

Supporting Information for

Dye-enhanced Self-Electrophoretic Propulsion of Light-Driven TiO₂-Au Janus Micromotors

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Videos

Video S1 The motion of TiO₂-Au Janus micromotors in 10⁻⁵ g L⁻¹ Methyl Blue (MB), 10⁻⁴ g L⁻¹ Cresol Red (CR), 10⁻⁴ g L⁻¹ Methyl Orange (MO) aqueous solutions and pure water, respectively

Video S2 The moving direction of TiO₂-Au Janus micromotors under UV in MB aqueous solution

Video S3 The motion of TiO₂-Au Janus micromotors in different concentration of MB aqueous solution

Video S4 The motion of TiO₂-Au Janus micromotors in different concentration of CR aqueous solution

Video S5 The motion of TiO₂-Au Janus micromotors in different concentration of MO aqueous solution

Video S6 Directional control of Au-Ni- TiO₂ Janus micromotors in MB aqueous solution

Video S7 The motion of Au-Ni- TiO₂ Janus micromotors in MB aqueous solution

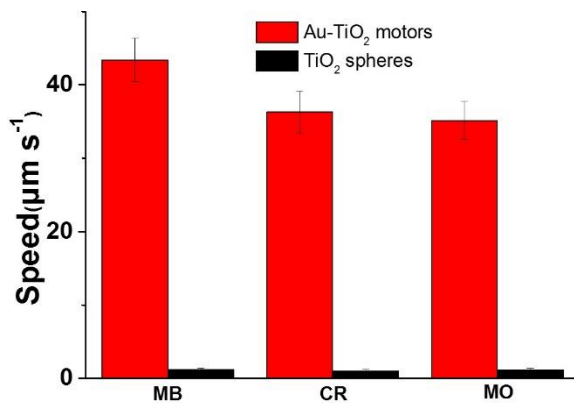


Fig. S1 The motion speeds of Au-TiO₂ motors (red) and pure TiO₂ spheres (black) in 10⁻⁵ g L⁻¹ Methyl Blue (MB), 10⁻⁴ g L⁻¹ Cresol Red (CR), 10⁻⁴ g L⁻¹ Methyl Orange (MO), respectively.

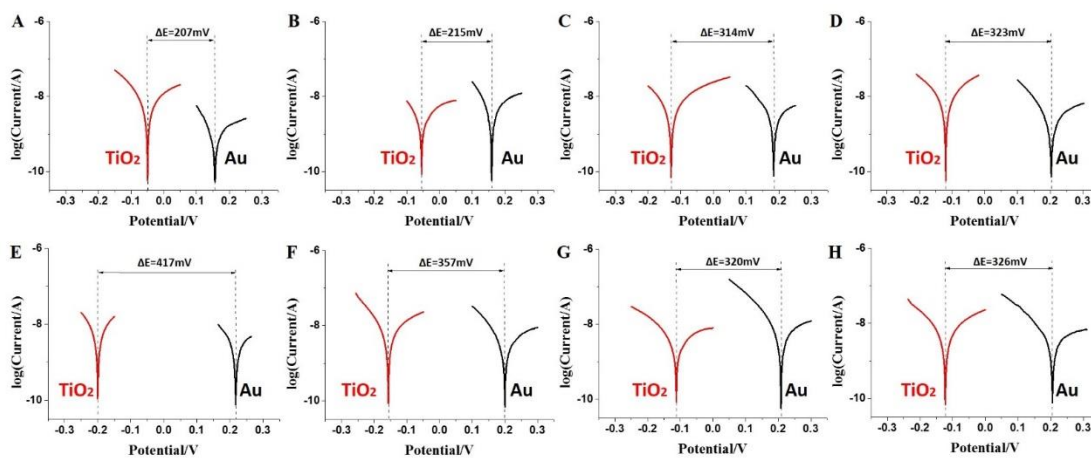


Fig. S2 Tafel plots of Au (black lines) and TiO₂ (red lines) under UV light in (A)-(H) 10⁻¹~10⁻⁸ g L⁻¹ MB aqueous solutions, respectively

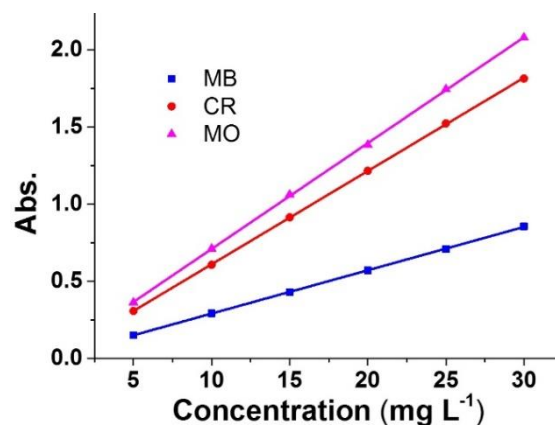


Fig. S3 Absorption–concentration standard curve of MB, CR and MO in aqueous solutions

The absorption intensity A of MB, CR, and MO in aqueous solutions were proportional to their concentrations C . The working curves were obtained by linearly fitting the data of A and C . The obtained linear equations are below:

$$A (MB) = 0.0281C + 0.0095 \quad (r^2 = 0.9999) \quad (1)$$

$$A (CR) = 0.0605C + 0.0049 \quad (r^2 = 0.9999) \quad (2)$$

$$A (MO) = 0.0687C + 0.0227 \quad (r^2 = 0.9998) \quad (3)$$