

Supporting Information for

Amphipathic Phenylalanine Induced Nucleophilic-Hydrophobic Interface towards Highly Reversible Zn Anode

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Supplementary Figures

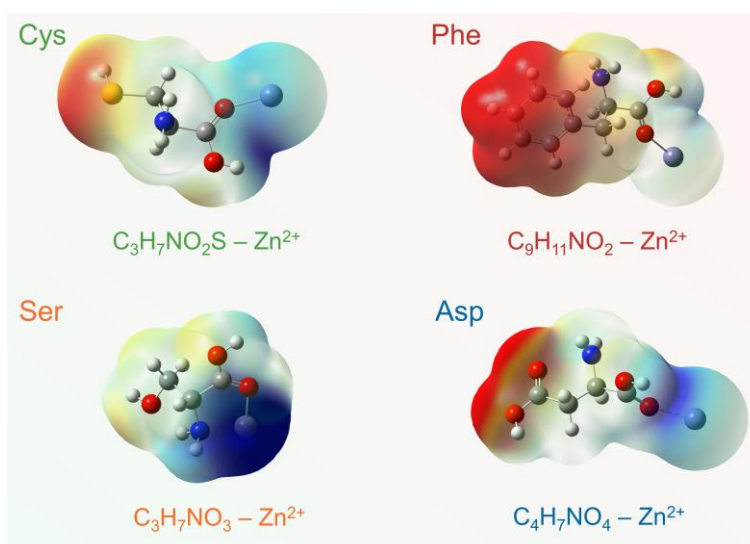


Fig. S1 Electrostatic potential mapping of Phe/Asp/Ser/Cys molecules within interaction with Zn²⁺

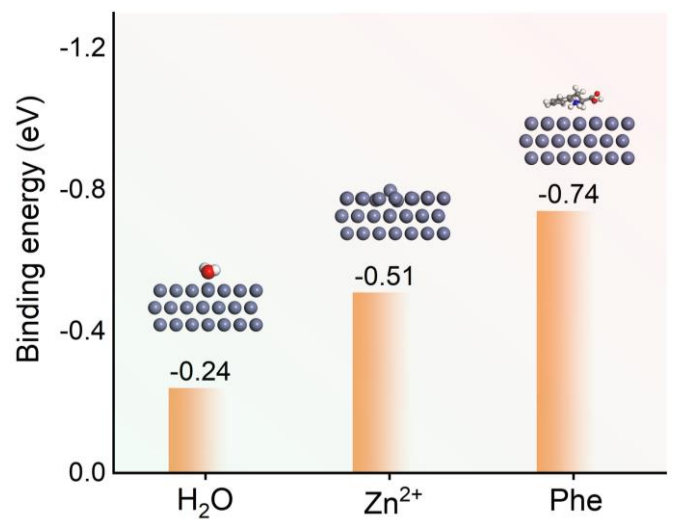


Fig. S2 The adsorption energy of Phe, Zn²⁺, and H₂O on the Zn (002) facet

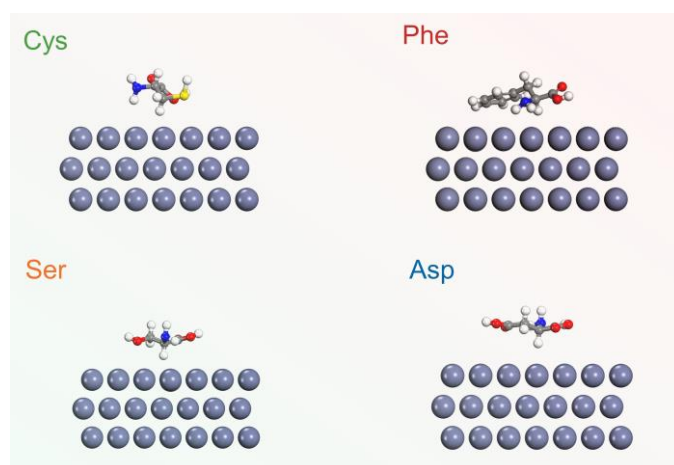


Fig. S3 The adsorption model of Phe/Asp/Ser/Cys molecules on Zn (002) facet

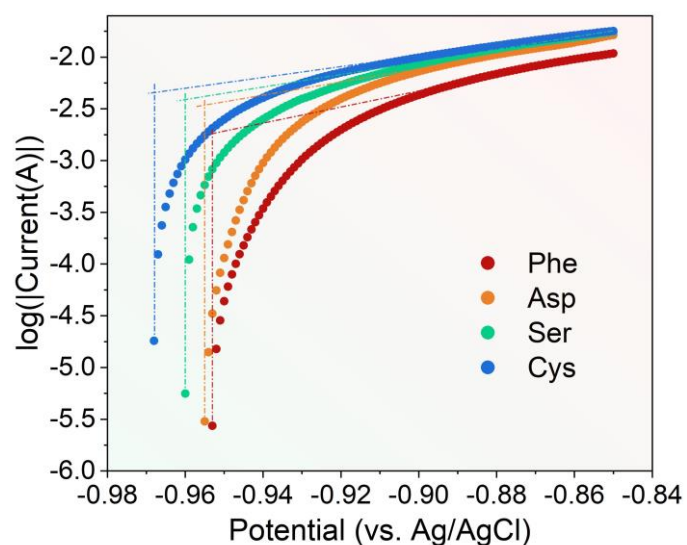


Fig. S4 Tafel plots of the Zn electrodes measured in ZSO electrolytes with 20 mmol L⁻¹ Phe/Asp/Ser/Cys additives

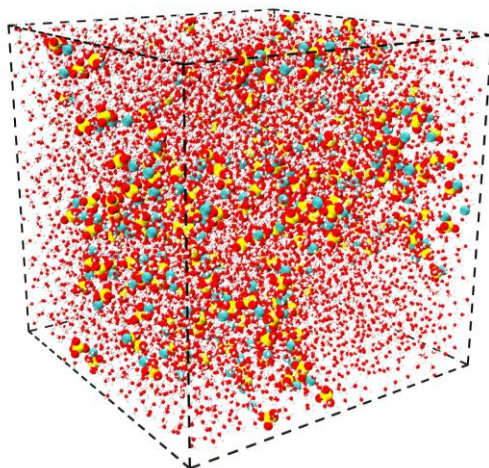


Fig. S5 Images of ZSO system obtained from molecular dynamics simulations

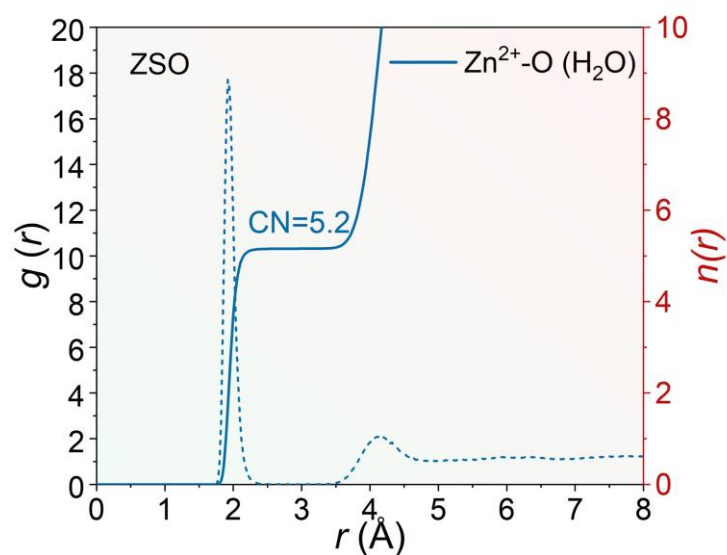


Fig. S6 The radial distribution functions for Zn^{2+} -O (H_2O) in ZSO electrolyte

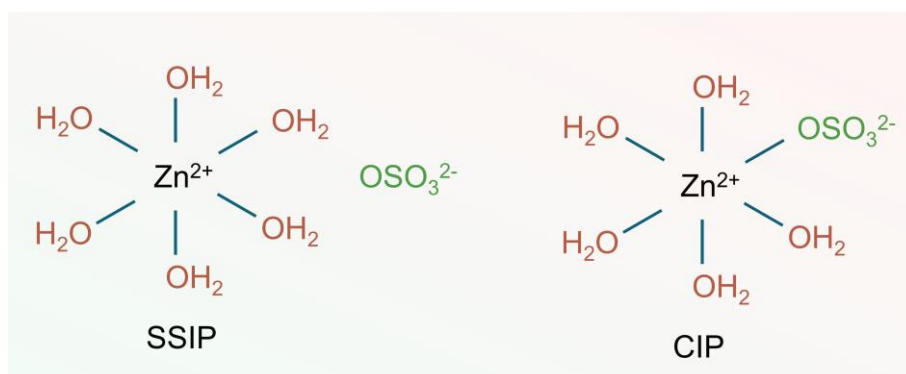


Fig. S7 Schematic illustration of Zn^{2+} solvation structures in ZSO/Phe electrolyte

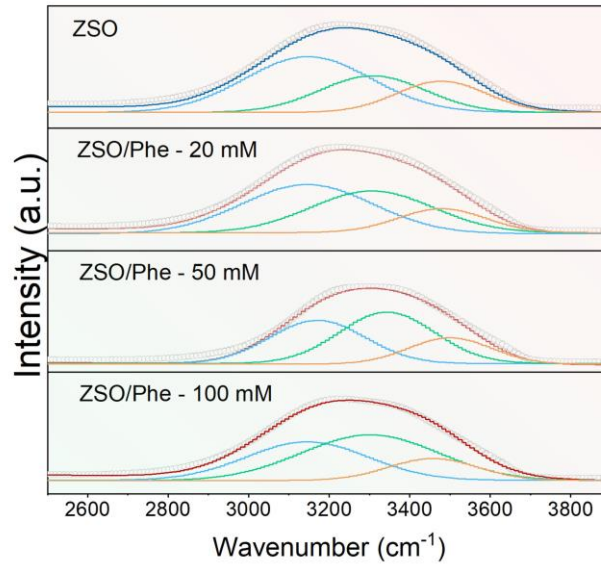


Fig. S8 FTIR spectra for characteristic peaks assigned to the O-H stretching vibration in ZSO and ZSO/Phe system

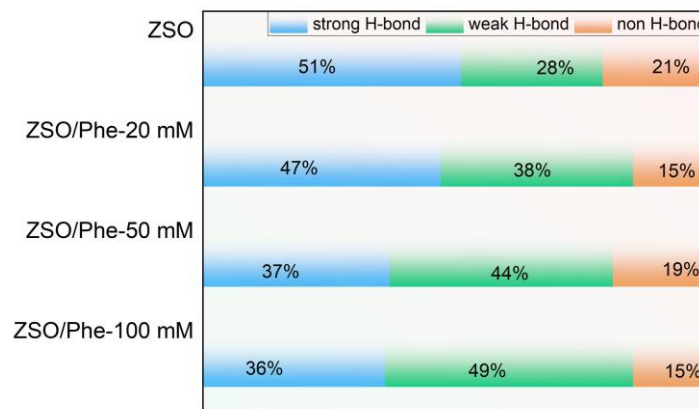


Fig. S9 The H-bond proportion graphs obtained from FTIR spectra

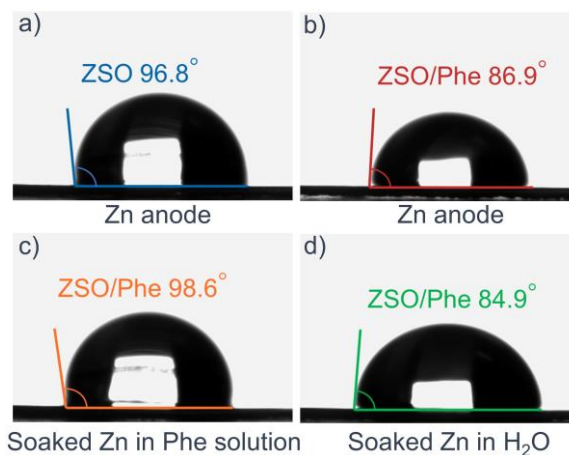


Fig. S10 Contact angles of ZSO electrolyte (a) and ZSO/Phe electrolyte (b) on Zn anode, and ZSO/Phe electrolyte on pre-soaked Zn anode of Phe-contained aqueous solution (c), and ZSO/Phe electrolyte on pre-soaked Zn anode of deionized water (d)

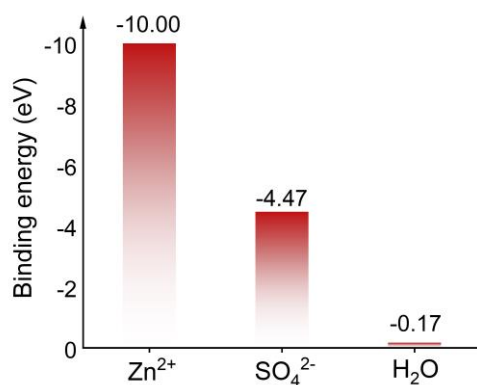


Fig. S11 Binding energy for Phe with SO₄²⁻, Zn²⁺ and H₂O molecule by the density functional theory calculation

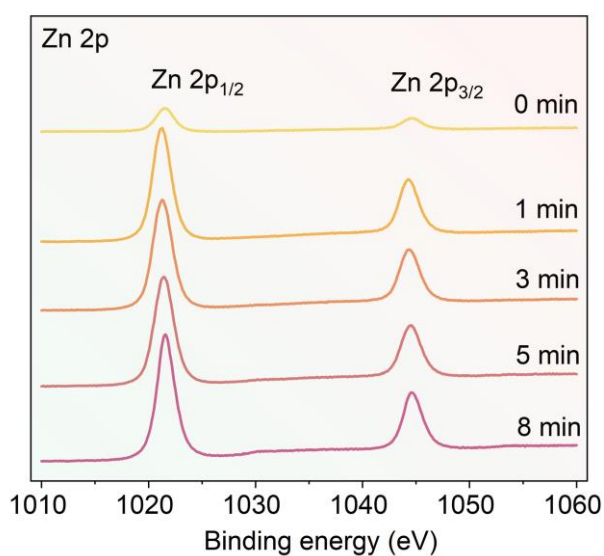


Fig. S12 XPS depth profile of Zn 2p for Zn anode cycled in ZSO/Phe electrolyte

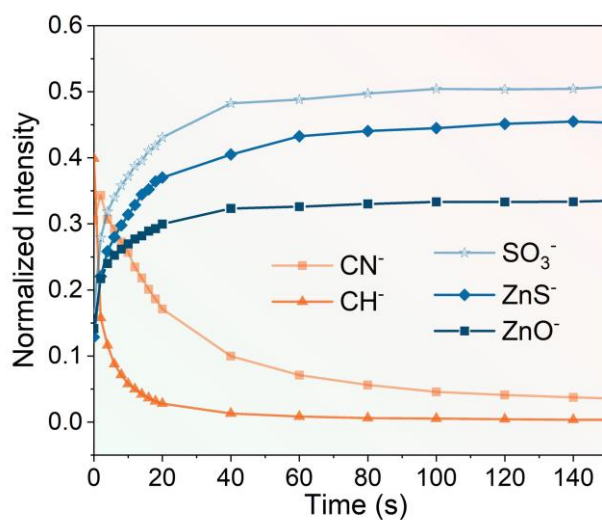


Fig. S13 The normalized intensity of organic CN⁻ and CH⁻ and inorganic ZnS⁻, ZnO⁻, and SO₃⁻ species

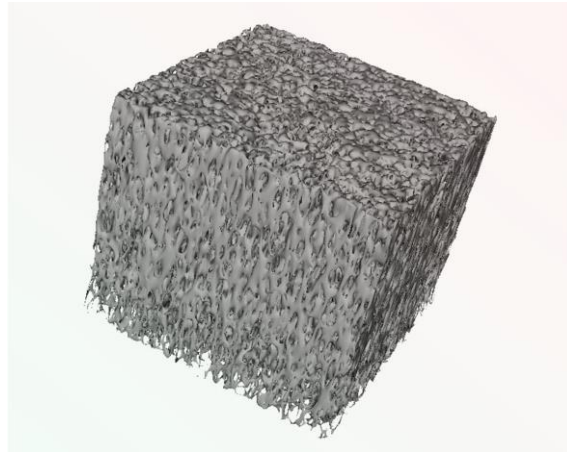


Fig. S14 3D visualization of TOF-SIMS for Zn

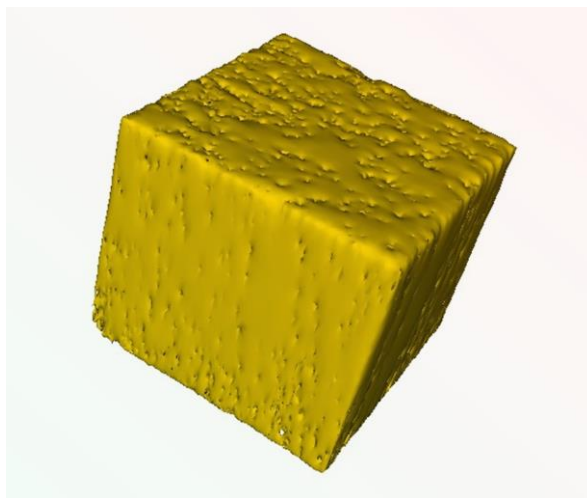


Fig. S15 3D visualization of TOF-SIMS for ZnO⁻

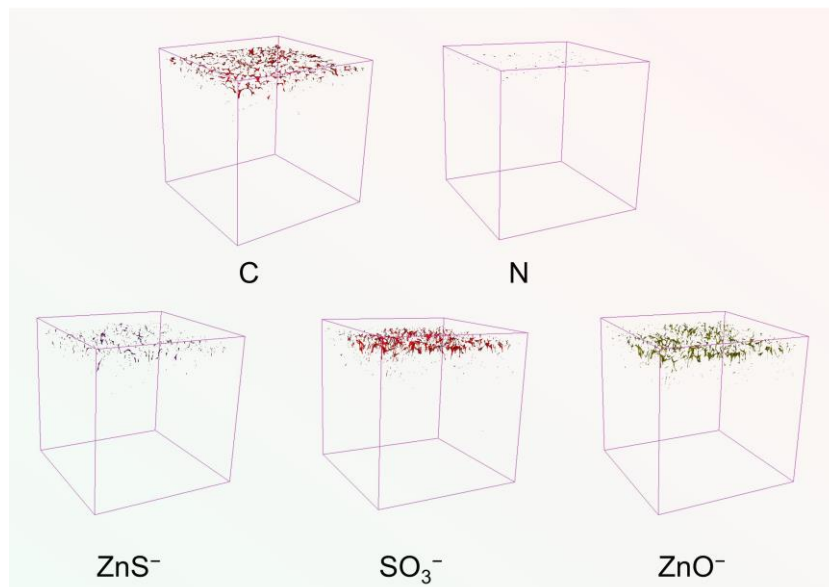


Fig. S16 3D visualization of TOF-SIMS for C, N elements, ZnS⁻, SO₃⁻, and ZnO⁻ species in ZSO electrolyte

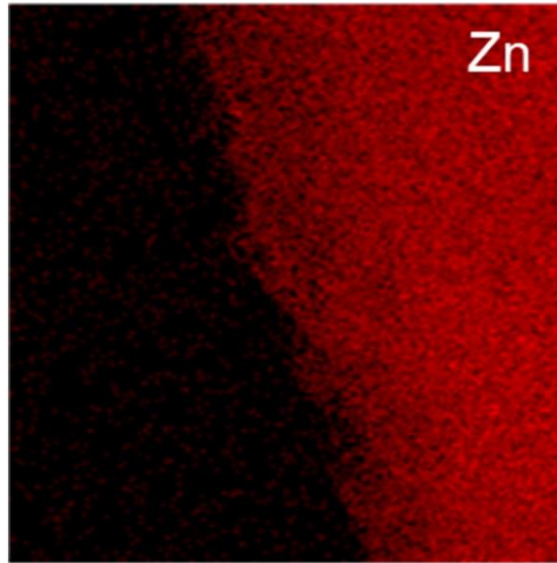


Fig. S17 HRTEM image of the electrode interface and the Zn elemental mapping

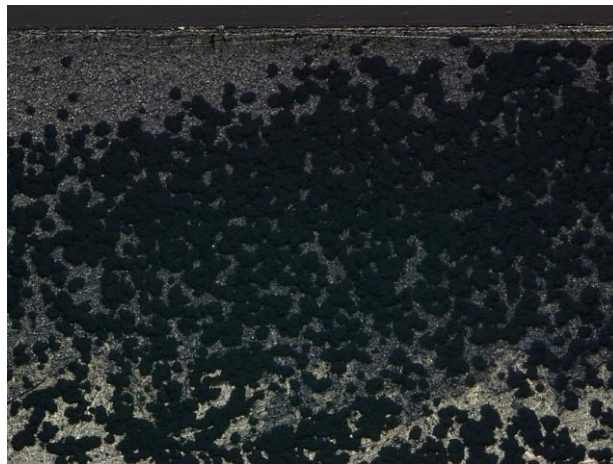


Fig. S18 The deposition morphology of Zn anode electrodeposited in ZSO electrolyte

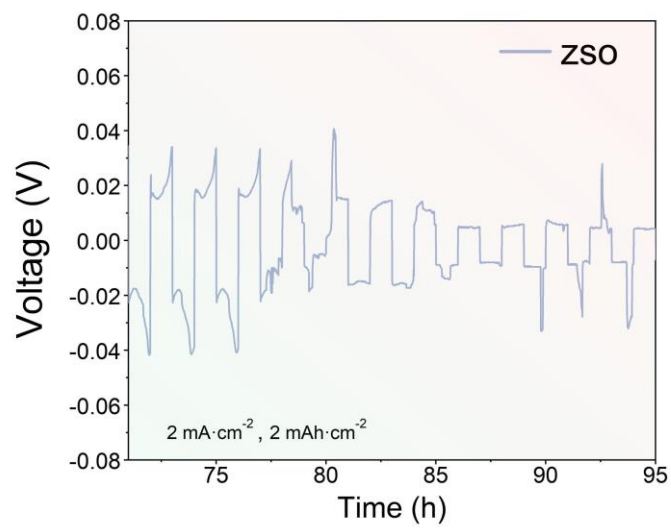


Fig. S19 The galvanostatic cycling of Zn||Zn cells in ZSO electrolyte

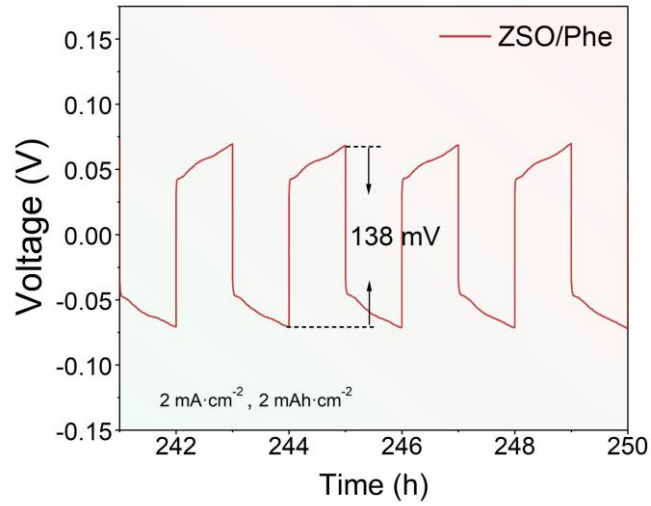


Fig. S20 The galvanostatic cycling of Zn||Zn cells in ZSO/Phe electrolyte

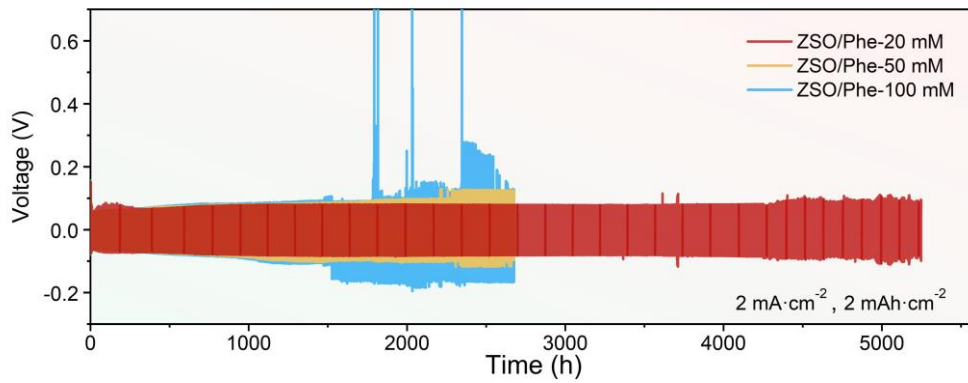


Fig. S21 The galvanostatic cycling of Zn||Zn cells in different Phe concentration electrolytes

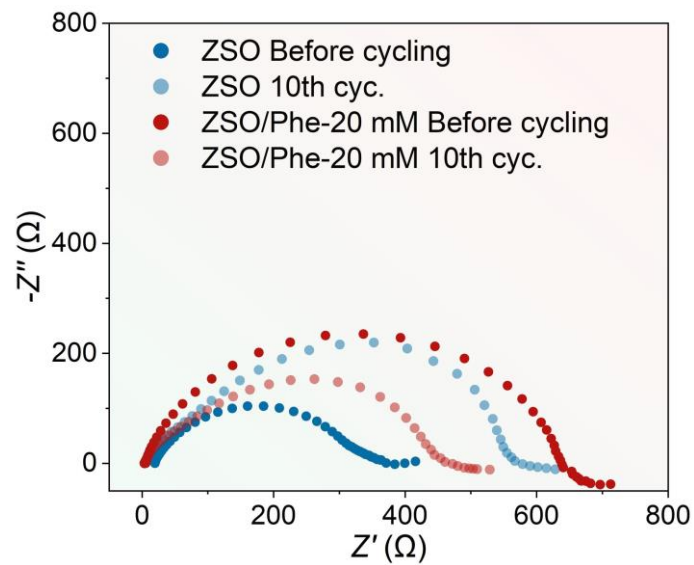


Fig. S22 EIS curves of Zn||Zn symmetric cells using ZSO and ZSO/Phe-20 mM electrolytes before and after 10 cycles

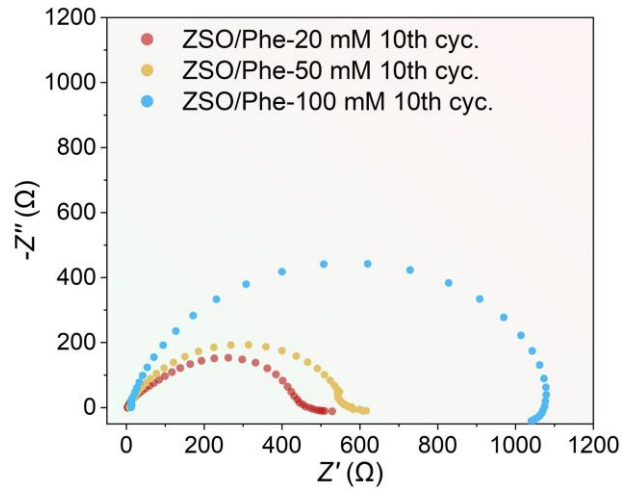


Fig. S23 EIS curves of Zn||Zn symmetric cells using ZSO/Phe-20 mM, ZSO/Phe-50 mM and ZSO/Phe-100 mM electrolytes after 10 cycles

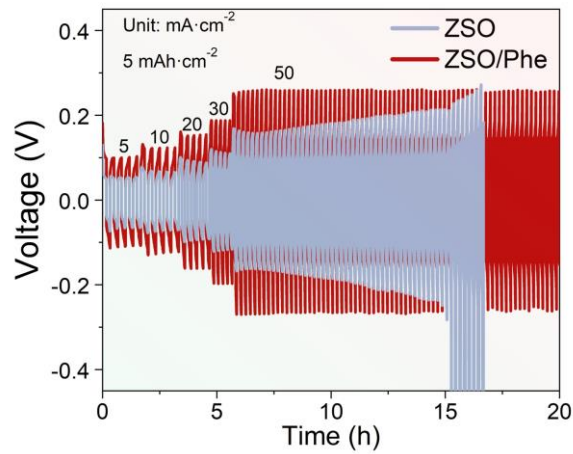


Fig. S24 The galvanostatic cycling of Zn||Zn cells under condition of 50 mA cm^{-2} , 5 mAh cm^{-2} in the initial stage

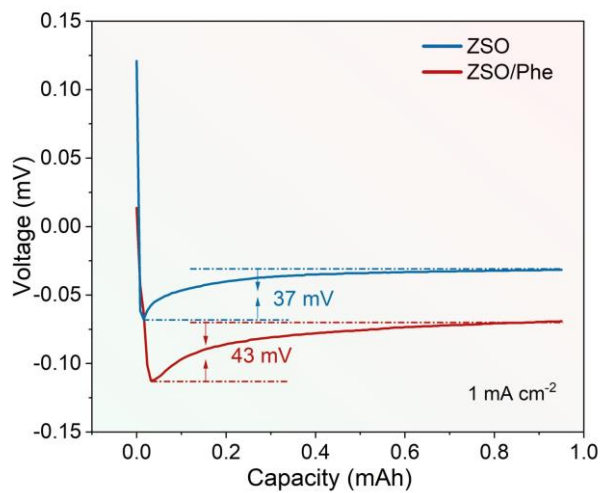


Fig. S25 The nucleation overpotential tested in Zn||Cu half cells

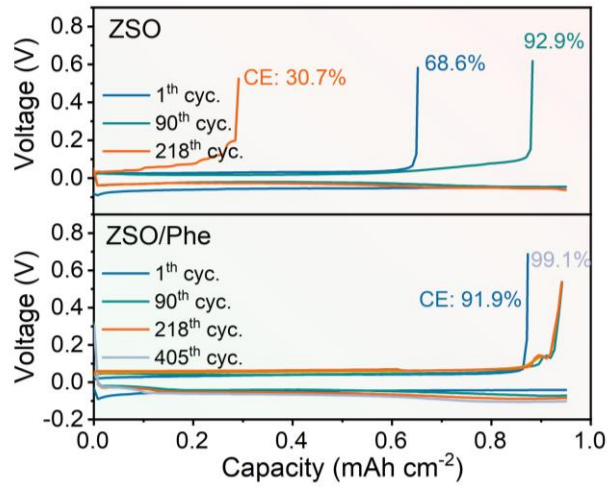


Fig. S26 The corresponding voltage profiles for CE measurement in Zn||Cu cells

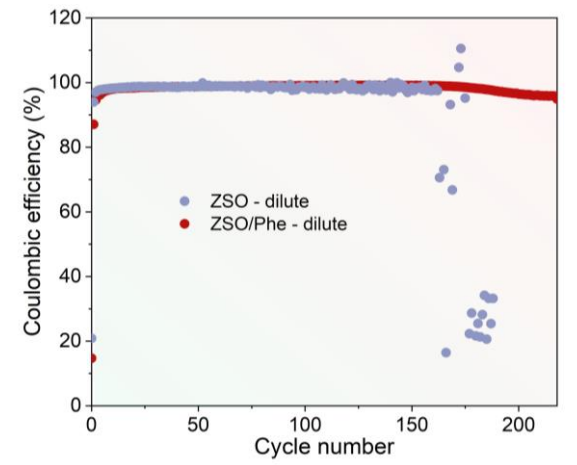


Fig. S27 The CE of the Zn plating/stripping in dilute electrolytes

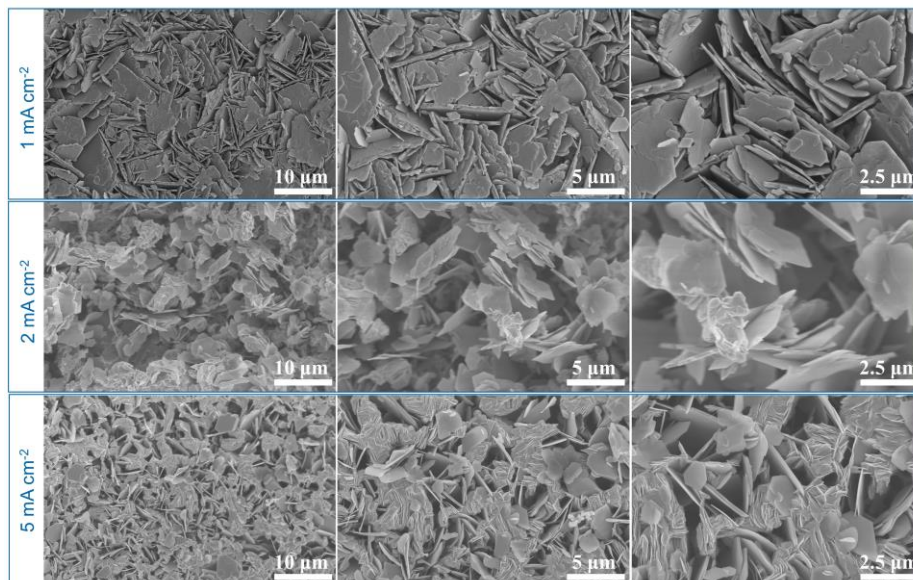


Fig. S28 SEM images of the deposited Zn metal for 5 cycles at 1 mA cm^{-2} , 2 mA cm^{-2} and 5 mA cm^{-2} in ZSO electrolyte

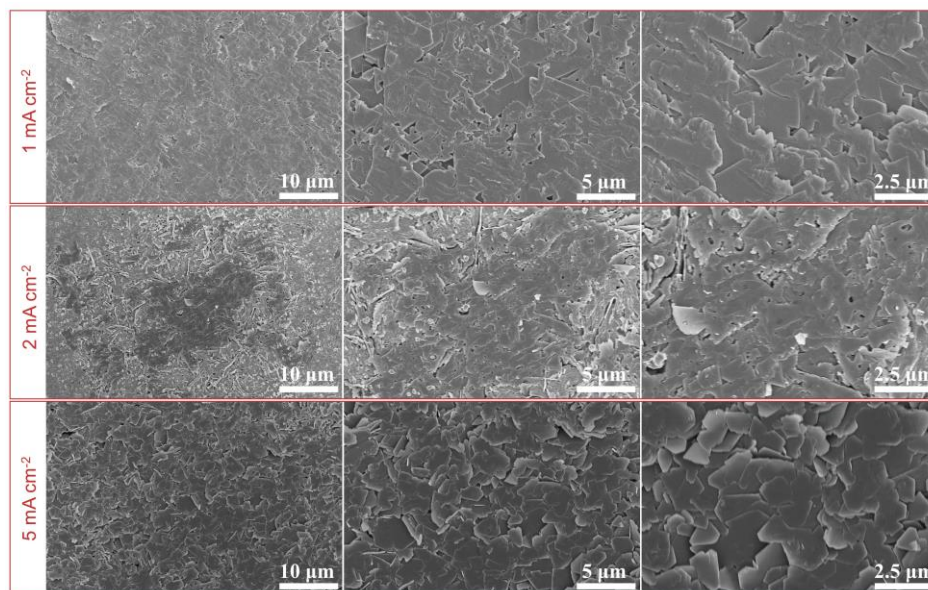


Fig. S29 SEM images of the deposited Zn metal for 5 cycles at 1 mA cm^{-2} , 2 mA cm^{-2} and 5 mA cm^{-2} in ZSO/Phe electrolyte

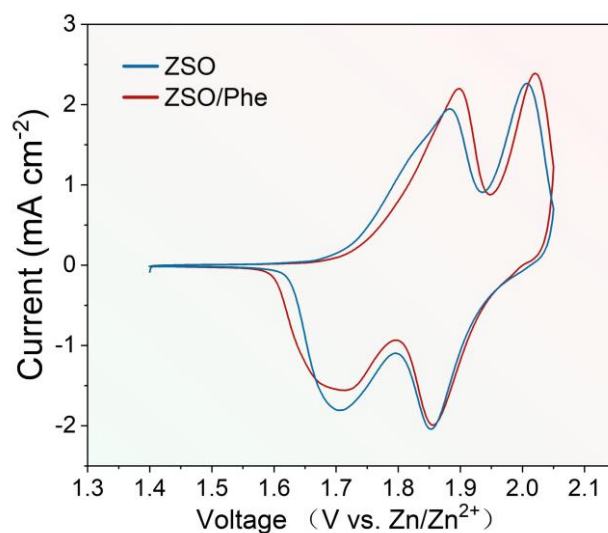


Fig. S30 CV curves of Zn||LMO full cells with ZSO and ZSO/Phe electrolytes

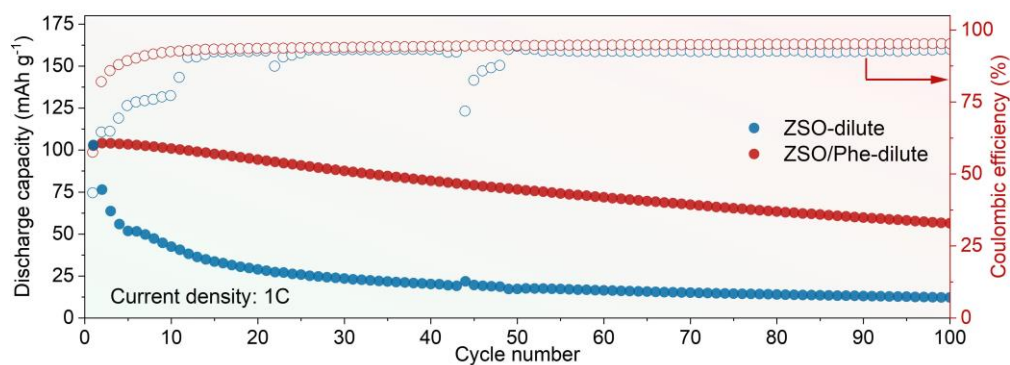


Fig. S31 Zn||LMO full cells cycling performance in dilute electrolytes

Table S1 Comparison of the cycling stability of Zn||Zn cells with previous reports

Additives	Test condition	Lifespan	Cumulative Capacity (Ah cm ⁻²)	References
Phe	2 mA cm ⁻² , 2 mAh cm ⁻²	5250 h	5.25 Ah cm ⁻²	This work
	5 mA cm ⁻² , 5 mAh cm ⁻²	2645 h	6.6 Ah cm ⁻²	
	10 mA cm ⁻² , 10 mAh cm ⁻²	990 h	4.95 Ah cm ⁻²	
	20 mA cm ⁻² , 10 mAh cm ⁻²	466 h	4.66 Ah cm ⁻²	
	30 mA cm ⁻² , 10 mAh cm ⁻²	338 h	5.07 Ah cm ⁻²	
	50 mA cm ⁻² , 5 mAh cm ⁻²	155 h	3.87 Ah cm ⁻²	
His	2 mA cm ⁻² , 2 mAh cm ⁻²	3000 h	3 Ah cm ⁻²	[54]
	5 mA cm ⁻² , 2 mAh cm ⁻²	2300 h	5.75 Ah cm ⁻²	
	10 mA cm ⁻² , 4 mAh cm ⁻²	670 h	3.35 Ah cm ⁻²	
Glu	10 mA cm ⁻² , 10 mAh cm ⁻²	520 h	2.6 Ah cm ⁻²	[56]
	20 mA cm ⁻² , 20 mAh cm ⁻²	176 h	1.76 Ah cm ⁻²	
Cys	2 mA cm ⁻² , 1 mAh cm ⁻²	2000 h	2 Ah cm ⁻²	[57]
	5 mA cm ⁻² , 5 mAh cm ⁻²	600 h	1.5 Ah cm ⁻²	
	10 mA cm ⁻² , 10 mAh cm ⁻²	260 h	1.3 Ah cm ⁻²	
Arg	5 mA cm ⁻² , 4 mAh cm ⁻²	2200 h	5.5 Ah cm ⁻²	[33]
	10 mA cm ⁻² , 4 mAh cm ⁻²	900 h	4.5 Ah cm ⁻²	
Gly	10 mA cm ⁻² , 10 mAh cm ⁻²	460 h	2.3 Ah cm ⁻²	[48]
	20 mA cm ⁻² , 20 mAh cm ⁻²	80 h	0.8 Ah cm ⁻²	
PASP	5 mA cm ⁻² , 2.5 mAh cm ⁻²	550 h	1.38 Ah cm ⁻²	[55]
	20 mA cm ⁻² , 1 mAh cm ⁻²	200 h	2 Ah cm ⁻²	
Ser	5 mA cm ⁻² , 5 mAh cm ⁻²	800 h	2 Ah cm ⁻²	[53]
SH	10 mA cm ⁻² , 10 mAh cm ⁻²	820 h	4.1 Ah cm ⁻²	[58]
	20 mA cm ⁻² , 20 mAh cm ⁻²	480 h	4.8 Ah cm ⁻²	
	30 mA cm ⁻² , 10 mAh cm ⁻²	180 h	2.7 Ah cm ⁻²	
Glucose	2 mA cm ⁻² , 2 mAh cm ⁻²	740 h	0.74 Ah cm ⁻²	[25]
	5 mA cm ⁻² , 5 mAh cm ⁻²	270 h	0.675 Ah cm ⁻²	