

Supporting Information for

N-Doped-Graphene Decorated NiCo Alloy Couple with Mesoporous NiCoMoO Nano-sheets Heterojunction for Enhanced Water Electrolysis Activity at High Current Density

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Supplementary Tables and Figures

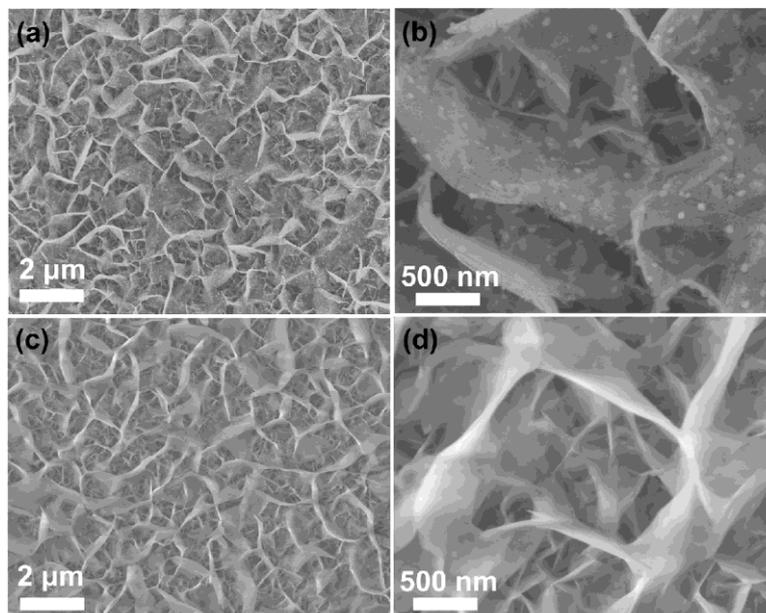


Fig. S1 SEM images of (a, b) precursors annealed at 450 °C and (c, d) precursors

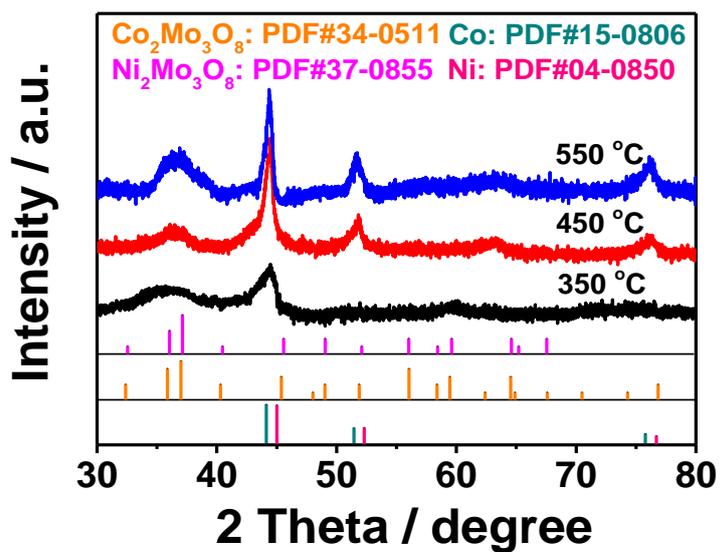


Fig. S2 XRD spectra of precursors annealed at different temperatures

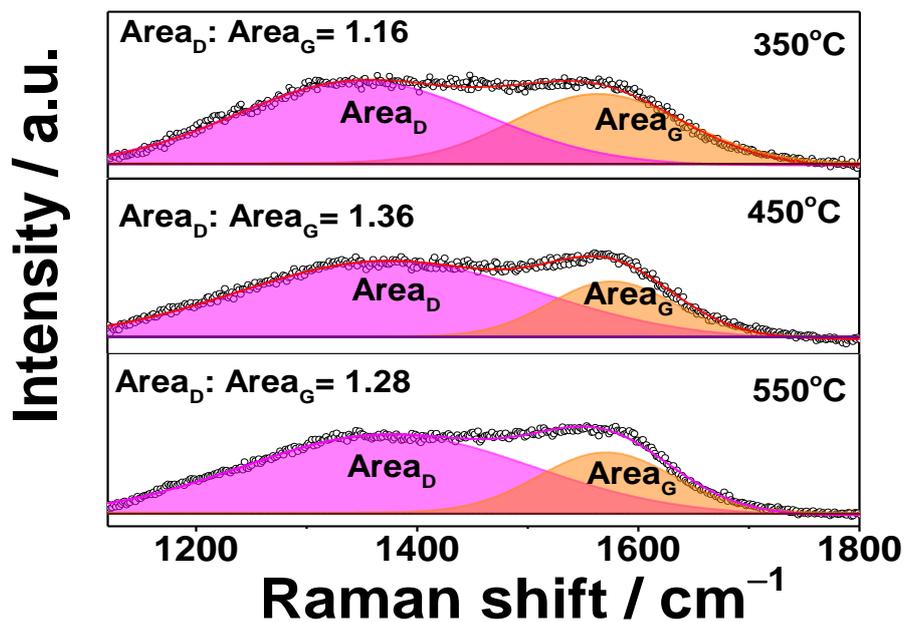


Fig. S3 Raman spectra of precursors annealed at different temperatures

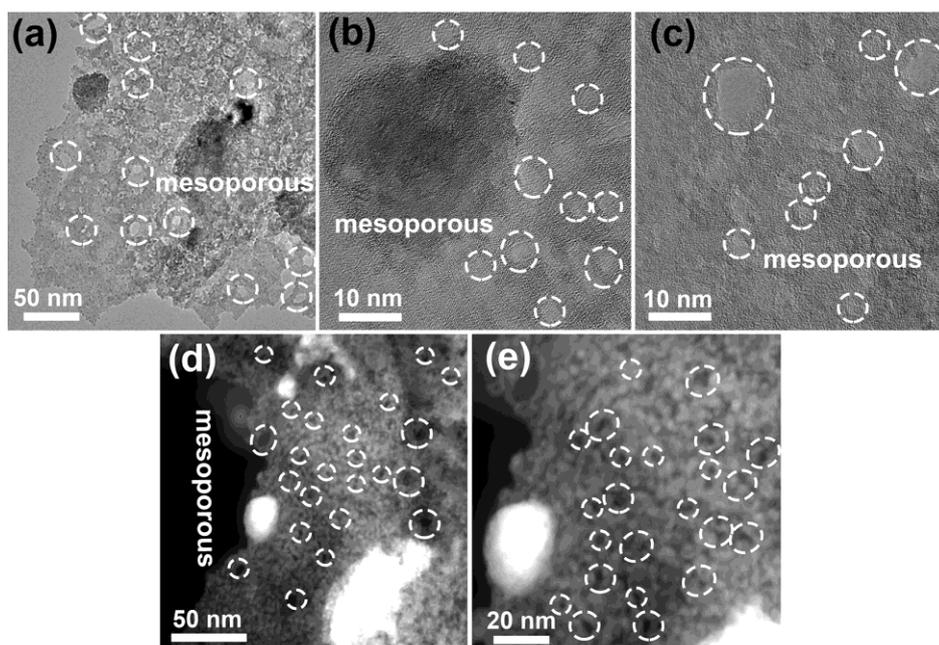


Fig. S4 (a-c) TEM and HRTEM and (d, e) HAADF-STEM images of NiCo@C-NiCoMoO/NF

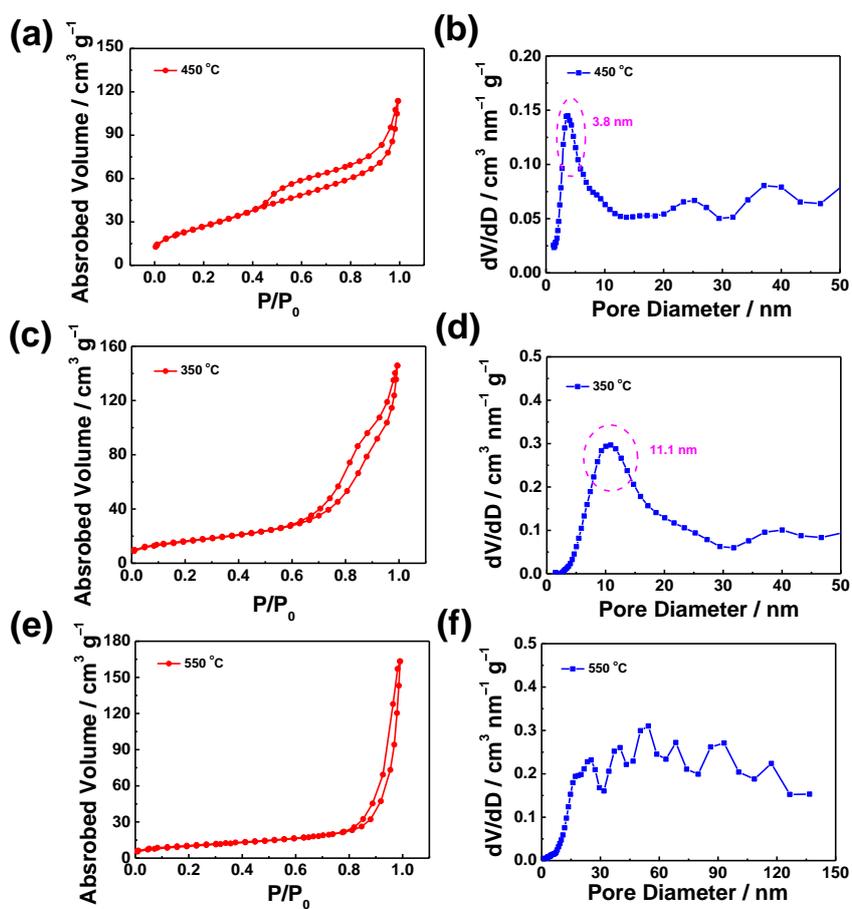


Fig. S5 (a, c, e) N_2 adsorption/desorption isotherms and (b, d, f) the corresponding pore size distributions of precursors annealed at different temperatures

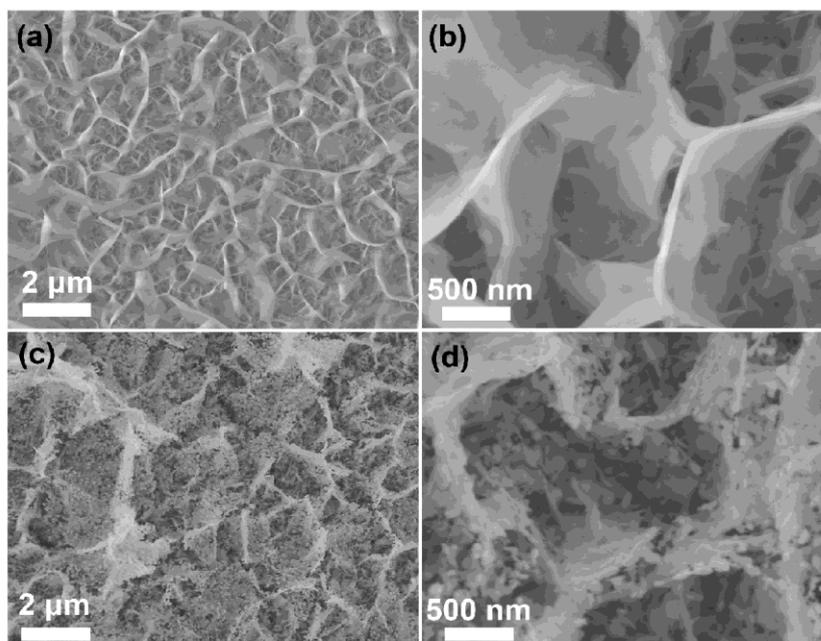


Fig. S6 SEM images of precursors annealed at (a, b) 350 °C and (c, d) 550 °C

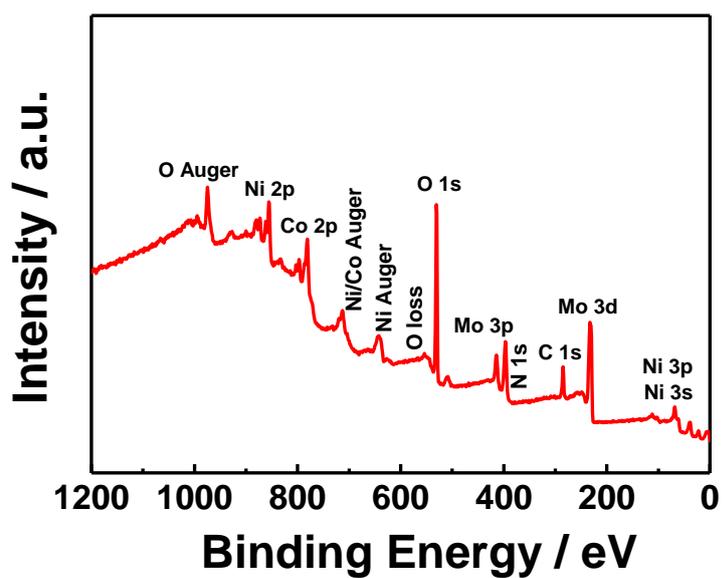


Fig. S7 XPS summary spectra of NiCo@C-NiCoMoO/NF

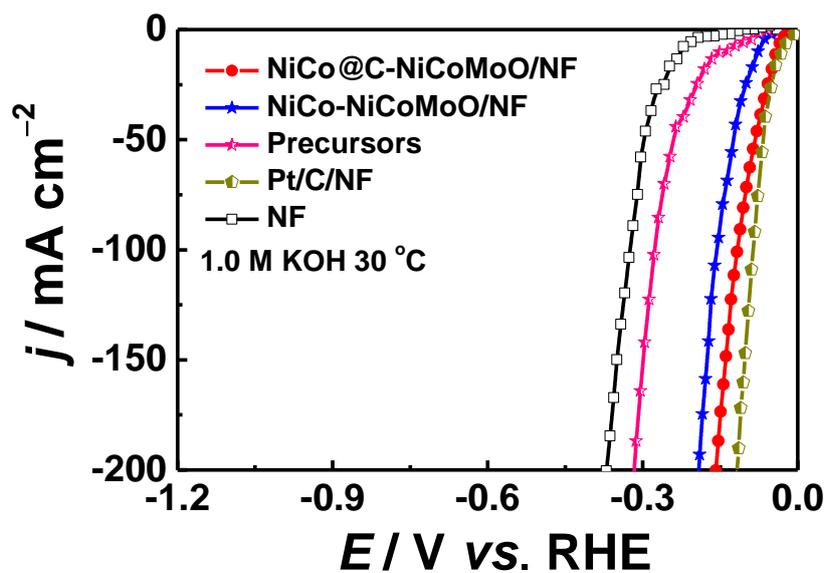


Fig. S8 LSV curves of HER for investigated samples

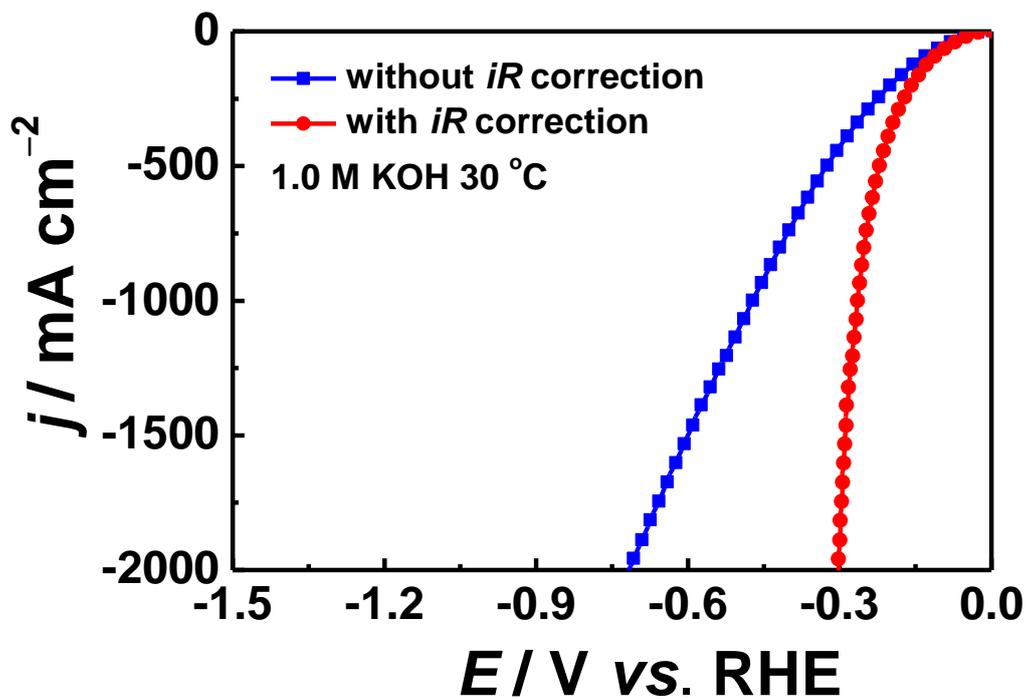


Fig. S9 LSV curves of NiCo@C-NiCoMoO/NF with/without iR correction

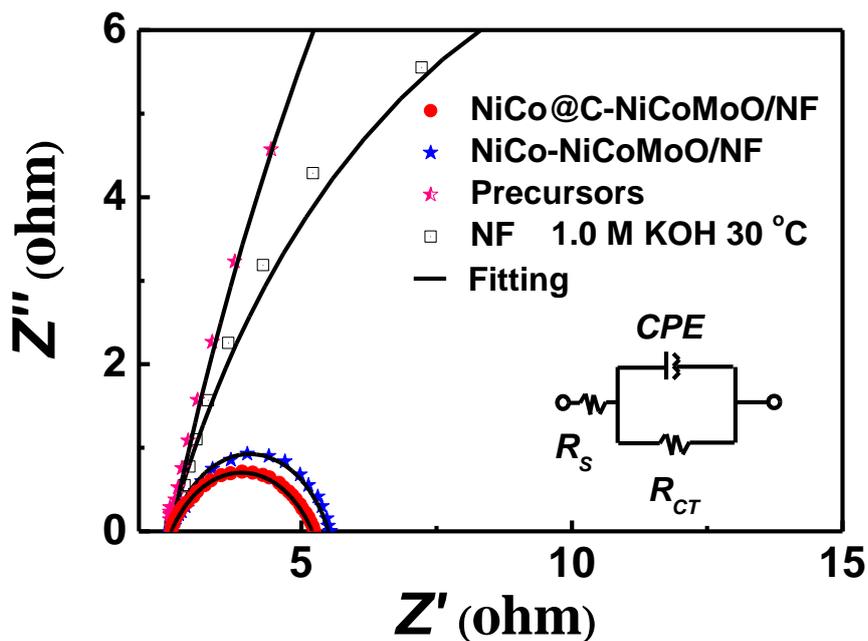


Fig. S10 Nyquist plots tested at -0.2 V for HER with a frequency from 100,000 to 0.1 Hz in 1.0 M KOH; inset is the equivalent circuit model

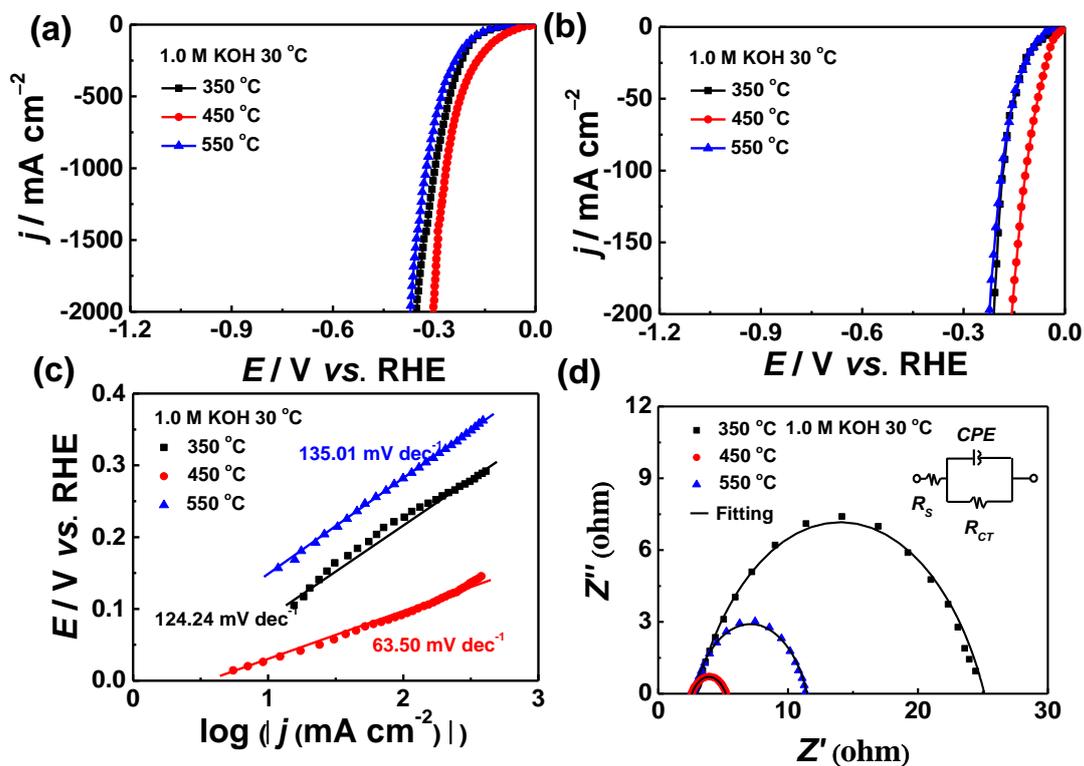


Fig. S11 (a-d) LSV curves of HER for precursors annealed at different temperatures

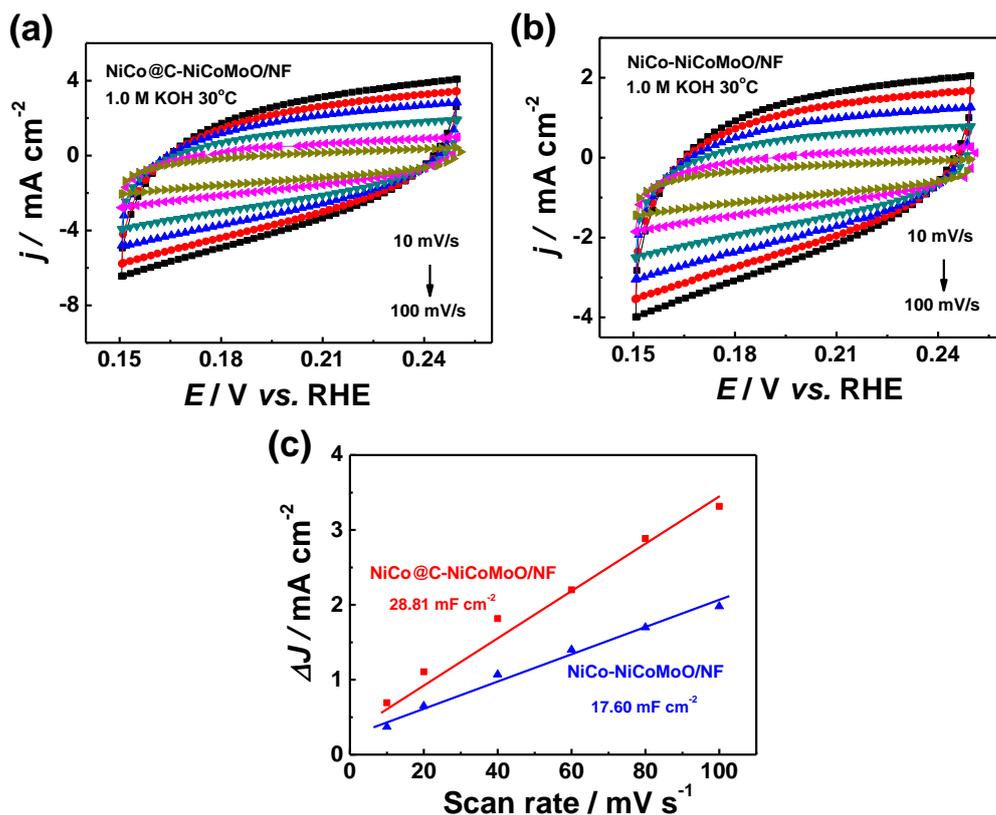


Fig. S12 (a, b) Typical CVs of the samples with scan rates ranging from 10 to 100 mV s^{-1} , the scanning potential range is from 0.15 V to 0.25 V; (c) Estimation of C_{dl} by plotting the capacitive current density against the scan rate to fit a linear regression

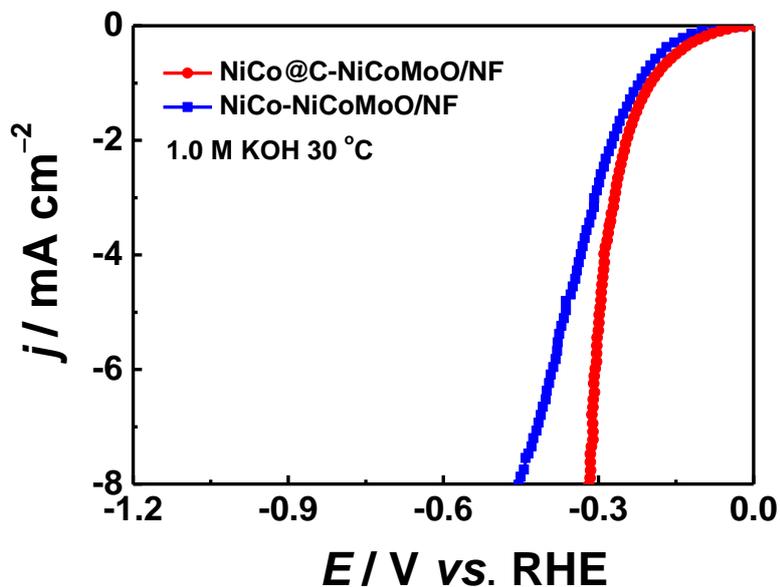


Fig. S13 LSV curves of HER for NiCo@C-NiCoMoO/NF and NiCo-NiCoMoO/NF normalized by EASAs

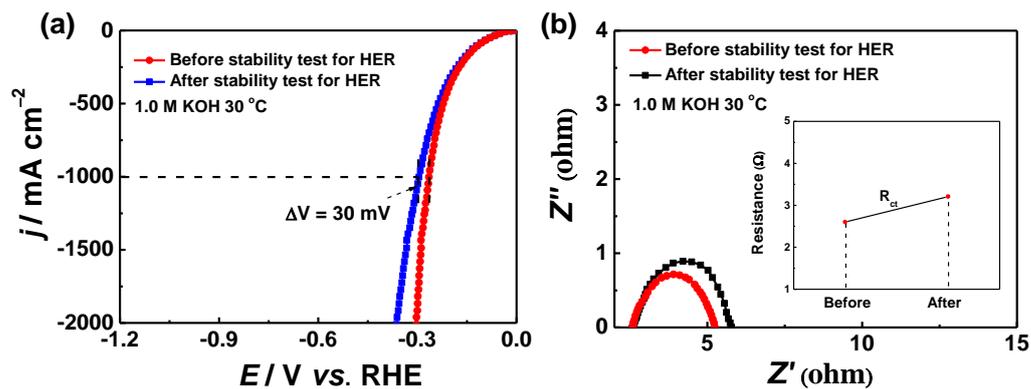


Fig. S14. (a) LSV curves and (b) R_{ct} of NiCo@C-NiCoMoO/NF before and after HER stability test

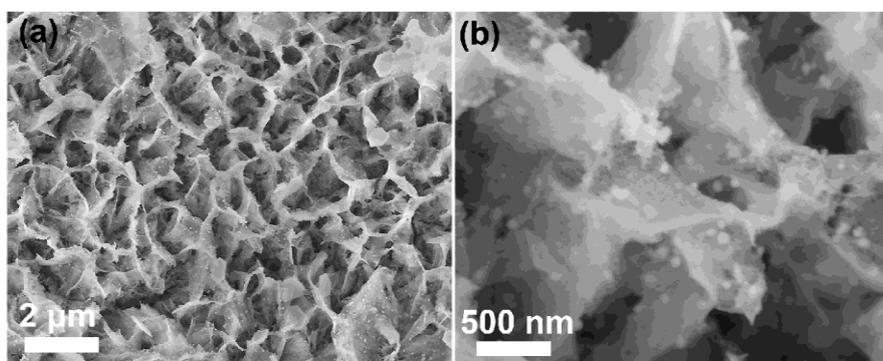


Fig. S15 (a, b) SEM images of NiCo@C-NiCoMoO/NF after HER stability test

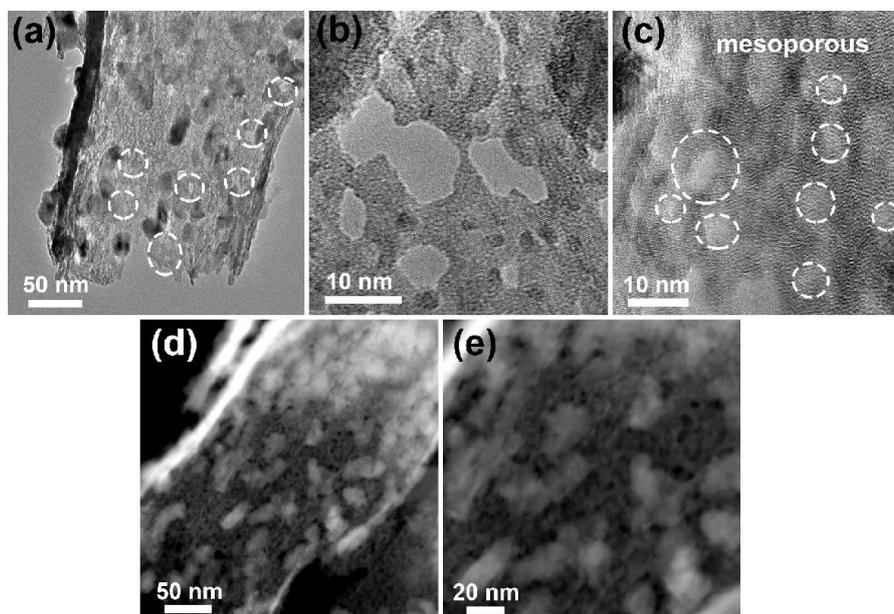


Fig. S16 (a-e) TEM, HRTEM, and HAADF-STEM images of NiCo@C-NiCoMoO/NF after HER stability test

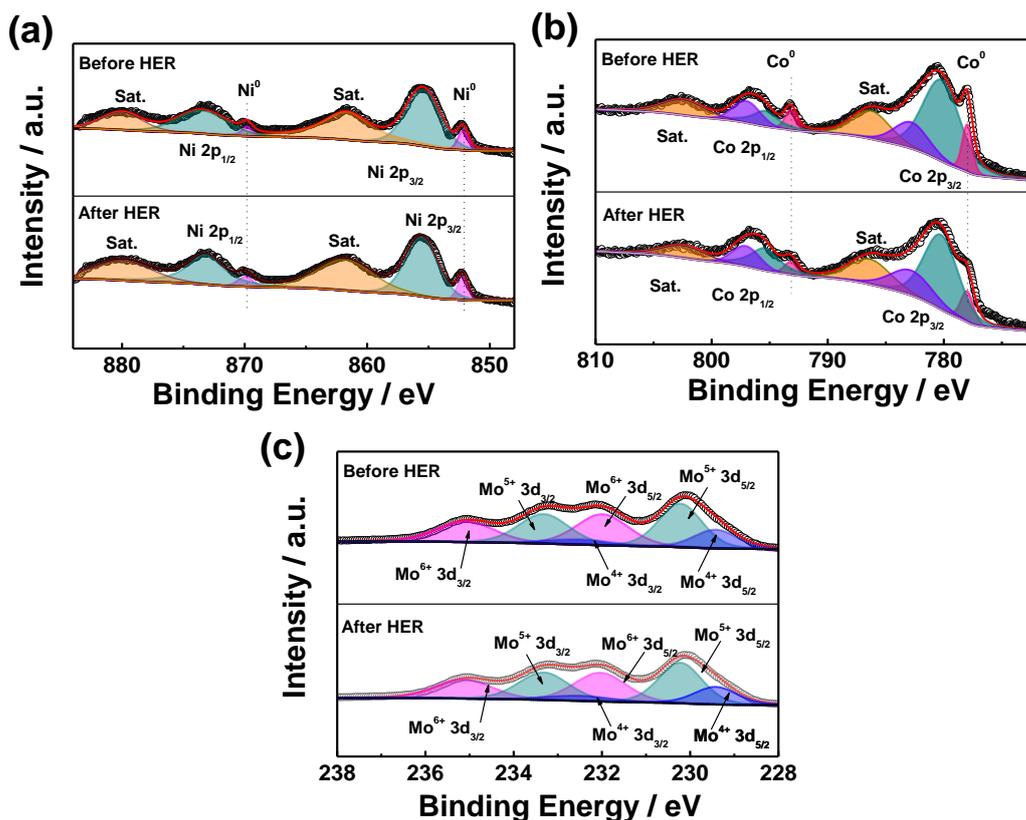


Fig. S17 (a-c) XPS spectra of NiCo@C-NiCoMoO/NF after HER stability test

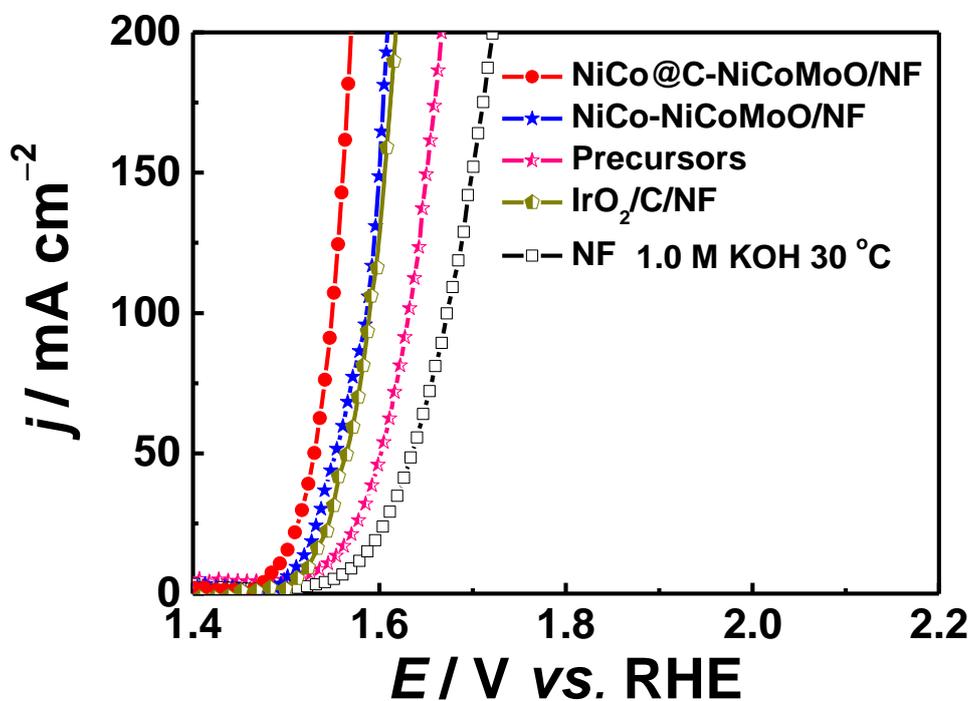


Fig. S18 LSV curves of OER for the investigated samples

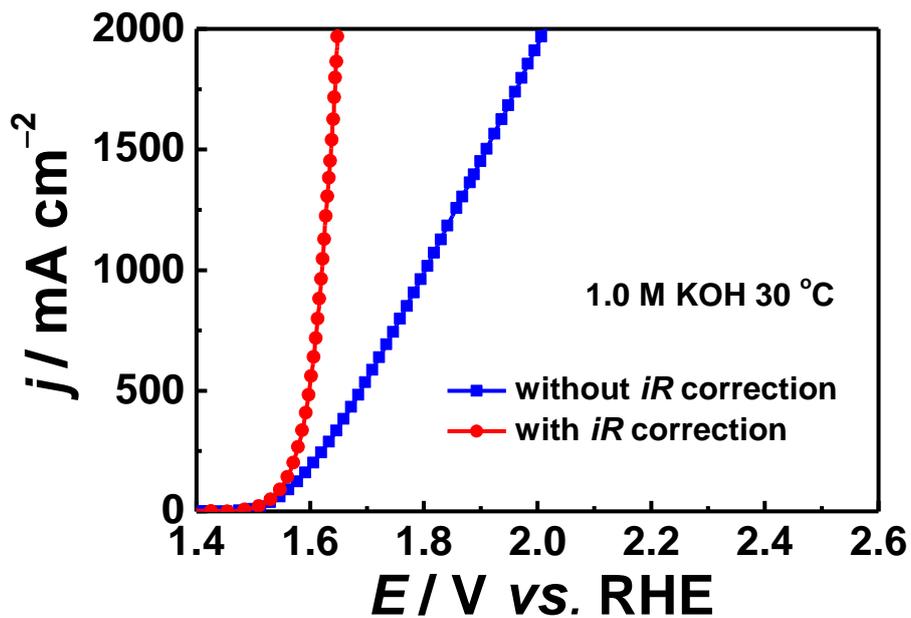


Fig. S19 LSV of NiCo@C-NiCoMoO/NF with/without iR correction

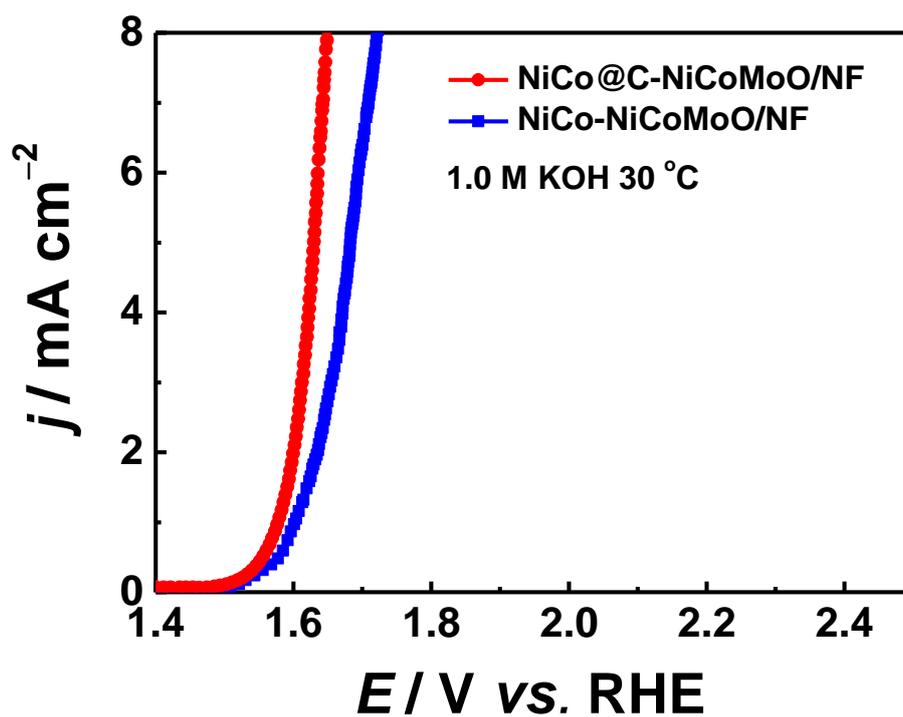


Fig. S20 LSV curves of OER for NiCo@C-NiCoMoO/NF and NiCo-NiCoMoO/NF normalized by EASAs

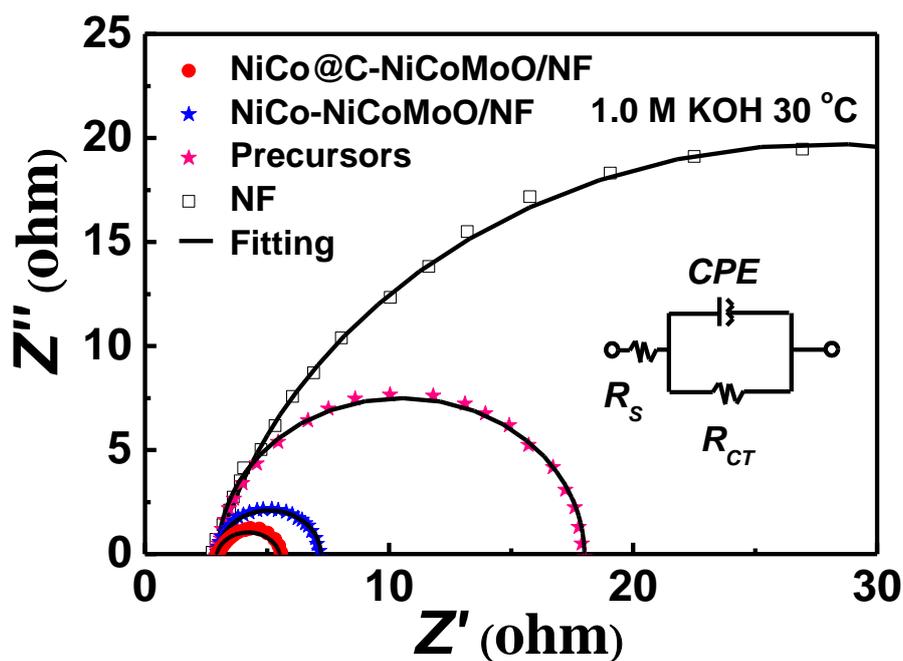


Fig. S21 Nyquist plots tested at 1.5 V for OER with a frequency from 100,000 to 0.1 Hz in 1.0 M KOH; inset is the equivalent circuit model

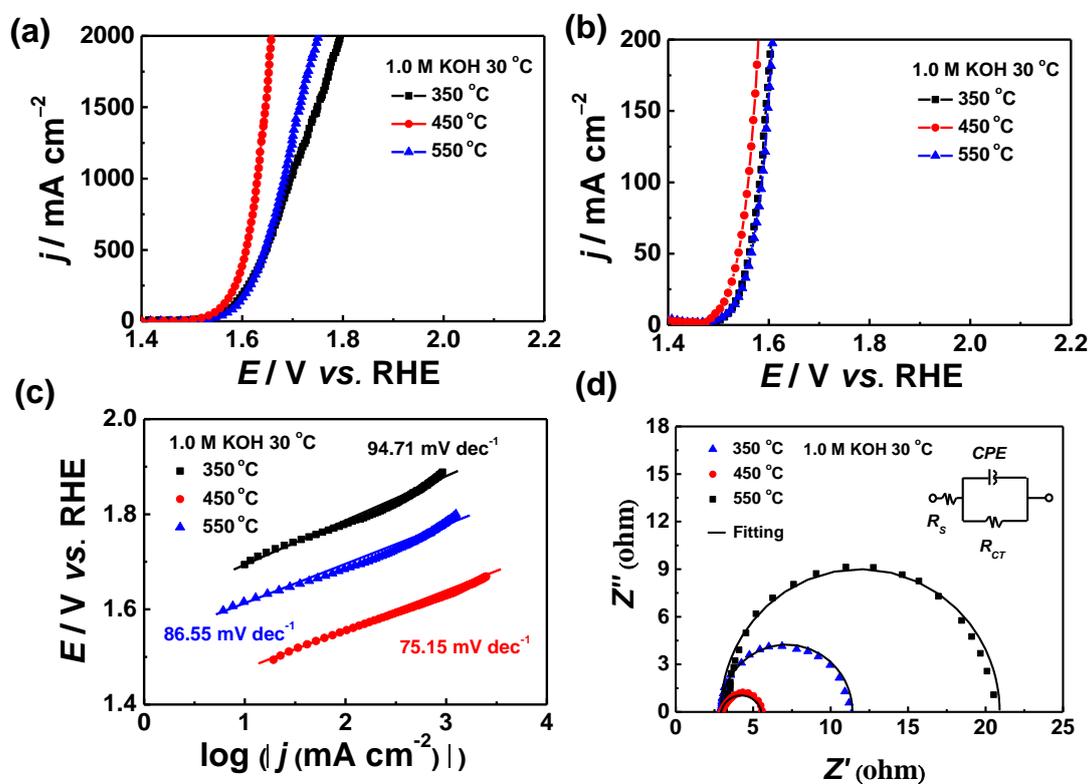


Fig. S22 (a-d) LSV curves, Tafel, and EIS of OER for precursors annealed at different temperatures

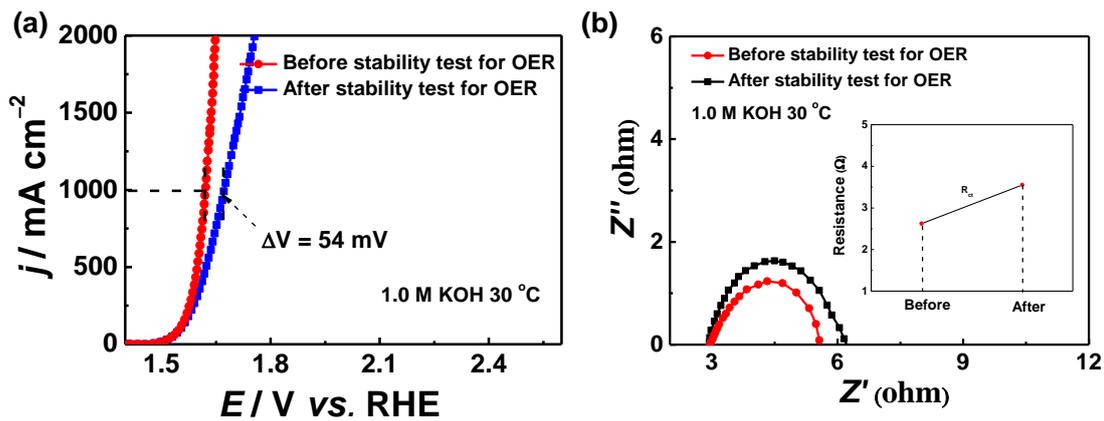


Fig. S23 (a) LSV curves and (b) R_{ct} of NiCo@C-NiCoMoO/NF before and after OER stability test

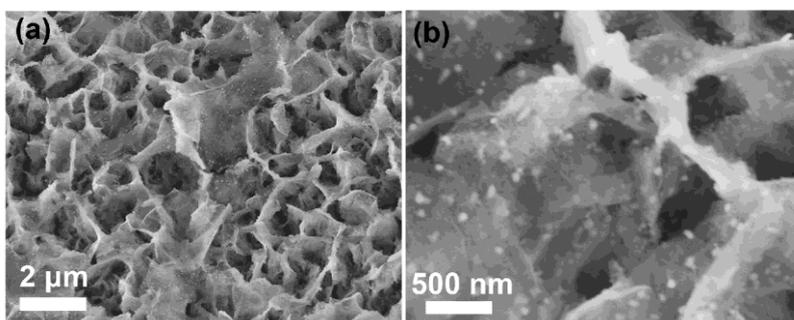


Fig. S24 SEM images of NiCo@C-NiCoMoO/NF after OER stability test

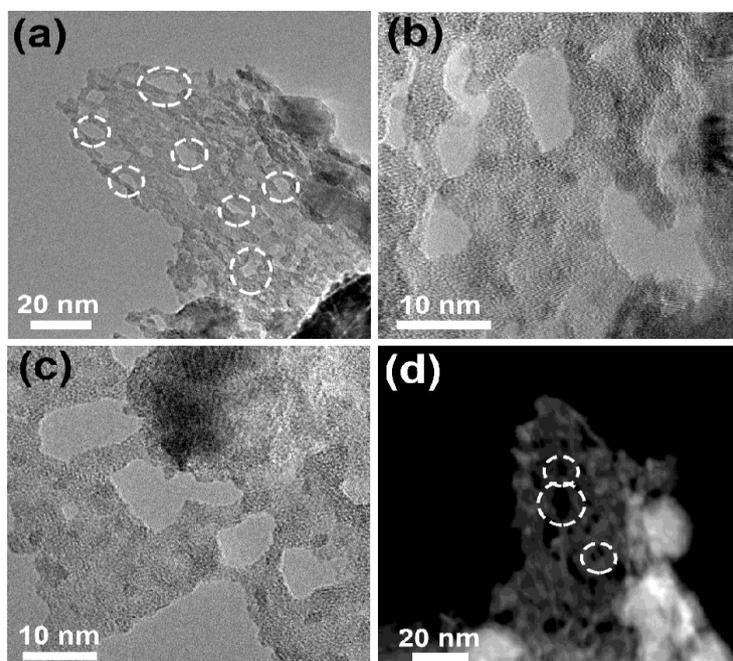


Fig. S25 (a-d) TEM, HRTEM, and HAADF-STEM images of NiCo@C-NiCoMoO/NF after OER stability test

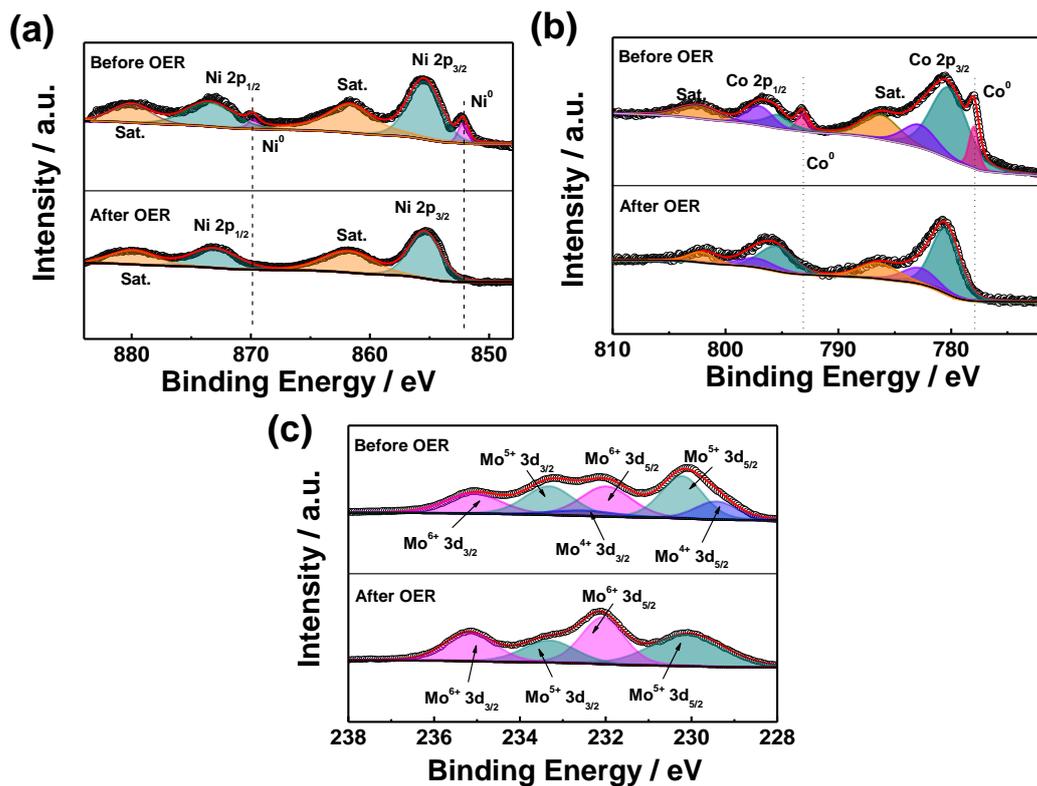


Fig. S26 (a-c) XPS spectra of NiCo@C-NiCoMoO/NF after OER stability test

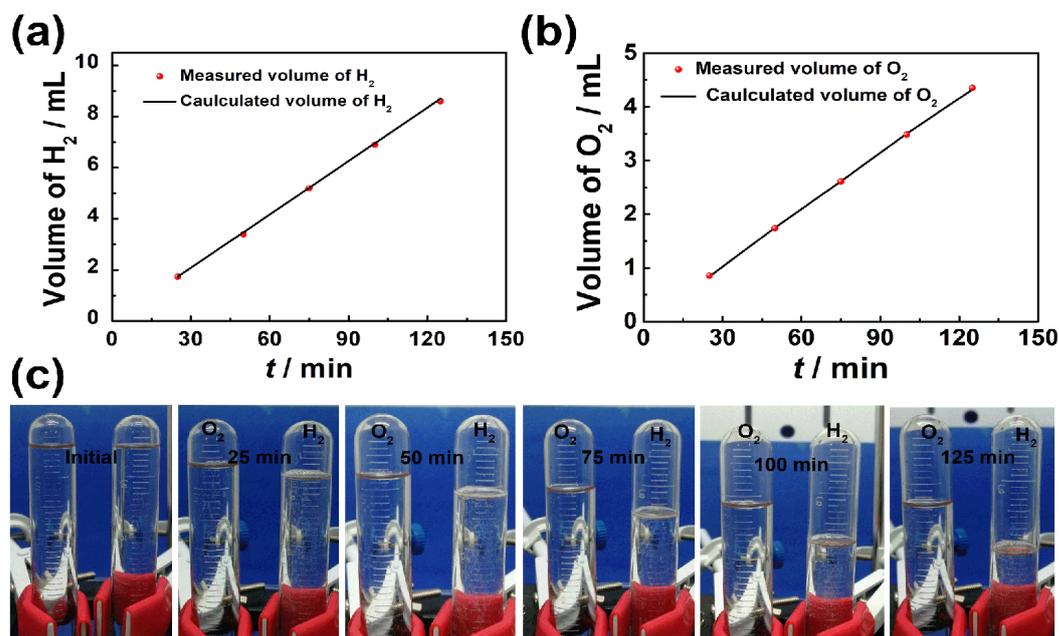


Fig. S27 (a, b) Volume of H₂ and O₂ theoretically calculated and actually measured at ± 10.0 mA versus time for NiCo@C-NiCoMoO/NF in 1.0 M KOH solution; (c) volume of H₂ and O₂ at 0, 25, 50, 75, 100, and 125 min

Table S1 BET results of precursors annealed at different temperatures

Catalysts	BET surface areas (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Pore size (nm)
350 °C	58.69	0.23	15.36
450 °C	102.96	0.18	6.83
550 °C	37.36	0.25	27.05

Table S2 Elemental composition of NiCo@C-NiCoMoO/NF obtained from ICP-MS

Element wt%	Ni	Co	Mo
NiCo@C-NiCoMoO/NF	31	15	22

Table S3 TOF and MA of NiCo@C-NiCoMoO/NF obtained at different overpotentials for HER

Overpotentials (mV)	TOF (s ⁻¹)	MA (mA g ⁻¹)
50	1.0×10 ⁻³	2.6×10 ³
100	4.0×10 ⁻³	1.2×10 ⁴
150	9.0×10 ⁻³	2.7×10 ⁴
200	2.0×10 ⁻²	5.6×10 ⁴

Table S4 Comparison TOF and MA of NiCo@C-NiCoMoO/NF for HER with other reported non-noble-metal catalysts

Catalysts	TOF (s ⁻¹ @ mV)	MA (mA g ⁻¹ @ mV)	References
NiCo@C-NiCoMoO/NF	4.0×10⁻³@100	1.2×10⁴@100	This work
NiSe ₂ -CoSe ₂ /NCF	2.13×10 ⁻³ @100	1.92×10 ³ @100	[S1]
Ni ₁₂ P ₅ -Ni ₄ Nb ₅ P ₄ /PCC	5.32×10 ⁻² @100	3.05×10 ³ @100	[S2]
NiSe	7.5×10 ⁻¹ @250	N/A	[S3]
Ni ₃ S ₂ -FeS-CoS/PNFCF	1.4×10 ⁻¹ @100	N/A	[S4]
Cr-doped FeNi-P/NCN	2.14×10 ⁻¹ @190	N/A	[S5]
Holey NCP	7.32×10 ⁻¹ @200	N/A	[S6]
Ni _{1.8} Cu _{0.2} -P/NF	1.2×10 ⁻³ @100	N/A	[S7]
N-NiCoP/NCF	4.958×10 ⁻² @200	N/A	[S8]
NiSe ₂ -FeSe ₂	1.6×10 ⁻³ @300	5.53×10 ² @100	[S9]
Mo-W-P/CC	1.9×10 ⁻² @100	N/A	[S10]
N-NiVFeP/NFF	3.867×10 ⁻² @180	N/A	[S11]
N-NiCoP _x /NCF	8.7×10 ⁻⁴ @40	3.14×10 ³ @100	[S12]
MoSe ₂ -NiSe ₂ -CoSe ₂ /PNCF	1.5×10 ⁻⁴ @80	1.38×10 ² @100	[S13]
CoP UPNSs	N/A	1.51×10 ⁵ @100	[S14]
2H-MoS ₂	N/A	~1.6×10 ⁴ @100	[S15]
2H-NbS ₂	N/A	4.315×10 ⁴ @250	[S16]
NiS/Ni ₂ P/CC	9.01×10 ⁻¹ @200	~5.8×10 ¹ @200	[S17]
NiCo ₂ P _x	2.6×10 ⁻² @100	N/A	[S18]

Table S5 TOF and MA of NiCo@C-NiCoMoO/NF obtained at different overpotentials for OER

Overpotentials (mV)	TOF (s ⁻¹)	MA (mA g ⁻¹)
250	4.0×10 ⁻⁴	3.0×10 ³
300	2.7×10 ⁻³	9.5×10 ³
350	1.5×10 ⁻²	4.5×10 ⁴

Table S6 Comparison TOF and MA of NiCo@C-NiCoMoO/NF for OER with other reported non-noble-metal catalysts

Catalysts	TOF (s ⁻¹ @ mV)	MA (mA g ⁻¹ @ mV)	References
NiCo@C-NiCoMoO/NF	2.7×10 ⁻³ @350	9.5×10 ³ @300	This work
NiSe ₂ -CoSe ₂ /NCF	4.55×10 ⁻¹ @300	5.28×10 ² @300	[S1]
NiSe	~3×10 ⁻¹ @320	N/A	[S3]
Amorphous NiFe-OH/NiFeP	3.6×10 ⁻² @250	~5.0×10 ³ @250	[S19]
Ni ₃ S ₂ -FeS-CoS/PNFCF	1.2×10 ⁻³ @180	N/A	[S4]
Cr-doped FeNi-P/NCN	1.06×10 ⁻¹ @140	N/A	[S5]
N-NiCoP/NCF	1.47×10 ⁻³ @200	N/A	[S8]
NiSe ₂ -FeSe ₂	1.09×10 ⁻⁴ @300	1.338×10 ² @100	[S9]
N-NiCoP _x /NCF	1.2×10 ⁻⁴ @300	1.014×10 ³ @400	[S12]
NiS/Ni ₂ P/CC	3.31×10 ⁻¹ @300	~6.0×10 ¹ @320	[S17]

Supplementary References

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