

Supporting Information for

Origin of Excellent Charge Storage Properties of Defective Tin Disulfide in Magnesium/Lithium-Ion Hybrid Batteries

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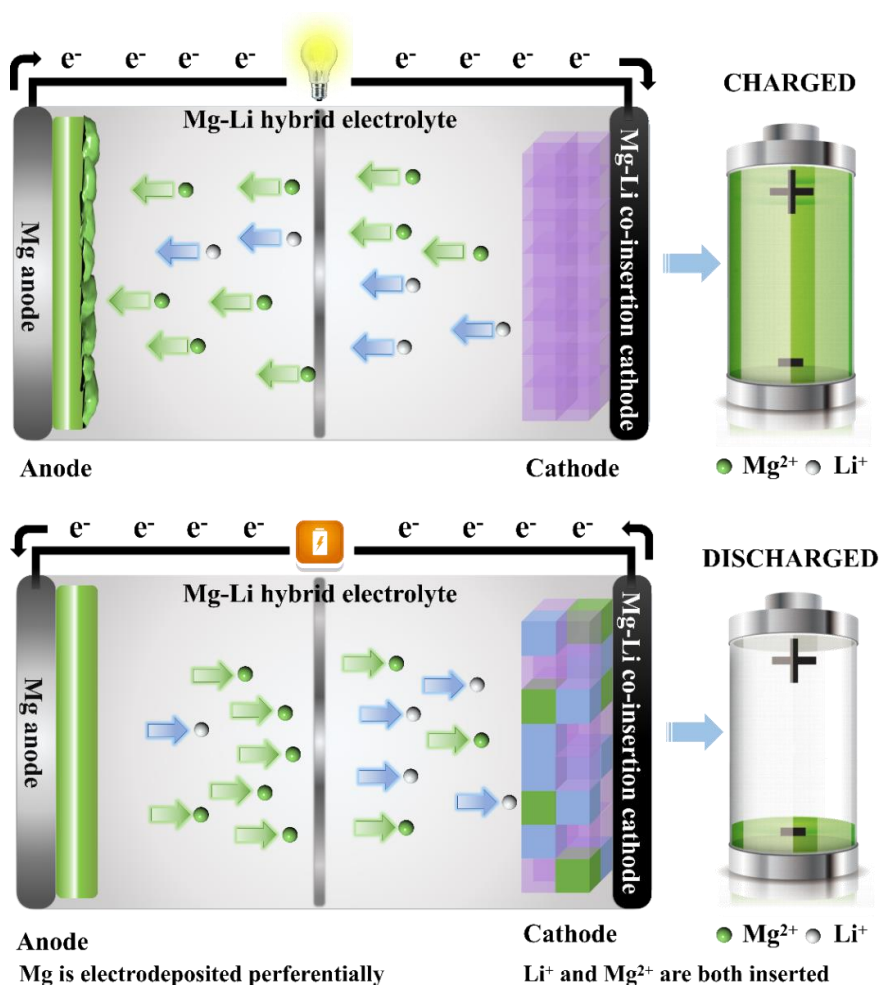
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Supplementary Figures and Tables



Scheme S1 Schematic illustration of the movement of Mg^{2+} and Li^+ during charge and discharge in a MLHB cell with Mg anode and a Mg^{2+}/Li^+ co-intercalation cathode

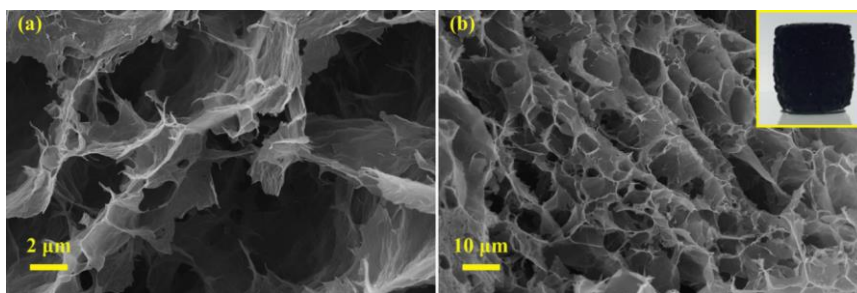


Fig. S1 FESEM images of holey graphene foams (HGF)

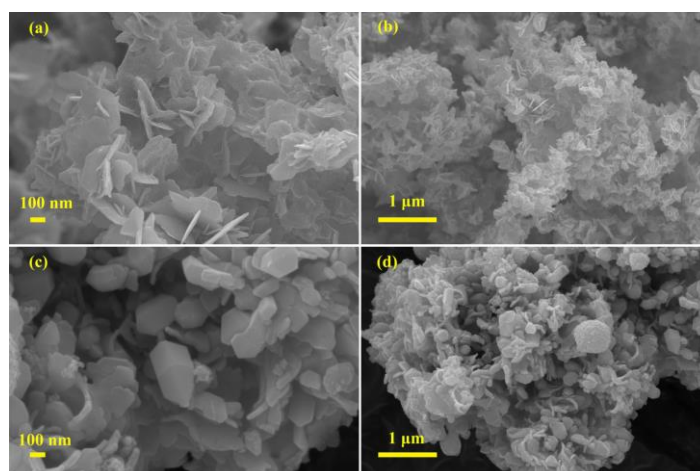


Fig. S2 FESEM images at different magnifications of SnS₂ (a, b) and SnS (c, d)

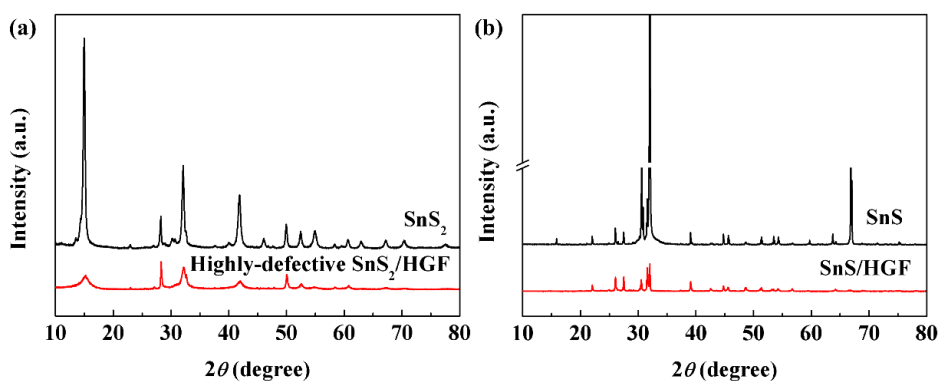


Fig. S3 XRD patterns of (a) SnS₂ and highly-defective SnS₂/HGF and of (b) SnS and defect-free SnS/HGF (b)

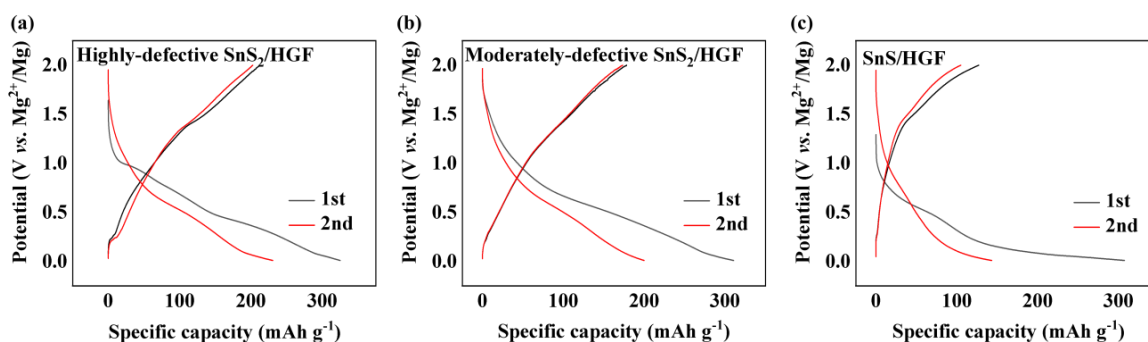


Fig. S4 Galvanostatic discharge and charge profiles of (a) highly-defective SnS₂/HGF, (b) moderately-defective SnS₂/HGF and (c) defect-free SnS/HGF cycled in a MIB at 50 mA g⁻¹ in the voltage window between 0.01 and 2.0 V vs. Mg²⁺/Mg

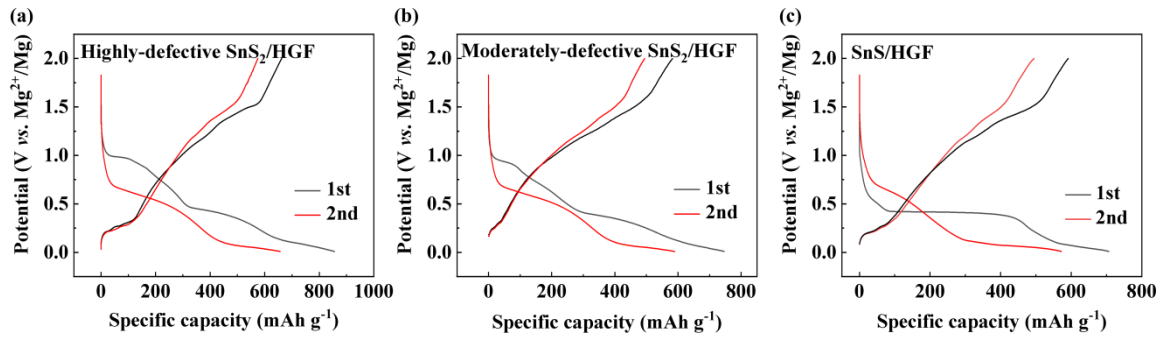


Fig. S5 Galvanostatic discharge and charge profiles of (a) highly-defective SnS₂/HGF, (b) moderately-defective SnS₂/HGF and (c) defect-free SnS/HGF cycled in a MLHB at 50 mA g⁻¹ in the voltage range 0.01-2.0 V vs. Mg²⁺/Mg

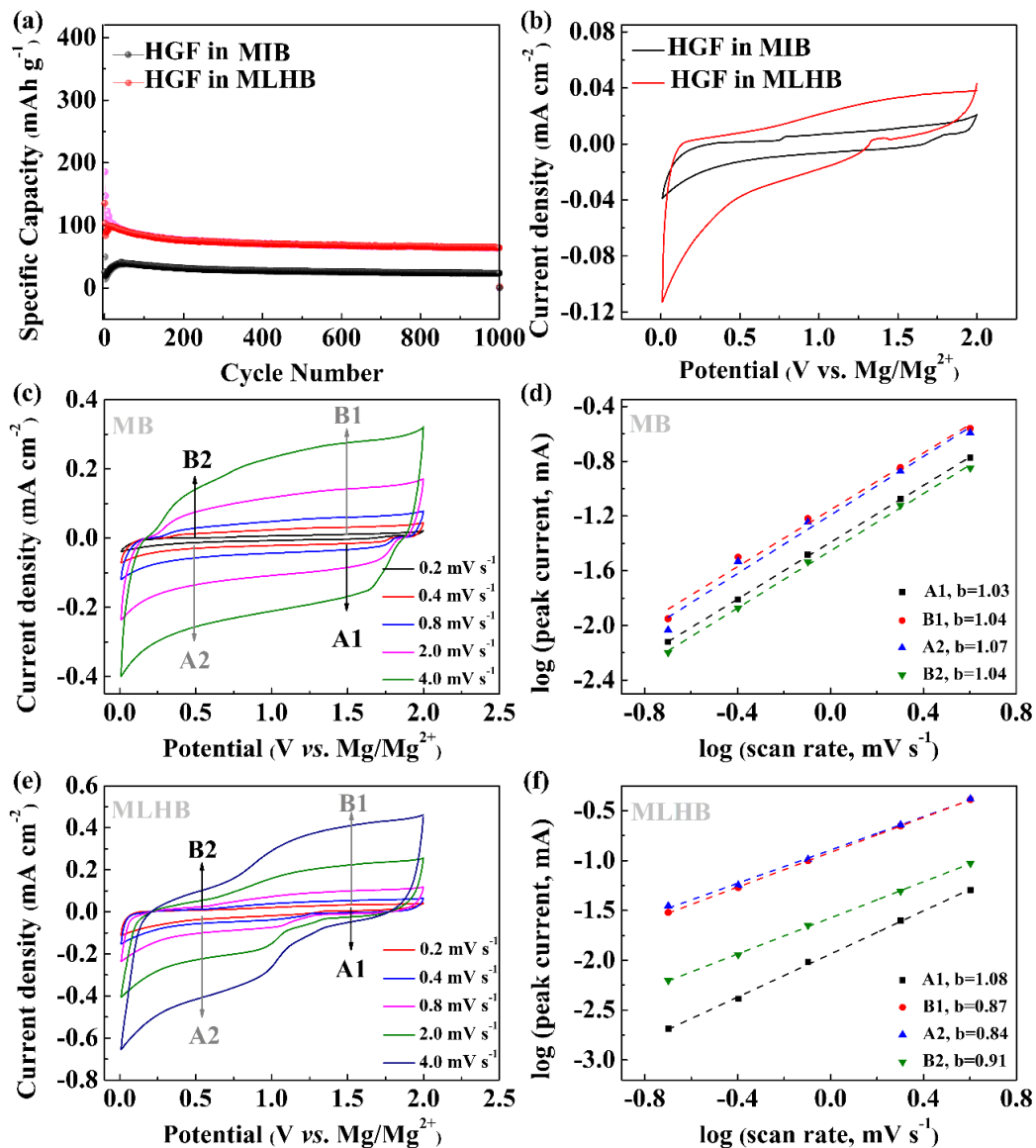


Fig. S6 (a) Cycling performance and (b) CV curves of HGF in MIBs and MLHBs, showing negligible capacity contribution from HGF. (c-f) CV curves at different scan rates to determine b -values at different potentials vs. Mg²⁺/Mg in MIBs (c, d) and MLHBs (e, f). Constant b can be obtained by plotting $\log i(V)$ vs. $\log v$ according to $i(V) = av^b$, where Constant a is a constant, indicating a capacity contribution mainly from charge transfer with surface/subsurface atoms

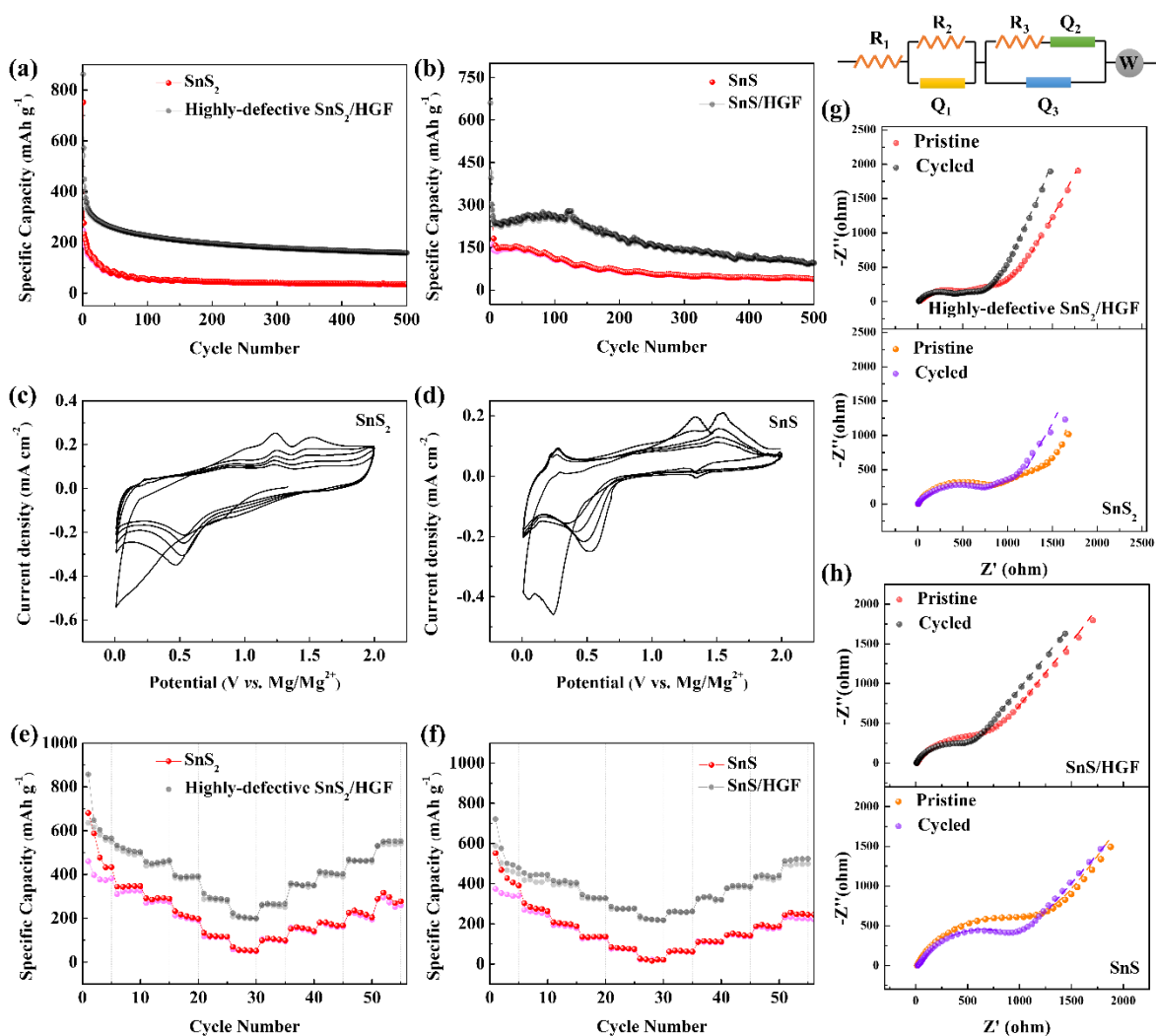


Fig. S7 Comparison of cycling stability between SnS₂ and highly-defective SnS₂/HGF (a) and between SnS and SnS/HGF (b) at 800 mA g⁻¹ in the voltage range between 0.01 and 2.0 V vs. Mg²⁺/Mg. Cyclic voltammograms of SnS₂ (c) and SnS (d) at the scan rate of 0.2 mV s⁻¹. Comparison of rate performance between SnS₂ and highly-defective SnS₂/HGF (e) and between SnS and SnS/HGF (f). Comparison of Nyquist plots between SnS₂ and highly-defective SnS₂/HGF (g) and between SnS and SnS/HGF (h) before and after first cycle at the OCP. The dotted lines were fitted data using the equivalent circuit shown above g

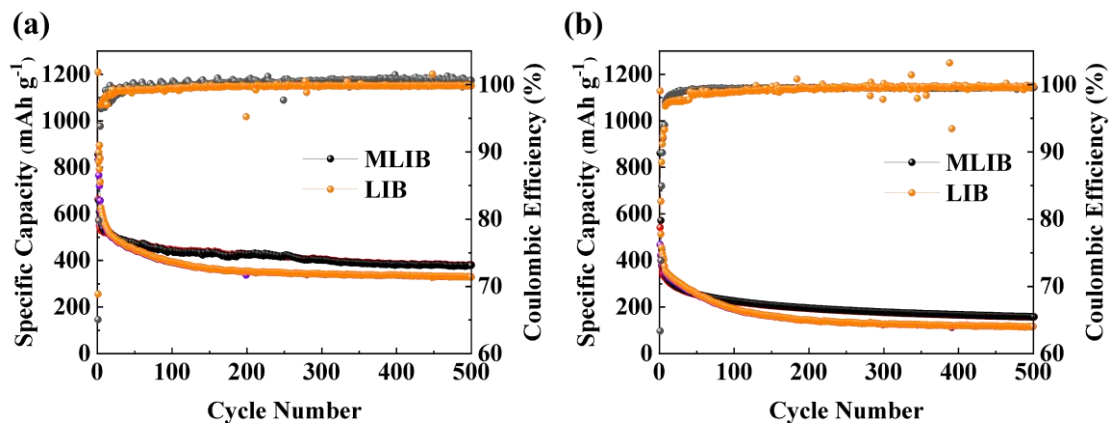


Fig. S8 Cycling performance of highly-defective SnS₂/HGF in LIB in the potential window range between 0.8 and 2.8 V vs. Li⁺/Li and in MLIB in the potential windows between 0.01 and 2.0 V vs. Mg²⁺/Mg at 50 mA g⁻¹ (a) and 800 mA g⁻¹ (b)

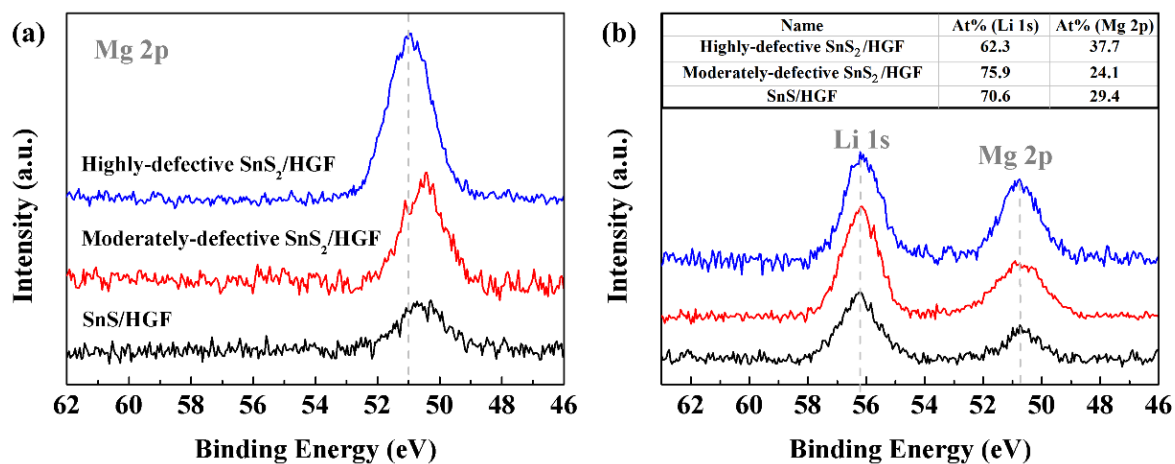


Fig. S9 Mg 2p and Li 1s XPS spectra for fully discharged defective SnS₂/HGF and defect-free SnS/HGF at 0.01 V vs. Mg²⁺/Mg in a MIB cell (a) and a MLHB cell (b)

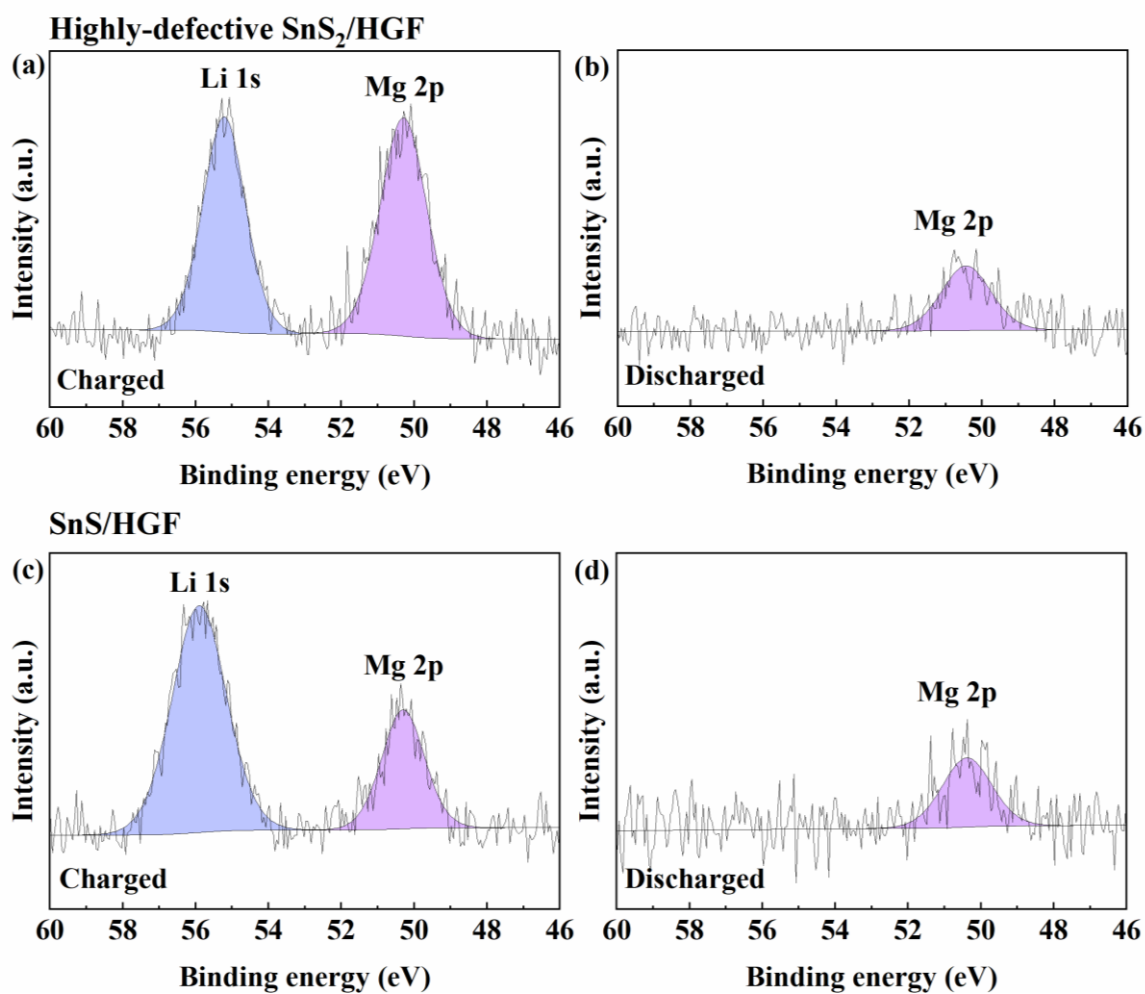


Figure S10 *Ex-situ* XPS spectra represent changes of Mg 2p and Li 1s of the (a, b) highly-defective SnS₂/HGF and (c, d) SnS/HGF after the first cycle at charged and discharged stages, respectively

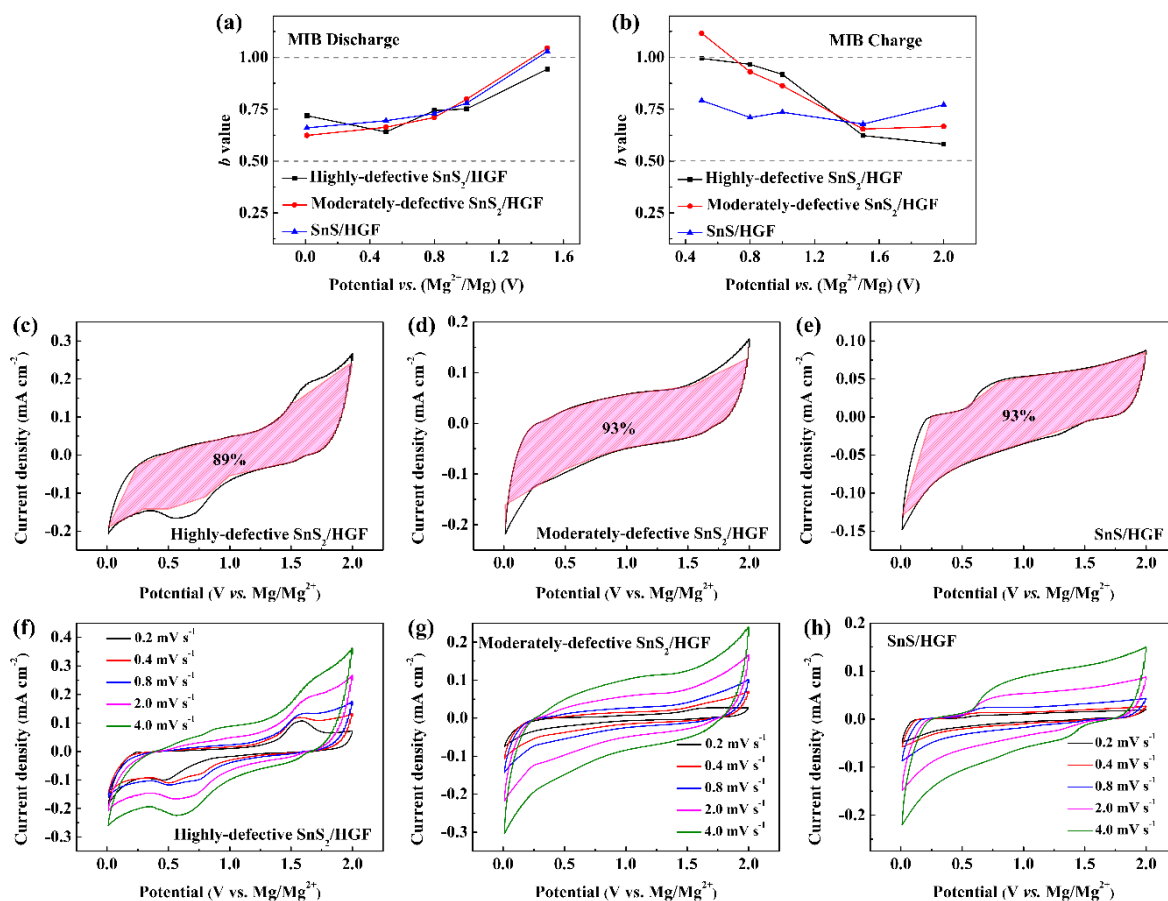


Fig. S11 Kinetics and quantitative analysis of SnS_x/HGF in MIBs. b -values of highly-defective SnS_2/HGF , moderately-defective SnS_2/HGF and SnS/HGF at different potentials vs. Mg^{2+}/Mg during (a) discharging and (b) charging processes. CV curves of (c) highly-defective SnS_2/HGF , (d) moderately-defective SnS_2/HGF and (e) SnS/HGF at various scan rates, and the corresponding capacitive charge storage contributions of (f) highly-defective SnS_2/HGF , (g) moderately-defective SnS_2/HGF and (h) SnS/HGF at a scan rate of 2.0 mV s^{-1}

Table S1 Comparison of MLHB cell performance of different electrodes

Sample	Electrolyte	Capacity (mAh g^{-1})/current density (mA g^{-1})	Refs.
MoS_2	0.25 M LiCl + 0.25 M APC ^a	301/20	[S1]
TiO_2	1.5 M LiBH_4 + 0.5 M $\text{Mg}(\text{BH}_4)_2/\text{TGM}^b$	140/20	[S2]
$\text{Li}_4\text{Ti}_5\text{O}_{12}/\text{Graphene}$	1.5 M LiBH_4 +0.4 M APC	147.5/20	[S3]
TiS_2	0.4 M LiCl +0.4 M APC	161/20	[S4]
Mo_6S_8	0.4 M LiCl +1 M APC	126/20	[S5]
MoSe_2/C	0.5 M LiCl +0.2 M APC	204/50	[S6]
VO_2	1 M LiCl +0.25 M APC	210.6/50	[S7]
$\text{VS}_2\text{-GO}$	1 M APC-LiC	235/90	[S8]
d- $\text{Ti}_3\text{C}_2\text{T}_x/\text{CNT}$	0.4 M LiCl +0.5 M APC	105/10	[S9]

TiNb ₂ O ₇	1.25 M LiCl + 0.4 M APC	240/7.75	[S10]
Cu ₂ Se	1 M LiCl + 0.4 M APC	239.7/26	[S11]
Li ₃ V ₂ (PO ₄) ₃	1 M LiCl + 0.4 M APC	147.8/50	[S12]
Li ₄ Mn ₅ O ₁₂	1 M LiCl + 0.25 M APC	155/16.3	[S13]
FeS ₂	1.5 M LiBH ₄ + 0.1 M Mg(BH ₄) ₂ (DGM ^c)	600/45	[S14]
FeS/CNF	0.4 M LiCl + 0.4 M APC	463/70	[S15]
Cu ₉ S ₅ -AEHPA ^d	1 M LiTFSI + 0.2 M Mg(HMDS) ₂ ^e -AlCl ₃ -MgCl ₂ (DGM)	280/50	[S16]
LiV ₃ O ₈ @GO	1 M LiCl + APC	245.9/50	[S17]
Cu ₂ Se/rGO	1 M LiCl + 0.4 M APC	243/26	[S18]
TiNb ₂ O ₇	1.5 M LiCl + 0.4 M APC	225/7.75	[S19]
Cu ₂ S@C	1 M LiCl + 0.4 M APC	393.2/16.84	[S20]
Ni-doped MnO ₂ /CNT	1 M LiCl + 0.4 M APC	175/20	[S21]
(NiMnCo) ₃ O ₄	1 M LiCl + 0.4 M APC	550/50	[S22]
Na ₂ C ₆ O ₆	1 M LiCl + 0.25 M APC	350/50	[S23]
SnS ₂ /HGF	0.25 M LiCl + 0.25 M APC	600/50	This work
SnS/HGF	0.25 M LiCl + 0.25 M APC	520/50	

* ^aAPC: All-phenyl complex. ^bTGM: Tetraglyme. ^cDGM: Diglyme. ^dAEHPA: Amino-ended hyperbranched polyamide. ^eMg(HMDS)₂: Bis-hexamethyl-disilazide magnesium.

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