

Supporting Information for

A High Capacity Ammonium Vanadate Cathode for Zinc-ion Battery

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Supplementary Figures and Table

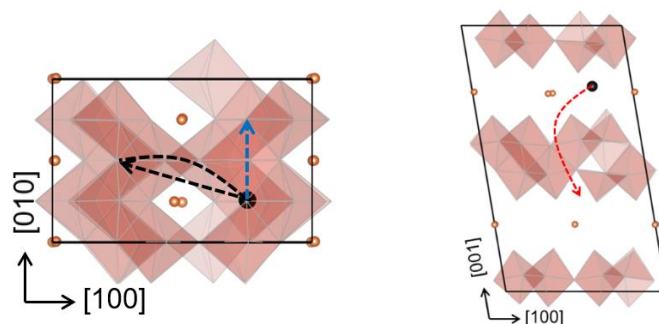


Fig. S1 Schematics of possible diffusion pathways for Zinc ion diffusion in monoclinic $\text{NH}_4\text{V}_4\text{O}_{10}$ viewed along the [100] (black arrows), [010] (blue arrow) and [001] (red arrow) direction. Orange and black balls indicate NH_4^+ ions and the most energetically

favorable location for Zn^{2+} intercalation, respectively. The VO layers are represented by red polyhedrons.

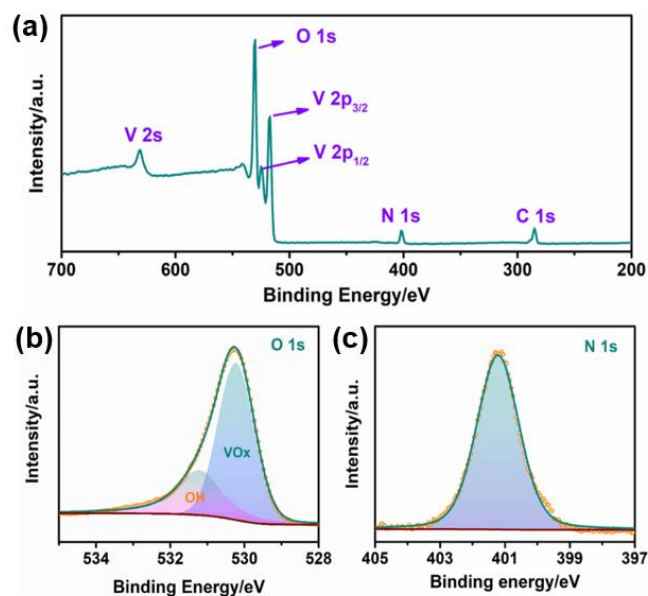


Fig. S2 The full XPS spectrum (a), and high-resolution XPS spectra of O (b) and N (c) species.

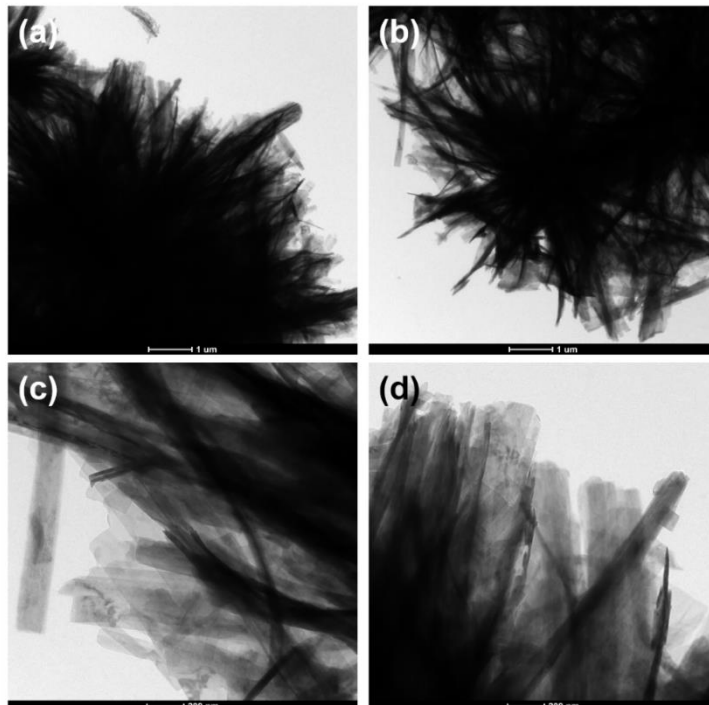


Fig. S3 TEM images of the 3D-NVO sample at different magnifications

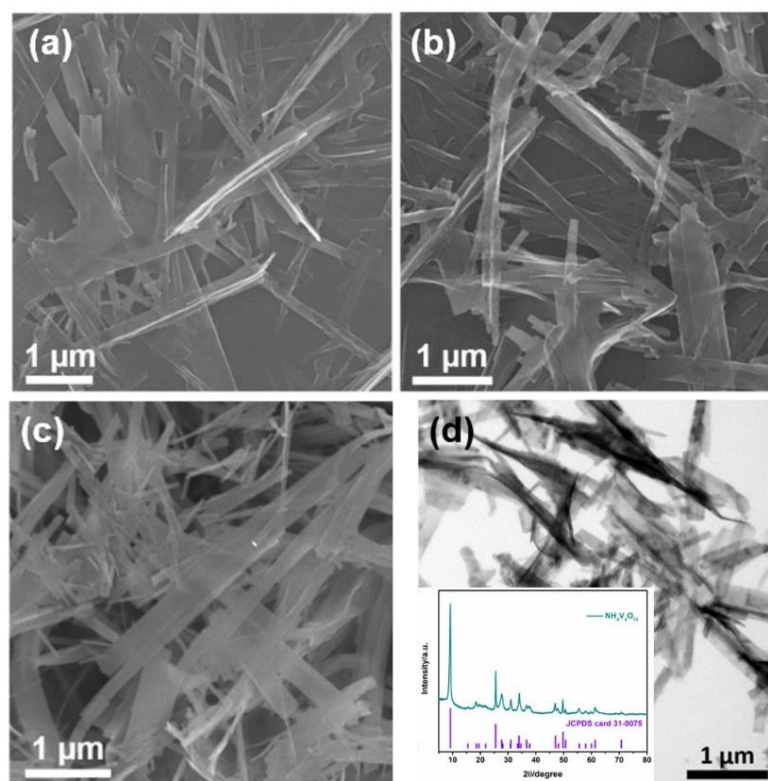


Fig. S4 (a-c) SEM and (d) TEM images of the sample prepared by conventional autoclave hydrothermal method at reaction time of 2 h (a), 6 h (b) and 12 h (c and d). Inset in (d) is the corresponding XRD pattern obtained at 12 h, which can be assigned to a pure monoclinic structure of NH₄V₄O₁₀ (JCPDS Card No. 31-0075).

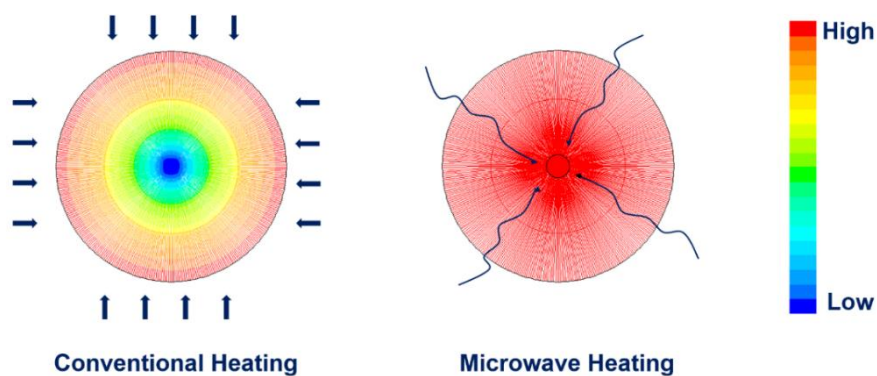


Fig. S5 Schematic illustration of the difference between microwave and conventional heating. In conventional heating system, there is a temperature gradient from the outside to the inner core.

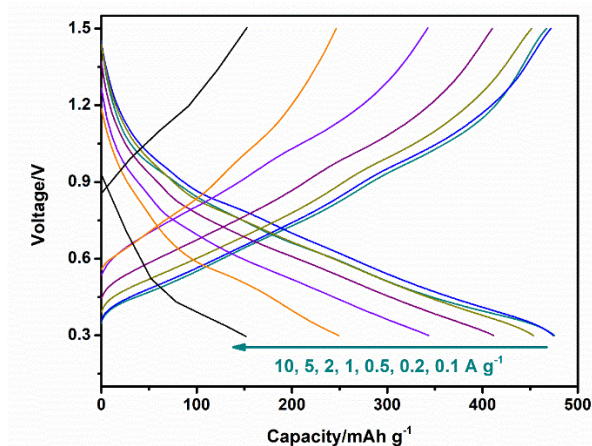


Fig. S6 Galvanostatic discharge and charge profiles of the 3D-NVO cathode at different current densities

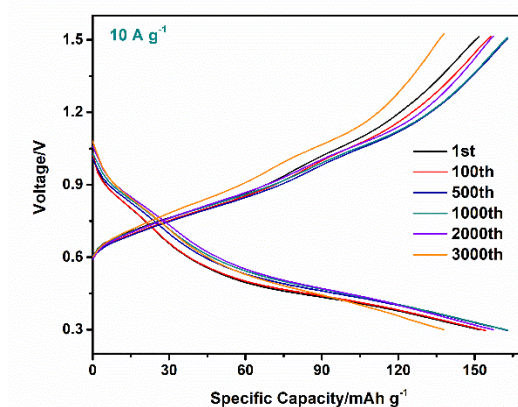


Fig. S7 Galvanostatic discharge and charge profiles of the 3D-NVO cathode during different cycles at 10 A g⁻¹

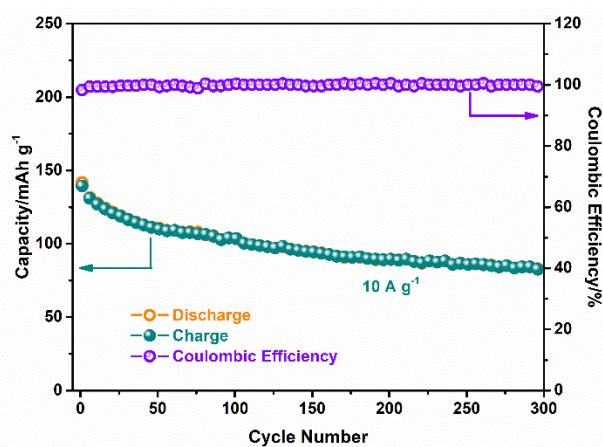


Fig. S8 Cycling performance of the NVO nanobelts obtained from the traditional hydrothermal method at 10 A g⁻¹

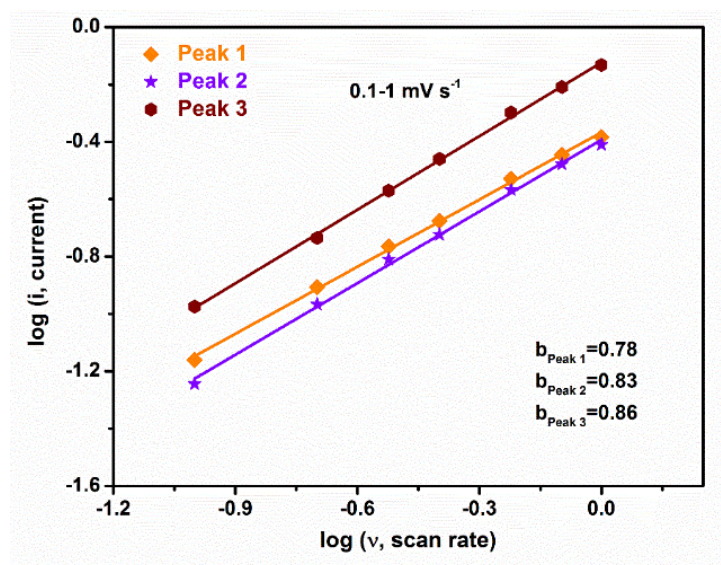


Fig. S9 Plots of $\log(i)$ vs. $\log(v)$ based on anodic and cathodic peaks in Fig. 3f

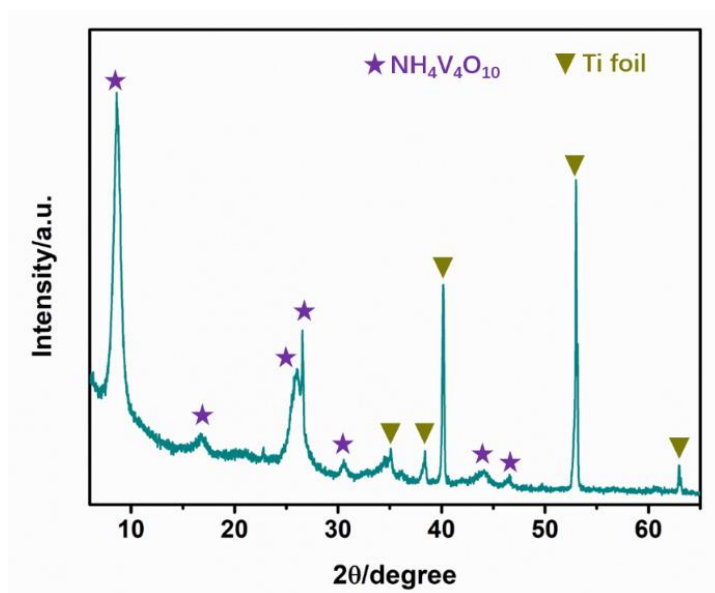


Fig. S10 The XRD pattern of the 3D-NVO cathode after 100 cycles at 10 A g^{-1}

Table S1 A comparison of the reversible capacity of our 3D-NH₄V₄O₁₀ cathode to previously reported superior ZIB cathodes

Cathode materials	Reversible capacity [mAh g ⁻¹]	Rate [mA g ⁻¹]	Number of cycles	Capacity retention
This work (3D-NH₄V₄O₁₀)	486	100	50	98%
α-MnO ₂ [S1]	233	83	50	63%
δ-MnO ₂ [S2]	225	83	100	50%
Mn ₃ O ₄ [S3]	195	200	70	67%
ZnMn ₂ O ₄ [S4]	106	100	300	59%
VO ₂ [S5]	357	100	50	99%
VS ₂ [S6]	138	200	200	80%
V ₂ O ₅ [S7]	215	100	160	95%
V ₂ O ₅ ·nH ₂ O [S8]	196	14.4	120	87%
H ₂ V ₃ O ₈ @Graphene [S9]	360	300	150	93%
LiV ₃ O ₈ [S10]	185	133	65	78%
Na _{0.33} V ₂ O ₅ [S11]	276	200	100	91%
Zn ₂ V ₂ O ₇ [S12]	210	300	200	94%
Zn ₃ V ₂ O ₇ (OH) ₂ ·2H ₂ O [S13]	149	200	300	68%
Na ₂ V ₆ O ₁₆ ·1.63H ₂ O [S14]	296	100	100	78%
Na _{1.1} V ₃ O _{7.9} @rGO [S15]	200	50	30	87%
V _{1-x} Al _x O _{1.52} (OH) _{0.77} [S16]	156	15	50	68%
Na ₃ V ₂ (PO ₄) ₃ @C [S17]	97	50	100	74%

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