Supporting Information for

Post-Synthetic and In-Situ Vacancy Repairing of Iron Hexacyanoferrate Towards Highly Stable Cathodes for Sodium-Ion Batteries

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Supplementary Table and Figures

Table S1 Elemental contents of FeHCF, FeHCF-P and FeHCF-I (wt%)

	Na	Fe	C	N
FeHCF	**12.39**	**32.79**	**18.92**	**21.93**
FeHCF-P	**11.30**	**33.26**	**19.51**	**22.98**
FeHCF-I	**14.13**	**33.31**	**20.14**	**23.24**

Table S2 Energy levels (eV) of Na^+ at 8c and 24d sites in FeHCF with different sodium concentrations

	Na-0	Na-10	Na-20
8c	**-3.26**	**-4.00**	**-3.80**
24d	**-3.44**	**-4.31**	**-4.36**

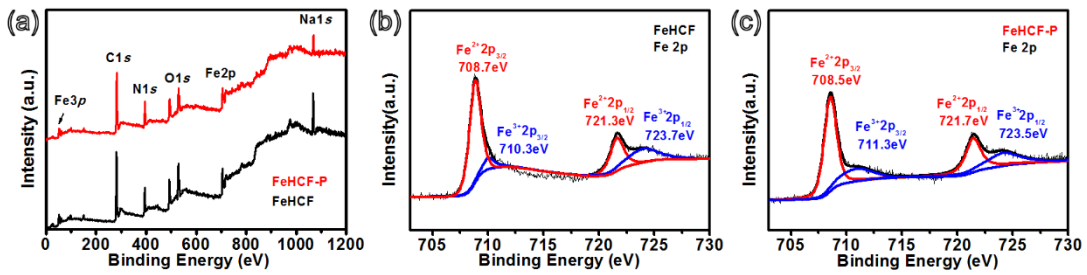

Fig. S1 a XPS survey spectra of FeHCF and FeHCF-P. Fe 2p XPS spectra of **b** FeHCF and **c** FeHCF-P

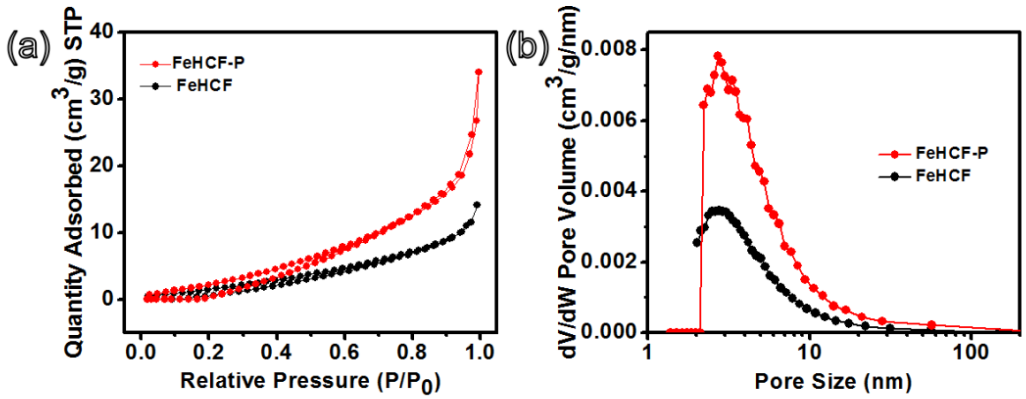

Fig. S2 a N_2 adsorption/desorption isotherm curves and **b** BJH pore-size distribution curves of FeHCF and FeHCf-P

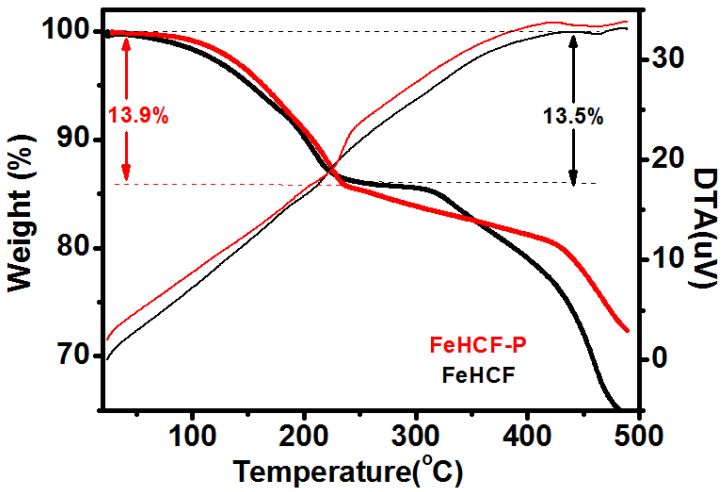

Fig. S3 TG and DTA curves of FeHCF and FeHCF-P in N_2

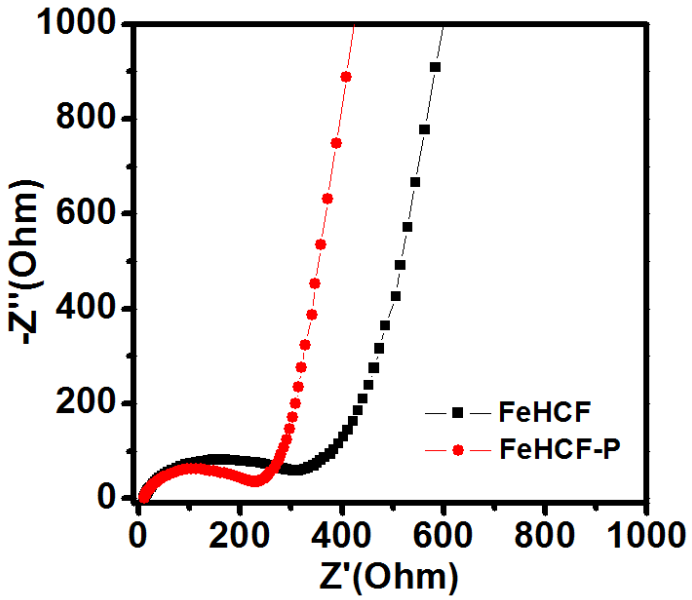

Fig. S4 Electrochemical impedance spectra of FeHCF and FeHCF-P

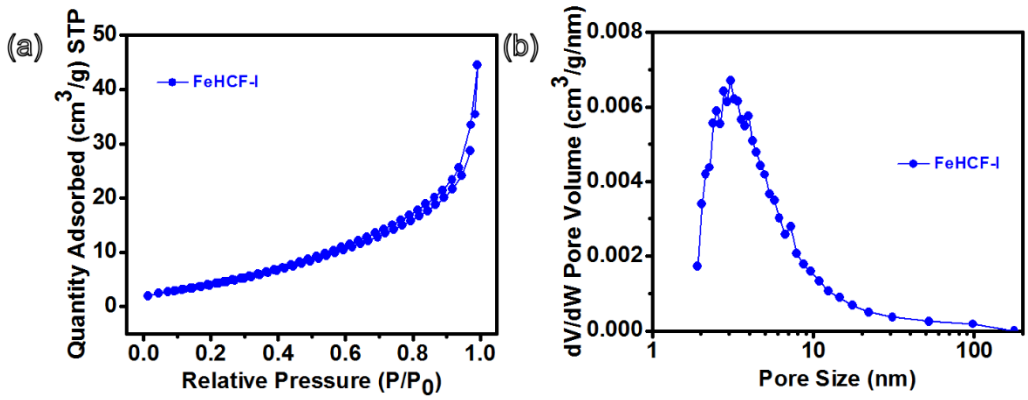

Fig. S5 a N_2 adsorption/desorption isotherm curve and **b** the BJH pore-size distribution curve for FeHCF-I

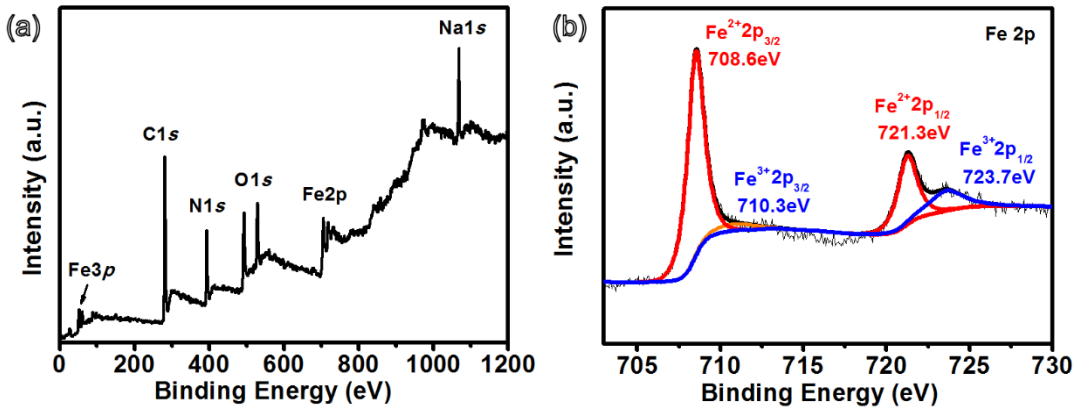

Fig. S6 a XPS survey spectrum and **b** Fe 2p XPS spectrum of FeHCF-I

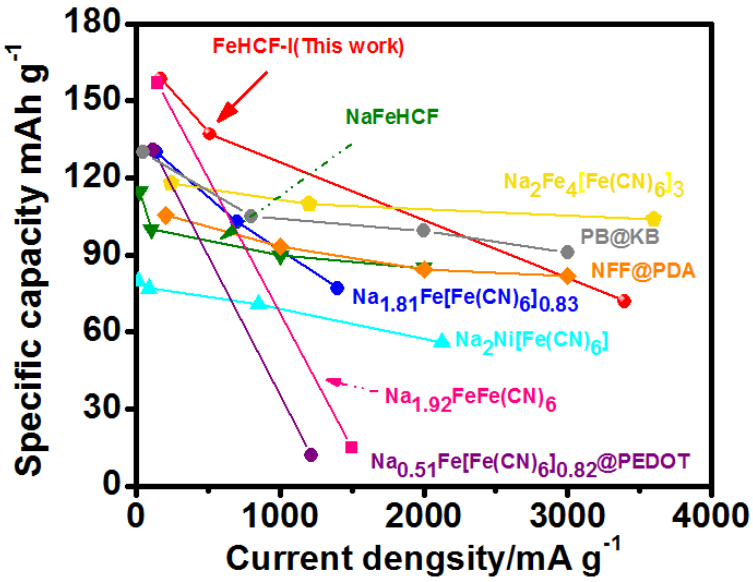

Fig. S7 Comparison of rate performance between FeHCF-I and reported PBAs cathode in previous literatures