

Supporting information

Surface-modified graphene oxide/lead sulfide hybrid film-forming ink for high efficiency bulk nano- heterojunction colloidal quantum dot solar cells

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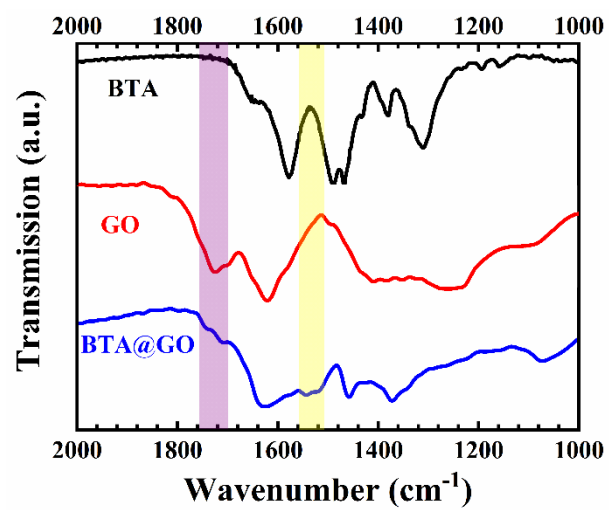


Fig. S1 FT-IR spectra of BTA, GO and GO@BTA.

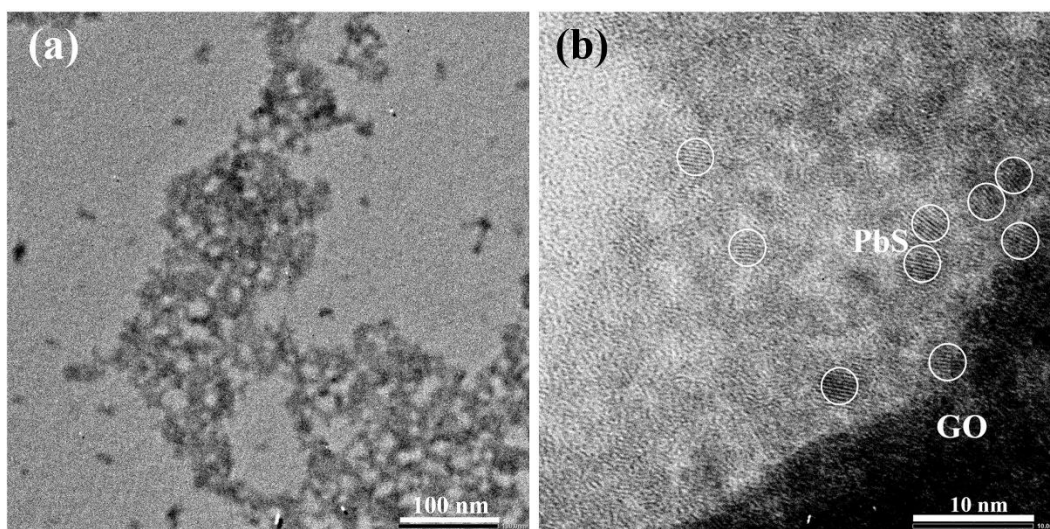


Fig. S2 (a) TEM image of PbS-PbX₂ CQDs and (b) HR-TEM image of PbS CQDs coupled with BTA@GO.

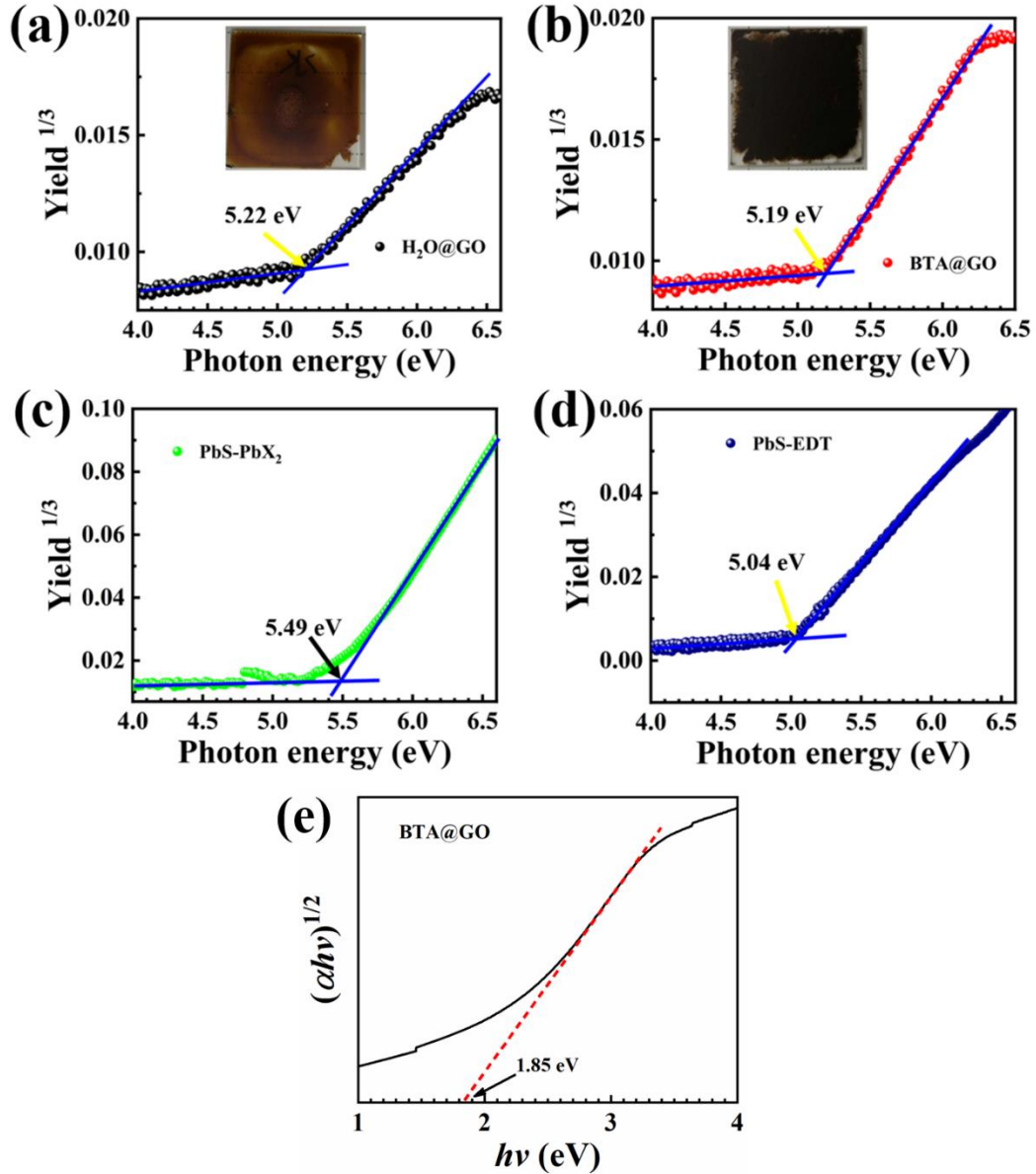


Fig. S3 PYS spectra of (a) H₂O@GO, (b) BTA@GO, (c) PbS-PbX₂ CQDs film, (d) PbS-EDT CQDs film, and (e) Tauc plot of $(\alpha h\nu)^{1/2}$ against the photon energy ($h\nu$) for BTA@GO, respectively. The inset pictures in (a) and (b) are the photographs of H₂O@GO and BTA@GO films, respectively.

GO is a heavily oxygenated monolayer material consisting of a variety of functional groups which has both sp² and sp³ clusters and sp² clusters mostly embedded inside sp³ cluster, thus the bandgap of GO like disorder structure. The optical bandgap estimated from Tauc plot considering an indirect bandgap is 1.85 eV for BTA@GO as shown in Fig. S3e.[1-2]

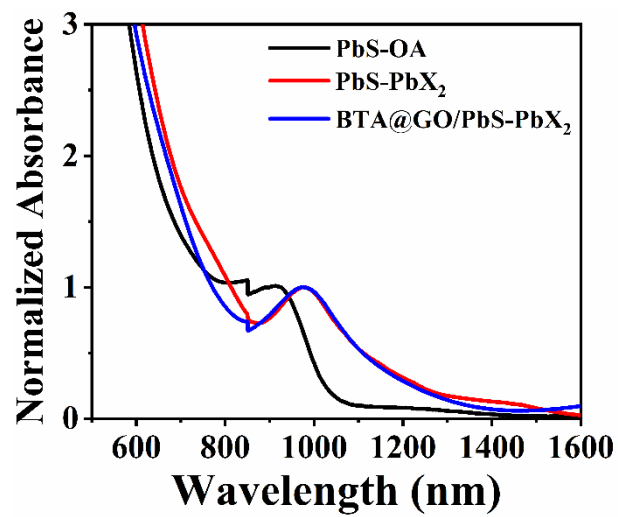


Fig. S4 Normalized absorption spectra of TA film samples.

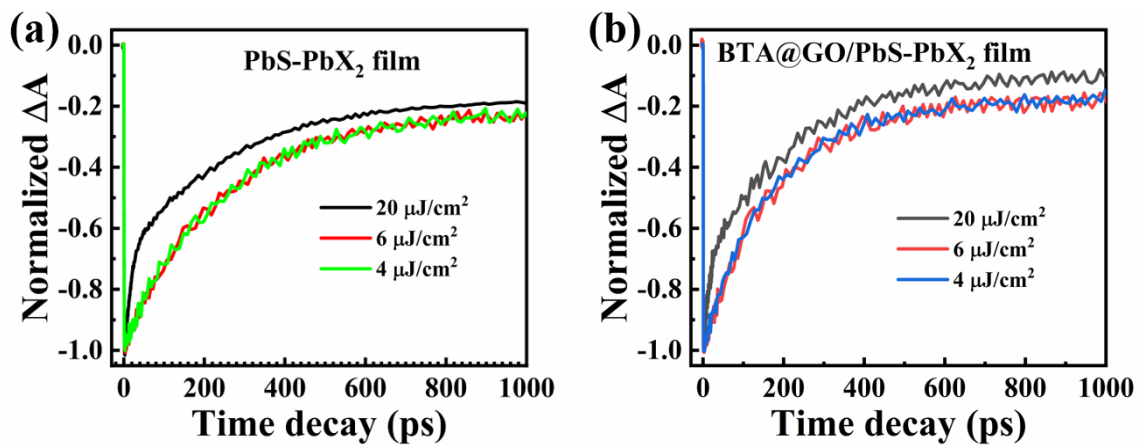


Fig. S5 Power dependent TA decay spectra of (a) PbSe-PbX₂ CQDs film and (b) BTA@GO/PbSe-PbX₂ hybrid CQDs film. The sample is pumped by 470 nm and probed at 970 nm. When the pump fluence reduced to 6 $\mu\text{J}/\text{cm}^2$, the signal of Auger recombination disappeared.

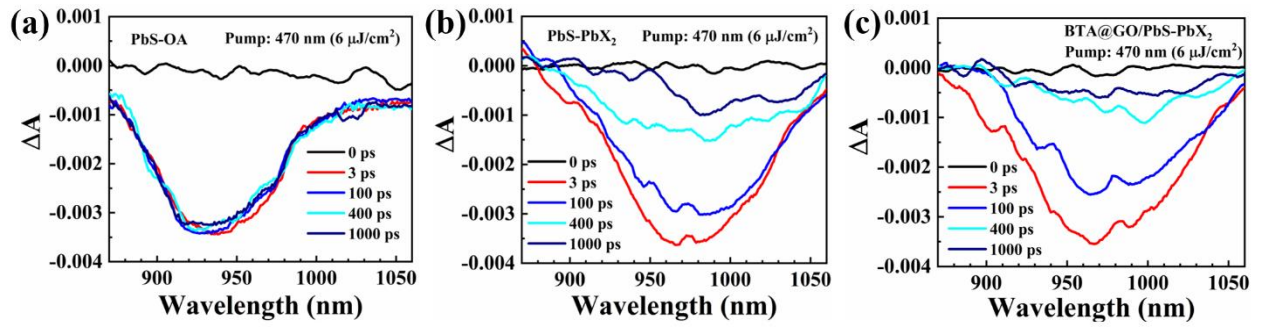


Fig. S6 TA spectra for (a) PbS-OA, (b) PbS-PbX₂ and (c) BTA@GO/PbS-PbX₂ CQD films at 0, 3, 100, 400, and 1000 ps. The pumped wavelength is 470 nm with a pulse fluence of $6 \mu\text{J}/\text{cm}^2$.

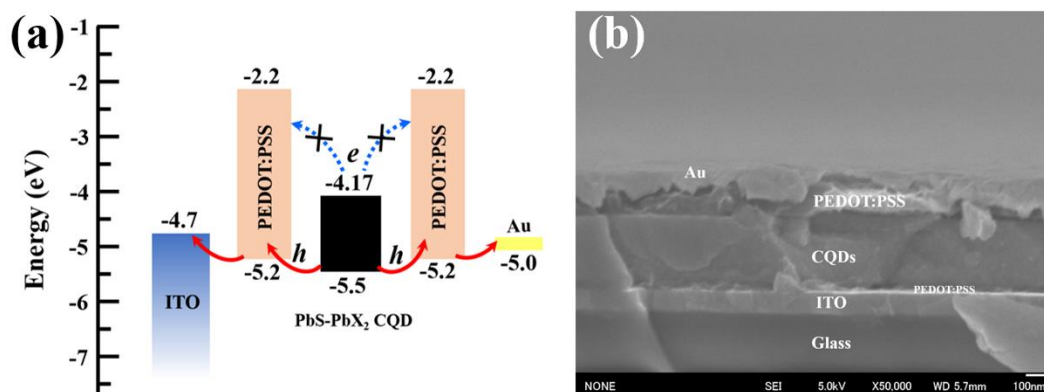


Fig. S7 (a) Schematic energy band diagrams of ITO, PEDOT:PSS, PbS-PbX₂ CQD and Au. (b) SEM cross section image of hole-only device.

The hole-only device with a structure of ITO/PEDOT/CQD/PEDOT/Au was fabricated by spin-coating a mixture of PEDOT:PSS solution and methanol with a volume ratio of 1:3 on ITO at 4000 r.p.m. After annealing ITO/PEDOT substrate at 150 °C for 15 min, the CQD ink was deposited on PEDOT:PSS layer by spin-coating. Finally, a thick PEDOT:PSS layer was deposited on CQD layer by spin-coating PEDOT:PSS solution/methanol mixture solution with a volume ratio of 1:1 followed by annealing in N₂ atmosphere.

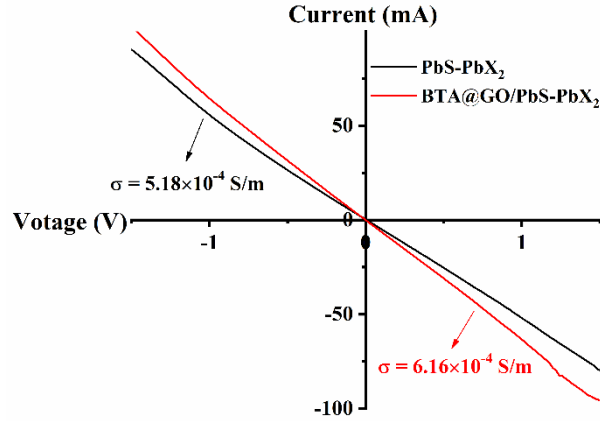


Fig. S8 I-V curves of Glass/Au/CQD film/Au devices based on PbS-PbX₂ and BTA@GO/PbS-PbX₂ CQD films which were measured under dark condition.

The values of resistance (R) fitted from the I-V curves are 18.2 Ω and 15.3 Ω for PbS-PbX₂ and BTA@GO/PbS-PbX₂ CQD films, respectively. The conductivity (σ) of the film can be calculated from following equation:

$$\sigma = \frac{d}{R \times S} \quad (\text{S1})$$

where d is the thickness of the film (about 330 nm) and S is the area of the device (0.35 cm²). The evaluated σ values for PbS-PbX₂ and BTA@GO/PbS-PbX₂ CQD films are 5.18×10^{-4} S/m and 6.16×10^{-4} S/m, respectively.

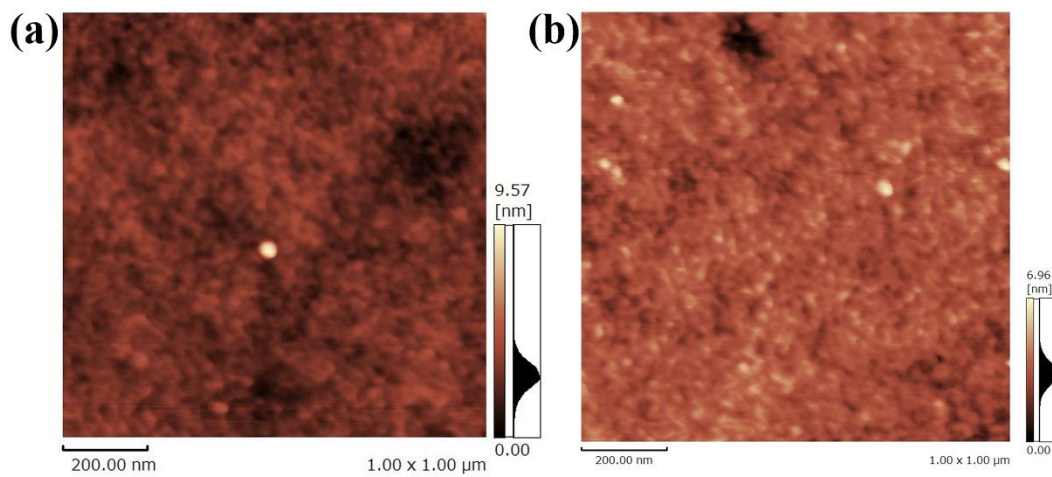


Fig. S9 AFM images of (a) PbS-PbX₂ and (b) BTA@GO/PbS-PbX₂ CQD films.

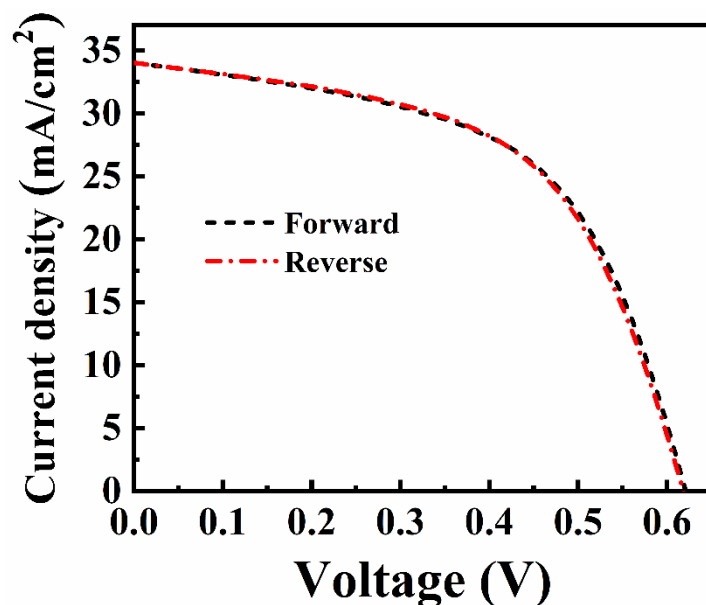


Fig. S10 *J-V* curves of BTA@GO/PbS-PbX₂ hybrid CQDs film (GO concentration is 0.2 mg/mL, CQD ink concentration is 300 mg/mL) based CQDSCs measured by forward (short circuit → open circuit) and reverse (open circuit → short circuit) scans with 0.01 V voltage steps and 200 ms delay times under AM 1.5G 100 mW/cm² illumination.

Table S1 Performance details of BTA@GO/PbS-PbX₂ hybrid CQDs film based CQDSCs as shown in Fig. S10.

	<i>J</i> _{sc} (mA/cm ²)	<i>V</i> _{oc} (V)	<i>FF</i> (%)	PCE (%)
Forward Scan	33.9	0.620	55.4	11.6
Reverse Scan	34.0	0.618	55.2	11.6
Average	34.0	0.619	55.3	11.6

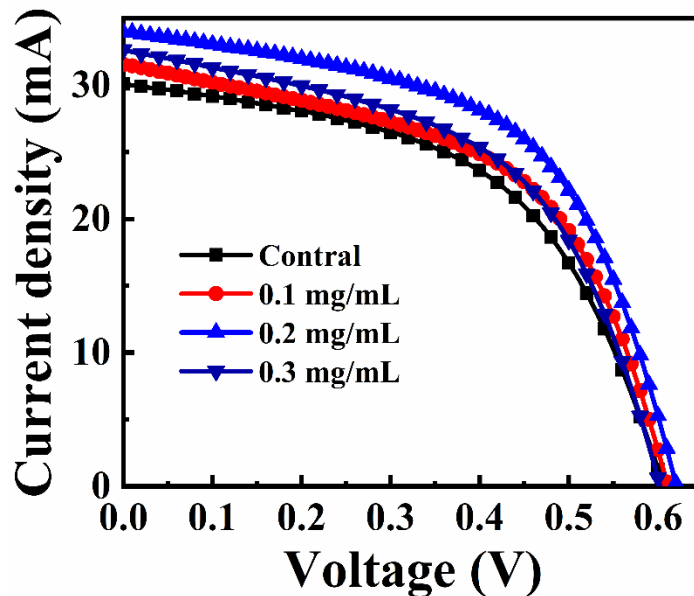


Fig. S11 J–V curves of the CQDSCs devices with the different concentrations of GO from 0 mg/mL to 0.3 mg/mL (CQD ink concentration is 300 mg/mL).

Table S2 Performance details of the PbS-PbX₂ CQD inks based devices which are fabricated by using different GO concentration inks as shown in Fig. S11.

GO concentration (mg/mL)	J_{sc} (mA/cm ²)	V_{oc} (V)	FF (%)	PCE (%)
0	30.1	0.606	52.3	9.5
0.1	31.5	0.611	53.2	10.3
0.2	33.9	0.621	55.4	11.7
0.3	32.6	0.602	52.4	10.3

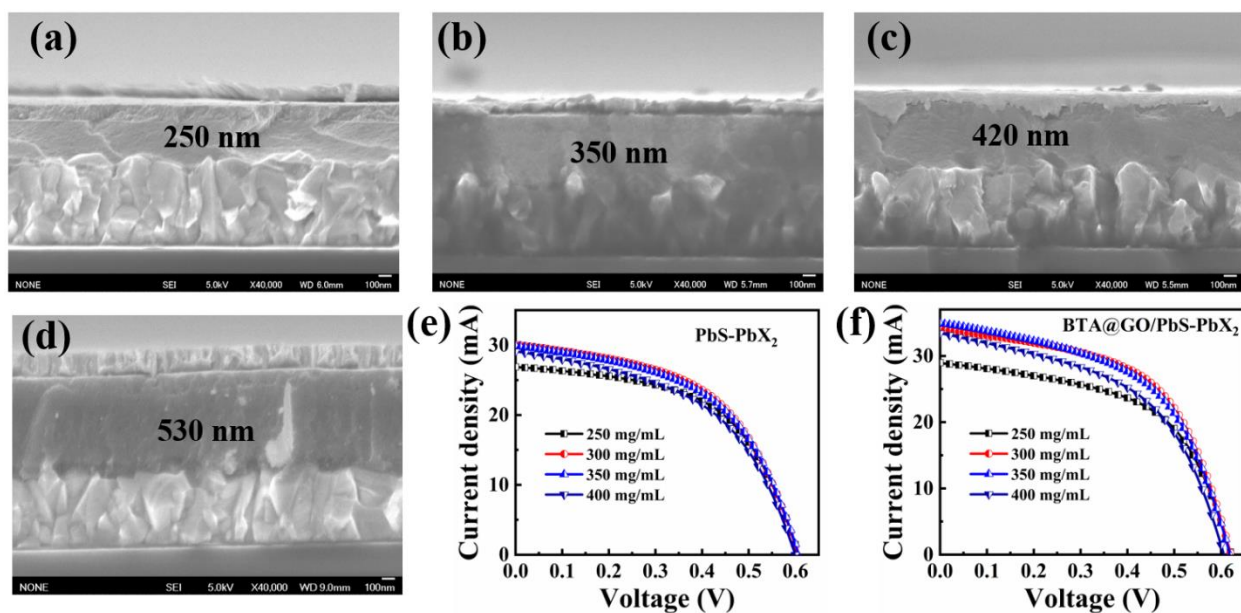


Fig. S12 SEM cross section images of CQDSCs with different BTA@GO/PbS-PbX₂ hybrid CQDs layer thicknesses (a, 250 mg/mL ink; b, 300 mg/mL ink; c, 350 mg/mL ink; d, 400 mg/mL ink), and *J-V* curves of the devices (e, without GO; f, with GO (0.2 mg/mL)).

Table S3. Performance details of the PbS-PbX₂ CQD ink based devices which are fabricated by using different CQD concentration inks as shown in Fig. S12e.

Ink concentration (mg/mL)	J_{sc} (mA/cm²)	V_{oc} (V)	FF (%)	PCE (%)
250	26.9	0.607	54.8	9.0
300	30.1	0.606	52.3	9.5
350	29.9	0.604	51.6	9.3
400	29.1	0.600	49.4	8.6

Table S4 Performance details of the BTA@GO/PbS-PbX₂ hybrid CQD inks based devices which are fabricated by using different CQD inks as shown in Fig. S12f.

Ink concentration (mg/mL)	J_{sc} (mA/cm²)	V_{oc} (V)	FF (%)	PCE (%)
250	29.0	0.621	54.9	9.9
300	34.0	0.621	55.4	11.7
350	34.9	0.619	52.5	11.4
400	33.4	0.606	50.5	10.2

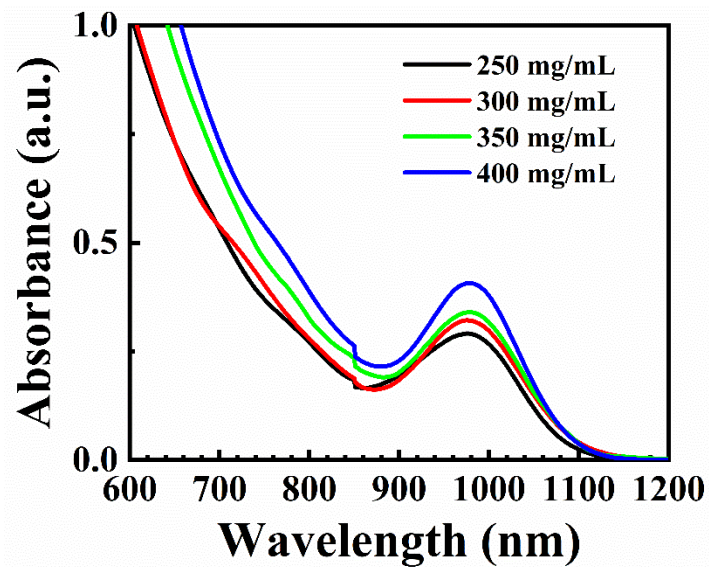


Fig. S13 The absorption spectra of BTA@GO/PbS-PbX₂ hybrid CQD inks based devices which are fabricated by using different CQD inks.

Table S5 Fitted resistances and capacitances from the EIS spectra of PbS-PbX₂ CQDs and BTA@GO/PbS-PbX₂ hybrid CQDs based devices as shown in Fig. 6b.

Device	R_s (Ω)	R_{low} (Ω)	R_{inter} (Ω)	C_{low} (F)	C_{inter} (F)
PbS-PbX ₂	12.9	1035	130	1.13E-7	1.34E-7
BTA@GO/PbS-PbX ₂	12.3	1215	196	1.1E-7	1.3E-7

The value of the R_{rec} was calculated by using follow equation:

$$R_{rec} = R_{low} + R_{inter} \quad (S2)$$

where R_{low} and R_{inter} are resistances at low and intermediate frequency, respectively. The values of R_{rec} are 1165 Ω and 1411 Ω for PbS-PbX₂ film and BTA@GO/PbS-PbX₂ hybrid CQDs film based devices, respectively.

The value of k_{rec} was calculated by using follow equation:

$$k_{rec-low} = \frac{1}{R_{low} \times C_{low}} \quad (S3)$$

$$k_{rec-inter} = \frac{1}{R_{inter} \times C_{inter}} \quad (S4)$$

where C_{low} and C_{inter} are capacitances at low and intermediate frequency, respectively. The values of $k_{rec-low}$ are $8.5 \times 10^3 \text{ s}^{-1}$ and $7.5 \times 10^3 \text{ s}^{-1}$ for PbS-PbX₂ film and BTA@GO/PbS-PbX₂ hybrid CQDs film based devices, respectively. And the values of $k_{rec-inter}$ for PbS-PbX₂ film and BTA@GO/PbS-PbX₂ hybrid CQDs film based devices are $5.7 \times 10^4 \text{ s}^{-1}$ and $3.9 \times 10^4 \text{ s}^{-1}$, respectively. In the solar cells device, the recombination rate mainly depends on the fast recombination process. Thus, the carrier recombination rates for PbS-PbX₂ film and BTA@GO/PbS-PbX₂ hybrid CQDs film based devices can be approximated as $5.7 \times 10^4 \text{ s}^{-1}$ and $3.9 \times 10^4 \text{ s}^{-1}$, respectively.

Reference

1. T.F. Yeh, F.F. Chan, C.T. Hsieh, H. Teng. Graphite Oxide with Different Oxygenated Levels for Hydrogen and Oxygen Production from Water under Illumination: The Band Positions of Graphite Oxide. *J. Phys. Chem. C*, **115**(45), 22587-22597 (2011). <https://doi.org/10.1021/jp204856c>
2. M.A. Velasco-Soto, S.A. Pérez-García, J. Alvarez-Quintana, Y. Cao, L. Nyborg, L. Licea-Jiménez. Selective Band Gap Manipulation of Graphene Oxide by Its Reduction with Mild Reagents. *Carbon*, **93**, 967-973 (2015). <https://doi.org/10.1016/j.carbon.2015.06.013>