

Supporting Information for

Nanocrystalline Iron Pyrophosphate Regulated Amorphous Phosphate Overlayer for Enhancing Solar Water Oxidation

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S1 Experimental Detail

S1.1 Chemicals

All materials generated in this study are available from the lead contact without restriction. Iron (III) chloride hexahydrate ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$; Sigma-Aldrich; AR 97.0%), urea ($(\text{NH}_2)_2\text{CO}$; Sigma-Aldrich; AR, 99%), hydrochloric acid (HCl ; Sigma-Aldrich; 36.5-38.0%), sodium hypophosphite monohydrate ($\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$; AR, 99%), sodium hydroxide (NaOH ; Sigma-Aldrich; AR, > 85%), sodium sulfite (Na_2SO_3 ; AR 97.0%), and fluorine-doped tin oxide glass (FTO, 15 Ω , thickness 2 mm) were used as received unless stated otherwise.

S1.2 Faradic Efficiency Calculation

The faradaic efficiency of O_2 production was calculated using the follow equation:

$$\text{Faradaic efficiency} = n\text{O}_2 / (Q/2F) \quad (\text{S1})$$

where $n\text{O}_2$ defines produced oxygen with the utilization of total charge Q , F refers to Faradic constant.

S1.3 DFT Calculations

DFT calculations were performed using the Vienna *ab initio* package (VASP) [S1, S2]. The core electrons were described by using the projector augmented wave (PAW) method [S3]. A generalized gradient approximation (GGA) with the Perdew-Burke-Ernzerhof (PBE) functional was employed for the plane-basis wave expansion [S4, S5]. A corrective Hubbard U correction (GGA+ U) method introduced by Dudarev *et al.* was included in the calculations to accurately describe the strongly correlated interaction of Fe 3d orbitals [S6]. U_{eff} was set to 5 eV for Fe atoms. The energy cut-off of 400 eV was used. Brillouin zones were U_{eff} was set to 5 eV for Fe atoms and an energy cut-off of 400 eV was used. Brillouin zones were sampled with a gamma-centered k-point grid of $1 \times 2 \times 1$ in the supercell of FePy, FePi, and $\alpha\text{-Fe}_2\text{O}_3(012)$ of which the lateral dimensions are $21.71 \times 9.68 \text{ \AA}^2$, $18.24 \times 18.22 \text{ \AA}^2$, and $19.39 \times 10.28 \text{ \AA}^2$, respectively [S7]. Large enough vacuum layers of at least 15 \AA were applied to minimize undesirable interactions between adjacent periodic cells. The Gaussian smearing scheme was applied with

a smearing width of 0.1 eV. The energy convergence criteria in the self-consistent field were set to 10^{-6} eV and geometry structures were fully relaxed with Hellman-Feynman forces with a tolerance of $0.1 \text{ eV } \text{\AA}^{-1}$.

The ΔG of the OER intermediates was defined as follows:

$$\Delta G = \Delta E + \Delta ZPE - T\Delta S + \Delta G_U + \Delta G_{pH} \quad (\text{S2})$$

where ΔE is the calculated total energy difference, ΔZPE and $T\Delta S$ are the zero-point energy correction and entropy terms, respectively, which can be determined by frequency calculations. ΔG_{pH} is the correction of the H^+ free energy by the concentration and ΔG_U represents the free energy terms related to the applied electrode potential, U .

S2 Supplementary Figures

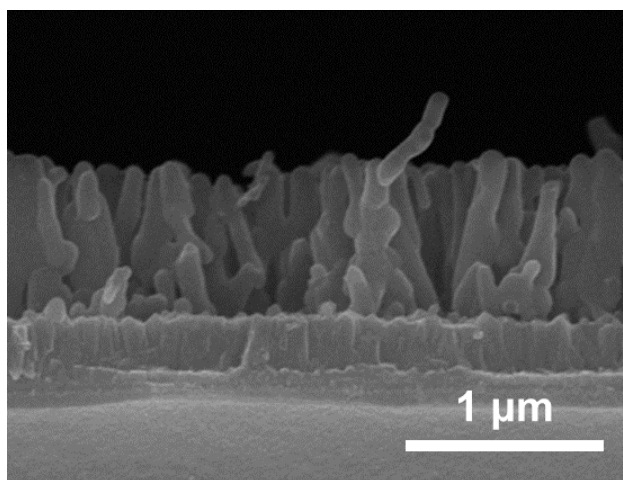


Fig. S1 Cross slide SEM image of FePy@FePi decorated nanorods

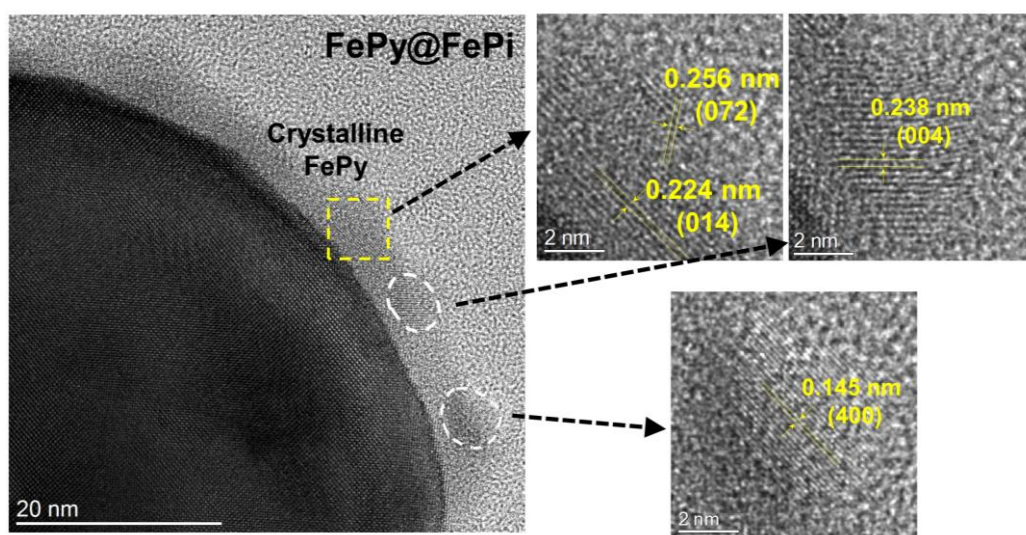


Fig. S2 HRTEM image of FePy@FePi hybrid overlayer and lattice fringe of $\text{Fe}_4(\text{P}_2\text{O}_7)_3$ crystalline phase

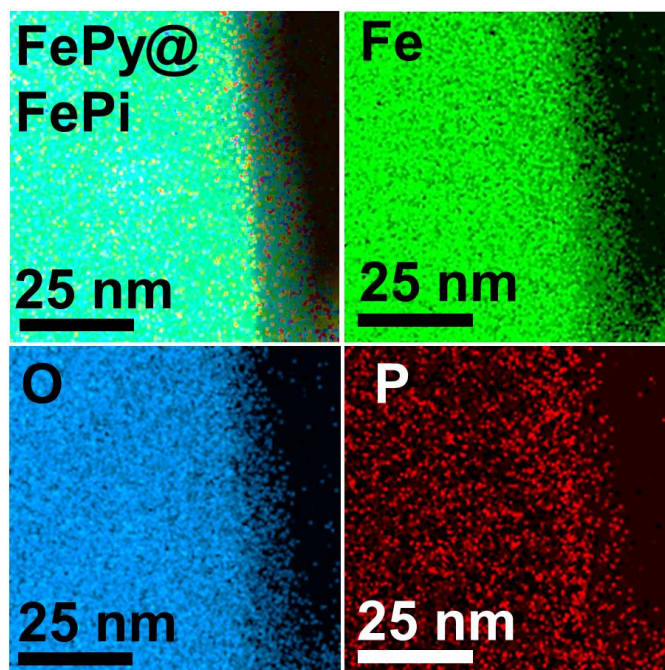


Fig. S3 EDX images of the FePy@FePi hybrid overlayer

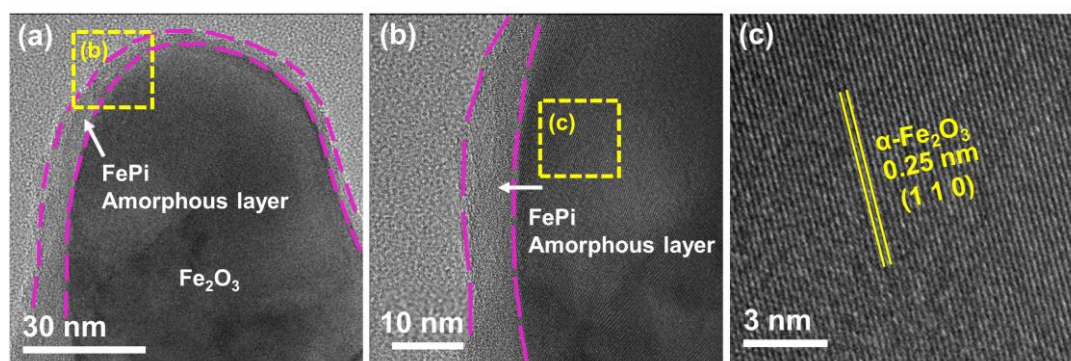


Fig. S4 a-c HR-TEM images of FePi at different magnifications

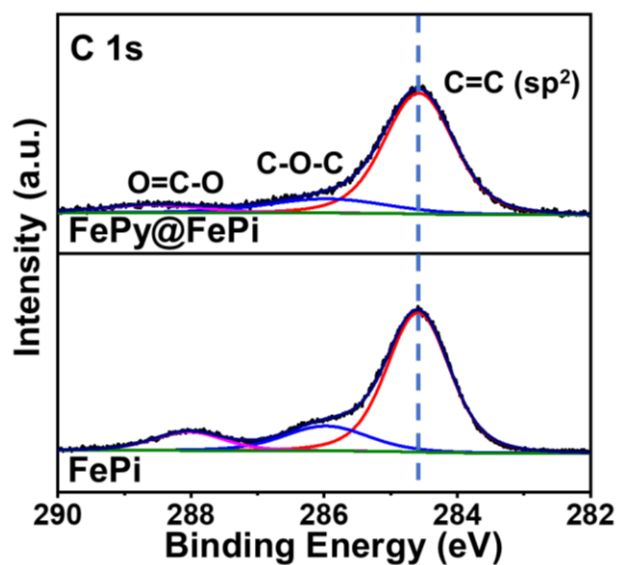


Fig. S5 XPS spectra of C 1s for FePi, and FePy@FePi

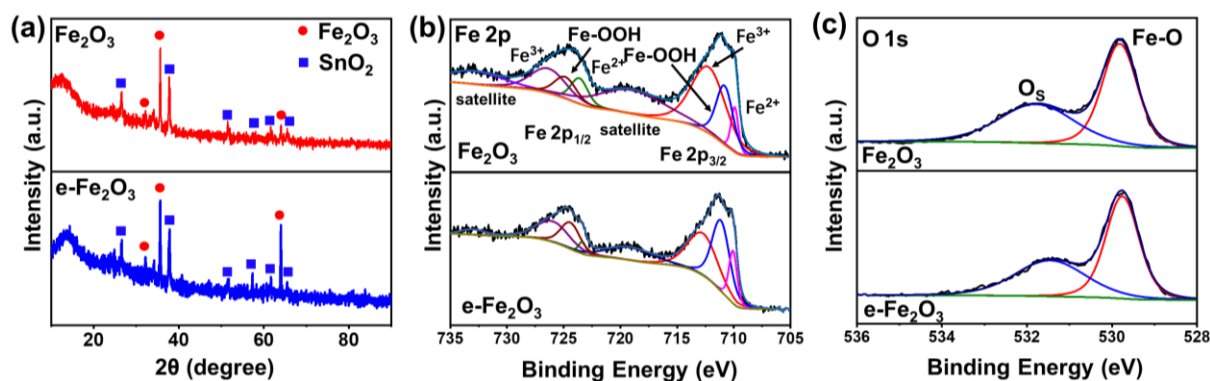


Fig. S6 **a** XRD patterns of Fe_2O_3 and $\text{e-Fe}_2\text{O}_3$. **b, c** XPS spectra of Fe 2p and O 1s of Fe_2O_3 and $\text{e-Fe}_2\text{O}_3$

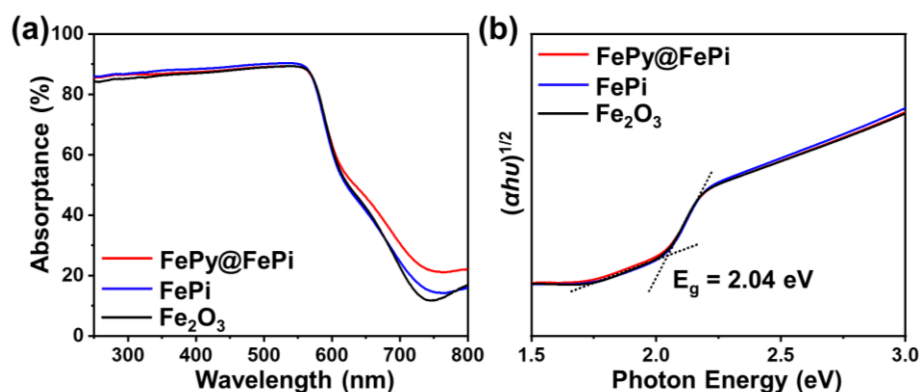


Fig. S7 **a** UV-vis absorbance spectra of Fe_2O_3 , FePi, and FePy@FePi decorated photoanode. The absorbance spectra (A) were obtained by considering the diffuse reflectance (R) and diffuse transmittance (T) spectra ($A = 100\% - R - T$). **b** Tauc plots of $\alpha\text{-Fe}_2\text{O}_3$ and FePi, and FePy@FePi decorated $\alpha\text{-Fe}_2\text{O}_3$

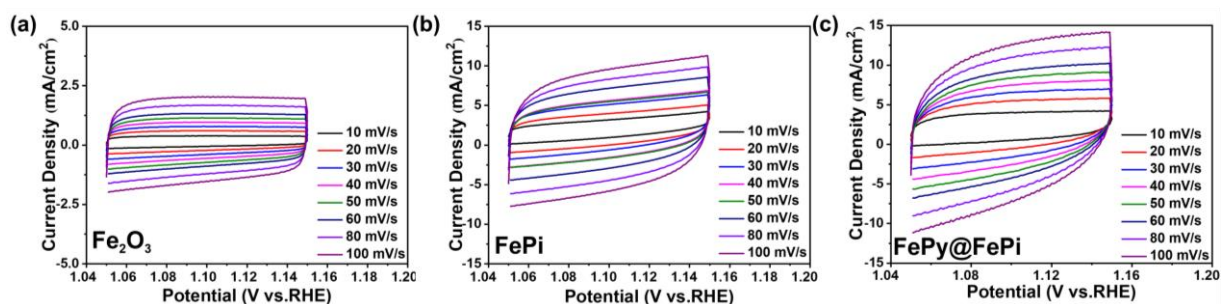


Fig. S8 **a-c** Cyclic voltammetry curves of Fe_2O_3 , FePi, and FePy@FePi decorated photoanode

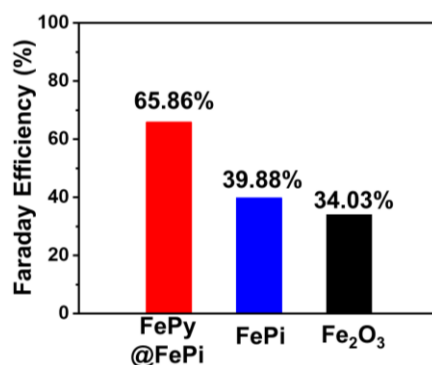


Fig. S9 Faraday Efficiency of Fe_2O_3 , FePi, and FePy@FePi decorated photoanode measured at 1.23 V vs. RHE at 1 sun illumination after measured for 1 h

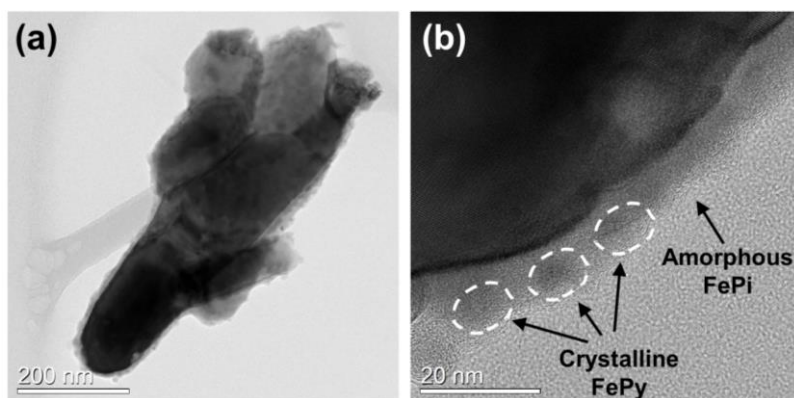


Fig. S10 **a** HRTEM image of FePy@FePi hybrid overlayer decorated α -Fe₂O₃ nanorod after PEC test. **b** HRTEM image of FePy@FePi hybrid overlayer after PEC test

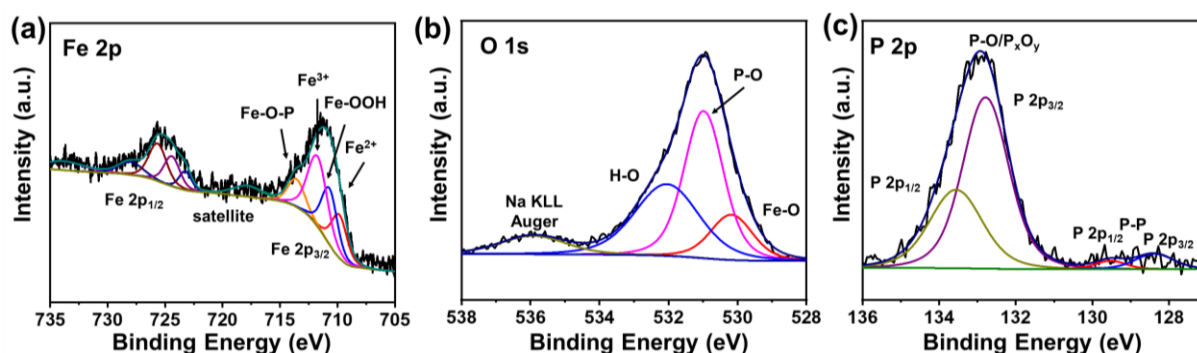


Fig. S11 a-c Fe 2p, O 1s, and P 2p XPS spectra of FePy@FePi decorated photoanode after stability test

Supplementary References

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