Supporting Information for

High Efficiency Wastewater Purification System Based on Coupled

Photo-Electric-Catalytic Action Provided by Triboelectric

Nanogenerator

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Supplementary Tables and Figures



Fig. S1 TEM images of CDs (a) and 0.2CDs/TNs (b)

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Fig. S2 XRD patterns of 3DGA@CDs-TNs and CDs-TNs after reaction **Table S1** Respective atomic content of 3DGA@CDs-TNs

-	Elements	Atomic %		
	С	69.65		
	Ν	1.18		
	Ο	24.38		
	Ti	4.79		



Fig. S3 EIS curves (**a**) and transient photocurrent density (**b**) of TNs and 0.2CDs/TNs under visible light irradiation

Dyes	Ppm	E _{TENG}	EPhotocatalysts	Eteng-3dga@cds-tnss
DB	5	59.2% (6h)	73.23% (3h)	89.6% (1.5h)
BG	20	59.59% (4h)	81.66% (2h)	88.26% (40min)

Table S2 Comparison of BG and DB degradation over different methods

Methods	k(10 ⁻³ /min)	Formula	R ²
TENG	3.41	y=-0.00341x+0.0009847	0.96148
3DGA@CDs-TNs	16.59	y=-0.01659x-0.27983	0.96638
TENG-Photocatalysts	27.03	y=-0.02703x-0.55186	0.53102

Table S3 Degradation kinetic parameters of BG over 3DGA@CDs-TNs, TENG and TENG/3DGA@CDs-TNs

Table S4 Degradation kinetic parameters of DB over 3DGA@CDs-TNs, TENG and TENG/3DGA@CDs-TNs

Methods	k(10 ⁻³ /min)	Formula	R ²
TENG	2.22	y=-0.00222x+0.00578	0.95317
3DGA@CDs-TNs	6.2	y=-0.0062x-0.03946	0.88643
TENG-Photocatalysts	23.72	y=-0.02372x-0.46223	0.83139



Fig. S4 Influence of different rotation rates on the removal of (a) BG and (b) DB

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	Retention time (min)	Theoretical mass (m/z)	Molecular Formula	Supposed structure
5B	5.259	992.89	C ₃₄ H ₂₄ N ₆ S ₄ O ₁₆ Na ₄	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
	7.997	903	$C_{34}H_{27}N_6S_4O_{16}$	$\overset{O}{\overset{O}{=}}\overset{NH_2}{\overset{O}{=}}\overset{O}{\overset{O}}{\overset{O}{=}}\overset{O}{\overset{O}{=}}\overset{O}{\overset{O}{=}}\overset{O}{\overset{O}{=}}\overset{O}{\overset{O}{=}}\overset{O}{\overset{O}{=}}\overset{O}{\overset{O}{=}}\overset{O}{\overset{O}}{O$
В	6.388	575.58	$C_{24}H_{23}N_4O_9S_2$	$H_{2N} \xrightarrow{H_{2N}} GH_{3C} \xrightarrow{H_{2N}} GH_{3HO} \xrightarrow{H_{2N}} SO_{3H}$
С	6.22	332.29	$C_{10}H_8N_2O_7S_2$	$\begin{array}{c} 0 & 0^{-} \\ H_2N & 0 \\ H_2N & 0 \\ H_2N & 0^{-} \\ H_2N & 0^{-} \end{array}$
D	5.548	286.27	$\mathrm{C_{10}H_6O_6S_2}$	
Е	1.669	239.25	C ₁₀ H ₉ NO ₄ S	HO ₃ S – OH
F	4.877	214.26	$C_{14}H_{14}O_2$	H ₂ N H ₃ CO
G	4.402	148.11	C ₈ H ₄ O ₃	C C C C C C C C C C C C C C C C C C C
Н	3.464	184.24	$C_{12}H_{12}N_2$	H ₂ NNH ₂
Ι	1.617	169.22	C ₁₂ H ₁₁ N	NH ₂
J	5.699	123.11	C ₆ H ₅ NO ₂	

 Table S5 Degradation intermediates of DB in the presence of TENG and 3DGA@CDs

 TNs under visible light irradiation, detected by LCMS in the positive ion mode

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К	6.822	93.14	C ₆ H ₇ N	NH ₂
L	6.752	110.1	$C_6H_6O_2$	но-Су-он
М	0.743	132.11	C ₅ H ₈ O	но
Ν	11.57	90.03	$C_2H_2O_4$	о он но о
0	0.83	104.06	$C_3H_4O_4$	но сон

Table S6 Degradation intermediates of DB in the presence of TENG and 3DGA@CDs-TNs under visible light irradiation, detected by LCMS in the positive ion mode

Intermediate products	Retention time (min)	Theoretical mass (m/z)	l Molecular Formula	Supposed structure
BR	5.259	482.63	C ₂₇ H ₃₄ N ₂ O ₄ S	H ₂ SO ₄
А	6.057	384	C ₂₇ H ₃₂ N ₂	
В	3.206	272.12	C19H16N2	H ₂ N NH ₂ +
С	1.205	198.21	$C_{13}H_{10}O_2$	O OH OH
D	1.184	182.22	C ₁₃ H ₁₀ O	
Е	1.238	106.12	C ₇ H ₆ O	O=CH

F	1.205	110.1	C ₆ H ₆ O ₂	$ \begin{array}{c} & () \\ HO \end{array} \\ () \\ () \\ () \\ OH \end{array} \\ () \\ () \\ OH \end{array} \\ () \\ () \\ () \\ () \\ OH \end{array} \\ () $
G	1.453	108.09	C ₆ H ₄ O ₂	$ \bigcirc^{\circ} \bigcirc^{\circ} \bigcirc$
Н	1.808	116.07	C4H4O4	
Ι	0.743	116.17	$C_6H_{12}O_2$	о
J	2.162	118.1	C4H6O4	но сон
К	1.281	90.03	C ₂ H ₂ O ₄	о он