

Supporting Information for

Boron Nanosheet-Supported Rh Catalysts for Hydrogen Evolution: A New Territory for the Strong Metal-Support Interaction Effect

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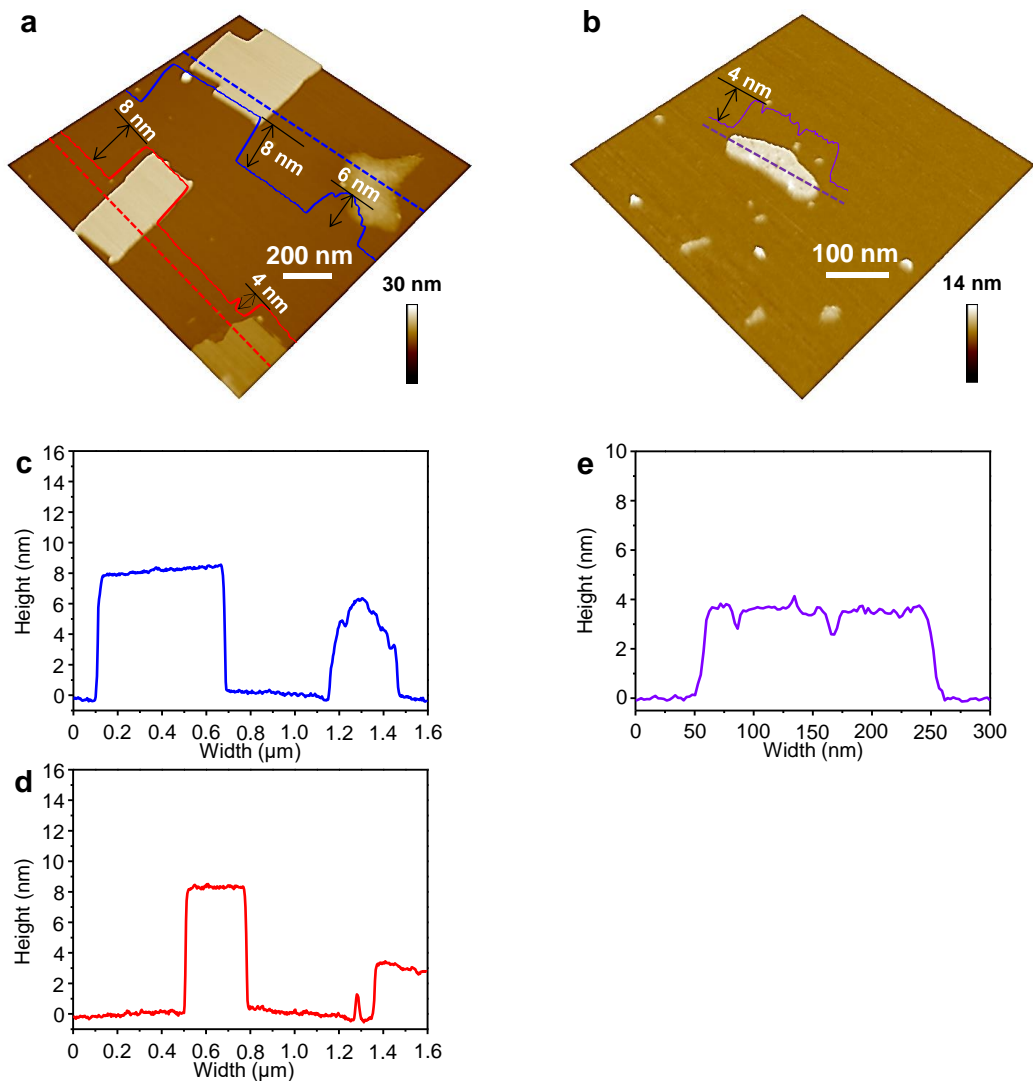


Fig. S1 **a, b** Three-dimensional AFM images of BNS and Rh NP@BNS. **c, d** The corresponding height profile for BNS. **e** The corresponding height profile for Rh NP@BNS.

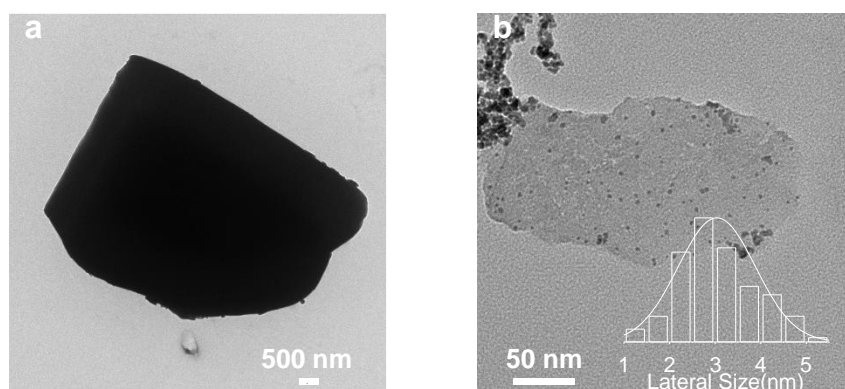


Fig. S2 **a** TEM images of Bulk B. **b** High-resolution TEM images of Rh NP@BNS

(inset: lateral size distribution of Rh NP).

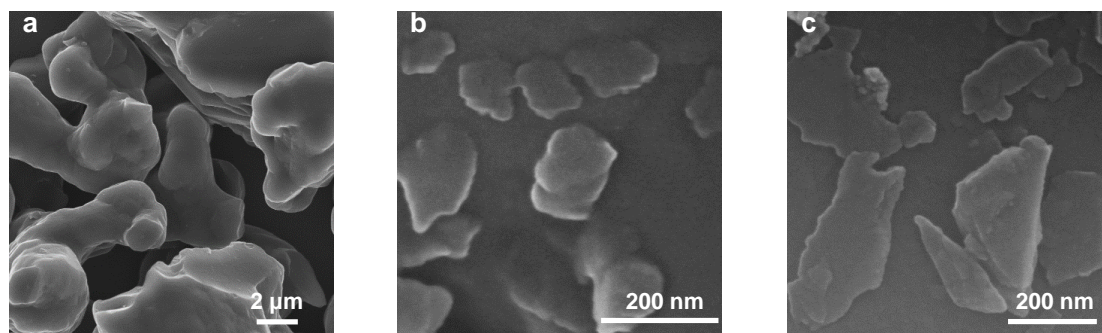


Fig. S3 SEM images of **a** Bulk B, **b** BNS, **c** Rh NP@BNS.

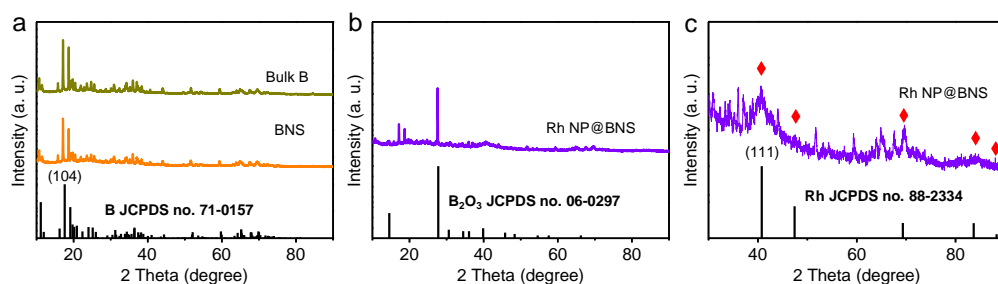


Fig. S4 a XRD patterns of Bulk B and BNS, Matching PDF card (JCPDS No. 71-0157). **b** XRD patterns of Rh NP@BNS, Compare with B₂O₃ PDF card (JCPDS No. 06-0297). **c** The enlarged part of the XRD pattern of Rh NP @ BNS from 30° to 90°, and the corresponding Rh PDF card (JCPDS No. 88-2334).

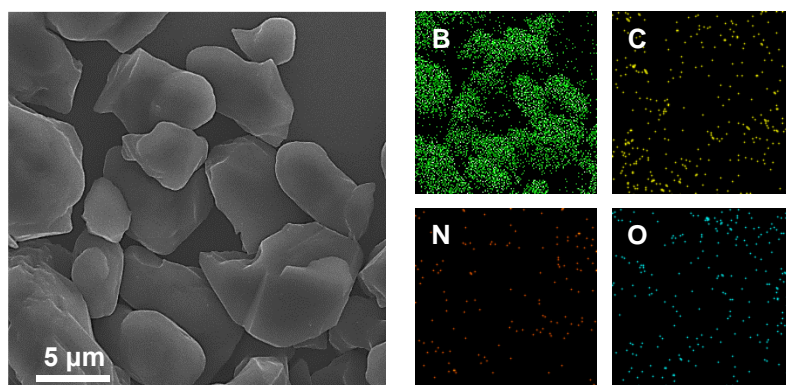


Fig. S5 Typical SEM image of Bulk B, corresponding elemental mapping images of B, C, N and O.

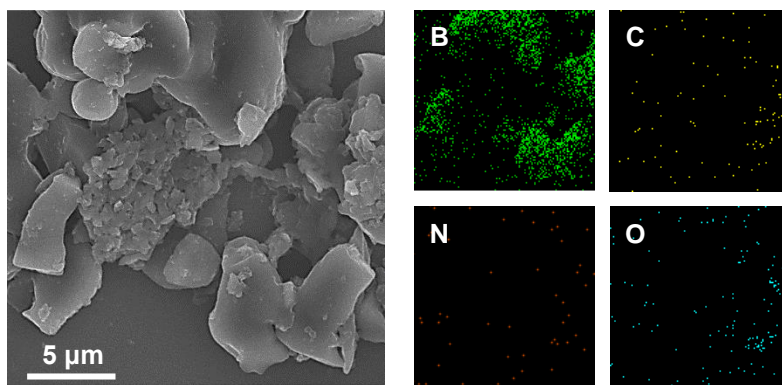


Fig. S6 Typical SEM image of BNS, corresponding elemental mapping images of B, C, N and O.

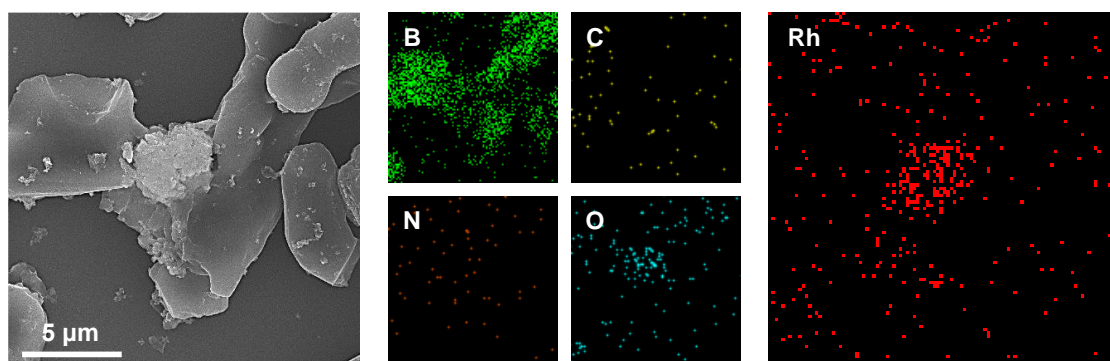


Fig. S7 Typical SEM image of Rh NP@BNS, corresponding elemental mapping images of B, C, N, O and Rh.

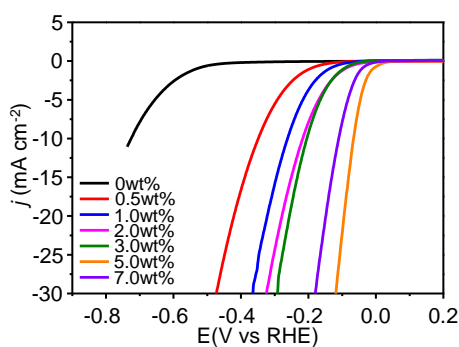


Fig. S8 HER polarization curves of x-Rh NP@BNS ($x=0, 0.5, 1.0, 2.0, 3.0, 5.0, 7.0$, x represents the mass percentage of Rh added during the preparation process) performed in 0.5 M H_2SO_4 electrolyte.

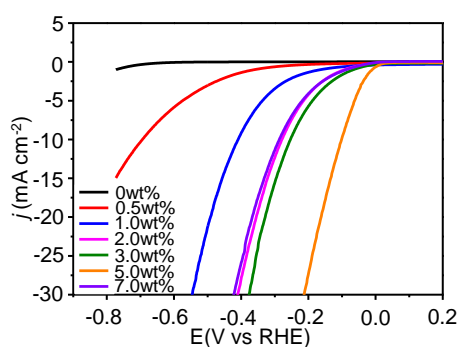


Fig. S9 HER polarization curves of x-Rh NP@BNS (x=0, 0.5, 1.0, 2.0, 3.0, 5.0, 7.0, x represents the mass percentage of Rh added during the preparation process) performed in 1.0 M KOH electrolyte.

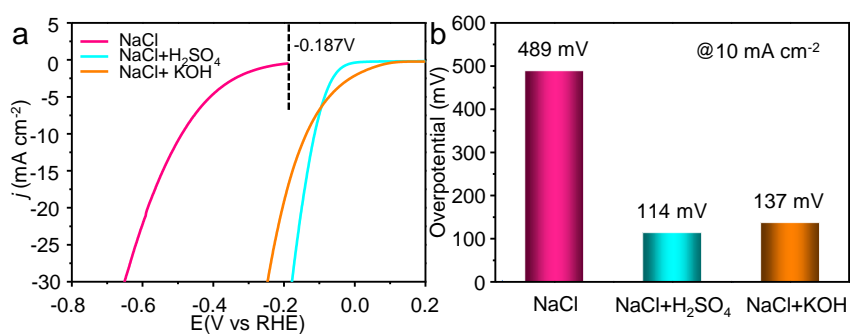


Fig. S10 a HER LSV curves of Rh NP@BNS performed in 1.0 M NaCl, 1.0 M NaCl+0.5 M H₂SO₄, and 1.0 M NaCl+1.0 M KOH electrolyte. **b** Comparison of the overpotentials at 10 mA cm².

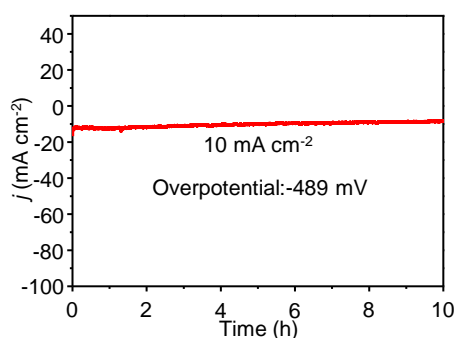


Fig. S11. Time dependence of current density at 489 mV versus RHE in 1.0 M NaCl electrolyte.

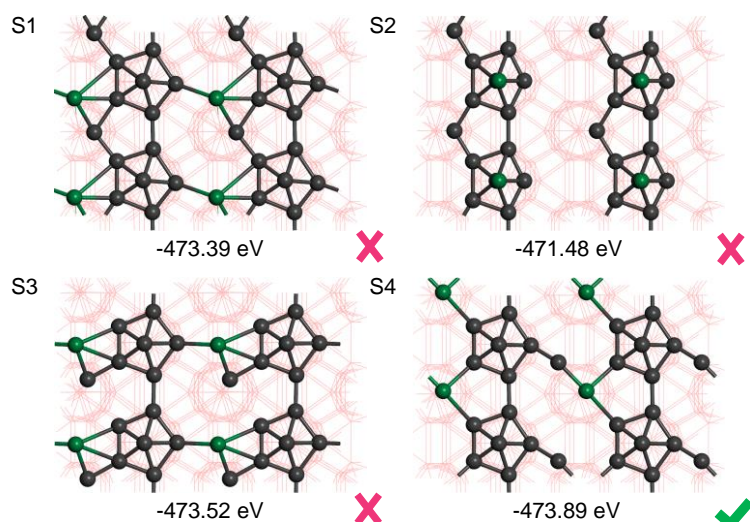


Fig. S12 Static optimization structure of Pt@B(104).

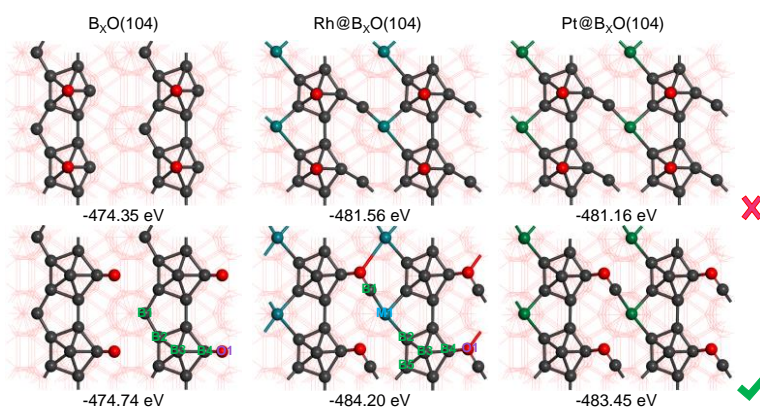


Fig. S13 Static optimization of surface oxidation structures of $B_xO(104)$, $Rh@B_xO(104)$ and $Pt@B_xO(104)$.

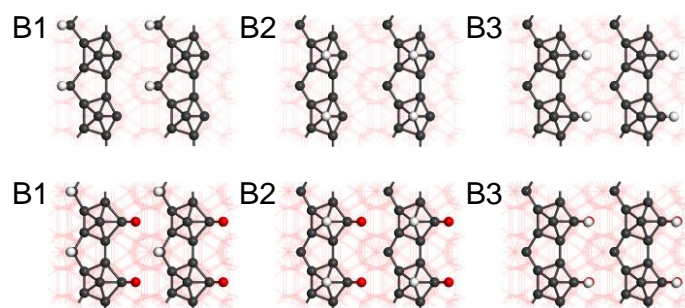


Fig. S14 Geometries Structures for the H adsorption of B(104) and $B_xO(104)$ Surface.

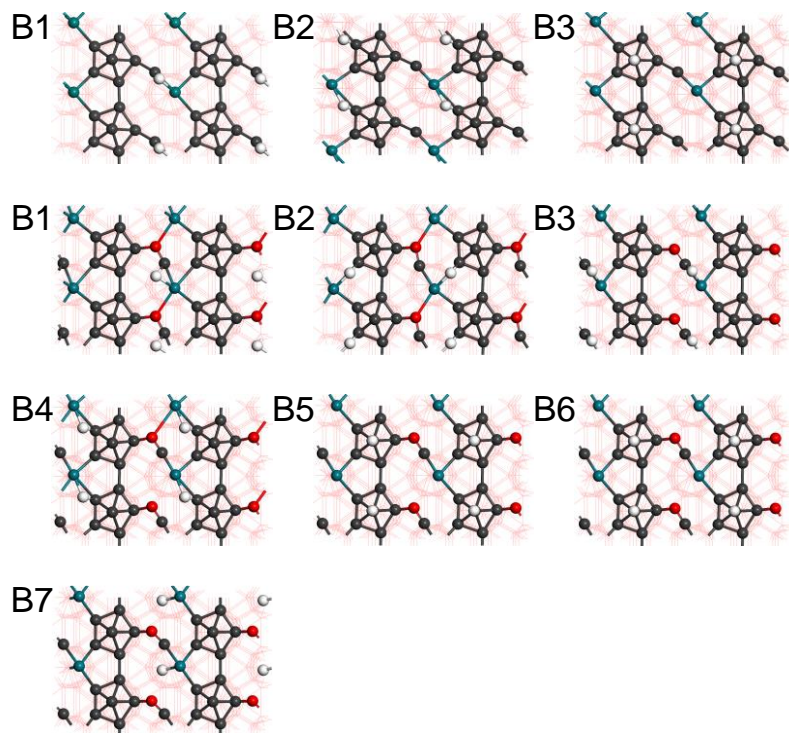


Fig. S15 Geometries Structures for the H adsorption of Rh@B(104) and Rh@B_xO(104) Surface.

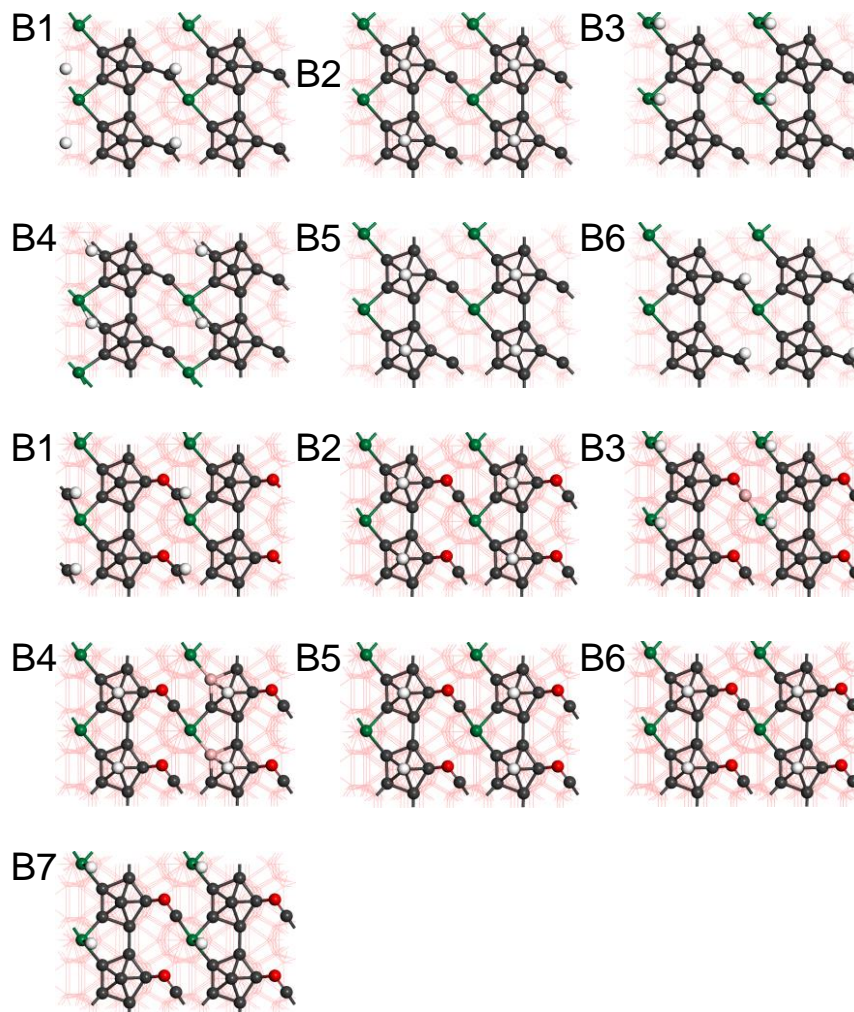


Fig. S16 Geometries Structures for the H adsorption of Pt@B(104) and Pt@B_xO(104) Surface.

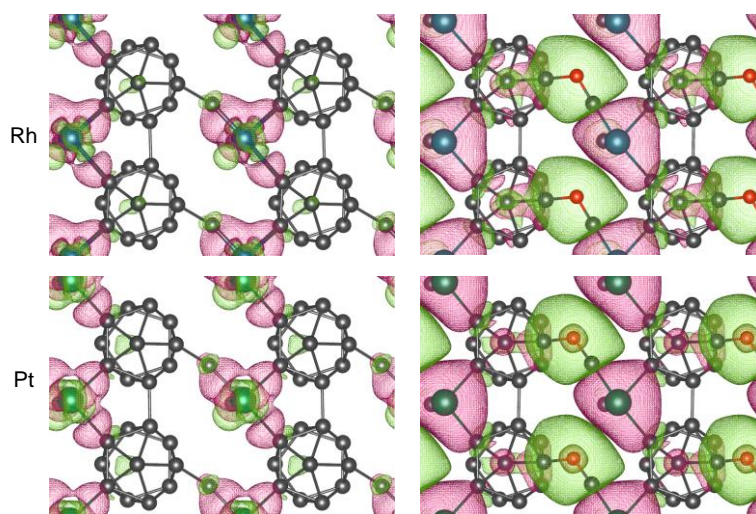


Fig. S17 Isosurface of charge density difference ($\Delta\rho$) for Rh@B_xO(104), Pt@B_xO(104).

Table S1 The total Rh content of the deposition was obtained by ICP-oes test.

Element	Weight/g	Volume/mL	Dilution factor	Instrument Readings mg/L	Concentration mg/kg	Percentage wt%
Rh	0.0244	50	1	5.4157	11097.6876	1.11

Table S2 DFT-calculated ΔGH^* of various Geometries Structures for the H adsorption of B(104), B_xO(104), M@B(104) and M@B_xO(104) Surface (M=Rh, Pt).

Doping structures	Sites	E _{DFT}	Ox	Sites	E _{DFT}
B(104)	1	-470.91	B _x O(104)	1	-478.30
	2	-472.49		2	-478.97
	3	-472.64		3	-480.22
Rh@B(104)	1	-478.56	Rh@B _x O(104)	1	-487.85
	2	-469.99		2	-487.02
	3	-478.56		3	-487.81
	4	-471.27		4	-487.95
	5	-479.60		5	-489.00
	6	-		6	-
	-	-		7	-487.29
Pt@B(104)	1	-478.22	Pt@B _x O(104)	1	-487.39
	2	-		2	-
	3	-477.44		3	-486.81
	4	-477.65		4	-
	5	-479.19		5	-488.84
	6	-		6	-
	-	-		7	-486.81

Table S3 DFT-calculated binding energy (BE) of various B₃₂ doping structures, spin magnetic moment (μ) of single metal atoms were also listed.

Doping structures	BE(eV)	$\mu(\mu_B)$
Ti	-3.45	4.00
Mn	-1.98	5.00
Fe	-3.22	4.00
Ni	-3.85	2.00
Cu	-2.03	1.00
Mo	-2.98	6.00
Pt	-3.49	2.00
Rh	-4.17	1.00
Pd	-1.68	0.00

Table S4 Comparison of HER activities of catalysts in acidic electrolytes (0.5 M H₂SO₄)

Catalyst	Rh metal content (wt%)	Rh nanoparticle size (nm)	Overpotential at 10 mA cm ⁻² (mV)	Tafel slope (mV dec ⁻¹)	Reference
Rh NP@BNS	1.11	~3	66	56	This work
Rh-Ag/SiNW	2.3	12.3	120	51	[S1]
Rh-Au/SiNW	2.2	15.8	62	24	[S2]
rGO/CoP-Rh	NA	12.86	72	43	[S3]
Rh/SWNTs	6.1	~2	25	20	[S4]
MoSe ₂ /Rh	NA	~8	192	47	[S5]
Rh/F-graphene	9.2	9.39	46	30	[S6]
Rh-CN	4.2	3.4	13	25	[S7]
Rh/Ni@NCNTs	2.84	1.92	45	37.2	[S8]
B-RhFe alloy	~18.29	4.01	25	32	[S9]
Rh-Rh ₂ P@C	NA	3.4	24	35.8	[S10]

Table S5 Comparison of HER activities of catalysts in alkaline electrolytes (1.0 M KOH)

Catalyst	Rh metal content (wt%)	Rh nanoparticle size (nm)	Overpotential at 10 mA cm ⁻² (mV)	Tafel slope (mV dec ⁻¹)	Reference
Rh NP@BNS	1.11	3	101	75	This work
rGO/CoP-Rh	NA	12.86	155	101	[S3]
MoSe ₂ /Rh	NA	~8	173	NA	[S5]
Rh/SWNTs	6.1	~2	25	20	[S4]
Rh–MoSe ₂ nanoflowers	8.2	2.5	73	118	[S11]
Rh-CN	4.2	3.4	55	44	[S7]
Rh/Ni@NCNTs	2.84	1.92	45	37.2	[S8]
Rh-Rh ₂ P@C	NA	3.4	37	32	[S10]
P-Rh/C	12.57	1.98	11	71.4	[S12]
Rh/N-CBs	3.5	~1.4	77	74.16	[S13]
Rh NSs	NA	18.4	43	107.2	[S14]

Supplementary References

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