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Ultrafine Vacancy-Rich Nb₂O₅ Semiconductors Confined in Carbon Nanosheets Boost Dielectric Polarization for High-Attenuation Microwave Absorption

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HIGHLIGHTS

- Ultrafine oxygen-vacancy-rich Nb₂O₅ semiconductors have been incorporated into carbon nanosheets for high-attenuation microwave absorption (-80.8 dB, > 99.999999% wave absorption).
- Nb₂O₅ semiconductors with abundant oxygen vacancies significantly facilitate polarization relaxation, electromagnetic response, dipole polarization, and interfacial polarization.
- Multiple reflections and scattering dissipation are enhanced by the unique 2D lamellar morphology.

ABSTRACT The integration of nano-semiconductors into electromagnetic wave absorption materials is a highly desirable strategy for intensifying dielectric polarization loss; achieving high-attenuation microwave absorption and realizing in-depth comprehension of dielectric loss mechanisms remain challenges. Herein, ultrafine oxygen vacancy-rich Nb₂O₅ semiconductors are confined in carbon nanosheets (ov-Nb₂O₅/CNS) to boost dielectric polarization and achieve high attenuation. The polarization relaxation, electromagnetic response, and impedance matching of the ov-Nb₂O₅/CNS are significantly facilitated by the Nb₂O₅ semiconductors with rich oxygen vacancies, which consequently realizes an extremely high attenuation performance of - 80.8 dB (> 99.999999% wave absorption) at 2.76 mm. As a dielectric polarization center, abundant Nb₂O₅-carbon heterointerfaces can intensify interfacial polarization loss to strengthen dielectric polarization, and the presence of oxygen vacancies endows Nb₂O₅ semiconductors with abundant charge separation sites to reinforce elec-



tric dipole polarization. Moreover, the three-dimensional reconstruction of the absorber using microcomputer tomography technology provides insight into the intensification of the unique lamellar morphology regarding multiple reflection and scattering dissipation characteristics. Additionally, ov-Nb₂O₅/CNS demonstrates excellent application potential by curing into a microwave-absorbing, machinable,

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and heat-dissipating plate. This work provides insight into the dielectric polarization loss mechanisms of nano-semiconductor/carbon composites and inspires the design of high-performance microwave absorption materials.

KEYWORDS Electromagnetic wave absorption; Nb_2O_5 semiconductor; Dielectric polarization loss; Oxygen vacancy; Nb_2O_5 - carbon hetero-interface

1 Introduction

With the flourishing development of wireless communication, particularly fifth-generation (5G) communication technology, serious concerns about electromagnetic radiation, electromagnetic interference, and electromagnetic pollution have arisen due to rampant electromagnetic wave (EMW) signals [1–3]. Microwave absorption materials (MAMs) with exceptional abilities to absorb and dissipate electromagnetic energy have non-negligible potential for addressing electromagnetic hazards [4–6]. The EMW attenuation performance characteristics of MAMs are closely associated with the dielectric/magnetic loss parameters (complex permittivity and permeability) and morphologies. Hence, constructing MAMs with elaborately optimized dielectric/ magnetic loss properties and absorption-promoted morphology has been considered an effective EMW absorber design criterion [7–9].

To obtain MAMs with excellent dielectric/magnetic loss characteristics, researchers have made great efforts to construct multicomponent nanocomposites [4, 10–12]. This effort is because the EMW absorption properties of nanocomposites can be significantly optimized by integrating additional dielectric/magnetic components, thereby introducing many new attenuation mechanisms. For instance, Gao et al. [13] reported that Cu-intercalated MoS₂ with carbon modification can achieve synergistic multiple polarization. Su et al. [14] demonstrated a ternary-alloy FeCo₂Ni/ carbon composite, by which EMW absorption can be enhanced through the integration of magnetic-dielectric complementary attenuation capabilities. Li et al. [15] fabricated NiO/Ni particles on N-doped hollow carbon spheres with improved conduction loss and polarization loss. Xie et al. [16] synthesized oxygen vacancy-containing MoO₃/ PPy composites for electromagnetic wave absorption, showing enhanced electrical conductivity and dielectric loss. Conversely, the structure and geometry significantly influence both the interfacial polarization losses of MAMs and the penetration and scattering paths of EMWs, playing essential roles in the attenuation of EMW energy [17, 18]. Therefore, to enhance interfacial polarization loss and prolong the energy attenuation path of EMW, researchers have made great efforts to design and construct absorbing materials with versatile structures in special dimensions [19–21]. For example, nanotubes in one-dimensional (1D) structures [22–24], nanosheets in two-dimensional (2D) structures [8, 18, 25], and three-dimensional (3D) hollow/ yolk-shell structures have been reported [26, 27]. Generally, constructing multicomponent nanocomposites and designing various nano/microstructures are the main aspects of MAM preparation.

As a typical class of carbonaceous MAMs, carbon nanosheets (CNSs) possess extensive 2D lamellar interfaces that facilitate interfacial polarizations and multiple scatterings and are suitable for active composition regulations [28–30]. Integrating dielectric-type components into CNSs always holds broad research interest [31, 32]. The dielectric components with complex loss mechanisms of dielectric resonance, multiple polarization relaxation and abundant dipole-induced polarization can positively boost EMW attenuation [33-36]. As a prominent dielectric material, semiconductors (e.g., TiO₂ [37], SiC [38], and Nb₂O₅ [38]) exhibit distinctive electronic properties, such as electron hopping and electron transport, which can be effectively modulated through particle size regulation, doping design, and defect engineering to tune the polarization loss characteristics [16, 27, 39]. Commonly, nano-semiconductors embedded in carbonaceous substrates tend to form abundant nano-heterointerfaces. The original electric balance is disrupted at these nano-interfaces due to the different electronic properties of interfacial atoms at heterointerfaces. As a result, an increased number of polarization loss sites are induced by the consequent local charge separation and dipole formation under the alternative field. Given the respective benefits of dielectric semiconductors and CNSs, nano-semiconductor-anchored CNS nanocomposites can be novel EMW absorbers. In this context, controllable synthesis and insight into the dielectric loss mechanisms of nano-semiconductor-anchored CNS nanocomposites are urgently required to meet the demands of real-world applications.

Herein, we present ultrafine (~10 nm) oxygen vacancyrich Nb₂O₅ nano-semiconductors confined in carbon nanosheets (ov-Nb₂O₅/CNS) as high-attenuation EMW absorption materials. The innovative synthesis route of direct carbonization of an organic Nb5+-gluconate precursor with spontaneous foaming characteristics promises the efficient constitution of a 2D morphology and produces abundant Nb resources to form Nb-based nanoparticles (Nb₂O₅ or NbC). Compared with NbC conductors, semiconductive Nb₂O₅ facilitates polarization relaxation and the electromagnetic response and triggers stronger local charge separation at the Nb₂O₅-carbon hetero-interface. Interestingly, the presence of oxygen vacancies endows Nb₂O₅ nanocrystals with extensive charge separation sites to reinforce electric dipole polarization inside the semiconductor. Moreover, due to the 2D lamellar conductive carbon skeleton, satisfactory conductive loss, abundant macrointerfacial polarization and multiple scattering are realized to attenuate EMW energy. Consequently, ov-Nb₂O₅/CNS with boosted dielectric polarization losses exhibits superior EMW absorption with an ultralow minimum reflection loss (RL_{min}) of -80.8 dB (>99.999999% wave absorption) at 7.11 GHz (2.76 mm) and a wide effective absorption bandwidth of 3.37 GHz at 1.30 mm. In addition, the composite possesses excellent application potential by curing with cyanate resin to form a microwave-absorbing ov-Nb₂O₅/ CNS-cyanate plate. These results verify that ov-Nb₂O₅/CNS is expected to be an excellent microwave absorber when practically applied.

2 Experimental Section

2.1 Materials

Ammonium niobate (V) oxalate hydrate, ammonium hydroxide solution, and anhydrous ethanol were purchased from Aladdin Co. Ltd. Gluconic acid solution was purchased from Energy Chemical. All reagents were used as received without further purification. Deionized water was used throughout the experiments.

2.2 Preparation of Nb⁵⁺-Gluconate Precursor

Nb⁵⁺-gluconate was obtained from a neutralization reaction of gluconic acid and Nb(OH)₅. Briefly, 24.71 g (≈0.08 mol) ammonium niobate (V) oxalate hydrate was added to 100 mL deionized water at 80 °C and continuously stirred until it was completely dissolved. Then, ammonium hydroxide solution (25 wt%) was added to the above solution to adjust the pH level to pH > 7. The resulting white Nb(OH)₅ precipitate was centrifuged and washed three times with deionized water. A total of 160 g gluconic acid solution (50 wt%) and 20 mL deionized water were added to the above fresh Nb(OH)₅ paste. After continuous stirring at 80 °C, Nb(OH)5 was completely dissolved, and a transparent Nb5+-gluconate solution was obtained. Finally, the obtained Nb5+-gluconate solution was added dropwise into anhydrous ethanol to crystallize. Nb⁵⁺-gluconate was separated from the ethanol solution by filtration, washed 3 times with anhydrous ethanol, and dried overnight.

2.3 Preparation of *a*-Nb₂O₅/CNS, *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS Composites

The Nb⁵⁺-gluconate precursor was simply carbonized at different temperatures (700, 800, 1000, and 1200 °C) for 2 h under a N₂ atmosphere with a heating rate of 3 °C min⁻¹ to obtain *a*-Nb₂O₅/CNS, *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS composites. *a*-Nb₂O₅/CNS, *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS denote amorphous Nb₂O₅/ CNS, oxygen vacancy Nb₂O₅/CNS, crystallized NbC/CNS, and well-crystallized NbC/CNS, respectively.

2.4 Material Characterizations

The morphologies of the samples were observed by field emission scanning electron microscopy (FESEM; Nova NanoSEM 450) and field emission transmission electron microscopy (FETEM; JEM-2100F). Off-axis electron holography was observed by Lorentz-transmission electron microscopy (JEM-2100F). Thermogravimetric analysis (TGA; PerkinElmer TGA 4000) was conducted at an N₂ or airflow of 50 mL min⁻¹, with a heating rate of 10 °C min⁻¹. The X-ray diffraction (XRD) patterns were obtained on a Rigaku D/max 2550 diffractometer operating at 40 kV and 20 mA using Cu K α radiation (λ =1.5406 Å). X-ray photoelectron spectrometry (XPS) (Thermo Scientific ESCALAB Xi+) was used to determine the surface elemental compositions of the samples. The Raman spectrum was tested by Invia Reflex with a wavelength of 532 nm. Nitrogen adsorption/desorption isotherms were measured at 77 K with a Quadrasorb SI analyzer. Electron paramagnetic resonance (EPR) spectra were measured on a Bruker EMXPLUS spectrometer at room temperature. Microcomputer tomography (micro-CT; Xradia 520 Versa) was used to acquire sliced images.

2.5 Electromagnetic Measurement

The samples were uniformly dispersed in a paraffin matrix at 40 wt%, which was fabricated into a cylindrical sample with an inner diameter of 3.04 mm, an outer diameter of 7.00 mm, and a thickness of 2.00 mm. A vector network analyzer (VNA; Agilent E5071C) was used to measure the complex permittivity and permeability characteristics in the range of 2–18 GHz with the coaxial line method.

2.6 Preparation of ov-Nb₂O₅/CNS-cyanate Plate

First, 2.04 g ov-Nb₂O₅/CNS was mixed with 10.20 g cyanate ester (ov-Nb₂O₅/CNS: cyanate ester = 1:5). Then, 0.05 g curing agent of dibutyltin dilaurate was added to the above mixture and continuously stirred to form a uniform mixture resin. The ov-Nb₂O₅/CNS-cyanate resin was curved in an 80 mm × 40 mm mold at 80 °C to form the ov-Nb₂O₅/CNS-cyanate plate. The plate was further machined to the desired thickness and size.

2.7 Density Functional Theory Calculations

First-principles calculations were employed to perform all density functional theory (DFT) calculations within the generalized gradient approximation (GGA) using the Perdew–Burke–Ernzerhof (PBE) formulation. The projected augmented wave (PAW) potentials were chosen to describe the ionic cores and consider valence electrons using a plane wave basis set with a kinetic energy cutoff of 500 eV (for the energy band structure) or 520 eV (for the charge density distribution). Partial occupancies of the Kohn–Sham orbitals were allowed by using the Gaussian smearing method with a width of 0.05 eV. The electronic energy was considered self-consistent when the energy change was smaller than 10^{-4} eV (for the energy band structure) or 10^{-5} eV (for the charge density distribution). A geometry optimization was considered convergent when the energy change was smaller than 0.04 \AA^{-1} (for the energy band structure) or 0.03 eV $Å^{-1}$ (for the charge density distribution). For the energy band structure, Grimme's DFT-D3 methodology was used to describe the dispersion interactions. For charge density distributions, in our structure, U correction was used for Nb (3.91 eV) atoms. The vacuum spacing in a direction perpendicular to the plane of the structure was 20 Å for the surfaces. Brillouin zone integration was performed using $3 \times 3 \times 2$ (for the energy band structure) or $3 \times 3 \times 1$ (for the charge density distribution) Monkhorst-Pack k-point sampling for a structure.

3 Results and Discussion

3.1 Preparation and Structure Characterizations

The synthesis of Nb₂O₅ and NbC nanoparticle/carbon nanosheet composites is achieved for the first time via direct carbonization of the organic Nb⁵⁺ gluconate (Nb-GlcA) precursor, as illustrated in Fig. 1. The Nb-GlcA precursor is tactfully prepared through the highly efficient neutralization reaction between gluconic acid and freshly precipitated Nb(OH)₅, as expounded upon in the Experiment section. Notably, the Nb-GlcA precursor exhibits a unique self-foaming characteristic, leading to a very high volume expansion upon carbonization, as shown in Fig. S1. During the initial thermal treatment stage (< 200 °C), the precursor undergoes a morphological transformation from compact bulks to intermediates with closed-cell structures (Fig. S2). The closed-cell foaming structures ultimately develop into 2D lamellar structures for the final product. With further high-temperature annealing (700-1200 °C), organic gluconic carbohydrates and Nb⁵⁺ ions undergo structural rearrangement and transform into carbonaceous nanocomposites embedded with different types of Nb-based nanoparticles. The contents of Nb-based nanoparticles in these samples are \approx 34–40 wt%, according to the thermogravimetric analysis results (Fig. S3). From scanning electron microscopy (SEM) observations, all the composites feature a fragmented lamellar morphology with a large planar size (> 50 mm) (Fig. S4).



Fig. 1 Schematic illustration of the synthesis of two-dimensional Nb₂O₅ and NbC nanoparticle/carbon nanosheet composites

Moreover, C, O, and Nb are homogeneously distributed in the 2D hybrid skeleton, according to X-ray energy-dispersive spectroscopy (EDS) (Fig. S5).

The annealing temperature determines the microstructure and crystallography characteristics of the as-obtained composites, which are investigated through transmission electron microscopy (TEM) (Fig. 2). For all composites, plenty of Nb-based nanoparticles are homogeneously distributed in the 2D lamellar amorphous carbon skeletons (Fig. 2a-d). As the annealing temperature increases from 700 to 1200 °C, the size of Nb-based nanoparticles grows from less than 5 nm to approximately 20 nm (Fig. 2e-h). Simultaneously, the Nb-based nanoparticles undergo phase transitions from amorphous Nb₂O₅ (700 °C) to crystallized Nb₂O₅ (800 °C) and are eventually reduced to NbC (>1000 °C), as evidenced by the high-resolution TEM (HRTEM) images in Fig. 2i–1. Specifically, the amorphous nature of Nb₂O₅ nanoparticles in a-Nb₂O₅/CNS (700 °C) is evidenced by the absence of distinct crystal lattice fringes, as depicted in Fig. 2i. Upon increasing the annealing temperature to 800 °C, the Nb₂O₅ nanoparticles in *ov*-Nb₂O₅/CNS are crystallized from the carbon skeleton, which demonstrates a lattice spacing of 0.39 nm corresponding to the (001) plane of the orthorhombic Nb₂O₅ crystal (Fig. 2j). The reduction in Nb₂O₅ nanoparticles by the carbon skeleton is facilitated by their ultrafine size, enabling the process to occur at a relatively low temperature of 1000 °C. Hence, the crystallized nanoparticles in *c*-NbC/CNS (1000 °C) and *wc*-NbC/ CNS (1200 °C) with an average lattice spacing distance of 0.22 nm are indexed to the crystalline cubic NbC (200) plane (Fig. 2k–l).

The microcrystalline structures of all the samples can be disclosed by the X-ray diffraction (XRD) patterns (Fig. 3a). No characteristic peaks of Nb₂O₅ phases can be discerned in *a*-Nb₂O₅/CNS, confirming the existence of amorphous nanoparticles. The diffraction peaks in *ov*-Nb₂O₅/CNS are assigned to orthorhombic crystal Nb₂O₅ with a space group of Pbam (PDF, card No. 30–0873). For both *c*-NbC/CNS and *wc*-NbC/CNS, a group of typical sharp peaks corresponds to the cubic structure of the NbC nanocrystals (PDF, card No. 65–7964). The full width at half maximum (FWHM) of the characteristic NbC peaks decreases with increasing annealing temperature, confirming the growth of NbC nanocrystals. These results are consistent with the previous results obtained by HRTEM.

To reveal the elemental compositions and chemical states of the samples, X-ray photoelectron spectroscopy (XPS) analysis is performed. The survey spectrum confirms the



Fig. 2 a-d TEM images, e-h high-resolution TEM images, and i-l lattice fringe images of *a*-Nb₂O₅/CNS, *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS composites

coexistence of C, O, and Nb in the composites (Fig. S7). In the high-magnification Nb 3d XPS spectra of $a-Nb_2O_5/$ CNS and ov-Nb₂O₅/CNS, two characteristic peaks of Nb₂O₅ at approximately 207.2 and 209.9 eV index to the doublet of Nb $3d_{5/2}$ and Nb $3d_{3/2}$, respectively (Fig. 3b) [40]. For c-NbC/CNS and wc-NbC/CNS, the formation of NbC can be determined by an extra peak at 204.7 eV in the Nb 3d spectra, which is attributed to the Nb-C bond. Additionally, after annealing at 800 °C, the presence of oxygen vacancies in Nb₂O₅ is confirmed by the sub-peak at approximately 531.2 eV in the O 1s spectra of ov-Nb₂O₅/CNS-800 (Fig. 3c) [41]. The oxygen vacancy defect sites are evidenced by electron paramagnetic resonance (EPR) analysis, as shown in Fig. 3d. A remarkable symmetric peak with a g-value of 2.002 ascribed to unpaired electrons can be observed for ov-Nb₂O₅/CNS [42], while no detectable EPR signal exists in Nb₂O₅/CNS-700. Moreover, the oxygen vacancy concentration of ov-Nb₂O₅/CNS quantitatively reaches 4.59×10^{15} spins mg⁻¹.

The 2D lamellar carbon skeleton of all the samples can produce confinement to Nb-based nanoparticles to avoid aggregation and act as an in situ reductant to reduce Nb₂O₅ at a relatively high temperature. The graphitization degrees and structural defects of the composite 2D carbon skeletons are evaluated by Raman spectroscopy. As demonstrated in Fig. 3e, two distinct peaks at approximately 1340 and 1600 cm⁻¹ are fitted into four sub-peaks to accurately identify the intensities of the D band and G band. The I_D/I_G (area) values of *a*-Nb₂O₅/CNS, *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS are 1.55, 1.51, 1.64, and 1.45, respectively. The variation trend of I_D/I_G is consistent with the intensity alteration of the C–O sub-peak in



Fig. 3 a X-ray diffraction (XRD) patterns and **b** high-resolution Nb 3d XPS spectra of all samples. **c** High-resolution O 1s XPS spectra and **d** EPR spectra of *a*-Nb₂O₅/CNS and *ov*-Nb₂O₅/CNS. **e** Raman spectra and **f** high-resolution C 1s XPS spectra images of all samples

the high-resolution C 1s spectrum (Fig. 3f). These results imply that the presence of a large amount of amorphous Nb_2O_5 and the reduction in Nb_2O_5 can increase the structural defects and decrease the graphitization degree of the carbon skeleton.

3.2 Electromagnetic Wave Absorption Properties

As-synthesized composites with monodispersed Nb-based nanoparticles and conductive 2D carbon skeletons have great potential in absorbing EMW. Therefore, the associated EMW absorption performance characteristics of $a-Nb_2O_5/CNS$, $ov-Nb_2O_5/CNS$, c-NbC/CNS, and wc-NbC/CNS composites are systematically evaluated (Fig. 4). The reflection loss (RL) coefficient based on transmission line theory is used to quantify the EMW absorption performance [43], which generally requires less than – 10 dB (>90% wave absorption) to meet the demands of real-world applications. Notably, the $ov-Nb_2O_5/CNS$ exhibits the most competitive absorption performance among the four samples, achieving an extremely low minimum RL (RL_{min}) value of – 80.8 dB (>99.999999% wave absorption) at 7.11 GHz, with a

thickness of 2.76 mm (Fig. 4b). Moreover, *c*-NbC/CNS and *wc*-NbC/CNS deliver a relatively satisfactory absorbing performance of RL_{min} = -52.5 dB (5.08 GHz, 4.00 mm) and RL_{min} = -31.6 dB (16.32 GHz, 1.24 mm) (Fig. 4c–d). Unfortunately, *a*-Nb₂O₅/CNS demonstrates a disappointing EMW absorption performance of -4.8 dB (Fig. 4a). The effective absorption bandwidth (EAB) values of *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS are 3.37 GHz (1.30 mm), 3.61 GHz (1.43 mm), and 4.41 GHz (1.30 mm), respectively (Fig. S8). Generally, *ov*-Nb₂O₅/CNS exhibits a high-attenuation *RL*_{min} of -80.8 dB and a satisfactory EAB of 3.37 GHz, reflecting its extraordinary potential for EMW absorption.

Furthermore, a comparison of the EMW absorption performance between the ov-Nb₂O₅/CNS composite and other absorbers reported in the literature is conducted (Fig. 4e–f, Tables S2 and S3) [7, 9, 27, 44–56]. The as-prepared ov-Nb₂O₅/CNS composite exhibits ultrahigh EMW absorption performance with a thin matching thickness and a wide EAB. This phenomenon gives the composite a significant advantage over many other reported electromagnetic wave absorbers, making it an excellent choice for demanding EMW absorption applications.

3.3 Electromagnetic Wave Absorption Mechanism

The EMW attenuation performance characteristics of MAMs are closely associated with the dielectric and magnetic loss parameters. As a pure dielectric system, the EMW absorbing performance characteristics of a-Nb2O5/CNS, ov-Nb2O5/CNS, c-NbC/CNS, and wc-NbC/CNS are primarily attributed to the value of the complex permittivity ($\varepsilon_r = \varepsilon' - j\varepsilon''$). The disappointing EMW absorption performance of a-Nb₂O₅/CNS can be reflected by the extremely low values of both ε' and ε'' (Fig. 5a-b). The ε' for all the samples exhibits a generally decreasing trend as the frequency increases, which can be interpreted as the frequency dispersion behavior of the increased polarization hysteresis during the high-frequency electric field variation (Fig. 5a). Moreover, the multiple fluctuation peaks in ε'' of $ov-Nb_2O_5/$ CNS, c-NbC/CNS, and wc-NbC/CNS are assigned to the presence of polarization relaxation (Fig. 5b). Polarization relaxation can be induced by the electric dipoles in the Nb₂O₅ (or NbC) nanocrystals and interfacial polarizations at Nb₂O₅-carbon (or NbC-carbon) heterointerfaces. As shown in Fig. S9, the dielectric loss factor (tan δ_e) is applied to represent the EMW energy loss ability. Notably, *wc*-NbC/CNS has the highest tan δ_e , which may lead to impedance mismatching, thus weakening the microwave absorption performance.

The overall dissipation capabilities of these four samples to the incident microwave are further estimated by an attenuation constant (α) [4]. As illustrated in Fig. 5c, all the samples deliver α values with an overall upward trend of increasing frequency. The ov-Nb2O5/CNS composite delivers a larger α value than *a*-Nb₂O₅/CNS and c-NbC/CNS, implying its superior attenuation capacity. Conversely, the wc-NbC/CNS with the highest permittivity demonstrates an excessively large α value, potentially causing impedance mismatching. Then, a delta value $(|\Delta|)$ is calculated to evaluate the impedance matching [6], which represents the efficiency of EMW transmission into the absorber. The $|\Delta|$ values for *a*-Nb₂O₅/CNS are all above 0.4, demonstrating the extremely disappointing impedance matching ability (Fig. S10). Moreover, wc-NbC/CNS presents poor impedance matching, as indicated by the lowest integration area of $|\Delta|$ values smaller than 0.4 (Fig. 5f). Notably, ov-Nb₂O₅/CNS demonstrates the largest proper impedance matching area (rainbow-colored region, $|\Delta| < 0.2$) (Fig. 5d), which suggests a high incident efficiency of EMW into the absorbers. Therefore, the superior impedance matching and enhanced attenuation capabilities of ov-Nb₂O₅/CNS are favorable for realizing high-attenuation EMW absorption.

The Nb-based nanoparticles in the composites operate as dielectric loss active components to improve the EMW absorption performance. To theoretically reveal the different dielectric behaviors of two different Nb-based species, first-principles calculations based on the energy band structure of Nb₂O₅ and NbC configurations are conducted (Fig. 5g). As shown in Fig. 5h, the electronic property of Nb₂O₅ is semiconductive, which can be identified by a forbidden gap with a small energy gap of 2.86 eV originating from the separated valence band and the conduction band. In contrast, the overlapping of the valence band and conduction band for NbC reveals its conductive nature (Fig. 5i). The poor impedance matching of wc-NbC/CNS may be attributed to the large amount of conductive NbC in the 2D carbon skeleton, which leads to excessively high permittivity parameters. Moreover, compared with NbC conductors,



Fig. 4 a-d Three-dimensional reflection loss diagrams of a-Nb₂O₅/CNS, ov-Nb₂O₅/CNS, c-NbC/CNS, and wc-NbC/CNS, respectively. Comparison of EMW absorption properties among ov-Nb₂O₅/CNS and other absorbers reported in the literature: **e** minimum reflection loss *versus* thickness and **f** minimum reflection loss *versus* effective absorption bandwidth

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Fig. 5 a Real part of permittivity, **b** imaginary part of permittivity, and **c** attenuation constant (α) of all samples. **d–f** 2D delta ($|\Delta|$) value maps of *ov*-Nb₂O₅/CNS, *c*-NbC/CNS, and *wc*-NbC/CNS, respectively. **g** Structures of the Nb₂O₅ and NbC configurations. Energy band structures of **h** Nb₂O₅ and **i** NbC. **j** Off-axis electron holograms and **k** stray field flux lines of *ov*-Nb₂O₅/CNS. **l** Off-axis electron holograms and **m** stray field flux lines of *wc*-NbC/CNS

 Nb_2O_5 semiconductors with reduced charge transfer ability slow the electrical neutralization of the dipole under an alternating electromagnetic field and hence reinforce dipole polarization relaxation. As a result, ov-Nb₂O₅/CNS with plenty of well-crystallized Nb₂O₅ semiconductors can achieve superior dielectric loss.

Nb₂O₅ and NbC with different dielectric behaviors can impact the electromagnetic responses of the composites. Off-axis electron holography is conducted to investigate the interaction of ov-Nb₂O₅/CNS and wc-NbC/CNS with the electromagnetic field (Fig. 5j-m). As shown in Fig. 5k, high-density stray field flux lines robustly penetrate from the inside of the ov-Nb₂O₅/CN nanosheet and radiate to the outside. This result confirms the significant interaction between the ov-Nb₂O₅/CNS composite and electromagnetic field. However, no obvious stray field flux lines are triggered to be released from wc-NbC/CNS (Fig. 5m). It can be speculated that the Nb₂O₅ nanoparticles produce plentiful dielectric polarization and activate a strong induced electromagnetic field to influence the EMW. Moreover, the divergence of stray field flux lines in ov-Nb₂O₅/CNS builds a multidimensional electromagnetic response network and effectively reinforces the attenuation of electromagnetic energy.

Moreover, the incorporation of Nb2O5 and NbC nanoparticles into the carbon nanosheets generates multiple heterointerfaces. Interfacial charge aggregation occurs due to the disrupted original electric balance at these heterointerfaces, which typically leads to interfacial polarization. Notably, Nb₂O₅ and NbC with different electronic properties can affect the charge aggregation at Nb₂O₅-carbon and NbC-carbon interfaces, respectively. Therefore, the charge density distributions of the Nb₂O₅-carbon and NbC-carbon configurations are investigated to theoretically reveal the interfacial polarization induced by Nb-based nanoparticles (Figs. 6a-f and S11). As depicted in Fig. 6a, b, d, e, charges are unevenly distributed at these heterointerfaces, where irregular yellow and blue regions correspond to the aggregation and dispersion of electrons, respectively. It is apparent that the electrons generally delocalize from the carbon and flow into Nb₂O₅ at the Nb₂O₅-carbon heterointerface, while the delocalized electrons flow from the carbon and NbC to the intermediate region at the NbC-carbon interface. These results are supported by the planar average potential of the Nb₂O₅-carbon and NbC-carbon interfaces (Fig. 6c, f). Moreover, the Nb_2O_5 -carbon heterointerface demonstrates a pronounced charge separation effect, which consequently enhances the interfacial polarization capability. The ov-Nb₂O₅/CNS with abundant well-crystallized Nb₂O₅ nanoparticles possesses numerous Nb2O5-carbon heterointerfaces. Consequently, intensified electromagnetic energy

dissipation through strengthened interfacial polarization loss can be achieved by *ov*-Nb₂O₅/CNS.

Complementary to excellent interfacial polarization loss, the presence of abundant oxygen vacancy defects in Nb₂O₅ nanoparticles can contribute to the excellent electric dipole polarization in ov-Nb₂O₅/CNS. The charge distributions of perfect Nb₂O₅ and oxygen vacancy Nb₂O₅ configurations are estimated by first-principles calculations to better understand the intensification of electric dipole polarization for ov-Nb₂O₅/CNS. Figure S12 presents the configurations of perfect Nb₂O₅ and oxygen vacancy Nb₂O₅ (ov-Nb₂O₅). The charge distribution in perfect Nb₂O₅ is relatively uniformly distributed (Fig. 6g). However, with the presence of two oxygen vacancies, electrons delocalize at the vacancy sites and flow into nearby oxygen atoms, resulting in charge separation (Fig. 6h). Subsequently, electric dipoles are generated in ov-Nb₂O₅, which can induce the formation of electronic dipole polarization oscillation in an external electromagnetic field. Therefore, Nb₂O₅ nanoparticles with abundant oxygen vacancies can act as electronic dipole polarization oscillation units, thereby efficiently enhancing the dielectric loss capability of ov-Nb₂O₅/CNS.

To gain a comprehensive understanding of the dielectric loss mechanisms, Cole–Cole plots based on Debye theory are further investigated. As depicted in Fig. S13, the rich polarization relaxation processes for the *ov*-Nb₂O₅/CNS are suggested by a more perfect Cole–Cole semicircle with a large diameter. In contrast, the highly distorted Cole–Cole trajectories observed for other composites suggest a weakened Debye dipolar relaxation.

Furthermore, the 2D lamellar morphology of ov-Nb₂O₅/ CNS is beneficial for establishing abundant nanosheet–paraffin interfaces, which can strengthen the multiple reflection and scattering dissipation characteristics. To enable 3D reconstruction of the phase diagram, the ov-Nb₂O₅/CNSparaffin absorber is scanned using a microcomputer tomography (micro-CT) device, as shown in Fig. 6i–j. Image slicing techniques are employed to acquire the nondestructive ov-Nb₂O₅/CNS phase (blue phase) and paraffin phase (transparent gray phase). The ov-Nb₂O₅/CNS nanosheets stack in the absorber and form numerous voids, and these voids are filled with paraffin wax. This unique structure can produce multiple macro-heterointerfaces to enhance interfacial polarization loss and prolong the reflection and scattering path of the EMW to enhance attenuation. Additionally, upon



Fig. 6 Charge density distribution of the **a–b** Nb₂O₅–carbon configuration and **d–e** NbC–carbon configuration. Planar average electrostatic potential of **c** Nb₂O₅–carbon and **f** NbC–carbon. **g** Charge density contour of perfect Nb₂O₅. **h** Charge density distribution of ov-Nb₂O₅ (yellow and blue regions represent aggregation and dispersion of electrons). **i-k** Micro-CT images of the ov-Nb₂O₅/CNS-paraffin absorber by 3D reconstruction (dark blue and gray regions refer to the ov-Nb₂O₅/CNS and paraffin wax phase, respectively). **l** Schematic illustrations of the multiple scattering and conductive loss mechanism of the 2D lamellar ov-Nb₂O₅/CNS. (Color figure online)

removing the paraffin phase, an interconnected 3D conductive ov-Nb₂O₅/CNS network is observed (Fig. 6k), which can realize intensive conductivity loss. Generally, these results suggest that the ov-Nb₂O₅/CNS in the absorber can efficiently strengthen the attenuation of EMW by intensifying interfacial polarization loss, multiple reflections, scattering dissipation, and conductivity loss.

3.4 Possible Application Prospects

In practical applications, EMW absorbing materials are mainly used as coatings or plates. To verify the application potential of ov-Nb₂O₅/CNS, the composite is mixed with EMW-transmitting cyanate ester to form a mixed resin. The functional resin can successfully cure into a rectangular plate

 $(80 \text{ mm} \times 40 \text{ mm} \times 2 \text{ mm})$, as shown in Fig. 7a. Moreover, the ov-Nb₂O₅/CNS-cyanate plate demonstrates excellent machinability, allowing it to be precisely cut to a size of 22.9 mm \times 10.2 mm \times 2 mm to investigate the microwave absorption performance using a waveguide method (Fig. S14). As shown in Figs. 7b-c and S15, the ov-Nb₂O₅/CNScyanate plate achieves a satisfying RL_{min} value of - 19.8 dB at 12.38 GHz with a thickness of 1.88 mm, and the EAB reaches 2.41 GHz with a thickness of 2.10 mm. Since the absorption mechanism of EMW absorbers is dominated by transforming electromagnetic energy into thermal energy, the heat dissipation capacity is an imperative factor in practical applications. When the ov-Nb₂O₅/CNS-cyanate plate is placed on a heating platform (160 °C), the temperature of the plate increases rapidly to 90.2 °C (56.4% of 160 °C) within 10 s and reaches a stable temperature of approximately 150 °C (93.8% of 160 °C) after 180 s (Figs. 7d-f and S16). Overall, ov-Nb₂O₅/CNS-800 demonstrates excellent application potential by curing into an excellent microwaveabsorbing, machinable, and heat-dissipating ov-Nb2O5/CNScyanate plate.

4 Conclusions

In conclusion, we demonstrate the successful synthesis of ultrafine (~10 nm) Nb₂O₅ nano-semiconductors with rich oxygen vacancies in carbon nanosheets (ov-Nb₂O₅/ CNS) for achieving high-attenuation electromagnetic wave absorption. Semiconductive Nb₂O₅ nanoparticles endow ov-Nb₂O₅/CNS with more strengthened charge polarization at the Nb₂O₅-carbon hetero-interface than NbC conductors, which facilitates interfacial polarization loss. Additionally, Nb₂O₅ nanocrystals with abundant oxygen vacancies reinforce electric dipole polarization inside the semiconductor. The 2D lamellar morphology of ov-Nb₂O₅/CNS strengthens the multiple reflection and scattering dissipation characteristics in the absorbers. Therefore, when integrated with outstanding polarization relaxation, intensified electromagnetic response, and excellent impedance matching, ov-Nb2O5/CNS achieves a superior EMW absorption capability with an unparalleled RL_{min} of -80.8 dB (> 99.999999% wave absorption) at 7.11 GHz (2.76 mm); additionally, it exhibits a wide



Fig. 7 a Digital photographs of the ov-Nb₂O₅/CNS-cyanate plate. b Three-dimensional reflection loss diagrams and c reflection loss contour maps of the ov-Nb₂O₅/CNS-cyanate plate. d-f Thermal infrared images of the ov-Nb₂O₅/CNS-cyanate plate on a heating platform (160 °C)

effective absorption bandwidth of 3.37 GHz at 1.30 mm. Moreover, the composite shows excellent application potential by curing into a microwave-absorbing, machinable, and heat-dissipating ov-Nb₂O₅/CNS-cyanate plate. The ov-Nb₂O₅/CNS-cyanate plate achieves a satisfactory RL_{min} value of – 19.8 dB at 12.38 GHz with a thickness of 1.88 mm, and the EAB reaches 2.41 GHz with a thickness of 2.10 mm. Our findings provide novel insights into the design and development of dielectric semiconductor-based carbon composites for electromagnetic wave absorption and related applications.

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Declarations

Conflict of interest The authors declare no conflict of interest. They have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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