Supporting Information for

TiN Paper for Ultrafast-Charging Supercapacitors

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Supplementary Figures



Fig. S1 SEM images of TiO₂ paper



Fig. S2 Optical image of a piece of flexible TiO₂ paper



Fig. S3 XRD pattern of a TiO_2 paper. The diffraction peaks are consistent with the values reported for monoclinic phase TiO_2 (TiO_2 -B) (JCPDS No. 74-1940)



Fig. S4 N₂ adsorption/desorption isotherms of TiO₂ nanobelts and TiN nanobelts



Fig. S5 Pore size distribution of TiO₂ nanobelts and TiN nanobelts



Fig. S6 XPS survey spectrum of TiN nanobelts



Fig. S7 A digital image of TiN papers as a part of the electrical connections to light a blue LED by a commercial 3V battery in dark environment



Fig. S8 CV curves of TiN paper electrodes obtained in electrolytes (1 M H₂SO₄, 1 M KOH, 3 M LiCl, and 0.5 M Na₂SO₄) with different pH values



Fig. S9 The capacitance of TiN paper electrodes obtained in different electrolytes are plotted as a function of scan rate



Fig. S10 SEM images of TiN pellet electrodes



Fig. S11 Galvanostatic charging and discharging curves of TiN paper SSC obtained at different current densities



Fig. S12 a CV curves and **b** galvanostatic charging and discharging curves of conventional TiN pellet SSC



Fig. S13 Plots of specific capacitance of TiN pellet SSC and TiN paper SSC as a function of scan rate



Fig. S14 Electrochemical impedance spectrum of TiN paper SSC. Inset shows the enlarged EIS spectrum at high frequencies



Fig. S15 Bode phase plot and imaginary capacitance (C") of TiN paper electrode obtained at different frequencies



Fig. S16 A plot compares the energy density and power density of TiN paper SSCs and the previous SCs



Fig. S17 A plot compares the energy density and power density of TiN paper SSCs and other metal nitride based SCs (corn-like TiN,^[S1] Nb₄N₅ nanochannels,^[S2] VO_x//VN,^[S3] VN/CNT,^[S4] TiN/MnO₂ nanowire,^[S5] TiN nanoarray,^[S6] TiN nanowire,^[S7] TiN//Fe₂N^[S8])



Fig. S18 Core level N 1s and Ti 2p XPS spectra collected for TiN paper electrodes after testing in different electrolytes (0.5 M Na₂SO₄,1 M H₂SO₄ and 1 M KOH) for 200,000 cycles



Fig. S19 Gravimetric capacitance and volumetric capacitance of TiN paper SSCs with different mass loadings of TiN at a high scan rate of 1 V s^{-1}

Materials	Highest scan rate	Cycling stability	Refs.
TiN nanotubes	200 mV s ⁻¹	98.5 % after 100 cycles at 2.5 mA cm ⁻²	[S6]
TiN nanowires	400 mV s ⁻¹	82 % after 15000 cycles at 100 mV s ⁻¹	[S7]
TiN/CNT	1 V s ⁻¹	90 % after 20000 cycles at 100 mV s ⁻¹	[S9]
PANI/TiN/PANI	200 mV s ⁻¹	83 % after 3000 cycles at 0.2 mA cm ⁻²	[S10]
TiN/PPy	200 mV s ⁻¹	72.6 % after 20000 cycles at 15 A g^{-1}	[S11]
TiN/NiCo ₂ O ₄	200 mV s ⁻¹	70 % after 20000 cycles at 10 mA cm ⁻²	[S12]
VN nanowires	100 mV s ⁻¹	95.3 % after 10000 cycles at 100 mV s ⁻¹	[S3]

Table S1 Summary of the rate and cycling performance of transition metal nitride electrodes

VN nanobelts/NC	500 mV s ⁻¹	91.8 % after 12000 cycles at 200 mV s ⁻¹	[S13]
VN/CNT	100 mV s ⁻¹	82 % after 10000 cycles at 0.2 A cm ⁻³	[S4]
VN/graphene	150 mV s ⁻¹	94 % after 2000 cycles at 30 mV s ⁻¹	[S14]
VN/Co(OH) ₂	100 mV s ⁻¹	86 % after 4000 cycles at 1 A g ⁻¹	[\$15]
Fe ₂ N/graphene	100 mV s ⁻¹	92.9 % after 20000 cycles at 4 A g ⁻¹	[S8]
GaN nanowires/graphene	100 V s ⁻¹	98 % after 50000 cycles at 10 mA cm ⁻²	[S16]
Co ₂ N/Ni-doped Co	100 mV s ⁻¹	82.5 % after 5000 cycles at 50 mV s ⁻¹	[S17]
Nb ₄ N ₅ nanobelts	1 V s ⁻¹	80 % after 1000 cycles	[S18]
Nb ₄ N ₅ nanochannels	200 mV s ⁻¹	70.9 % after 2000 cycles at 50 mV s ⁻¹	[S2]
MoN nanosheets	10 V s ⁻¹	95 % after 25000 cycles at 100 mV s ⁻¹	[S19]
Mo ₂ N nanobelts	200 mV s ⁻¹	91 % after 1000 cycles at 100 mV s ⁻¹	[S20]
Mo ₂ N nanobelts/graphene	1 V s ⁻¹	85.7 % after 4000 cycles at 0.57 A cm ⁻³	[S21]
TiN paper	100 V s ⁻¹	102.2 % after 200,000 cycles at 1 V s ⁻¹	this work

Supplementary References

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