

Supporting Information for

## Engineering Mesoporous Structure in Amorphous Carbon Boosts Potassium Storage with High Initial Coulombic Efficiency

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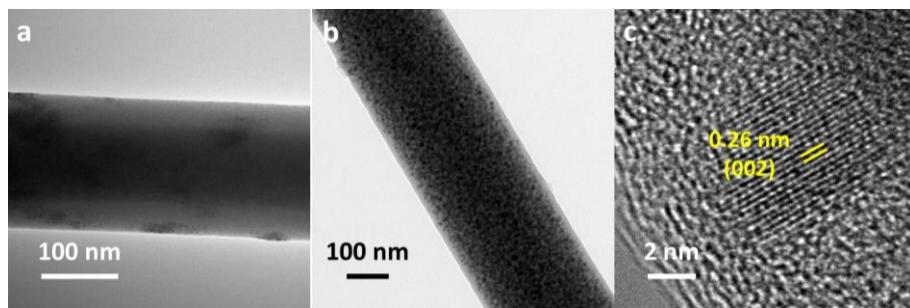
### S1 Diffusion Coefficients Calculation

The K<sup>+</sup> diffusion coefficients were calculated by using the equation based on Fick's second law [S1]:

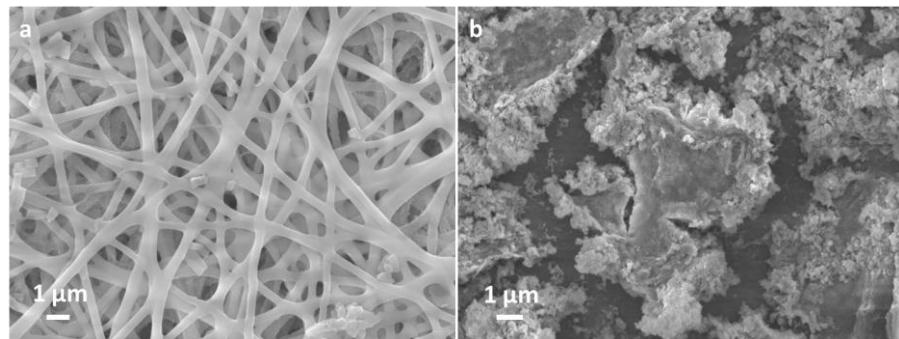
$$D = \frac{4}{\pi\tau} \left( \frac{n_B V_M}{S} \right)^2 \left( \frac{\Delta E_S}{\Delta E_\tau} \right)^2$$

where  $D$  is the K<sup>+</sup> diffusion coefficient,  $\tau$  is the current pulse time (s),  $n_B$  is the amount of substance,  $V_M$  is the molar volume of the active material, and  $S$  is the area of the electrodes.  $\Delta E_S$  is the potential difference of two adjacent steady-states and  $\Delta E_\tau$  is the potential change due to the pulse current.

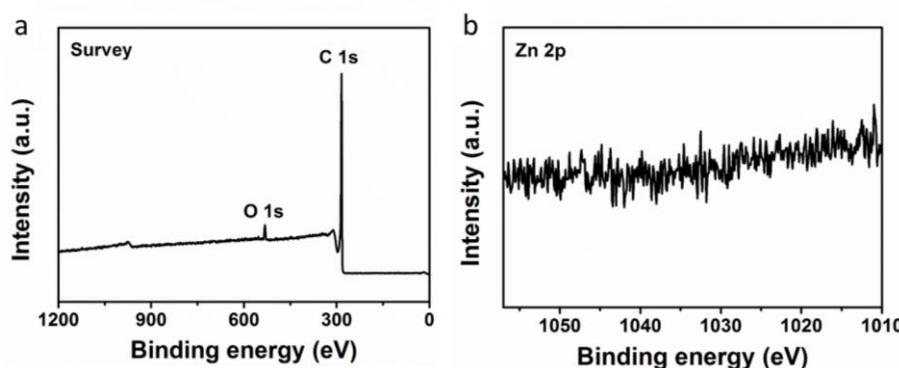
### S2 Supplementary Figures and Tables



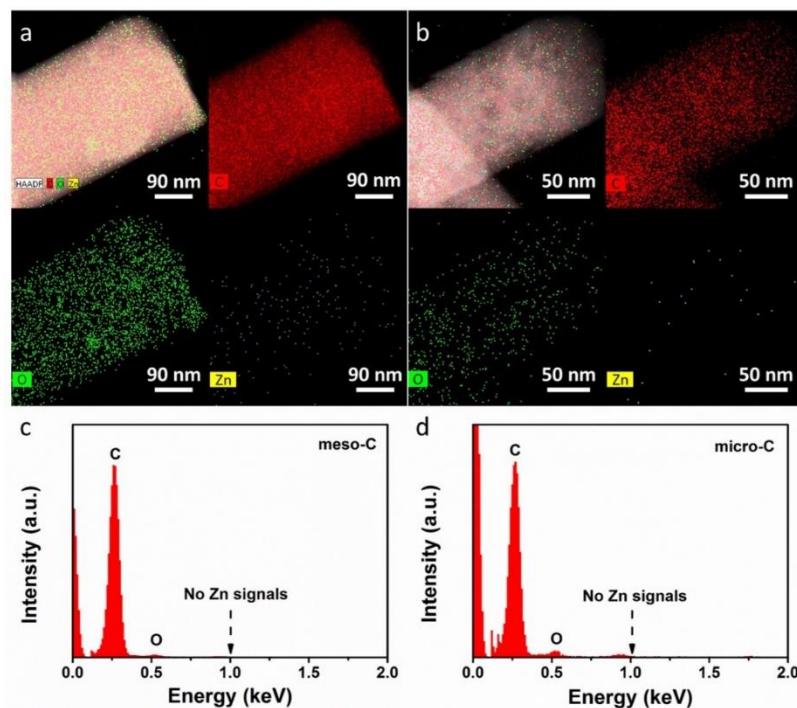
**Fig. S1** (a) TEM image of preheated Zn(Ac)<sub>2</sub>/PVA. (b, c) TEM and HRTEM images of ZnO@C



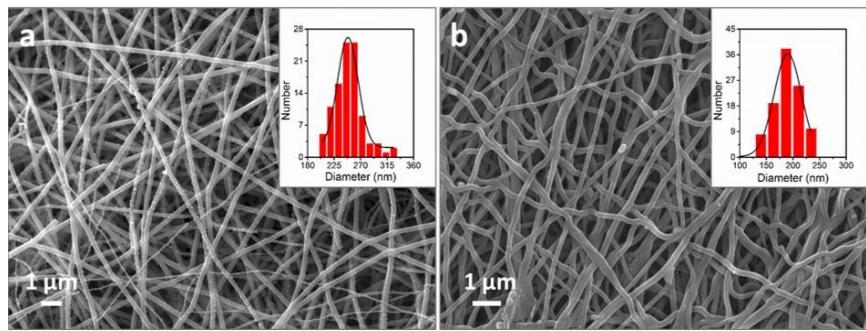
**Fig. S2** SEM images of (a) preheated Zn(Ac)<sub>2</sub>/PVA and (b) untreated Zn(Ac)<sub>2</sub>/PVA after immersing in deionized water for 24 h



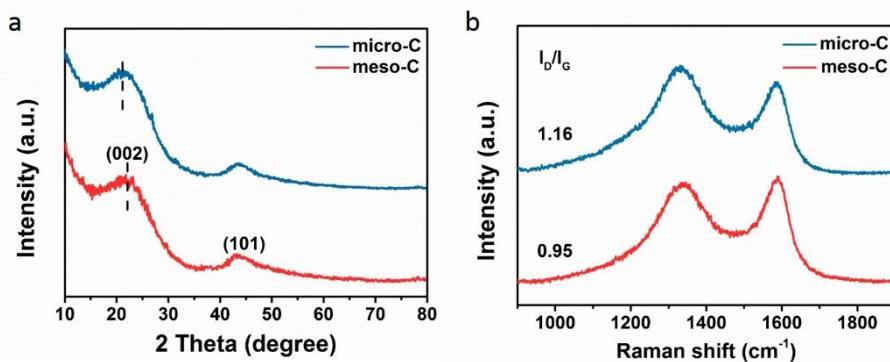
**Fig. S3** (a) XPS survey spectrum and (b) high-resolution Zn 2p XPS spectrum for meso-C sample



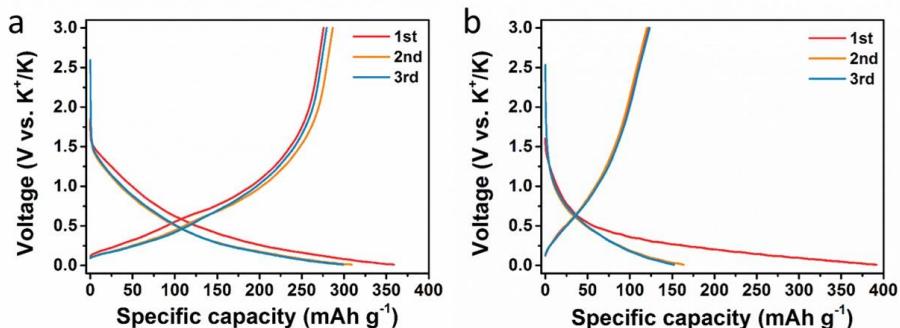
**Fig. S4** STEM-EDX mappings of (a) meso-C and (b) micro-C, and corresponding EDX spectra of (c) meso-C and (d) micro-C



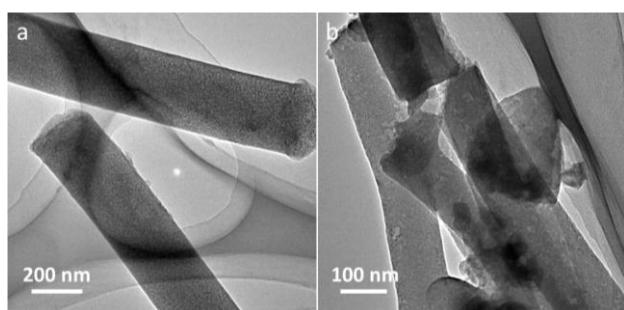
**Fig. S5** SEM images and the corresponding diameter distributions (insets) of (a) meso-C and (b) micro-C nanowires



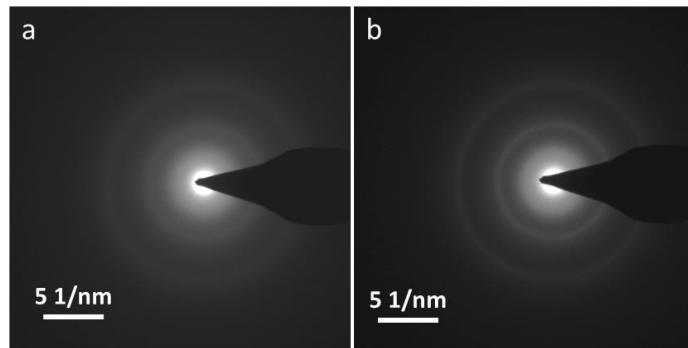
**Fig. S6** (a) XRD patterns and (b) Raman spectra of meso-C and micro-C



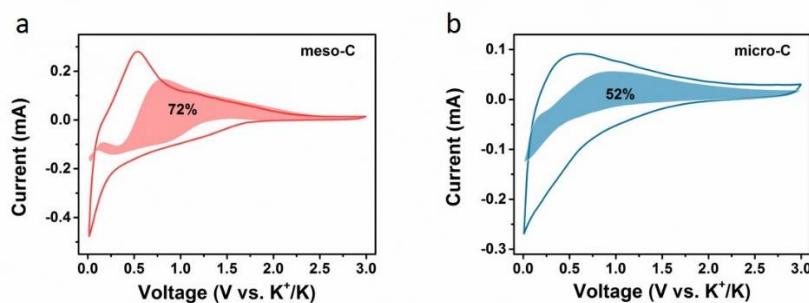
**Fig. S7** Galvanostatic charging/discharging profiles of (a) meso-C and (b) micro-C tested at 0.1 A g<sup>-1</sup>



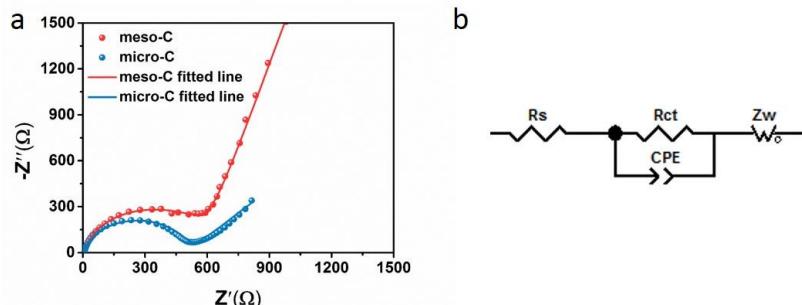
**Fig. S8** TEM images of (a) meso-C and (b) micro-C after long-term cycling at 1 A g<sup>-1</sup>



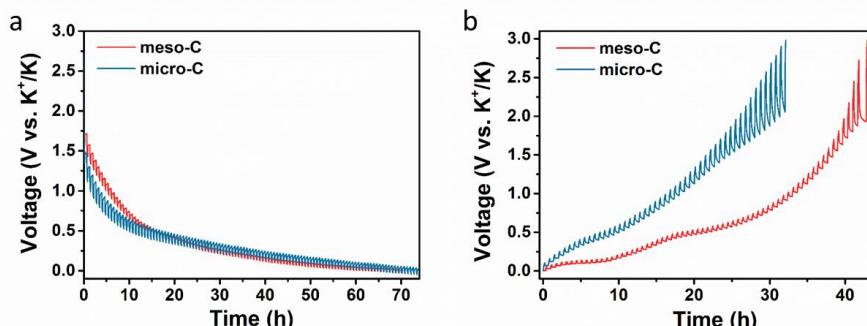
**Fig. S9** Ex situ SAED patterns of meso-C (a) after discharging to 0.01 V and (b) charging to 3 V



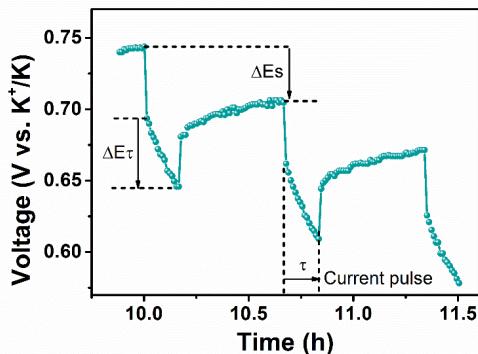
**Fig. S10** Contribution of the surface process at  $0.4 \text{ mV s}^{-1}$  for (a) meso-C and (b) micro-C samples



**Fig. S11** (a) Electrochemical impedance spectroscopy curves for fresh cells at OCV (V vs.  $\text{K}^+/\text{K}$ ). (b) The equivalent circuit used to fit the original data



**Fig. S12** (a, b) GITT curves of the discharging and charging processes, respectively



**Fig. S13** Regional GITT potential response with time

**Table S1** STEM-EDX mapping elemental atomic components of meso-C and micro-C samples

Samples	Element content (at.%)		
	C	O	Zn
meso-C	99.71	0.29	0.00
micro-C	99.02	0.89	0.00

**Table S2** K-storage performance comparison of meso-C anode in this work with other reported carbonaceous anodes

Samples	Current density (mA g <sup>-1</sup> )	Capacity after cycling (mAh g <sup>-1</sup> )	Cycle number	ICE (%)	References
Hollow interconnected neuron-like carbon	140	250	150	72.1	[S2]
Hard-soft composite carbon	279	200	200	67	[S3]
S doped RGO-600	50	361	50	65	[S4]
Graphite	93	255	2000	64	[S5]
Ordered mesoporous carbon	50	257.4	100	63.6	[S6]
Sulfur-grafted hollow carbon spheres	200	300	250	51.4	[S7]
NCNFs	25	248	100	49	[S8]
SMCF@CNTs	279	193	300	48	[S9]
N/O dual-doped carbon network	50	303	50	47.1	[S10]
Graphitic carbon nanocage	55.8	212	100	40	[S11]
CNF-O	279	160	300	35	[S12]
NCNTs	50	254.7	300	24.4	[S13]
Porous carbon nanofiber	200	211	1200	24.1	[S14]
N-doped hollow carbon	200	225.4	1000	16	[S15]
Graphitic nanocarbon	200	189	200	15.5	[S16]
meso-C nanowires	100	231	250	76.7	This work

## Supplementary References

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