices with normal and triple TBP content in  $\text{N}_2$  atmosphere at room temperature

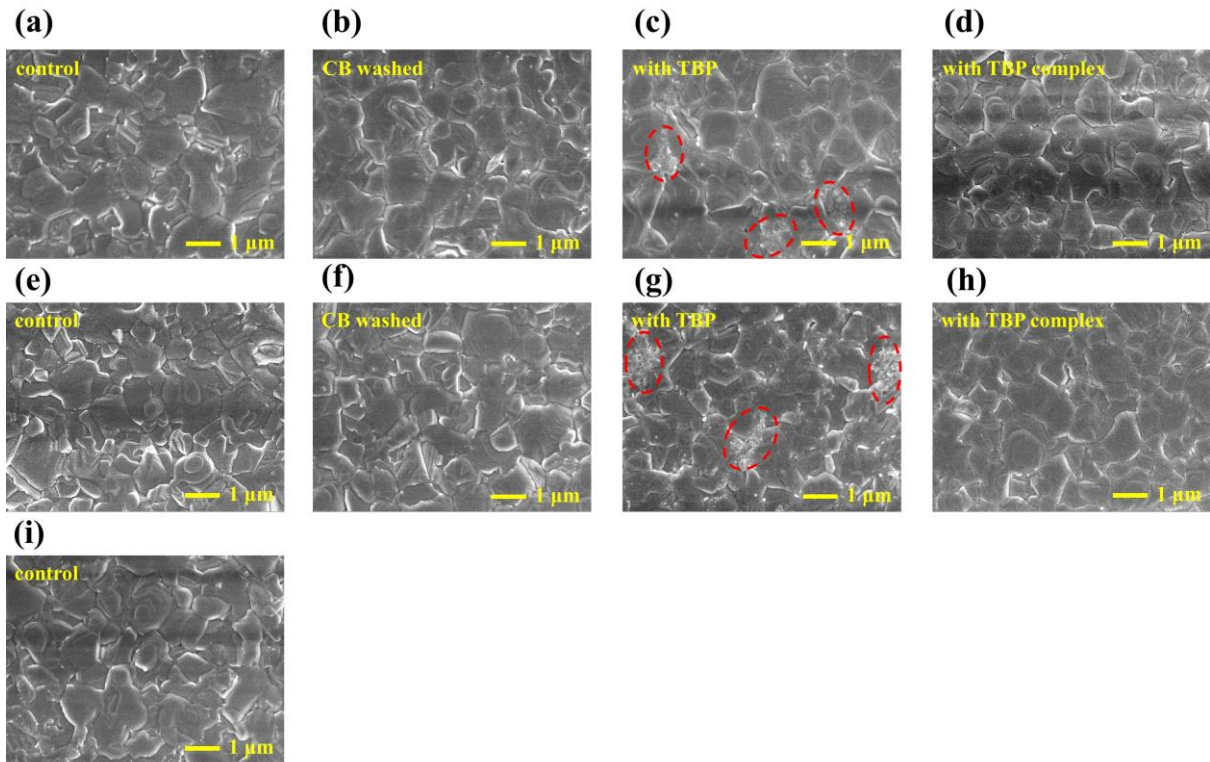


**Fig. S6** **a** J-V curves of the devices with triple TBP and triple TBP complex measured by forward and reverse scans. **b** PCE evolution of the unencapsulated devices with triple TBP and triple TBP complex in N<sub>2</sub> atmosphere at room temperature

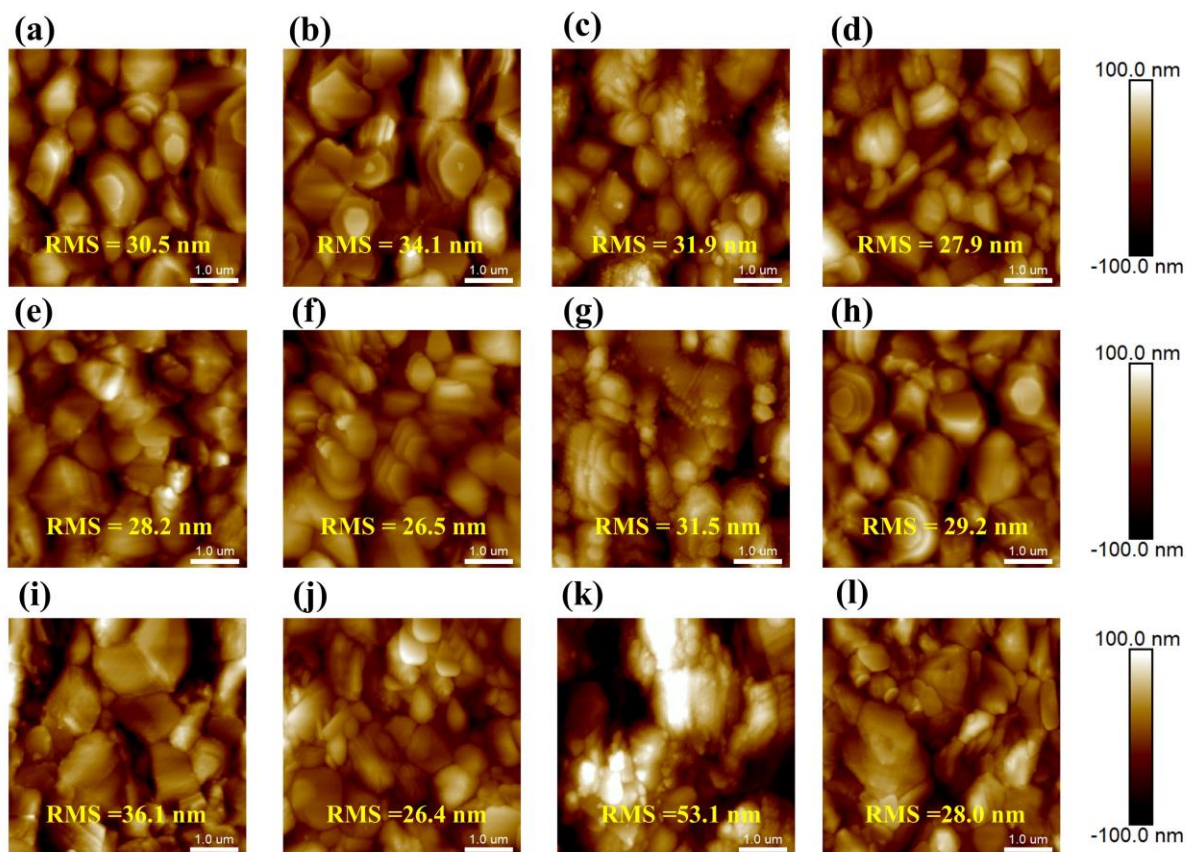
Although the TBP complex improves the performance and stability of the PSCs, the efficiency of the device with triple TBP is lower than that of the device with normal TBP because the three-fold usage is too high. In terms of the stability, the former is slightly better than that of latter.



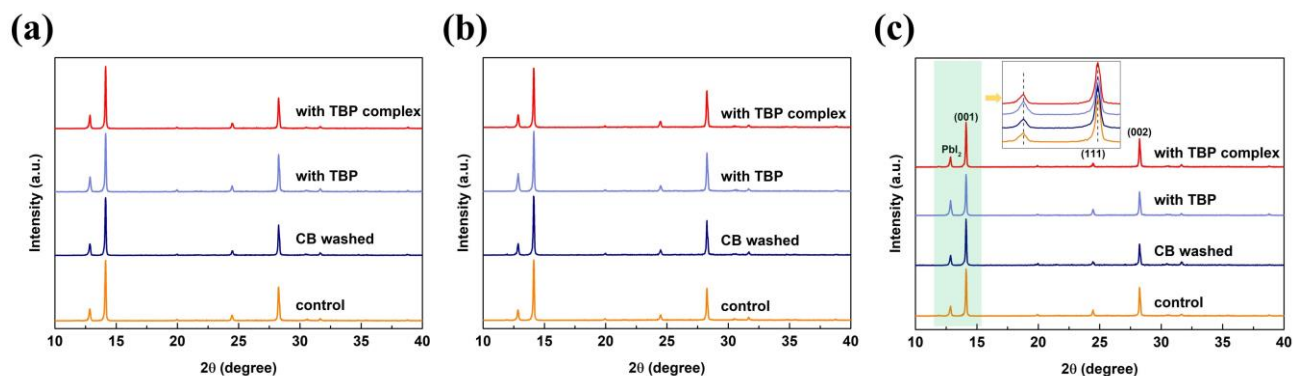
**Fig. S7** **a** XPS survey of the PbI<sub>2</sub> and PbI<sub>2</sub>-TBP. **b** High-resolution XPS spectra depicting the Pb 4f peaks of PbI<sub>2</sub> and PbI<sub>2</sub>-TBP



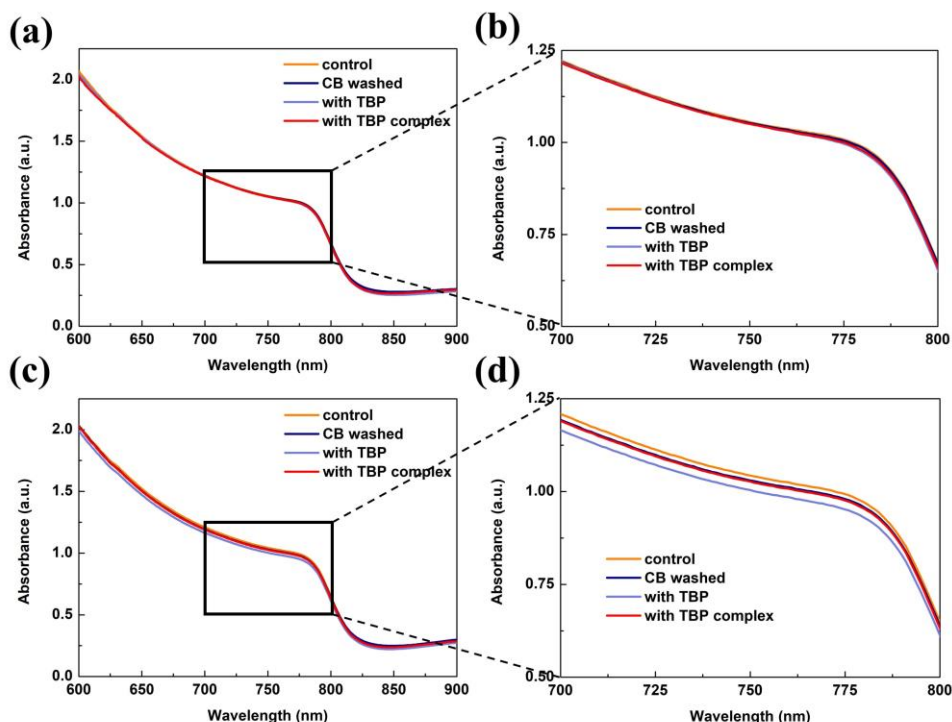
**Fig. S8** SEM images of the control film **a** without and **b** with CB washing, perovskite films once coated with spiro layer with **c** TBP and **d** TBP complex just after oxidation. SEM images of the control film **e** without and **f** with CB washing, perovskite films once coated with spiro layer with **g** TBP and **h** TBP complex after 3 days of storage in N<sub>2</sub>. **i** SEM image of the control film after 7 days of storage in N<sub>2</sub>



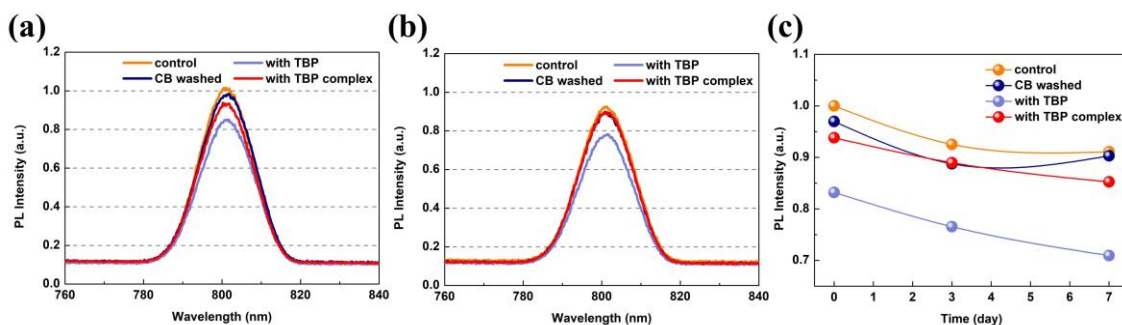
**Fig. S9** AFM images of the control film **a** without and **b** with CB washing, perovskite films once coated with spiro layer with **c** TBP and **d** TBP complex just after oxidation. AFM images of the control film **e** without and **f** with CB washing, perovskite films once coated with spiro layer with **g** TBP and **h** TBP complex after 3 days of storage in  $N_2$ . AFM images of the control film **i** without and **j** with CB washing, perovskite films once coated with spiro layer with **k** TBP and **l** TBP complex after 7 days of storage in  $N_2$



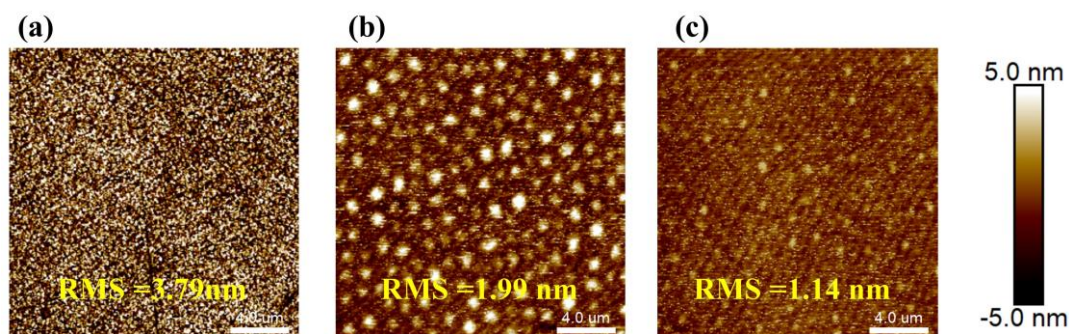
**Fig. S10** XRD patterns of the samples **a** just after oxidation, **b** stored in  $N_2$  for 3 days and **c** stored in  $N_2$  for 7 days, including the control film without and with CB washing, perovskite films once coated with spiro layer with TBP and TBP complex. The inset is the partial magnifications of the XRD spectra



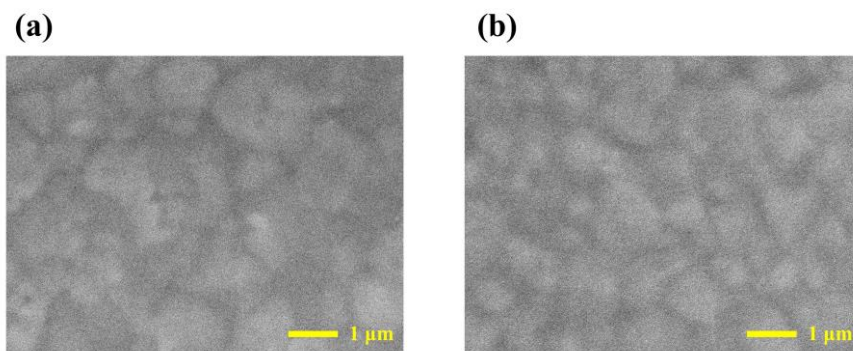
**Fig. S11** Absorption spectra of the samples **a** just after oxidation, **c** stored in  $N_2$  for 3 days, including the control film without and with CB washing, perovskite films once coated with spiro layer with TBP and TBP complex. **b** and **d** are the partial magnifications of the spectra



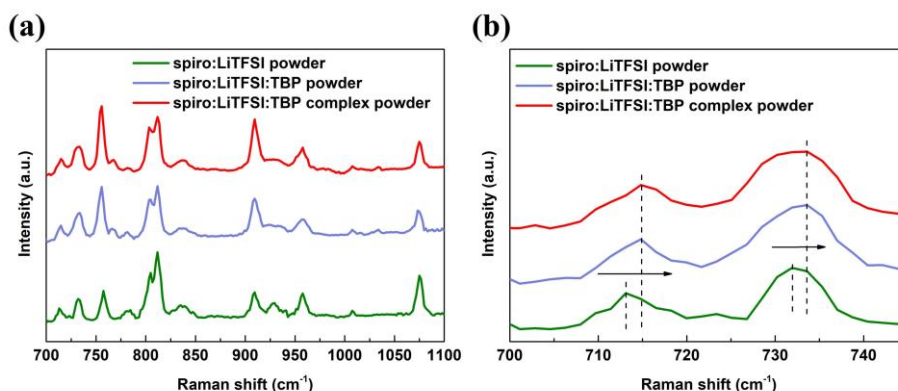
**Fig. S12** Steady-state PL spectra of the samples **a** just after oxidation, **b** stored in  $N_2$  for 3 days, including the control film without and with CB washing, perovskite films once coated with spiro layer with TBP and TBP complex. **c** PL intensity of these samples as the function of time



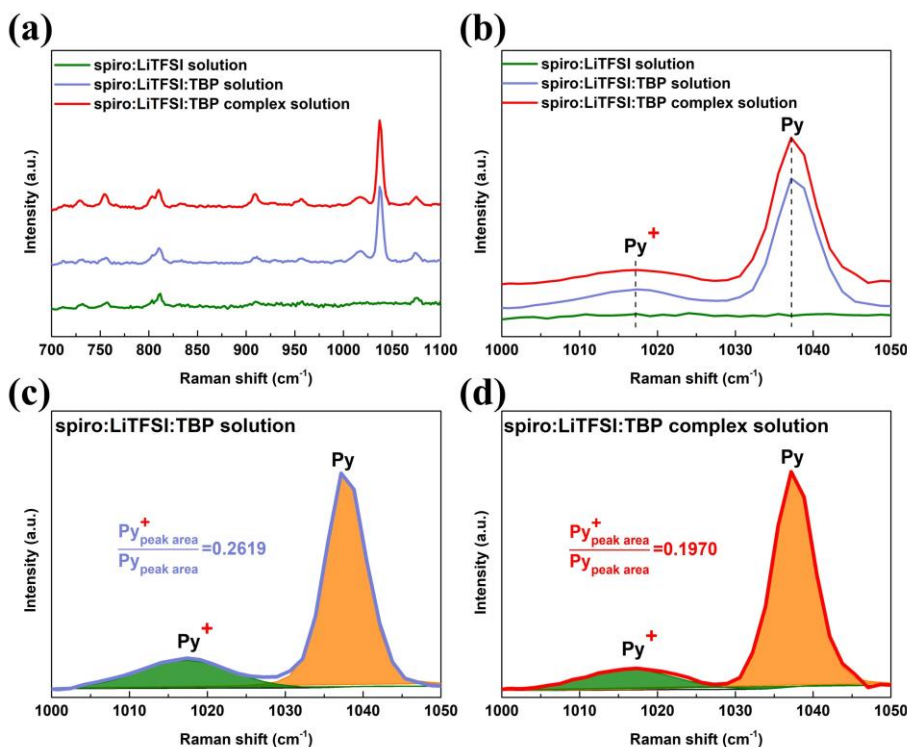
**Fig. S13** AFM images of the **a** ITO substrate, spiro films with **b** TBP and **c** TBP complex. The RMS of the ITO substrate is 3.79 nm. The spiro films with TBP or TBP complex were spin-coated on the ITO substrate directly because the RMS of the perovskite film is high.



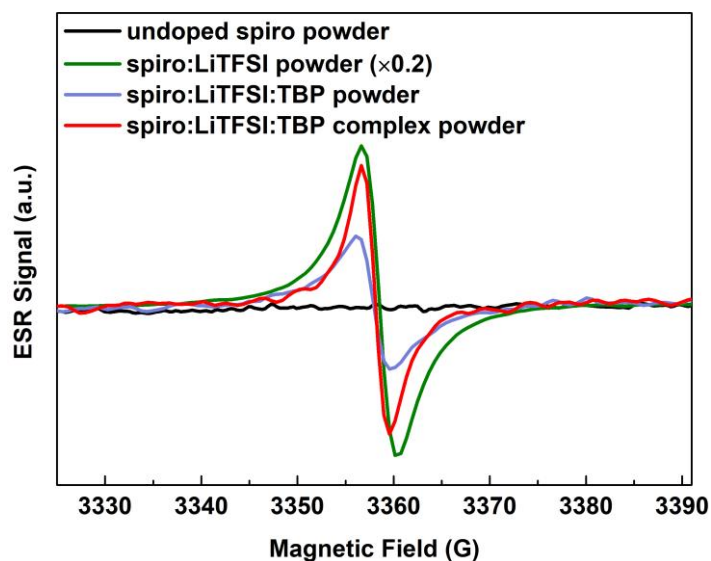
**Fig. S14** SEM images of the spiro film with **a** TBP and **b** TBP complex



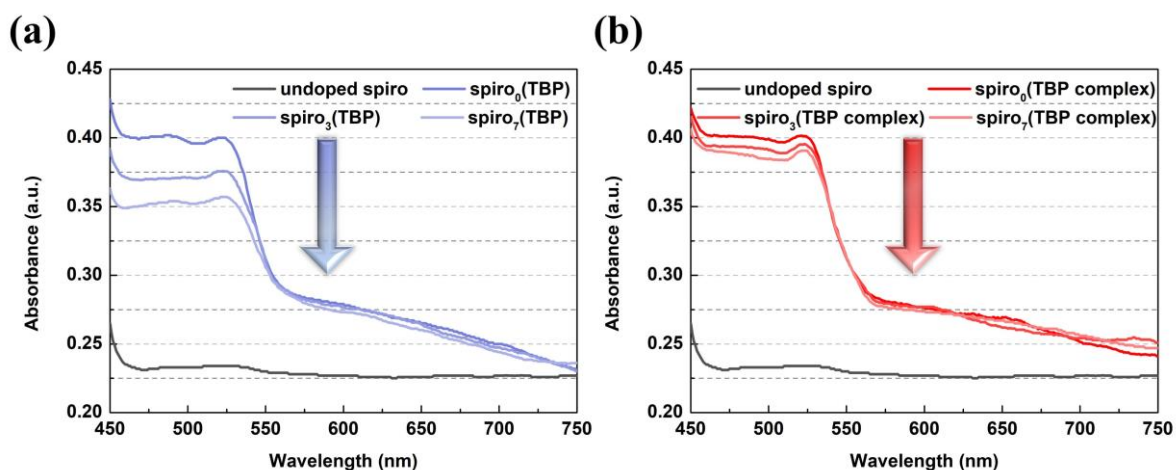
**Fig. S15** Raman spectra of the different spiro-based powders in the **a** full range and **b** 700-745 cm<sup>-1</sup> range. The peak around 732 cm<sup>-1</sup> is assigned to the a1 fundamental mode of fluorenes, whose shift represents the change of lattice parameters [S2]. The analysis for the peaks around 715 cm<sup>-1</sup> is difficult owing to the strong signal of TBP at this position [S3].



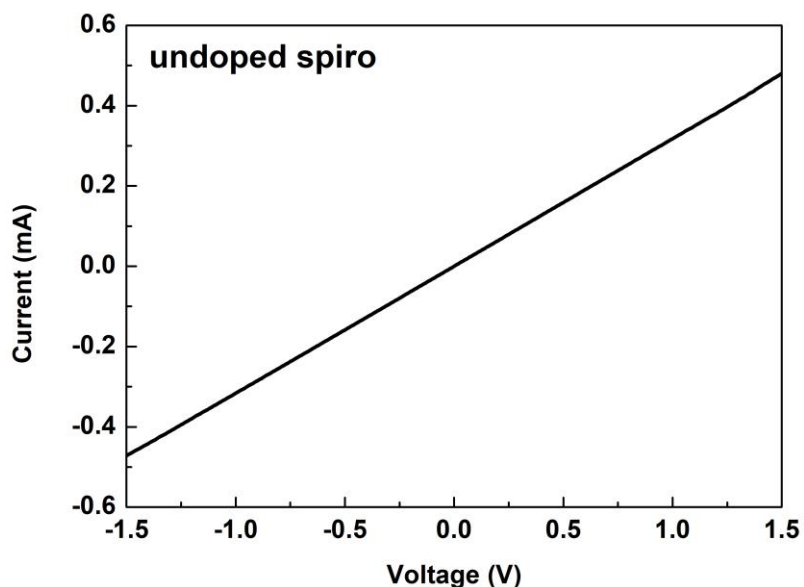
**Fig. S16** Raman spectra of the different spiro-based solutions in the **a** full range and **b** 1000-1050 cm<sup>-1</sup> range. The peak area ratio of Py<sup>+</sup> and Py for **c** spiro:LiTFSI:TBP solution and **d** spiro:LiTFSI:TBP complex solution



**Fig. S17** ESR spectroscopy of the different spiro-based powders. The spiro:LiTFSI signal is reduced by a factor of 5 to improve data visualization

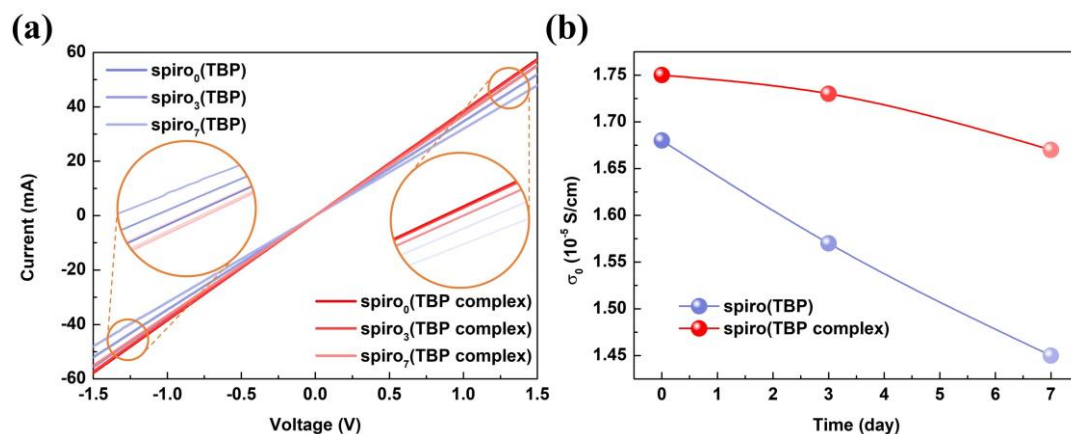


**Fig. S18** Absorption spectra of the different spiro solutions with **a** TBP and **b** TBP complex

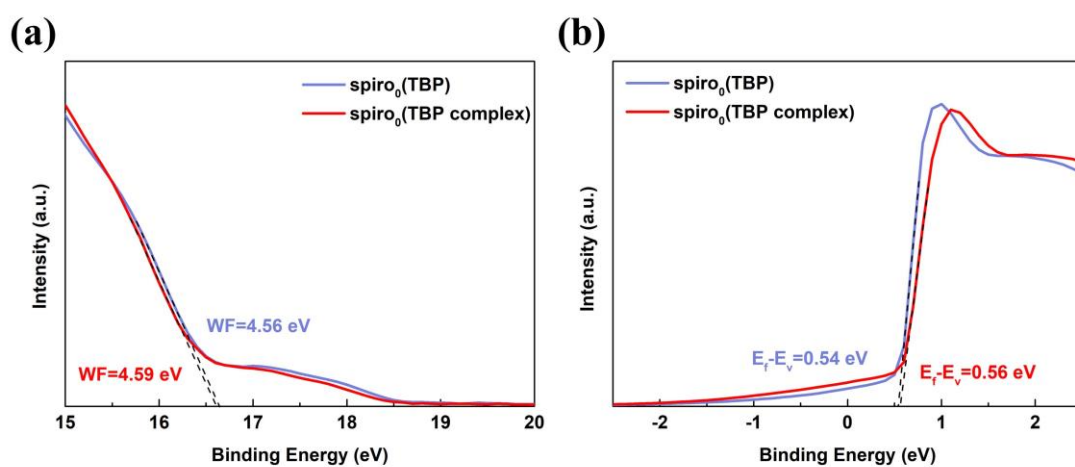


**Fig. S19** *I-V* curves of the device with the structure of ITO/undoped spiro/Ag

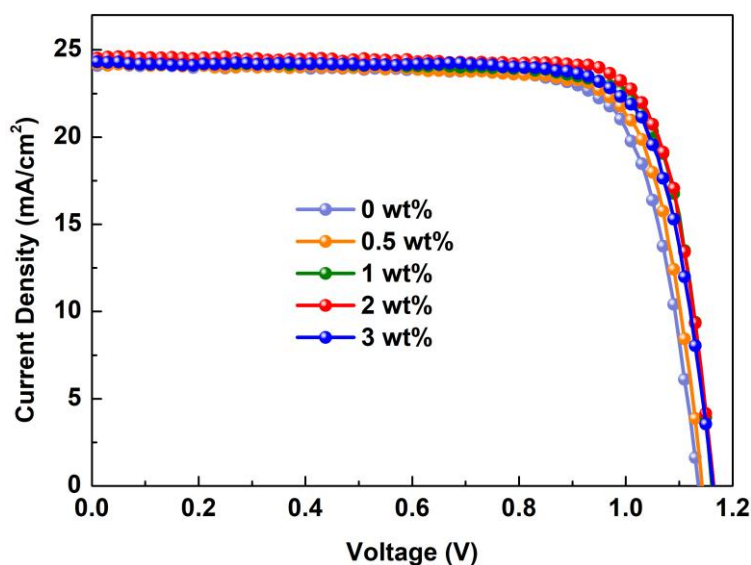




**Fig. S20** **a** *I-V* curves of the devices with the structure of ITO/different spiro-based films/Ag. **b** Variation of the  $\sigma_0$  of different spiro-based films with time



**Fig. S21** UPS spectra exhibiting the **a** secondary-electron cut-off and **b** valence band (VB) region of the spiro<sub>0</sub>(TBP) and spiro<sub>0</sub>(TBP complex) films. The UPS measurement is commonly utilized to obtain the work function (WF) and the difference between the Fermi level ( $E_f$ ) and the VB maximum for the film samples [S4]



**Fig. S22** *J-V* curves of the PSCs with different contents of 1,4-DITFB

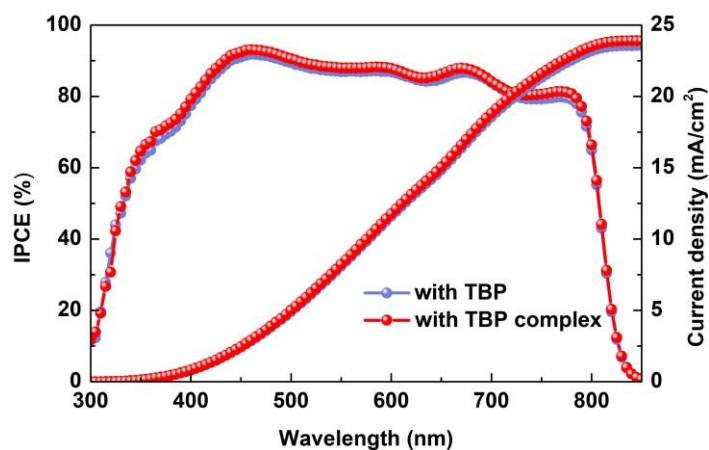


Fig. S23 IPCE spectra of the devices with TBP and TBP complex

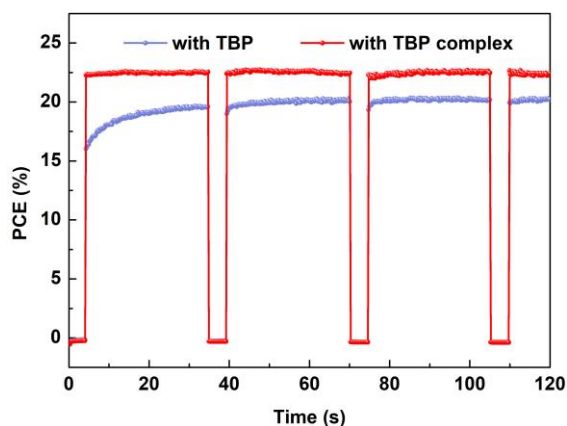


Fig. S24 PCE response tests of the devices with TBP and TBP complex

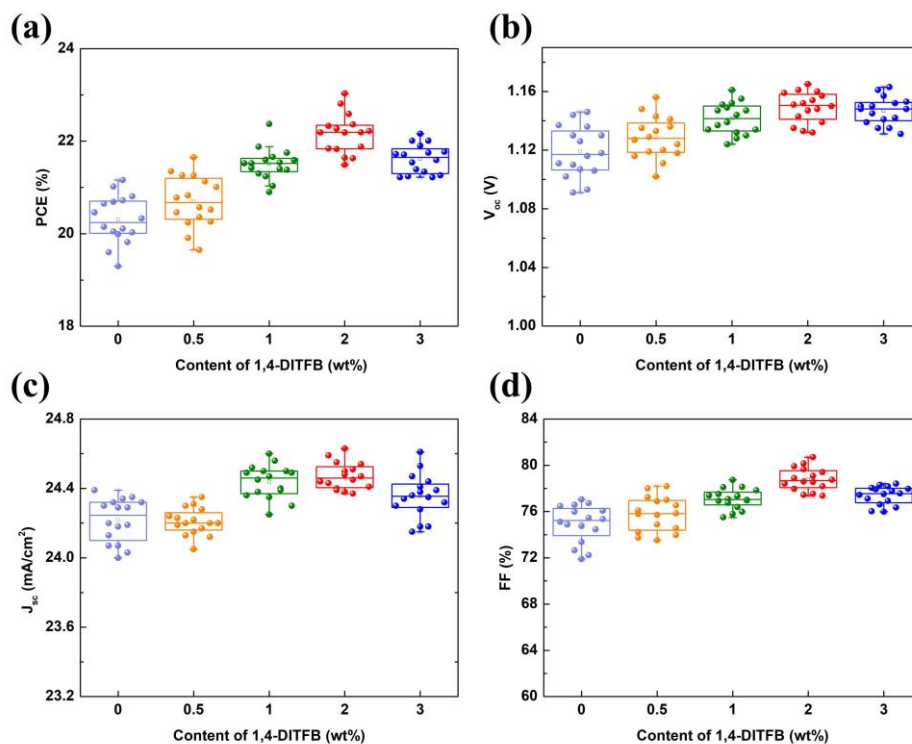
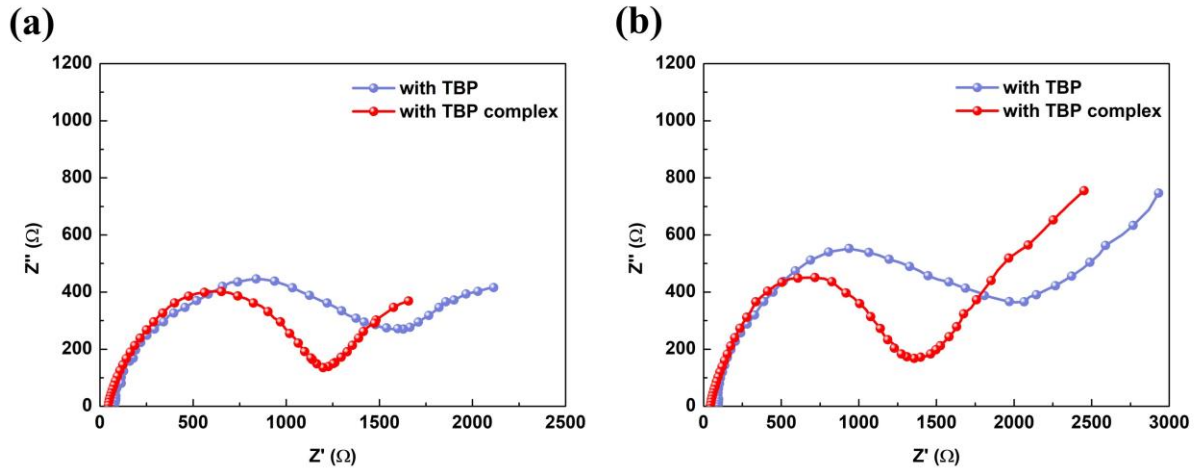
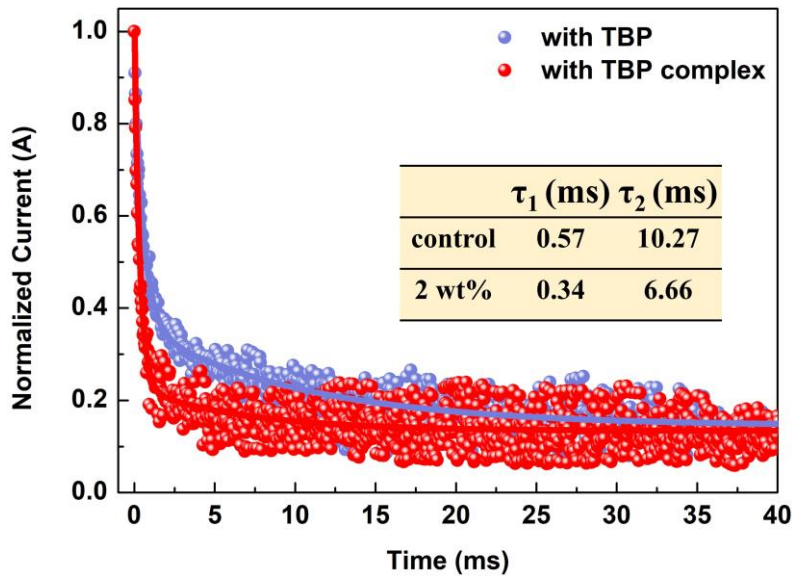


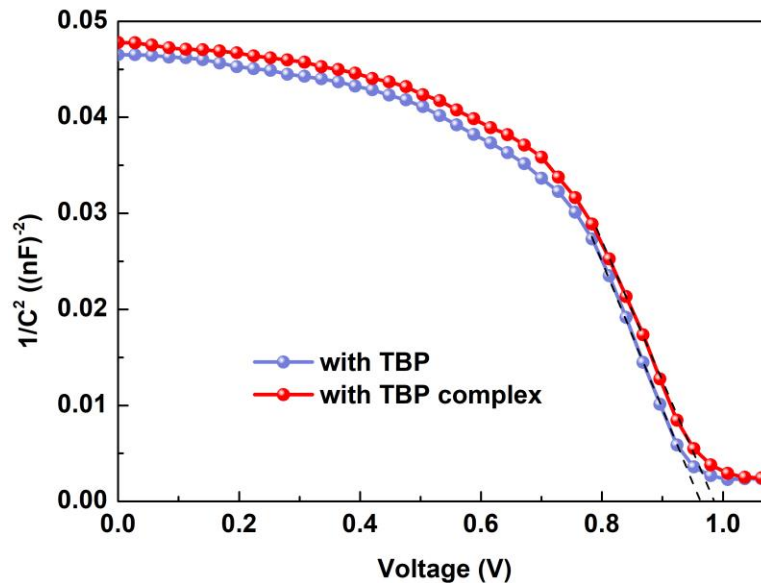
Fig. S25 Distribution of the **a** PCE, **b**  $V_{oc}$ , **c**  $J_{sc}$  and **d** FF of the devices with different content of 1,4-DITFB (16 individual cells are included)



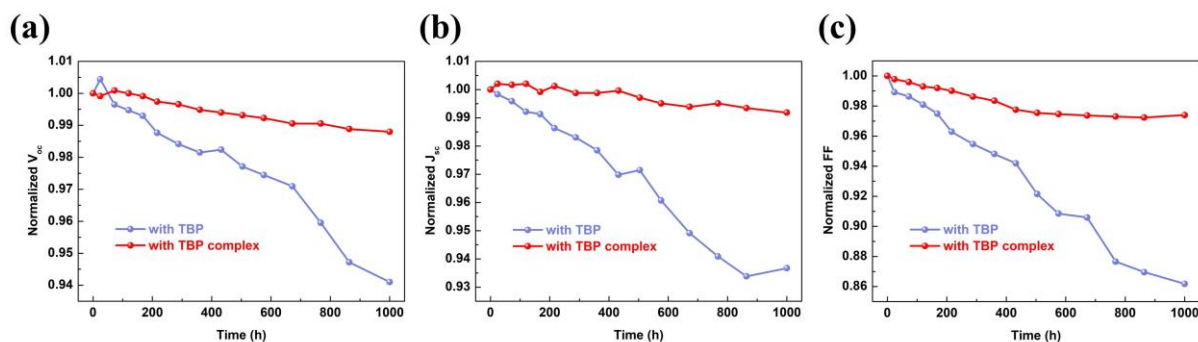
**Fig. S26** Nyquist plots of the devices with TBP and TBP complex measured under **a** 0.6 V bias voltage and **b** 0.4 bias voltage



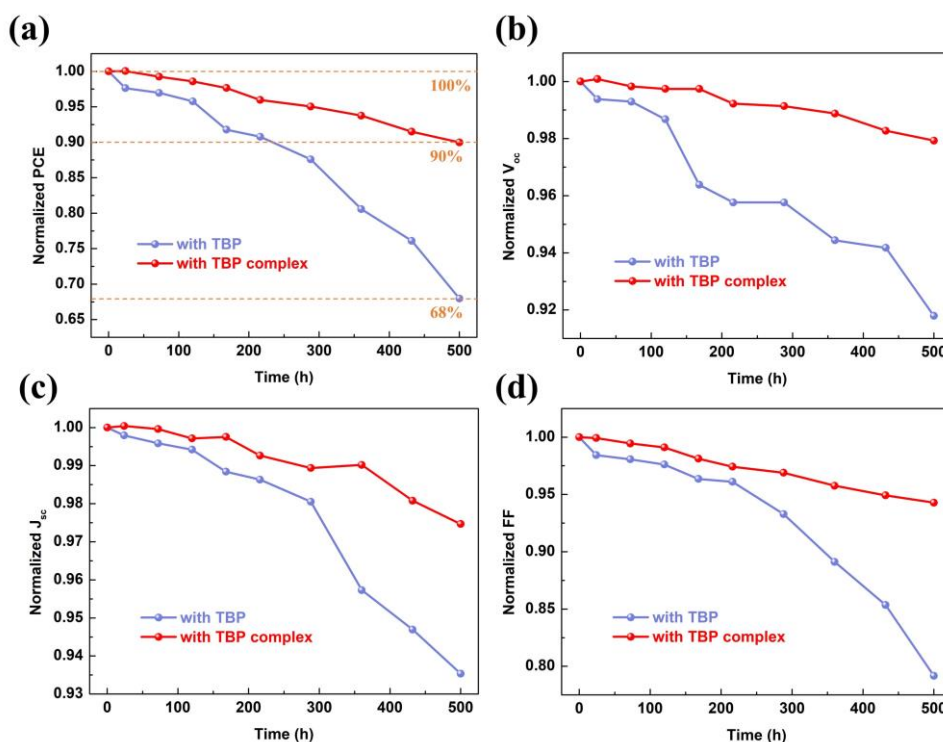
**Fig. S27** TPC based on the devices with TBP and TBP complex



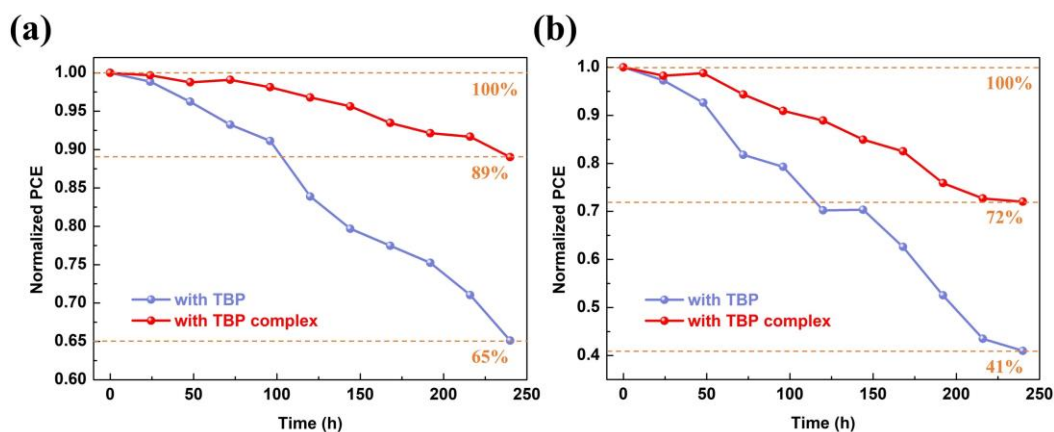
**Fig. S28** C-V characteristic curves of the devices with TBP and TBP complex



**Fig. S29** **a**  $V_{oc}$ , **b**  $J_{sc}$  and **c** FF evolution of the devices with TBP and TBP complex without encapsulation in  $N_2$  atmosphere at room temperature



**Fig. S30** **a** PCE, **b**  $V_{oc}$ , **c**  $J_{sc}$  and **d** FF evolution of the devices with TBP and TBP complex without encapsulation in  $40 \pm 5\%$  RH at room temperature



**Fig. S31** PCE evolution of the devices with TBP and TBP complex without encapsulation in **a**  $60 \pm 5\%$  RH and **b** about 85% RH at room temperature

In stability tests, the Ag electrode was replaced with the Au electrode to exclude the influence of I<sup>-</sup> ions migration and their reaction with Ag [S5].

**Table S1** The photovoltaic parameters of the PSCs with normal TBP, triple TBP and triple TBP complex measured under different scanning directions, including open-circuit voltage ( $V_{oc}$ ), short-circuit current density ( $J_{sc}$ ), fill factor (FF) and PCE

Device	scanning direction	$V_{oc}$ (V)	$J_{sc}$ (mA/cm <sup>2</sup> )	FF (%)	PCE (%)
normal TBP	reverse	1.118	24.05	77.10	20.73
	forward	1.102	24.01	76.01	20.11
triple TBP	reverse	1.141	22.78	65.60	17.05
	forward	1.133	22.77	62.25	16.06
triple TBP complex	reverse	1.154	23.36	73.66	19.86
	forward	1.152	23.29	72.34	19.41

**Table S2** The variation of the  $\sigma_0$  of different spiro-based films with time

sample	Time (day)		
	0	3	7
spiro	$1.68 \times 10^{-5}$	$1.57 \times 10^{-5}$	$1.45 \times 10^{-5}$
spiro(1,4-DITFB)	$1.75 \times 10^{-5}$	$1.73 \times 10^{-5}$	$1.67 \times 10^{-5}$
undoped spiro	$1.44 \times 10^{-7}$	$1.44 \times 10^{-7}$	$1.44 \times 10^{-7}$

**Table S3** The champion photovoltaic parameters of the PSCs with different content of 1,4-DITFB

Content (wt%)	$V_{oc}$ (V)	$J_{sc}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)
0	1.136	24.18	77.05	21.16
0.5	1.143	24.22	78.19	21.65
1	1.161	24.47	78.74	22.37
2	1.165	24.50	80.70	23.03
3	1.163	24.38	78.17	22.16

**Table S4** The photovoltaic parameters of the champion PSCs with TBP and TBP complex

Device	scanning direction	$V_{oc}$ (V)	$J_{sc}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)	HI (%)	$J_{sc}$ from IPCE
with TBP	reverse	1.136	24.18	77.05	21.16	4.58	23.56
	forward	1.116	24.15	74.92	20.19		
with TBP complex	reverse	1.165	24.50	80.70	23.03	1.09	23.90
	forward	1.162	24.50	80.03	22.78		

## Supplementary References

- [S1] T. Matsui, T. Yamamoto, T. Nishihara, R. Morisawa, T. Yokoyama et al., Compositional engineering for thermally stable, highly efficient perovskite solar cells exceeding 20% power conversion efficiency with 85 °C/85% 1000 h stability. *Adv. Mater.* **31**(10), 1806823 (2019). <https://doi.org/10.1002/adma.201806823>
- [S2] S.Y. Lee, B.H. Boo, Density functional theory study of vibrational spectra of fluorene. *J. Phys. Chem.* **100**(21), 8782-8785 (1996). <https://doi.org/10.1021/jp960020g>
- [S3] S. Yurdakul, M. Bahat, Fourier transform infrared and Raman spectroscopic studies on 4-tert.-butylpyridine and its metal(II) tetracyanonickelate complexes. *J. Mol. Struct.* **412**(1),

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[S4]X. Liu, Z. Liu, B. Sun, X. Tan, H. Ye et al., 17.46% efficient and highly stable carbon-based planar perovskite solar cells employing Ni-doped rutile TiO<sub>2</sub> as electron transport layer. *Nano Energy* **50**, 201-211 (2018). <https://doi.org/10.1016/j.nanoen.2018.05.031>

[S5]Y. Kato, L.K. Ono, M.V. Lee, S. Wang, S.R. Raga et al., Silver iodide formation in methyl ammonium lead iodide perovskite solar cells with silver top electrodes. *Adv. Mater. Interfaces* **2**(13), 1500195 (2015). <https://doi.org/10.1002/admi.201500195>