

Supporting Information for

Lightweight and High-Performance Microwave Absorber Based on 2D WS₂–RGO Heterostructures

Deqing Zhang^{1, 5, †}, Tingting Liu^{1, †}, Junye Cheng^{2, *}, Qi Cao³, Guangping Zheng⁴,
Shuang Liang¹, Hao Wang^{2, *}, Mao-Sheng Cao^{5, *}

¹School of Materials Science and Engineering, Qiqihar University, Qiqihar 161006, People's Republic of China

²Guangdong Provincial Key Laboratory of Micro/Nano Optomechatronics Engineering, College of Mechatronics and Control Engineering, Shenzhen University, Shenzhen 518060, People's Republic of China

³Department of Chemistry, School of Science, and Department of Mechanical Engineering, School of Engineering, The University of Tokyo, Tokyo 113-8656, Japan

⁴Department of Mechanical Engineering, Hong Kong Polytechnic University, Hung Hom, Kowloon, Hong Kong, People's Republic of China

⁵School of Materials Science and Engineering, Beijing Institute of Technology, Beijing 100081, People's Republic of China

†These authors contribute equally to this work.

*Corresponding authors. E-mail: whao@szu.edu.cn (Hao Wang); jy Cheng4-c@my.cityu.edu.hk (Junye Cheng); caomaosheng@bit.edu.cn (Mao-Sheng Cao)

Supplementary Figures and Tables

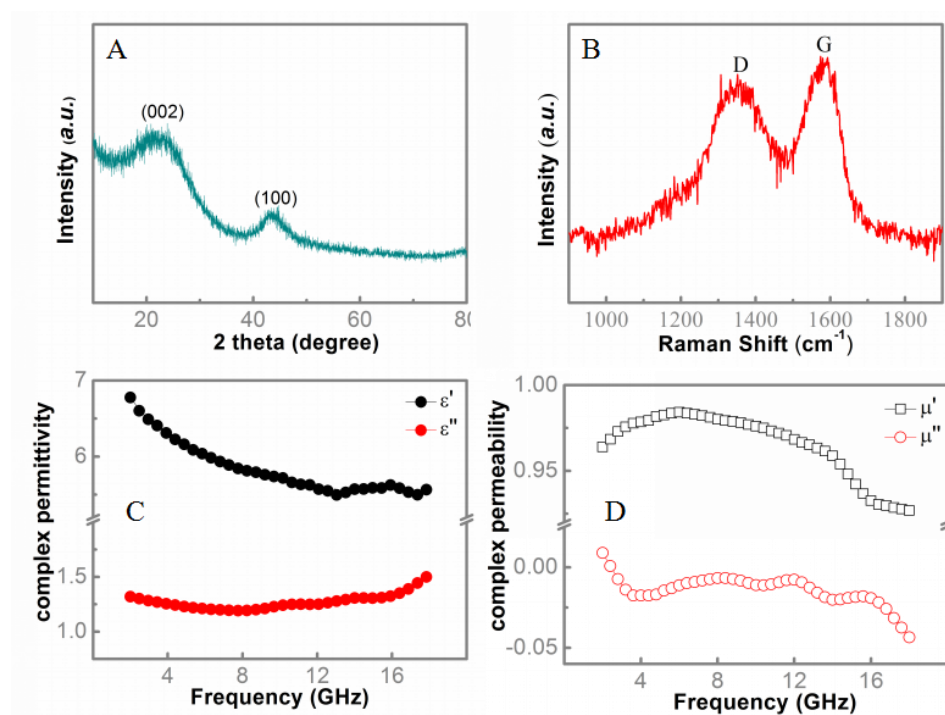


Fig. S1 A XRD patterns, B Raman spectrum, C Complex permittivity profiles, and D Complex permeability profiles of rGO

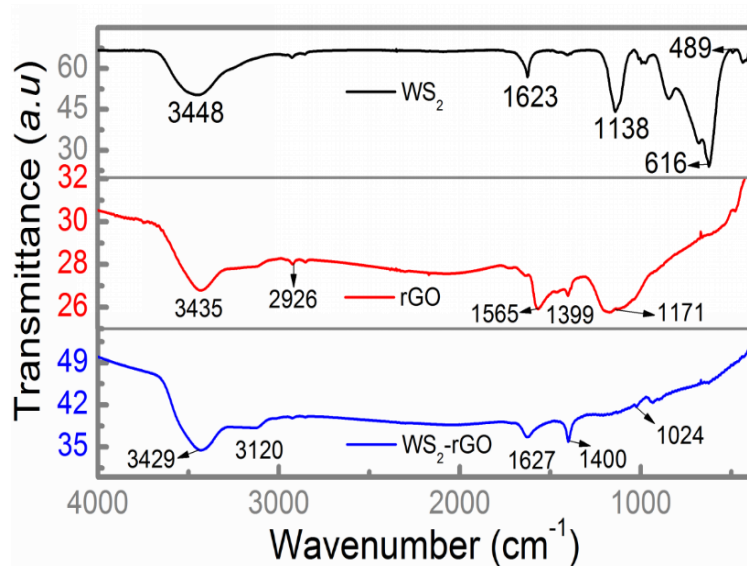


Fig. S2 FI-IR spectra of WS₂, rGO, and WS₂-rGO

The results confirm that, in the hybrid material, GO was well reduced to rGO. For rGO, the stretching vibration peak of -OH appears at 3435 cm⁻¹, the C-H stretching

vibration peak appears at 2926 cm^{-1} , the C-C stretching vibration absorption peak appears at 1565 cm^{-1} , and the O-H stretching vibration peak appears at 1399 cm^{-1} , the C-O stretching vibration peak appears at 1171 cm^{-1} . After the complexation, the peaks of the WS_2 -rGO heterostructure nanosheets are shifted, for example, the stretching vibration peak of -OH appears at 3429 cm^{-1} , the stretching vibration peak of C-H appears at 3120 cm^{-1} , the O-H stretching vibration peak appears at 1400 cm^{-1} , and the C-O stretching vibration peak appears at 1024 cm^{-1} . The results suggest that there are interactions among functional groups of rGO and WS_2 crystals and WS_2 -rGO heterostructure is well formed. At the same time, there are several characteristic peaks of tungsten disulfide for the WS_2 -rGO heterostructure nanosheets. The peak close to at 3448 cm^{-1} is ascribed to the W-S vibration; The peak close to 1623 cm^{-1} is the stretching vibration of -OH resulting from the combination of water.

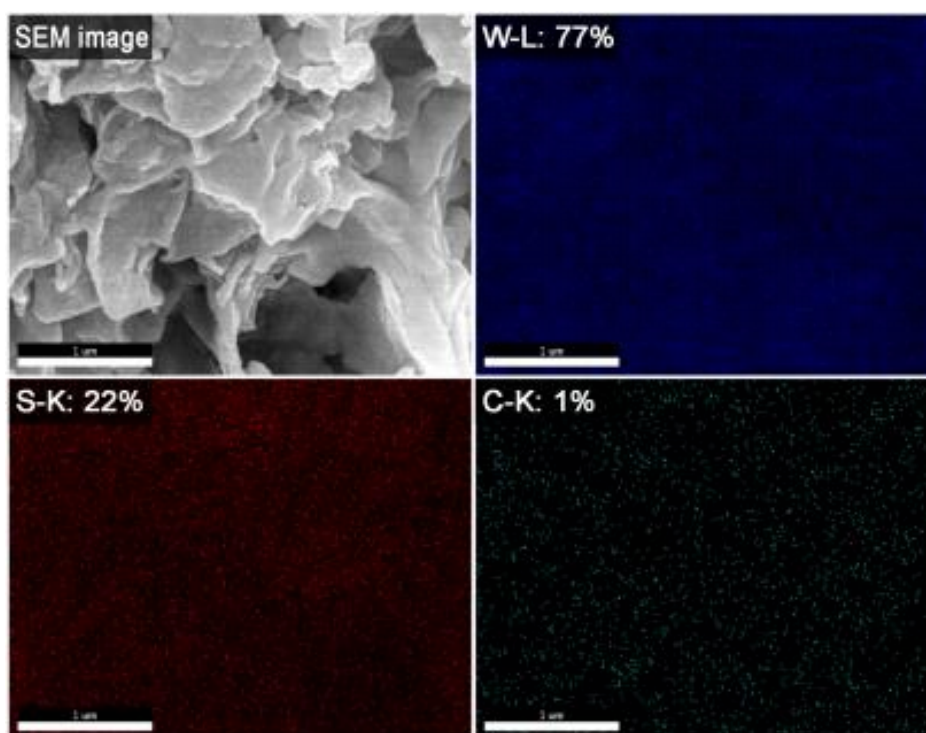


Fig. S3 SEM-EDS elemental mapping profiles of the as-synthesized WS_2 -rGO heterostructure nanosheet powders

The results confirm that the elements W and S (in WS_2), and also C (in rGO) distribute uniformly in the as-synthesized WS_2 -rGO heterostructure nanosheet powders.

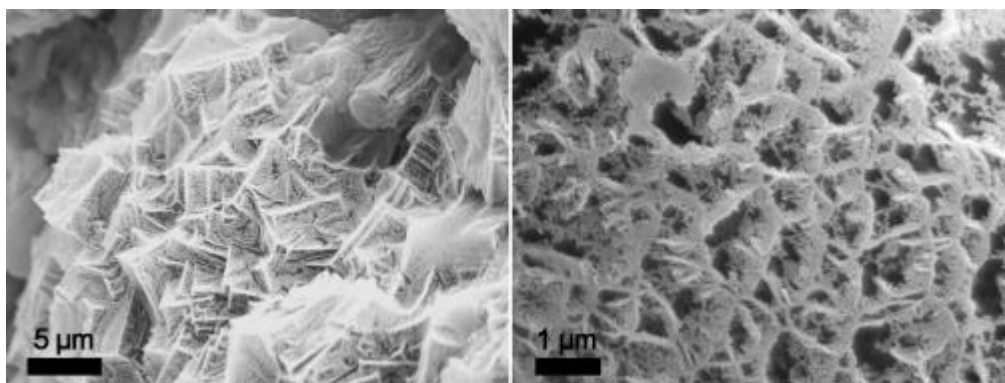


Fig. S4 Large-scale SEM images of the as-synthesized WS₂-rGO heterostructure nanosheet powders at different magnifications

The images reveal the 3D interconnected network structure formed from the as-synthesized WS₂-rGO heterostructure nanosheets at micrometer scales, which might be favorable for enhancing the multiple scattering-induced microwave dissipation, as illustrated in Fig. 8d in the main text.

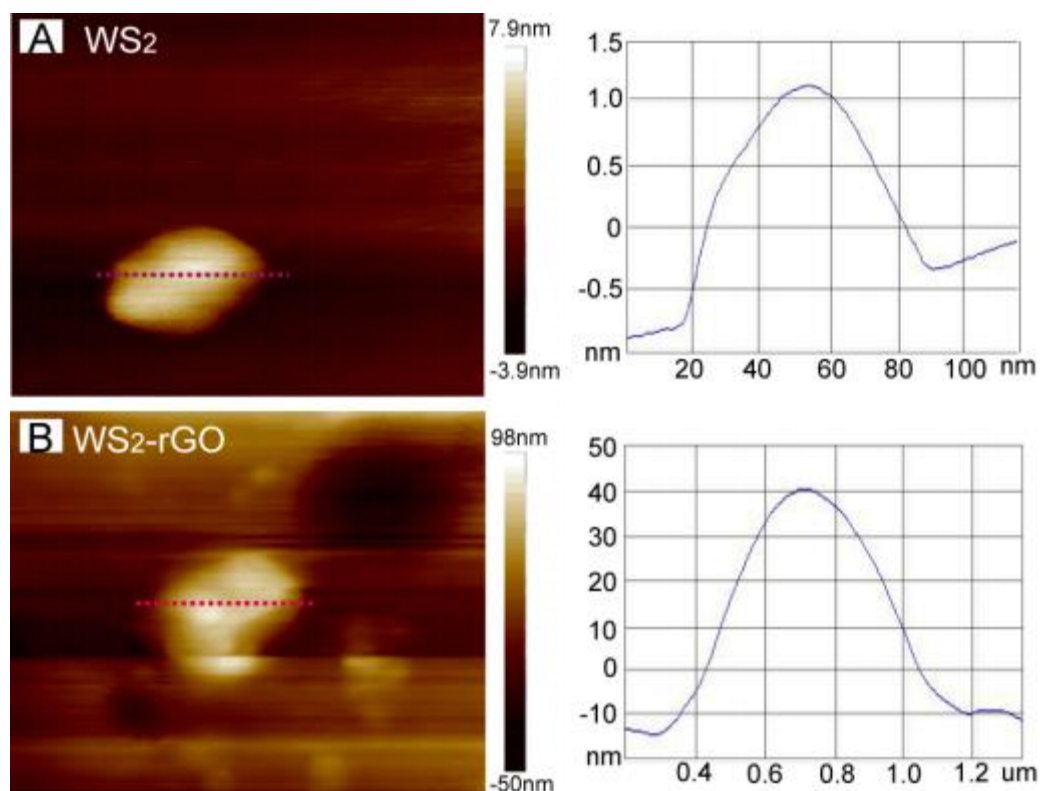


Fig. S5 AFM images of **A** WS₂, **B** WS₂-rGO

Compared to that of pristine tungsten disulfide nanosheets, the thickness of WS₂-rGO is increased, while it is only about 50 nm. The results prove that the WS₂-rGO heterostructure nanosheet prepared is a particularly thin material.



Fig. S6 Optical image of a piece of sample made from the as-synthesized WS₂-rGO heterostructure nanosheets, which is put on a *Setaria viridis*. The inset displays the image of the as-synthesized WS₂-rGO powders

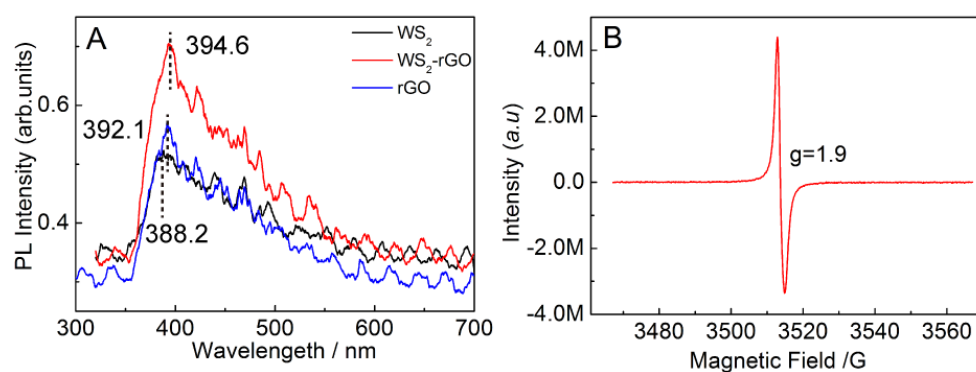


Fig. S7 **A** Photoluminescence spectra of WS₂, rGO, and the WS₂-rGO heterostructure nanosheets; $\lambda_{exc} = 280$ nm; **B** EPR signals of WS₂-rGO heterostructure nanosheets at room temperature

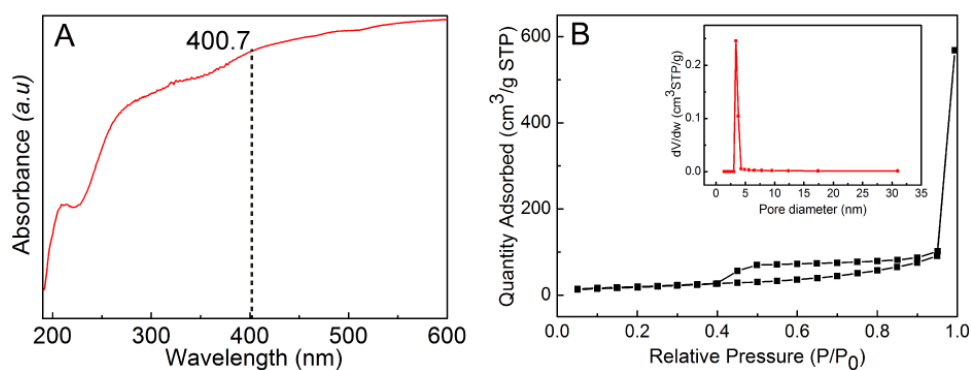


Fig. S8 **A** UV-Vis spectra of the WS₂-rGO; **B** Nitrogen adsorption-desorption isotherms and the corresponding pore size distribution curves for the WS₂-rGO

Table S1 The measured conductivity of WS₂, rGO, and WS₂-rGO

| Sample | WS ₂ | rGO | WS ₂ -rGO |
|--------------------------------|-----------------|-------|----------------------|
| σ (s cm ⁻¹) | 3.33 | 10.86 | 10 |

The results demonstrate that WS₂ and rGO have different electrical conductivity, and the electrical conductivity of the WS₂-rGO heterostructure nanosheets is higher than that of the WS₂ nanosheets, which also prove the successful combination of WS₂ and rGO.

Table S2 Microwave Absorption Performance of MA materials

| Samples | Matrix | Minimum RL value (dB) | Thickn ess (mm) | Loading ratio (wt%) | Frequency range (GHz) | Effective bandwidth (GHz) | Refs. |
|--------------------------------------|--------|-----------------------|-----------------|---------------------|-----------------------|---------------------------|-----------|
| WS ₂ -rGO | wax | -41.5 | 1.5 | 40 | 4.38-18.0 | 13.62 | This work |
| RGO | PVDF | -25.6 | 4.0 | 3 | 8.48-12.8 | 4.32 | [S1] |
| RGO/CoFe ₂ O ₄ | wax | -38.5 | 2.0 | 50 | 9.2-15.0 | 5.8 | [S2] |
| MoS ₂ -NS | wax | -38.42 | 2.4 | 60 | 9.6-13.76 | 4.16 | [S3] |
| Co ₃ O ₄ /RGO | wax | -31.7 | 2.5 | 20 | 5.50-16.00 | 10.50 | [S4] |
| MoS ₂ /RGO | wax | -31.57 | 2.5 | 10 | 6.5-18.0 | 11.5 | [S5] |
| MoS ₂ /CNT | wax | -46 | 2.9 | 50 | 3.4-13.9 | 10.5 | [S6] |

Supplementary References

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