

Supporting Information for

## Self-Modifying Nanointerface Driving Ultrahigh Bi-Directional Thermal Conductivity Boron Nitride-Based Composite Flexible Films

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### S1 Materials and Characterizations

**Materials:** Boric acid (AR grade, 99.9%) and melamine (AR grade, 97.0%) were obtained from Aladdin chemistry Co., Ltd, China. Commercial BN powder with average size in range of 10-15  $\mu\text{m}$  was purchased from Qinghuangdao Eno High Tech Material Development Co., Ltd. PVA (molecular weight: 13000-23000 and 98% hydrolyzed) was from Sigma-Aldrich. SA (98 wt%, viscosity  $200\pm 20$  mpa·s, 1%, 2 °C), citric acid (AR grade, 99.5%) were supplied from Aladdin chemistry Co., Ltd, China. Calcium chloride (AR Grade, 96.0%), 2-propanoal (IPA, AR Grade, 99.5%), ethanol (AR Grade, 99.7%) and acetone (AR Grade, 99.5%) were obtained from Sinopharm Chemical Reagent Co., Ltd., China. All chemicals were received and used without further purification.

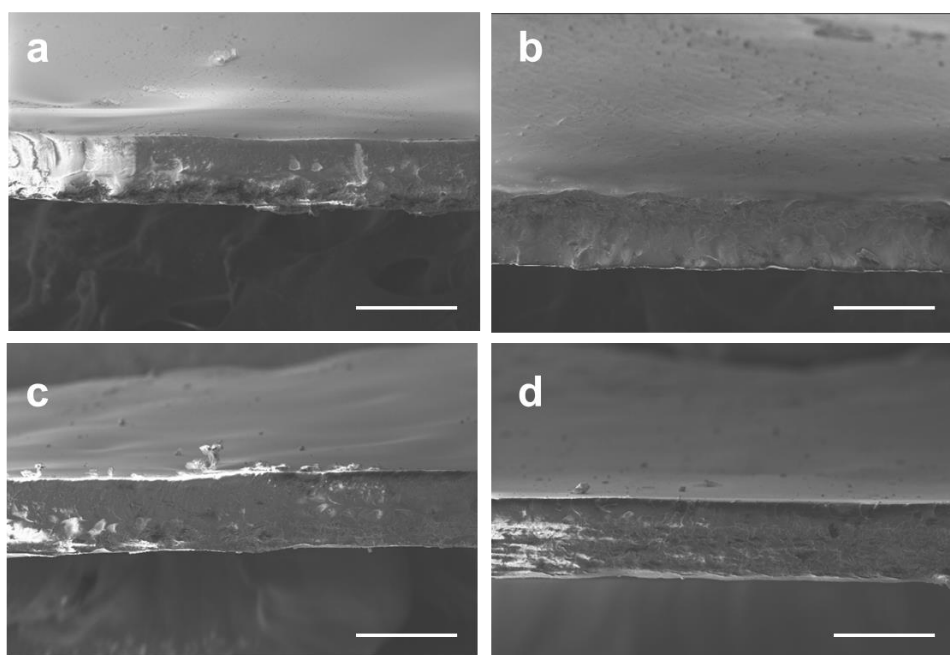
**Characterization:** The morphologies of BNNC, s-BN and composite film were observed by field emission SEM (FESEM, SEM500, Zeiss), AFM (Multimode 8, Bruker) and high resolution transmission electron microscope (HRTEM, JEM-2100F, JEOL). 3D structure of BN scaffolds was imaged using Nano X-ray Computed Tomography (nano-CT, SkyScan 2211, Bruker). The composition and bonding states of h-BN and s-BN were analyzed by an X-ray photoelectron spectrometer (XPS, K-Alpha, Thermo Scientific) X ray diffraction (XRD, D8 ADVANCE, Bruker) curves were recorded by Cu K $\alpha$  radiation and a scanning speed of 2 °/min to characterize crystalline structure of h-BN and s-BN. The oxygen contents of h-BN and s-BN were measured by pulse heating inert gas fusion-infra-red absorption method. The FTIR test was measured by a Fourier transform infrared (FT-IR) spectra meter ((FTIR, Nicolet 6700, Thermofisher). The mechanical properties of composite films were measured on a universal material testing machine (UTM, 5567A, Instron). The thermal conductivity ( $K$ ,  $\text{W m}^{-1}\text{K}^{-1}$ ) was calculated by multiplying thermal diffusivity ( $\alpha$ ,  $\text{mm}^2 \text{s}^{-1}$ ), density ( $\rho$ ,  $\text{g cm}^{-3}$ ) and specific heat ( $C_p$ ,  $\text{J g}^{-1}\text{K}^{-1}$ ).

$$K = \alpha \times C_p \times \rho \quad (\text{S1})$$

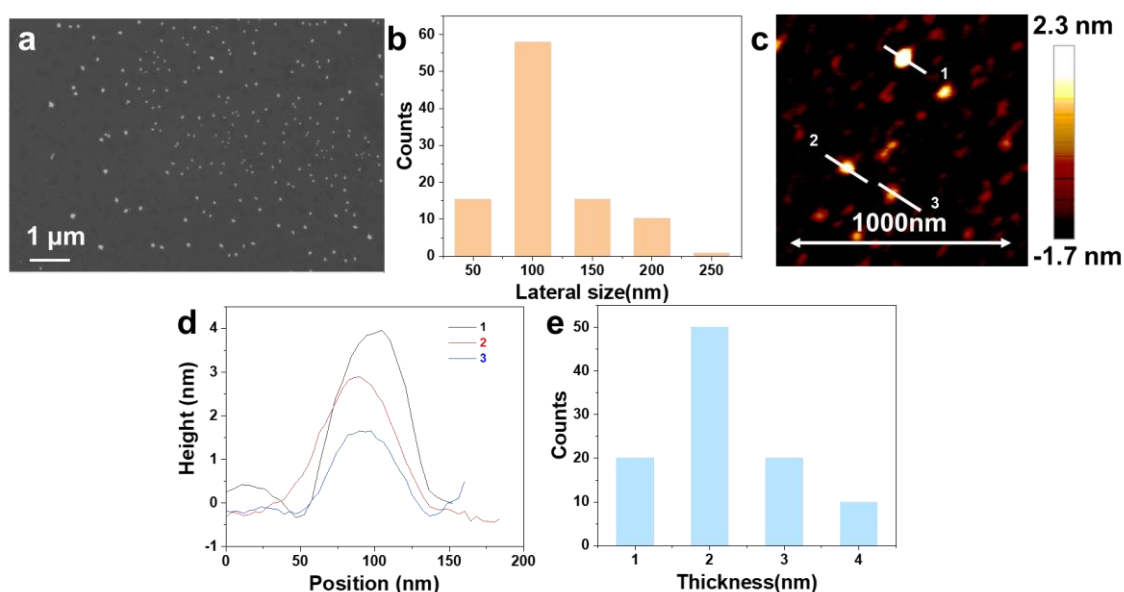
Thermal diffusivity was measured through the laser flash technique (LFA 467 HT Hyperflash, Nanoflash, Netzsch). The thermal conductivity of graphene oxide and BNNCs were measured by hot disk method (Hot Disk TPS2500S). Differential scanning calorimetry (DSC Q200, TA instruments) was used to measure specific heat capacity at heating rate of  $10 \text{ }^\circ\text{C min}^{-1}$ . At least three specimens were tested for each sample to obtain reliable data. Volume resistivity was measured by ZC-36 high resistance meter (Shanghai No. 6 Meter Factory Co., Ltd.). Thermal

expansion coefficient was measured by TMA (TA Q400, TA instruments) at heating rate of  $5\text{ }^{\circ}\text{Cmin}^{-1}$  from 40 to  $150\text{ }^{\circ}\text{C}$ . Thermal gravimetric (TG) analyses were evaluated by TGA (TGA 1, Mettler Toledo) at heating rate of  $20\text{ }^{\circ}\text{C min}^{-1}$  from 30 to  $800\text{ }^{\circ}\text{C}$ . The breakdown strength was obtained by a dielectric strength tester (HCDJC-100KV, Huace). The frequency-depended dielectric constant and dielectric loss were measured by a precision impedance analyzer (Agilent 4294A, Melrose) at room temperature with the frequency range of  $10^2$ - $10^6$  Hz.

## S2 Supplementary Figures



**Fig. S1** Cross-sectional SEM images of polymer/s-BN composite film with different filler content. **a** 0 wt%, **b** 5 wt%, **c** 15 wt% and **d** 35 wt%. Scale bar:  $100\text{ }\mu\text{m}$



**Fig. S2** Characterization of BNNCs. **a** SEM image, **b** lateral size histogram, **c** AFM image, **d** corresponding height variations and **e** thickness histogram of BNNCs

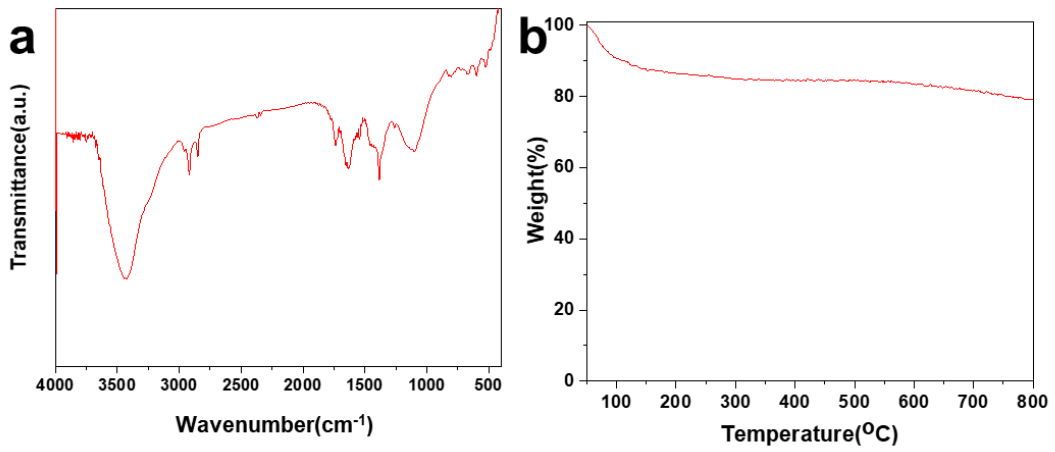


Fig. S3 FTIR transmittance spectrum and TGA curve for the BNNCs

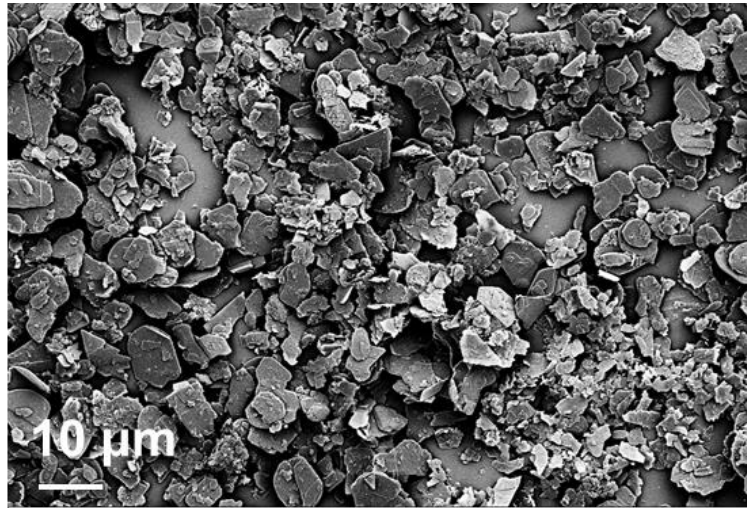


Fig. S4 SEM image of pristine h-BN powder with lateral sizes of 10-15 μm

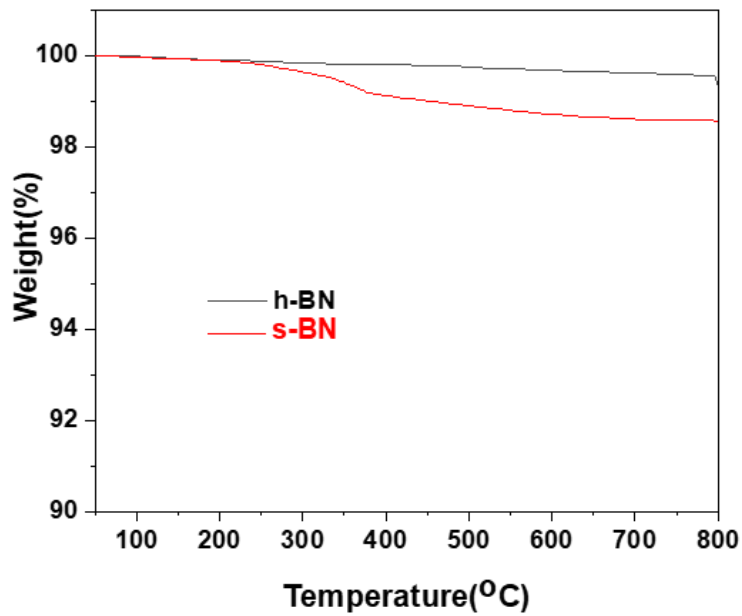
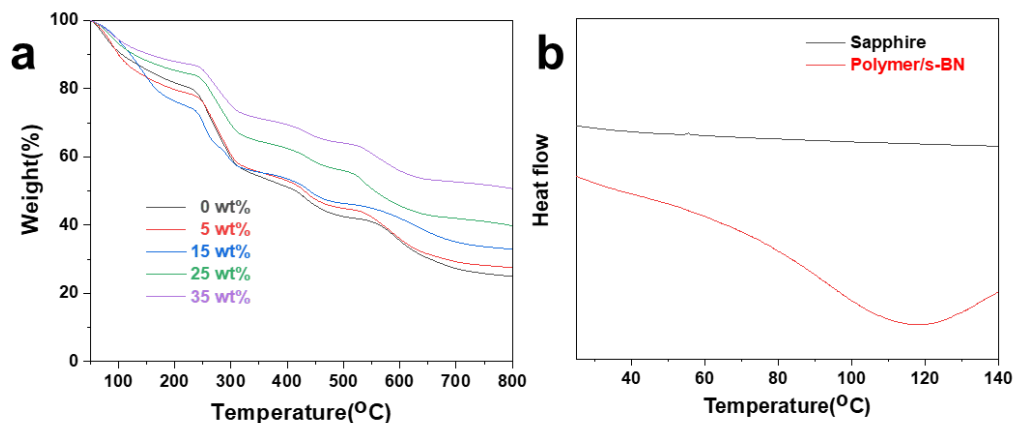
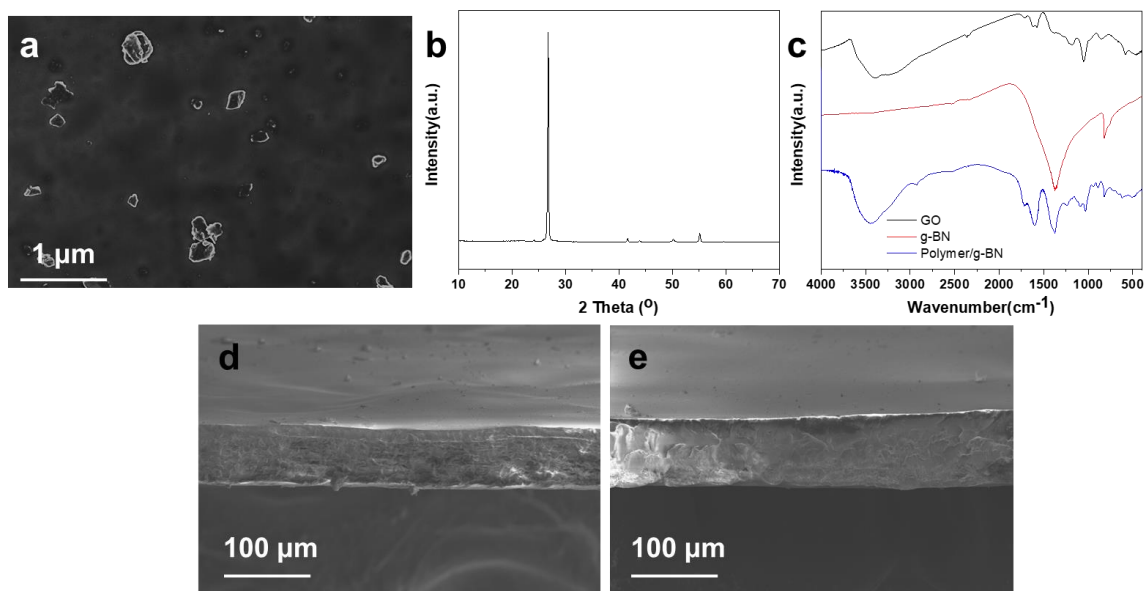


Fig. S5 TGA curves of h-BN and s-BN



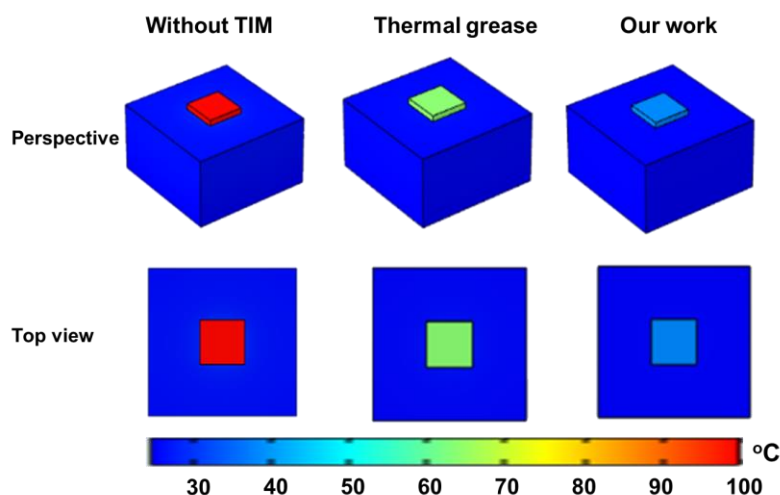
**Fig. S6** Thermodynamic analyses of composite films. **a** TGA curves of composite films with different BN contents. **b** DSC curves of composite films (25 wt% BNNS) and Sapphire



**Fig. S7** Basic characterization of the polymer/g-BN composite film. **a** SEM image of graphene oxide nanosheets (GO) with lateral sizes of 100-500 nm. **b** XRD pattern of the g-BN, presenting good crystallinity. **c** FTIR spectrum evolutions of GO, g-BN and polymer/g-BN. Cross-sectional SEM images of polymer/h-BN **d** and polymer/g-BN **e**



**Fig. S8** Digital photo of CPU



**Fig. S9** The temperature distribution of the simulated system without TIM, using thermal grease and the polymer/s-BN composite film

### S3 Supplementary Tables

**Table S1** The data for crosslinking degree of the SA in the dried composite film\*

Calcium ion concentration (wt%)	Film casting thickness( $\mu\text{m}$ )	Freezing time (min)	Sample weight before soaked (g)	Sample weight after soaked (g)
0.5	200	10	0.1522	0.1044
0.8	200	10	0.1322	0.1320
3	200	10	0.1345	0.1345
5	200	10	0.1424	0.1424
3	400	10	0.1352	0.1352
3	600	10	0.1534	0.1534
3	800	10	0.1743	0.0825
3	200	1	0.1328	0.0943
3	200	3	0.1453	0.1453
3	200	6	0.1374	0.1374
3	200	8	0.1394	0.1393

\* The crosslinking degree of the SA in the dried composite film is determined by comparing its weight change before and after complete soaking in water. If the weight of SA film obtained by ice-press assembly strategy didn't change before and after soaking in water for 24 h, it was completely cross-linked by calcium ions.

**Table S2** The detailed information of Fig. 3b

Filler content	$K_{\text{through-plane}}$ ( $\text{Wm}^{-1} \text{K}^{-1}$ )	STC ( $\text{Wm}^{-1} \text{K}^{-1}/\text{wt}\%$ )	Shape	Method	Refs.
27.05vol%	5.13	0.13	Bulk	wet-spinning	[S1]
21.9vol%	4.22	0.13	Bulk	Modified filtration	[S2]
18.2wt%	4.53	0.24	Bulk	Ice-template	[S3]
60vol%	6.11	0.07	Bulk	Sacrificial template	[S4]
8.35vol%	3.87	0.32	Bulk	Ice template	[S5]
4.4vol%	1.56	0.24	Bulk	Ice-template	[S6]
50wt%	9.8	0.20	Bulk	Freeze-drying	[S7]
53wt%	7.3	0.14	Bulk	CVD	[S8]
40wt%	3.85	0.096	Bulk	Hot-press	[S9]
60wt%	7.62	0.13	Bulk	Layer-by-layer stacking	[S10]
16wt%	1.15	0.07	Film	Tetrahedral skeleton template	[S11]
40wt%	7.29	0.18	Film	Vacuum filtration	[S12]
15wt%	9.6	0.64	Film	Ice-press assembly	[S13]
25wt%	21.3	0.85	Film	Ice-press assembly	Our work



**Table S3** The detailed parameters of the components for simulated analysis

Components	Size ( $\mu\text{m}\times\mu\text{m}$ )	Thermal conductivity ( $\text{W m}^{-1} \text{K}^{-1}$ )
Filler	10 $\times$ 0.01	300(in-plane); 10(through-plane)
Polymer	100 $\times$ 20	0.18

Thin layers with a thickness of 1 nm are added on the interfaces between the filler and the polymer, and the thermal resistances value of the thin layer are set according to the different fillers. The heat sources are attached to the bottom surface of the composite and the other boundaries have an initial value at 20 °C. All the models are set as the same conditions and the heating time is 0.0006 s.

**Table S4** The detailed parameters of the components in the simulated cooling system

Components	Size ( $\text{cm}\times\text{cm}\times\text{cm}$ )	Materials	Thermal conductivity ( $\text{W m}^{-1} \text{K}^{-1}$ )	Specific heat capacity ( $\text{J g}^{-1} \text{K}^{-1}$ )
CPU heater	3 $\times$ 3 $\times$ 0.1	Silicon	130	0.7
TIM	3 $\times$ 3 $\times$ 0.08	/	/	/
Heat sink	6 $\times$ 6 $\times$ 3	Aluminum alloy	238	0.9

This system simulates the heat transfer process from a CPU (54 W) to the heat sink kept at 25 °C. The steady-state temperature profiles of the simulated system are calculated according to the setting TIMs.

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