

Supporting Information for

High-Transconductance, Highly Elastic, Durable and Recyclable All-Polymer Electrochemical Transistors with 3D Micro-Engineered Interfaces

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Supplementary Figures and Table

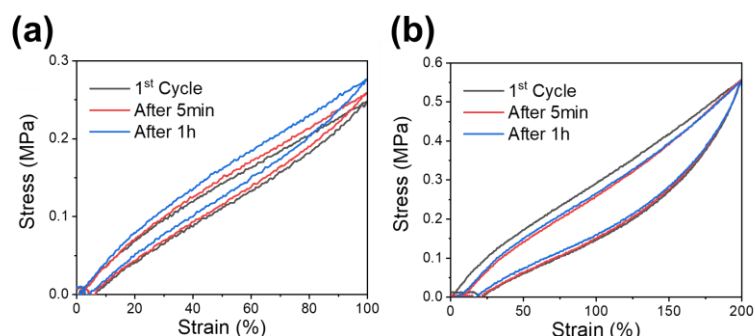


Fig. S1 Cyclic loading–unloading curves at 100% (a) and 200% (b) strains for 25% GEL–60% GLY/Na₃Cit, with diminished residual strain and recovery of dissipated energy observed after 5 min

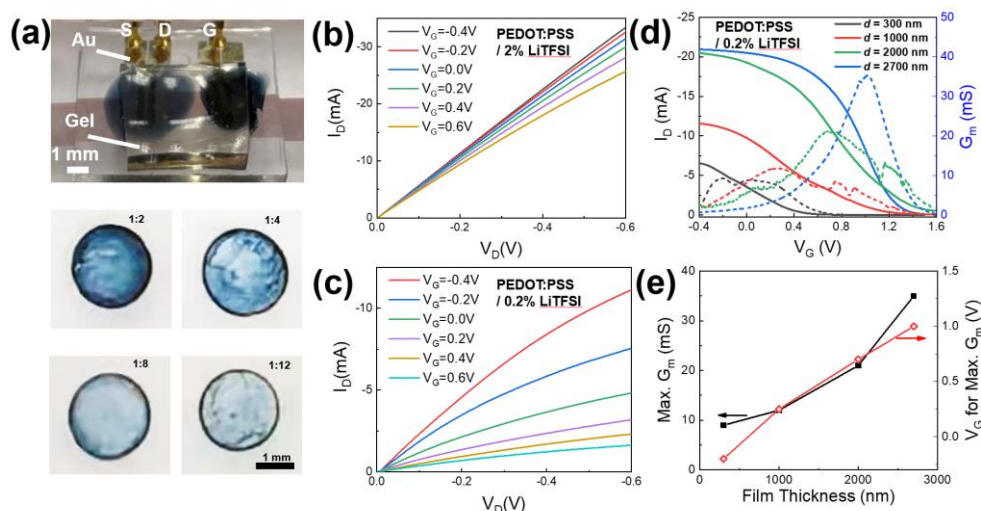


Fig. S2 a Photos of the OECT measurement on Au electrodes ($W/L = 5000/200 \mu\text{m}$) and PEDOT:PSS/LiTFSI with varied thickness through controlling the dilution ratio. b–c Typical output curves of PEDOT:PSS doped with 2 wt% LiTFSI (b) and 0.2 wt% LiTFSI (c) with

similar thickness (1:8 dilution). **d** Control of the transfer and G_m characteristics of PEDOT:PSS/0.2% LiTFSI channels by adjusting the thickness as measured by atomic force microscopy (AFM). **e** The corresponding plots of the max. G_m and the V_G for the max. G_m with channel thickness

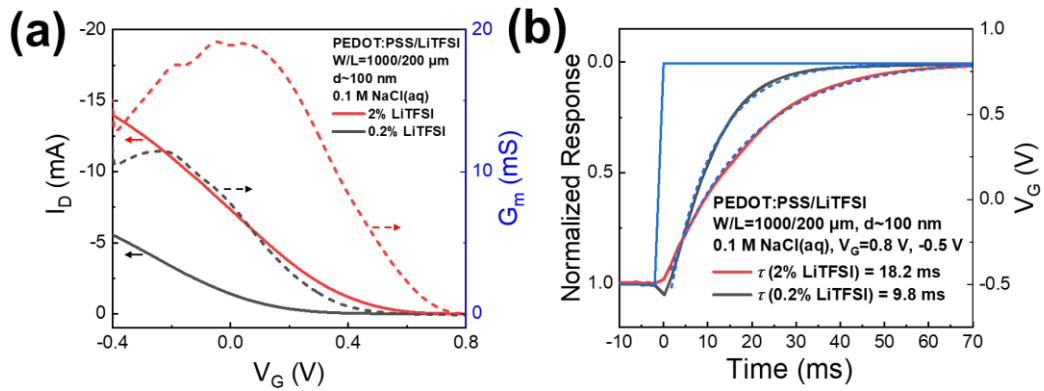


Fig. S3 **a** Transfer and G_m curves and **b** normalized temporal responses of PEDOT:PSS channels (~100-nm thick) with 0.2 wt% LiTFSI and 2 wt% LiTFSI, respectively, which were drop cast on Au electrodes ($W/L = 1000/200 \mu\text{m}$) and gated with Ag/AgCl through 0.1 M NaCl aqueous solution. The blue curve in (b) shows the applied V_G . The normalized transconductances were calculated as ~ 220 and $\sim 380 \text{ S cm}^{-1}$ for channels with 0.2 wt% LiTFSI and 2 wt% LiTFSI, respectively



Fig. S4 **a** Contact angle measurements of hydrophobic PDMS substrate before and after oxygen plasma treatment. **b** Fabrication process of patterned hydrophilic and hydrophobic PDMS surface using oxygen plasma under a stainless steel mask to form the PEDOT:PSS/LiTFSI patterns on PDMS by dewetting, followed by transfer printing to obtain stretchable microelectrodes on the gel substrate. **c** Resistance values of the PEDOT:PSS/LiTFSI electrode before (left) and after (right) transfer onto GEL-GLY/ Na_3Cit . The soft carbon nanotube (CNT) thin films were used for contact pads

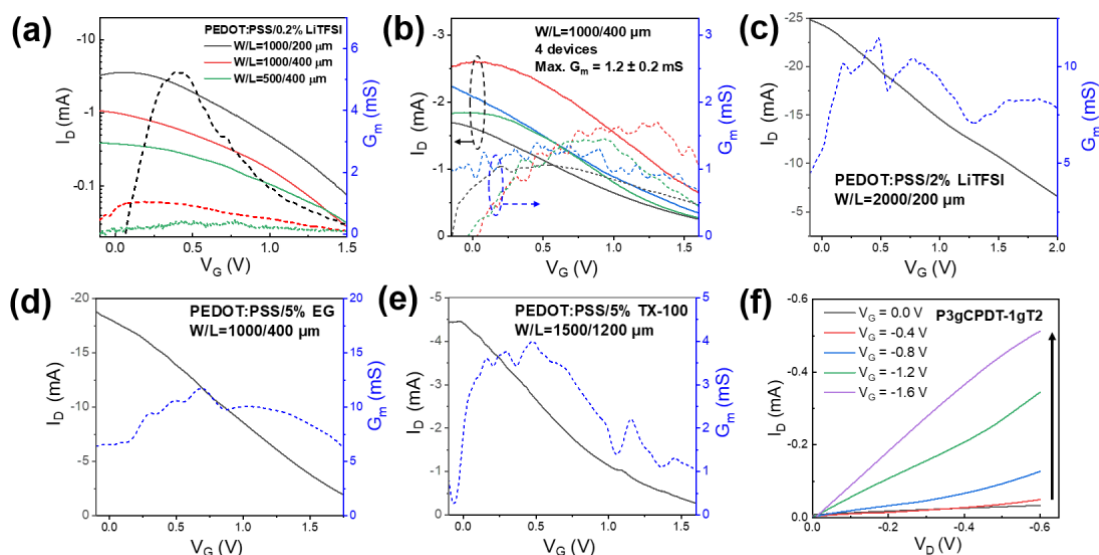


Fig. S5 **a** Transfer and G_m curves of all-polymer OECTs with controlled W/L ratios. The channels were PEDOT:PSS/0.2% LiTFSI with $d < 1 \mu\text{m}$. **b** Transfer and G_m curves of 4 devices with the constant electrode design of $W/L = 1000/400 \mu\text{m}$ to show the performance variation of the all-polymer OECTs from one batch fabrication. **(c-e)** Transfer and G_m curves of the all-polymer OECTs consisting of PEDOT:PSS/2% LiTFSI electrodes and thin channels of PEDOT:PSS/2% LiTFSI (**c**), PEDOT:PSS/5% ethylene glycol (EG, **d**) and PEDOT:PSS/5% TritonX-100 (TX-100, **e**). **f** Output characteristics of the enhancement-mode all-polymer OECT using P3gCPDT-1gT2 as the channel

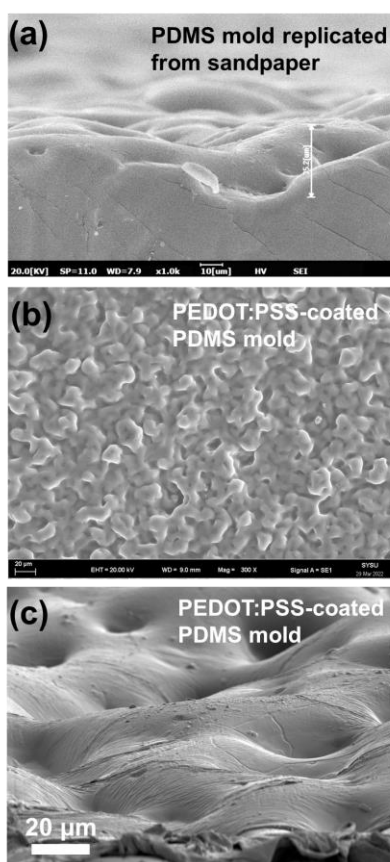


Fig. S6 **a** Cross-sectional scanning electron microscopy (SEM) image of PDMS mold replicated from a sandpaper (400 mesh). **b-c** SEM top (**b**) and side (**c**) view images of the PDMS mold after spin coating PEDOT:PSS/LiTFSI thin film

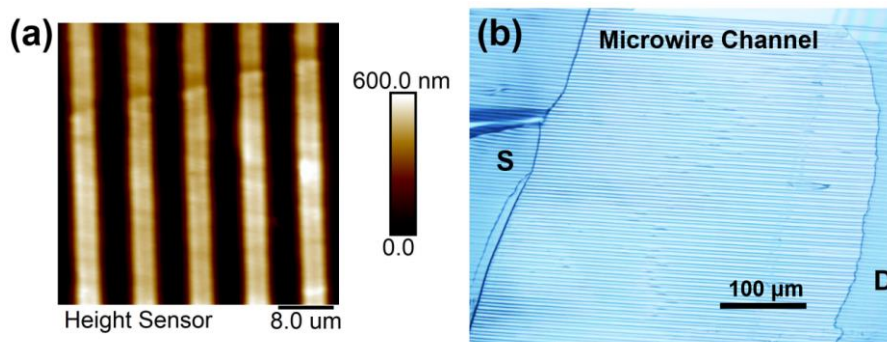


Fig. S7 **a** Typical AFM topography image of the printed PEDOT:PSS/LiTFSI microwire arrays on the gel electrolyte. **b** Optical microscope image of the microwire channel by printing two polymer electrodes to contact with the microwire arrays

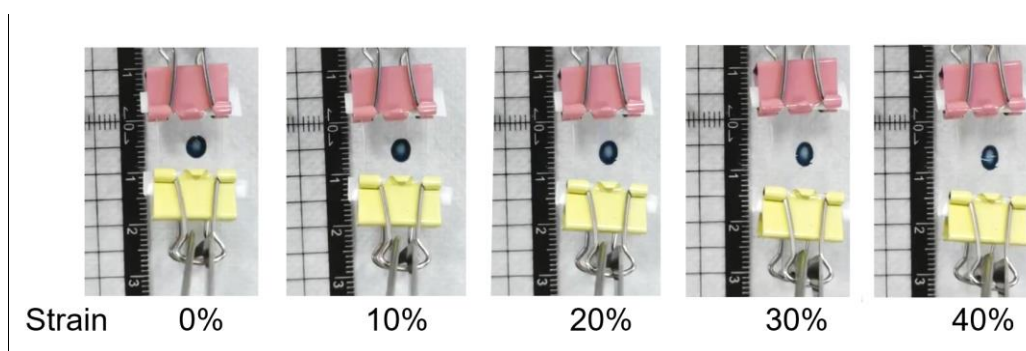


Fig. S8 Photos of PEDOT:PSS/2% LiTFSI film attached on the gel electrolyte under strains from 0-40%, showing the macroscopic cracks appearing at ~40% strain

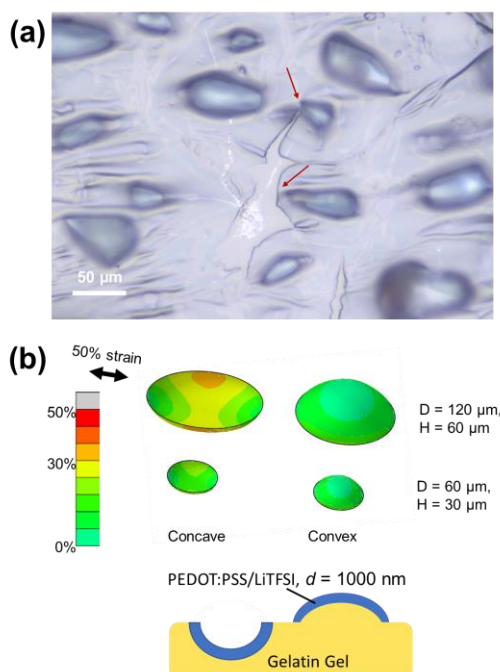


Fig. S9 **a** Magnified optical microscope image of the microstructured PEDOT:PSS/LiTFSI thin film on the gel substrate under ~50% strain, with the arrows indicating the prevented crack propagation at the bump regions. **b** Finite element simulations of strain distribution on PEDOT:PSS/LiTFSI thin layer coated on concave and convex microstructures of gelatin-based gel electrolyte, with an overall strain of 50%. It indicates that the convex structure and the smaller diameter may decrease the strain on the PEDOT:PSS/LiTFSI

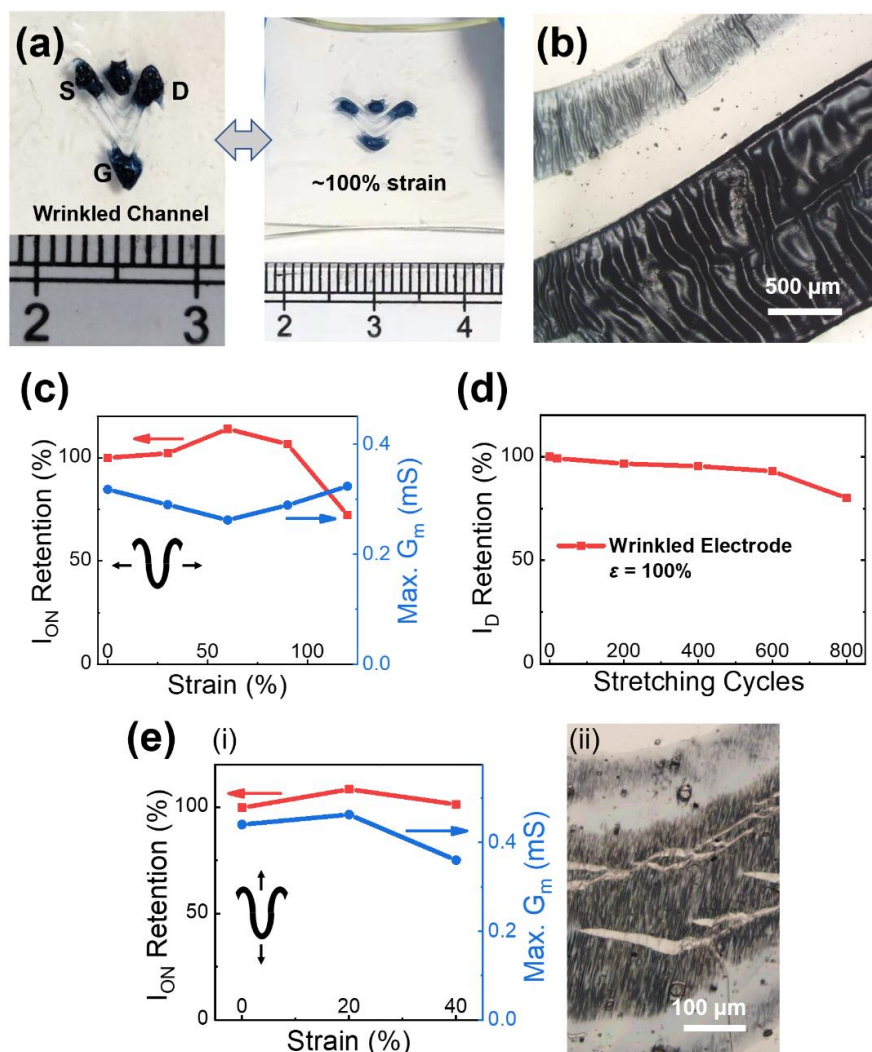


Fig. S10 **a** Photographs of the all-polymer OECT with wrinkled channel and electrodes (100% prestrain) under relaxed and stretched states. **b** Optical microscope image of the wrinkled channel and electrode. **c** I_{ON} retention and the maximum G_m of the stretchable all-polymer OECT during stretching from 0 to 100% strain parallel to the prestretched direction. **d** I_D retention of the wrinkled electrode under repeated stretching for 800 cycles at 100% strain. **e** OECT performance under strains of 0-40% in the perpendicular direction (i). The corresponding optical microscope image (ii) shows the channel cracking after perpendicular stretching

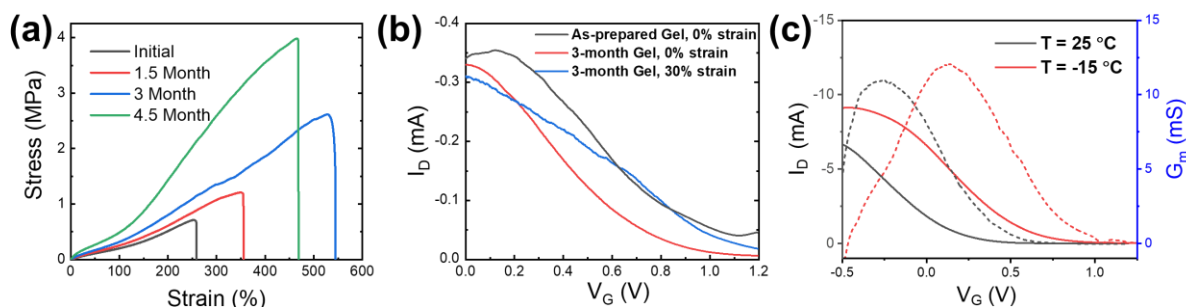


Fig. S11 **a** Typical tensile stress–strain curves of 20% GEL-60% GLY/Na₃Cit under ambient storage for up to 4.5 months. **b** Transfer curves of OECT prepared with fresh and aged gel electrolytes. **c** Transfer curves of gel-based OECT under varied temperature conditions

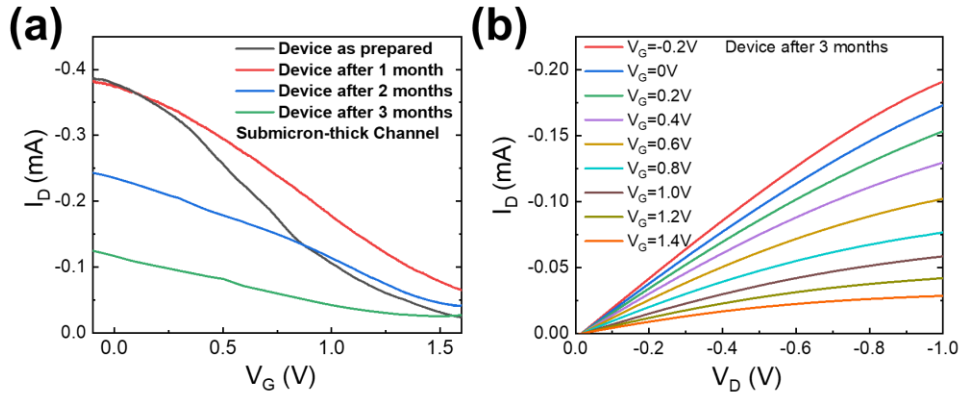


Fig. S12 a Transfer curves of an all-polymer OECT with submicron-thick channel during the storage for 3 months. **b** Output curves of the OECT after 3 months

Table S1 Comparison among stretchable OECTs using PEDOT:PSS channels

Materials	Device Dimensions	G_m	Stretch-ability	Long-term Operation	Refs.
PEDOT:PSS/glycerol/Capstone FS-30 Electrodes: wrinkled Au, carbon gate Electrolyte: PAM hydrogel	W/L=2000/ 8000 μm d=400 nm	1.1 mS Normalized: 110 $\text{S}\cdot\text{cm}^{-1}$	50 %	1000 cycles at 30 % strain	[S1]
PEDOT:PSS/EG/GOPS Electrodes: Au Electrolyte: aqueous solution	W/L=50/50 μm d<200 nm	1 mS Normalized: >50 $\text{S}\cdot\text{cm}^{-1}$	15 %	1000 cycles at 15 % strain	[S2]
PEDOT:PSS/EG/GOPS Electrodes: wrinkled Au Electrolyte: PAM/PVA/glycerol organohydrogel	W/L=50/50 μm d<200 nm	1.62 mS Normalized: >80 $\text{S}\cdot\text{cm}^{-1}$	30%	1000 cycles at 20% strain, storage: 8 days	[S3]
PEDOT:PSS/EG/GOPS Electrodes: microcracked Au Electrolyte: aqueous solution Wavy	W/L=630/ 130 μm d<200 nm	0.54 mS Normalized: >6 $\text{S}\cdot\text{cm}^{-1}$	100%	1000 cycles at 50 % strain	[S4]
PEDOT:PSS/glycerol/Zonyl fluoro-surfactant Electrodes: wavy Au/PEDOT:PSS Electrolyte: aqueous solution	NA	~1 mS	biaxial 30%	NA	[S5]
PEDOT:PSS/glycerol/Capstone FS-30 Electrodes: Au, carbon gate Electrolyte: aqueous solution	W/L=2000/ 8000 μm d=50 nm	0.2 mS Normalized: 160 $\text{S}\cdot\text{cm}^{-1}$	30 %	60 cycles at 30 % strain	[S6]
PEDOT:PSS/glycerol/Capstone FS-30/PEG 400 Electrodes: Au, carbon gate Electrolyte: aqueous solution	W/L=2000/ 8000 μm d=300 nm	0.1 mS Normalized: 13 $\text{S}\cdot\text{cm}^{-1}$	45 %	100 cycles at 45 % strain	[S7]

PEDOT:PSS/PEG/divinyl sulfone	W/L=2000/ 200 μm	27.43 mS Normalized: 62.3 S $\cdot\text{cm}^{-1}$	biaxial 30%	NA	[S8]
Electrodes: AuNPs-AgNWs Electrolyte: ionic liquid	d=440 nm				
PEDOT:PSS/PAMPS/ionic liquid	W/L=1000/ 250 μm	12.95 mS Normalized: 162 S $\cdot\text{cm}^{-1}$	biaxial 100%	NA	[S9]
Electrodes: AgNWs, SWCNT gate Electrolyte: P(VDF-HFP) ionogel	d=200 nm				
Microstructured PEDOT:PSS/LiTFSI	W/L=2000/ 200 μm	12.7 mS Normalized: 25.4 S $\cdot\text{cm}^{-1}$	biaxial 100%, uniaxial 120%	1000 cycles at 80 % strain, storage: >4 months	This Work
Electrodes: wrinkled PEDOT:PSS/LiTFSI Electrolyte: gelatin-glycerol/Na ₃ Cit organohydrogel	d=500 nm				

Supplementary References

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