

Supporting Information for

Flexible Conductive Anodes Based on 3D Hierarchical Sn/NS CNFs@rGO Network for Sodium-Ion Batteries

Linqu Luo¹, Jianjun Song^{1,*}, Longfei Song¹, Hongchao Zhang¹, Yicheng Bi², Lei Liu³, Longwei Yin⁴, Fengyun Wang^{1,*}, Guoxiu Wang^{5,*}

¹College of Physics and State Key Laboratory of Bio-Fibers and Eco-Textiles, Qingdao University, Qingdao 266071, People's Republic of China

²College of Electromechanical Engineering, Qingdao University of Science & Technology, No.99 Songling Road, Qingdao, Shandong 260061, People's Republic of China

³School of Materials Science and Engineering, Shandong University of Science and Technology, Qingdao 266590, People's Republic of China

⁴School of Materials Science and Engineering, Shandong University, Jinan 250061, People's Republic of China

⁵Centre for Clean Energy Technology, University of Technology Sydney, Broadway, Sydney, NSW 2007, Australia

*Corresponding authors. E-mail: jianjun.song@qdu.edu.cn (Jianjun Song); fywang@qdu.edu.cn (Fengyun Wang); Guoxiu.Wang@uts.edu.au (Guoxiu Wang)

Supplementary Figures and Table

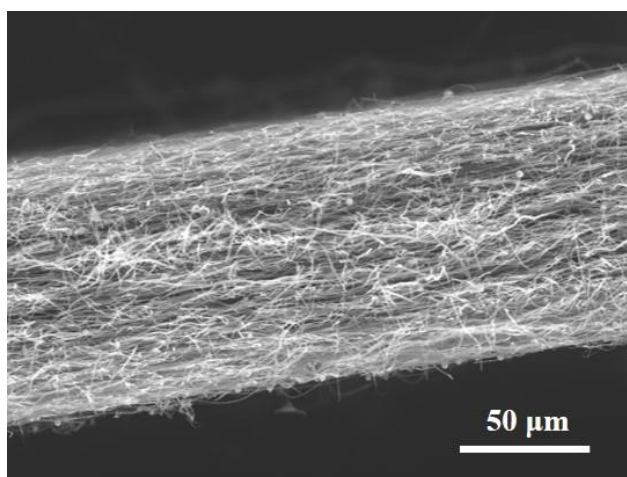


Fig. S1 Cross-sectional SEM image of the Sn/N-CNFs membrane with thickness of 109 μm

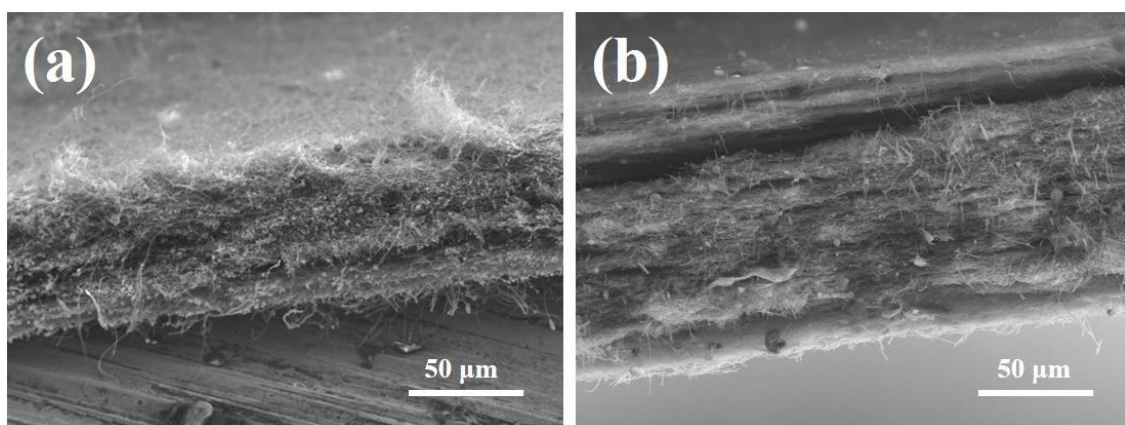


Fig. S2 Cross-sectional SEM image of the Sn/N-CNFs membrane with different thickness of **a** 78 μm , **b** 139 μm

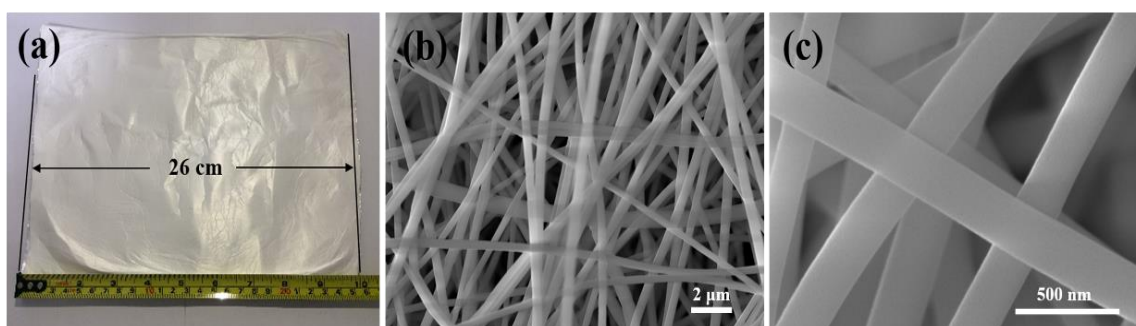


Fig. S3 **a** Digital photos of as-spun PAN/SnCl₂/thiourea nanofibers membrane; **b** and **c** SEM images of PAN/SnCl₂/thiourea precursor nanofibers

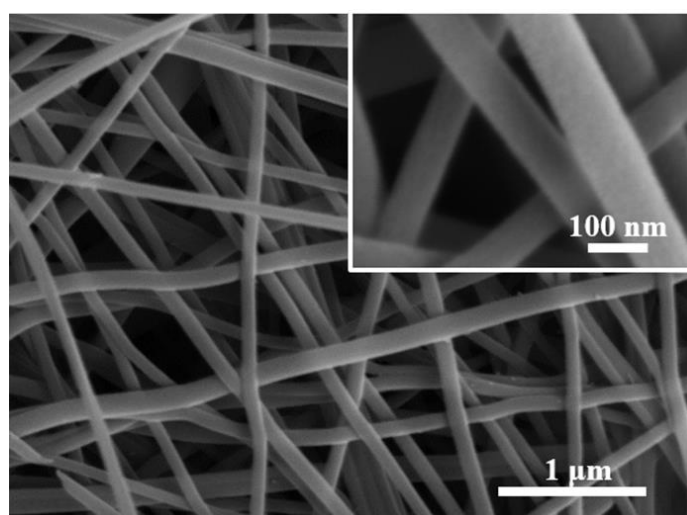


Fig. S4 SEM images of Sn/NS-CNFs

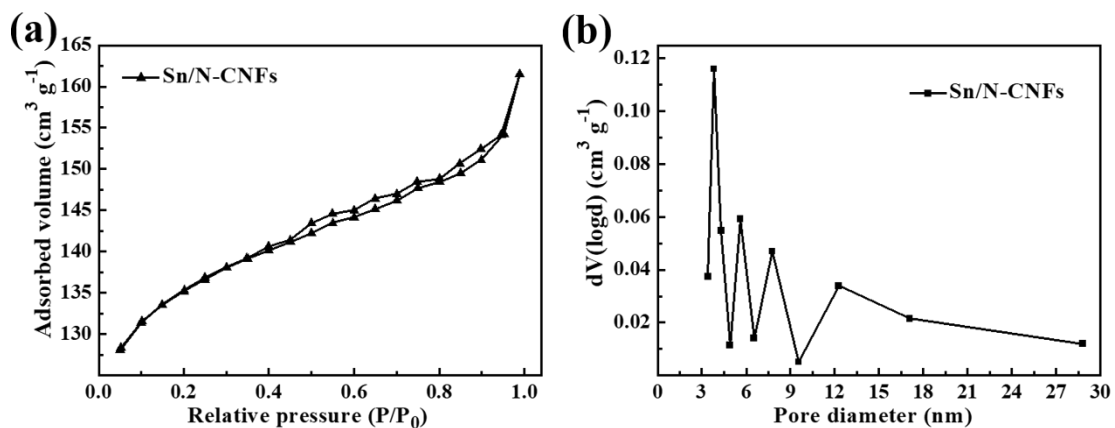


Fig. S5 **a** N_2 adsorption–desorption isotherms and **b** the corresponding pore size distribution curves of Sn/N-CNFs

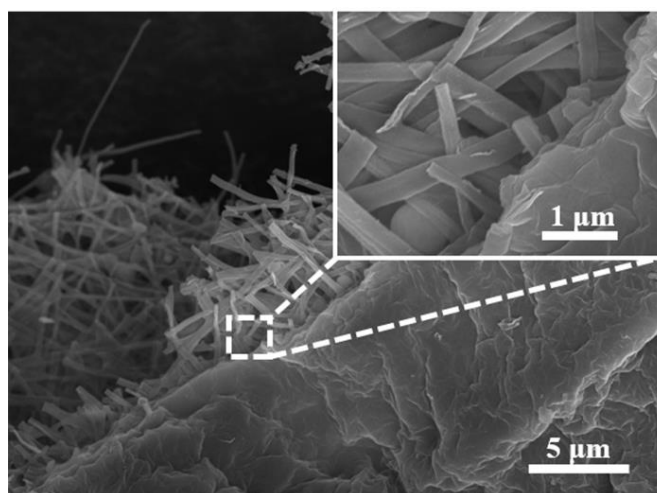


Fig. S6 SEM images of Sn/N-CNFs@rGO

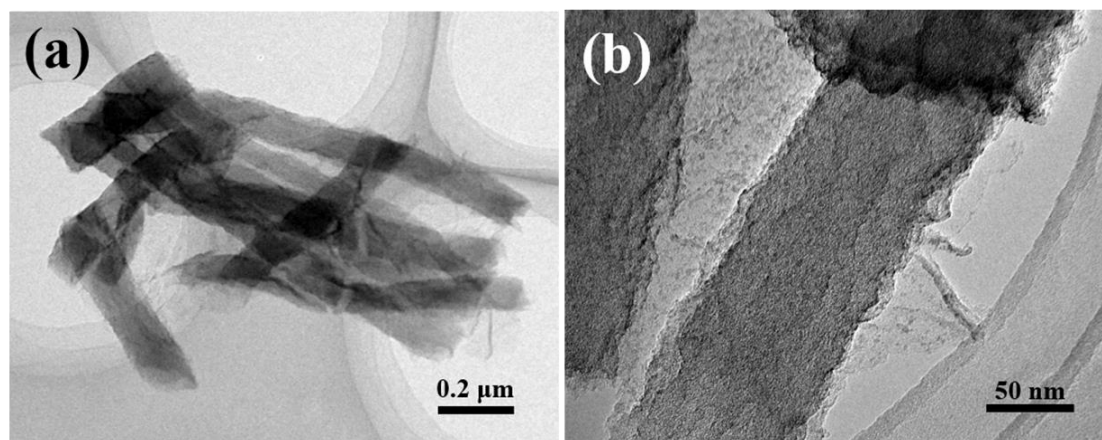


Fig. S7 **a** Low- and **b** high-magnification TEM images of Sn/NS-CNFs@rGO

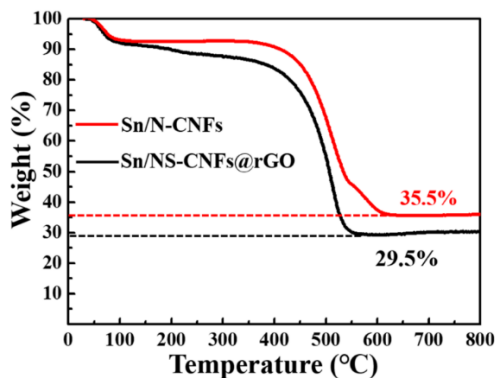


Fig. S8 The TGA data of Sn/N-CNFs and Sn/NS-CNFs@rGO

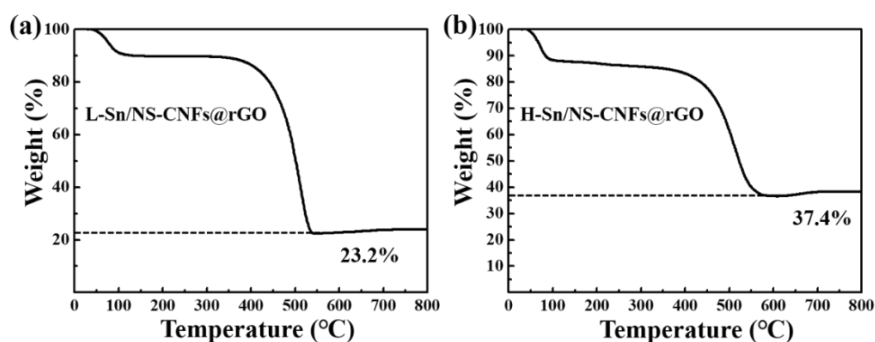


Fig. S9 a The TGA data of Sn/NS-CNFs@rGO with low Sn content (L-Sn/NS-CNFs@rGO), and **b** high Sn content (H-Sn/NS-CNFs@rGO)

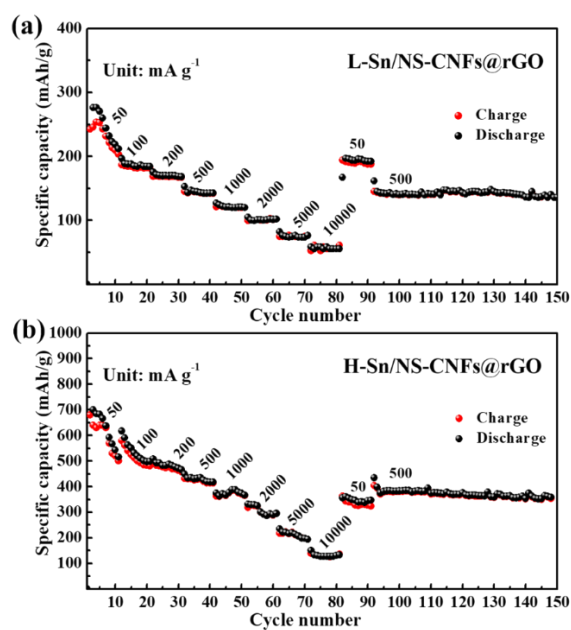


Fig. S10 a Rate capability and cycling performance of L-Sn/NS-CNFs@rGO electrode, and **b** H-Sn/NS-CNFs@rGO electrode

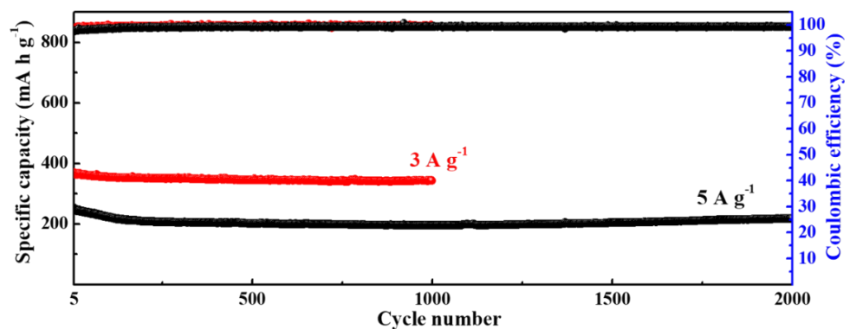


Fig. S11 Long-term cycling performance of the Sn/NS-CNFs@rGO electrodes at 3 A g^{-1} and 5 A g^{-1} , and the corresponding coulombic efficiency

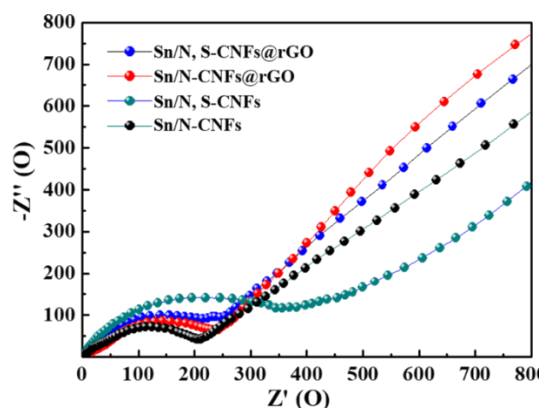


Fig. S12 EIS curves of Sn/N-CNFs, Sn/NS-CNFs, Sn/N-CNFs@rGO, and Sn/NS-CNFs@rGO electrodes after 5 cycles

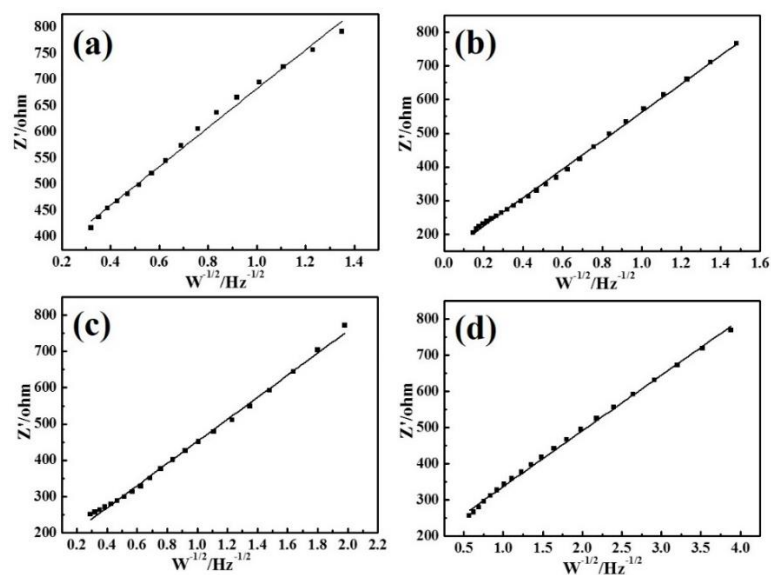


Fig. S13 Real parts of the impedance (Z') versus the reciprocal square root of angular frequency (ω) in the low frequency region of **a** Sn/N-CNFs, **b** Sn/NS-CNFs, **c** Sn/N-CNFs@rGO, and **d** Sn/NS-CNFs@rGO

The sodium ion diffusion coefficients (D_{Na}) is an important parameter of kinetics for an electrochemical reaction. It is calculated using Eq. S1:

$$D_{Na} = \frac{R^2 T^2}{2A^2 n^4 F^4 C^2 \sigma^2} \quad (S1)$$

where R is the gas constant, T is the absolute temperature, A is the surface area of the electrode, n is the number of electrons per molecule during oxidization, F is the Faraday constant, C is the concentration of sodium ion, and σ is the Warburg factor which has relationship with Z' :

$$Z' = R_D + R_C + \sigma \omega^{-\frac{1}{2}} \quad (S2)$$

Figure S13 shows the relationship between Z' and square root of frequency ($\omega^{-1/2}$) in the low-frequency region. The diffusion coefficient of sodium ion is calculated based on Eqs. S1 and S2. The calculated sodium ion diffusion coefficients of Sn/N-CNFs, Sn/NS-CNFs, Sn/N-CNFs@rGO, and Sn/NS-CNFs@rGO are 2.72×10^{-13} , 2.08×10^{-13} , 3.99×10^{-13} , 1.58×10^{-12} $\text{cm}^2 \text{s}^{-1}$, respectively. Obviously, the sodium ion diffusion ability of Sn/NS-CNFs@rGO is greatly enhanced compared with Sn/N-CNFs, Sn/NS-CNFs, Sn/N-CNFs@rGO.

Table S1 Comparisons of the sodium storage properties for previously reported 3D free-standing electrodes

Structures	Electrochemical performance	Refs.
MoS ₂ /Graphene	230 h g ⁻¹ after 20 cycles at 25 mA g ⁻¹	[S1]
Nitrogen-Doped Carbon Sheets	76 mAh g ⁻¹ after 2000 cycles at 4.5 C	[S2]
CC@CN@MoS ₂	265 mAh g ⁻¹ after 1000 cycles at 1 A g ⁻¹	[S3]
FeS@C/carbon cloth	365 mAh g ⁻¹ after 100 cycles at 0.15 C	[S4]
TiO ₂ -Sn@CNFs	413 mAh g ⁻¹ after 400 cycles at 100 mA g ⁻¹	[S5]
Hydrogen substituted graphdiyne	360 mAh g ⁻¹ after 1000 cycles at 1 A g ⁻¹	[S6]
ReS ₂ /N-CNfS	245 mAh g ⁻¹ after 800 cycles at 100 mA g ⁻¹	[S7]
Fe ₃ O ₄ @MoS ₂ -GP	388 mAh g ⁻¹ after 300 cycles at 100 mA g ⁻¹	[S8]
MoO _{3-x} grown on flexible carbon cloth	156 mAh g ⁻¹ after 200 cycles at 100 mA g ⁻¹	[S9]
SnS ₂ -RGONRP	334 mAh g ⁻¹ after 1500 cycles at 1 A g ⁻¹	[S10]
Free-standing fluorine and nitrogen co-doped graphene paper	203 mAh g ⁻¹ after 100 cycles at 50 mA g ⁻¹	[S11]
NCF@rGO-TiO ₂	214 mAh g ⁻¹ after 150 cycles at 1C	[S12]
CoSe ₂ /CNFs	430 mAh g ⁻¹ after 400 cycles at 200 mA g ⁻¹	[S13]
MoS ₂ -F	243 mAh g ⁻¹ after 1100 cycles at 1 A g ⁻¹	[S14]
SnS/C NFs	481 mAh g ⁻¹ after 100 cycles at 50 mA g ⁻¹ ; 349 mAh g ⁻¹ after 500 cycles at 200 mA g ⁻¹	[S15]
Sn/NS-CNfS@rGO	454 mAh g ⁻¹ after 200 cycles at 100 mA g ⁻¹ ; 373 mAh g ⁻¹ after 5000 cycles at 1 A g ⁻¹ ; 189 mA h g ⁻¹ at 10 A g ⁻¹	This work

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