

Supporting Information for

**MoS₂ Nanosheets Arrays Rooted on Hollow rGO Spheres as
Bifunctional Hydrogen Evolution Catalyst and Supercapacitor
Electrode**

Shizheng Zheng¹, Lijun Zheng¹, Zhengyou Zhu¹, Jian Chen¹, Jianli Kang², Zhulin Huang³, Dachi Yang^{1,*}

¹Key Laboratory of Photoelectronic Thin Film Devices and Technology of Tianjin; Department of Electronics, College of Electronic Information and Optical Engineering, Nankai University, Tianjin 300350, People's Republic of China

²School of Material Science and Engineering, Tianjin Polytechnic University, Tianjin 300387, People's Republic of China

³Key Laboratory of Materials Physics, and Anhui Key Laboratory of Nanomaterials and Nanotechnology, Institute of Solid State Physics, Chinese Academy of Sciences, Hefei 230031, People's Republic of China

*Corresponding author. E-mail: yangdachi@nankai.edu.cn (Dachi Yang)

Supplementary Figures and Tables

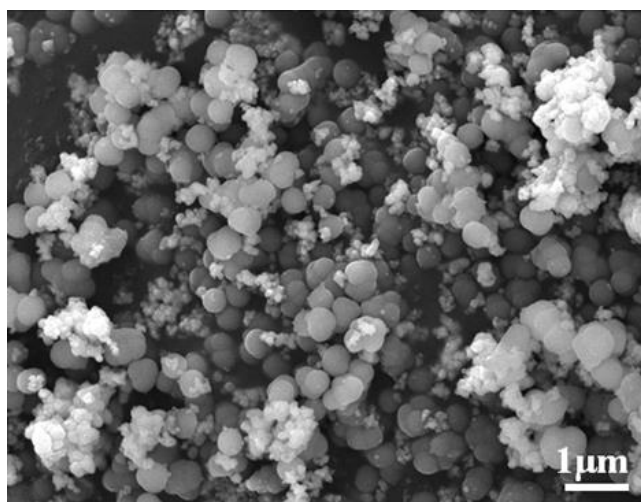


Fig. S1 SEM image of SiO₂/MoS₂ composites

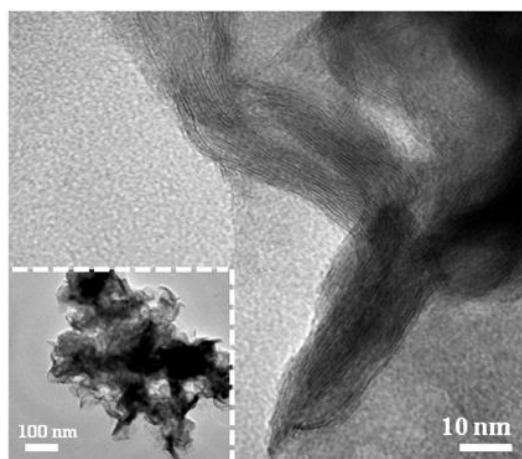


Fig. S2 TEM images of pristine MoS₂. The inset is the low magnification TEM image

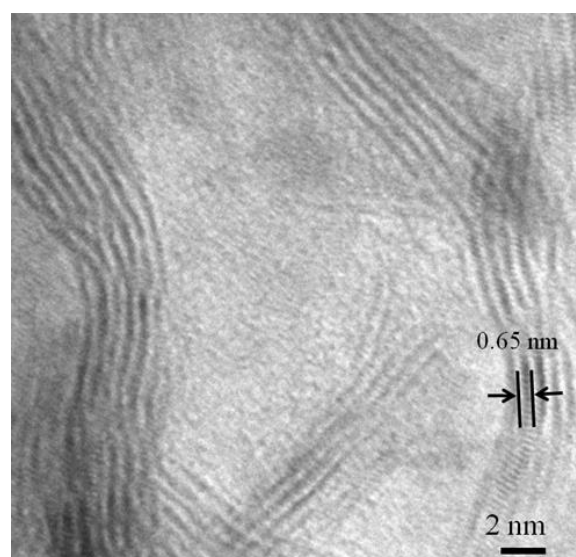


Fig. S3 HRTEM of h-rGO@MoS₂ with expanded (002) interlayer spacing

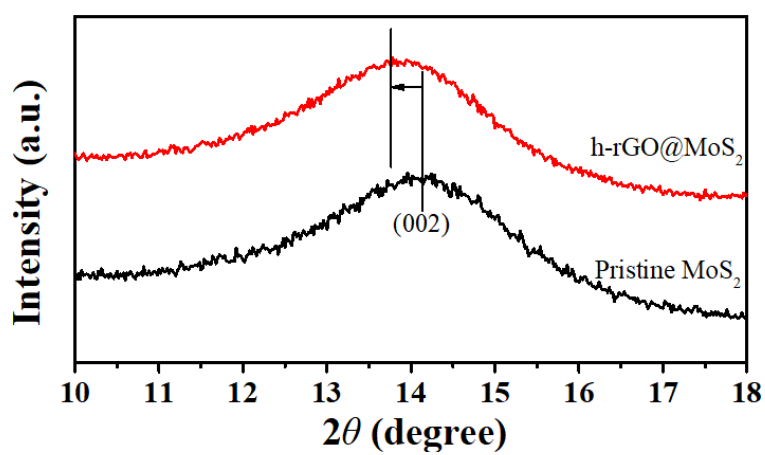


Fig. S4 XRD patterns of the (002) plane peaks of pristine MoS₂ and h-rGO@MoS₂

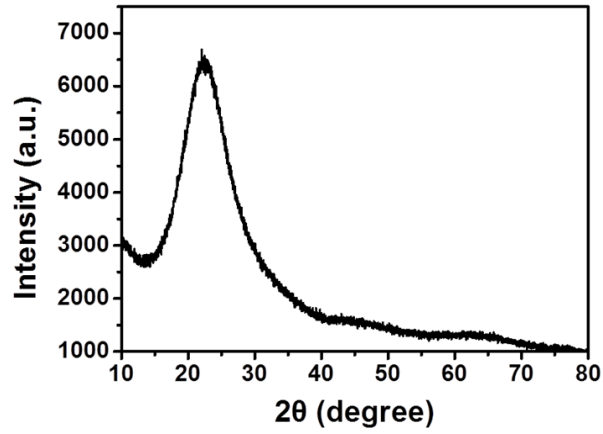


Fig. S5 XRD patterns of amorphous SiO₂ template

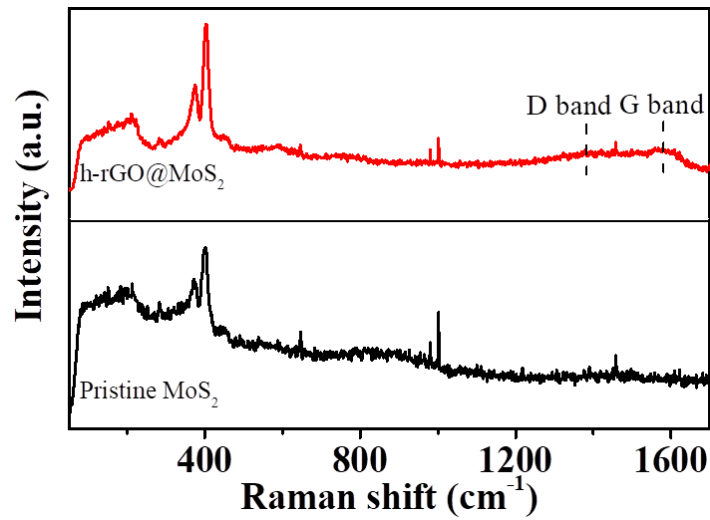


Fig. S6 Raman spectra of pristine MoS₂ and h-rGO@MoS₂

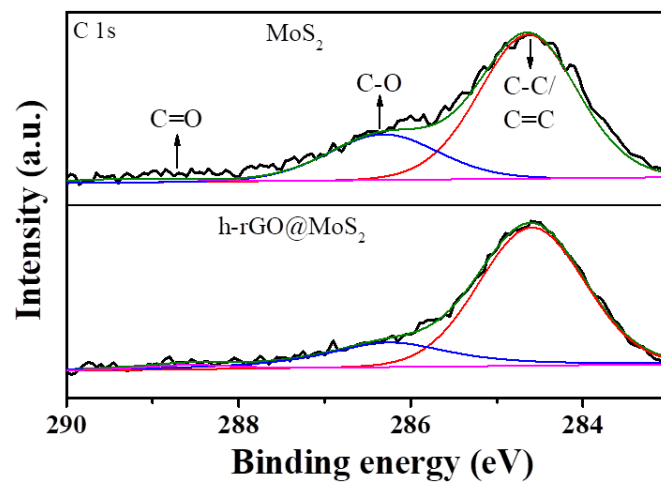


Fig. S7 The high resolution C 1s XPS spectra of pristine MoS₂ and h-rGO@MoS₂

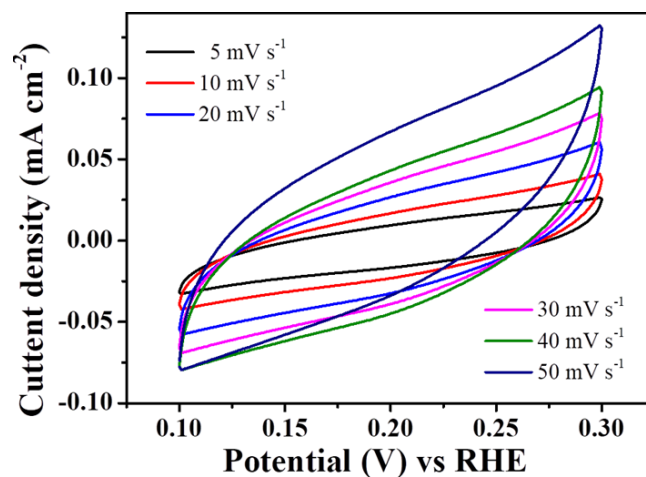


Fig. S8 Electrochemical double-layer capacitances of pristine MoS₂ ranged from 0.1 to 0.3 V at various scan rates (5, 10, 20, 30, 40, and 50 mV s⁻¹)

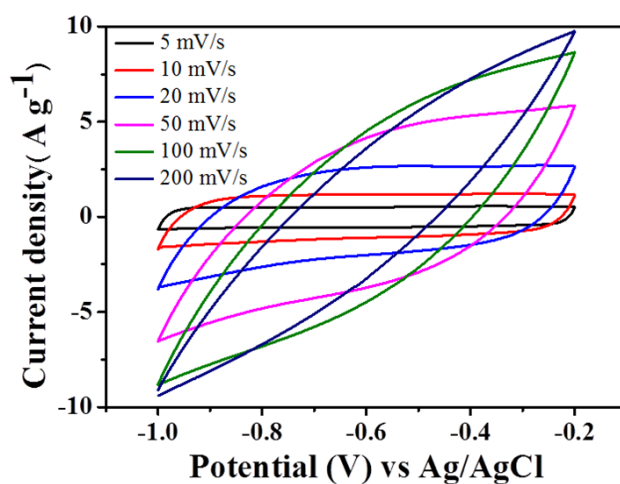


Fig. S9 CV curves of pristine MoS₂ at different sweep rates

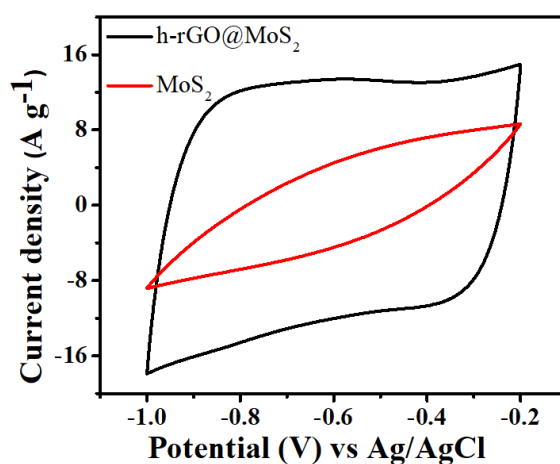


Fig. S10 CV curves contrast between h-rGO@MoS₂ and pristine MoS₂ at scan rate of 100 mV s⁻¹

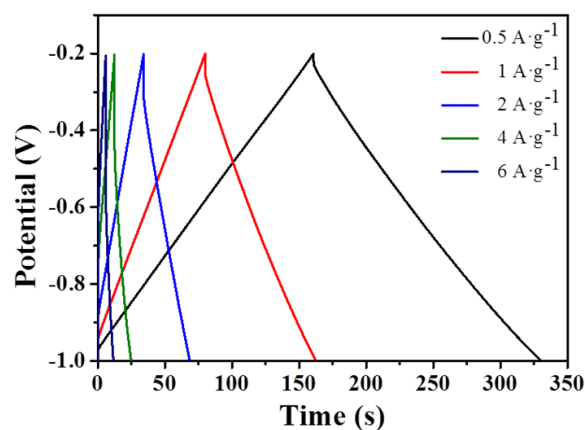


Fig. S11 GCD curves of pristine MoS₂ at various current densities

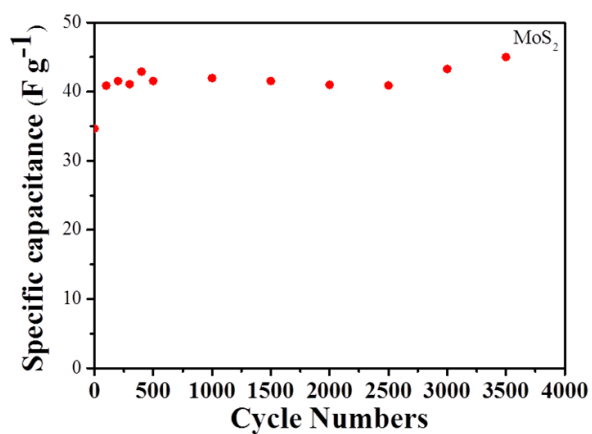


Fig. S12 3500 loops cycle stability of MoS₂ at scan rate of 100 mV s⁻¹

Table S1 MoS₂/graphene electrocatalysts and their performance comparison in onset overpotential and Tafel slope

Electrocatalysts	Onset overpotential (mV)	Tafel slope (mV/decade)	Ref.
(0D/3D) MoS ₂ on porous graphene	150	56	1
MoS ₂ /GO hybrid	150	91.9	2
MoS ₂ /N-doped graphene nanosheet Aerogels	236	230	3
Microwave-assisted synthesized MoS ₂ / graphene	170	80	4
MoS ₂ nanoflower-decorated rGO paper	190	90	5
3D MoS ₂ /rGO hierarchical frameworks	121	46.5	6
MoS ₂ vertically rooted on hollow rGO	105	105	This work

Table S2 MoS₂/graphene electrode materials and their capacitive performance in various electrolytes

Electrode material	Specific capacitance (F g ⁻¹)	Current density (A g ⁻¹)	Electrolyte	Ref.
hollow MoS ₂	142	1.18	1 M KCl	7
Layered MoS ₂ -graphene	243	1	1M Na ₂ SO ₄	8
MoS ₂ /RGO	216	1	1 M H ₂ SO ₄	9
MoS ₂ /N-doped graphene	245	0.25	1M KOH	10
Assembled MoS ₂ microflower	167.7	1	3M KOH	11
h-rGO@MoS ₂	238	0.5	1 M Na ₂ SO ₄	This work

Reference

- [1] Y. Liu, Y. Zhu, X. Fan, S. Wang, Y. Li, F. Zhang, G. Zhang, W. Peng, (0D/3D) MoS₂ on porous graphene as catalysts for enhanced electrochemical hydrogen evolution. *Carbon* **121**, 163-169 (2017). <https://doi.org/10.1016/j.carbon.2017.05.092>
- [2] E.G. Firmiano, M.A. Cordeiro, A.C. Rabelo, C.J. Dalmaschio, A.N. Pinheiro, E.C. Pereira, E.R. Leite, Graphene oxide as a highly selective substrate to synthesize a layered MoS₂ hybrid electrocatalyst. *Chem. Commun.* **48**(62), 7687-7689 (2012). <https://doi.org/10.1039/c2cc33397j>
- [3] Y. Hou, B. Zhang, Z. Wen, S. Cui, X. Guo, Z. He, J. Chen, A 3D hybrid of layered MoS₂/nitrogen-doped graphene nanosheet aerogels: An effective catalyst for hydrogen evolution in microbial electrolysis cells. *J. Mater. Chem. A* **2**(34), 13795-13800 (2014). <https://doi.org/10.1039/c4ta02254h>
- [4] J. Cao, X. Zhang, Y. Zhang, J. Zhou, Y. Chen, X. Liu, Free MoS₂ nanoflowers grown on graphene by microwave-assisted synthesis as highly efficient non-noble-metal electrocatalysts for the hydrogen evolution reaction. *PLoS One* **11**(8), e0161374 (2016). <https://doi.org/10.1371/journal.pone.0161374>
- [5] C.B. Ma, X. Qi, B. Chen, S. Bao, Z. Yin et al., MoS₂ nanoflower-decorated reduced graphene oxide paper for high-performance hydrogen evolution reaction. *Nanoscale* **6**(11), 5624-5629 (2014). <https://doi.org/10.1039/c3nr04975b>
- [6] W. Zhou, K. Zhou, D. Hou, X. Liu, G. Li, Y. Sang, H. Liu, L. Li, S. Chen, Three-dimensional hierarchical frameworks based on MoS₂ nanosheets self-assembled on graphene oxide for efficient electrocatalytic hydrogen evolution. *ACS. Appl. Mater. Interfaces* **6**(23), 21534-21540 (2014). <https://doi.org/10.1021/am506545g>

- [7] L. Wang, Y. Ma, M. Yang, Y. Qi, Hierarchical hollow MoS₂ nanospheres with enhanced electrochemical properties used as an electrode in supercapacitor. *Electrochim. Acta* **186**, 391-396 (2015).
<https://doi.org/10.1016/j.electacta.2015.10.130>
- [8] K.-J. Huang, L. Wang, Y.-J. Liu, Y.-M. Liu, H.-B. Wang, T. Gan, L.-L. Wang, Layered MoS₂-graphene composites for supercapacitor applications with enhanced capacitive performance. *Int. J. Hydrog. Energy* **38**(32), 14027-14034 (2013). <https://doi.org/10.1016/j.ijhydene.2013.08.112>
- [9] X. Li, C. Zhang, S. Xin, Z. Yang, Y. Li, D. Zhang, P. Yao, Facile synthesis of MoS₂/reduced graphene oxide@polyaniline for high-performance supercapacitors. *ACS Appl. Mater. Interfaces* **8**(33), 21373-21380 (2016).
<https://doi.org/10.1021/acsami.6b06762>
- [10] B. Xie, Y. Chen, M. Yu, T. Sun, L. Lu, T. Xie, Y. Zhang, Y. Wu, Hydrothermal synthesis of layered molybdenum sulfide/N-doped graphene hybrid with enhanced supercapacitor performance. *Carbon* **99**, 35-42 (2016).
<https://doi.org/10.1016/j.carbon.2015.11.077>
- [11] L.-M. Xu, L. Ma, X.-Y. Xu, X.-P. Zhou, L.-L. Zhang, W.-X. Chen, Molybdenum disulfide microflowers assembled by few-layered nanosheets and their electrochemical performance for supercapacitor. *Mater. Lett.* **173**, 84-87 (2016).
<https://doi.org/10.1016/j.matlet.2016.03.027>